



Te.M.P.E.

Istituto per la Tecnologia dei Materiali e dei Processi Energetici

ATLAS OF FLUORESCENCE SPECTRA OF AROMATIC HYDROCARBONS

Vol. 1

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<http://www.tempe.mi.cnr.it/zizak/atlas/cld-atlas-eng.html>

Editorial group & Graphic design: S. De Iuliis, D. Ferretti

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Introduction

This Atlas is the electronic version and extension of a collection of fluorescence spectra of aromatic hydrocarbons previously realized on paper support during the period 1991-1992. Subsequent modifications and requests of copies of the Atlas by colleagues, induced the authors to make their work more available to the public by realizing this web version for easy download. In the following there are some introductory remarks for the users of the Atlas.

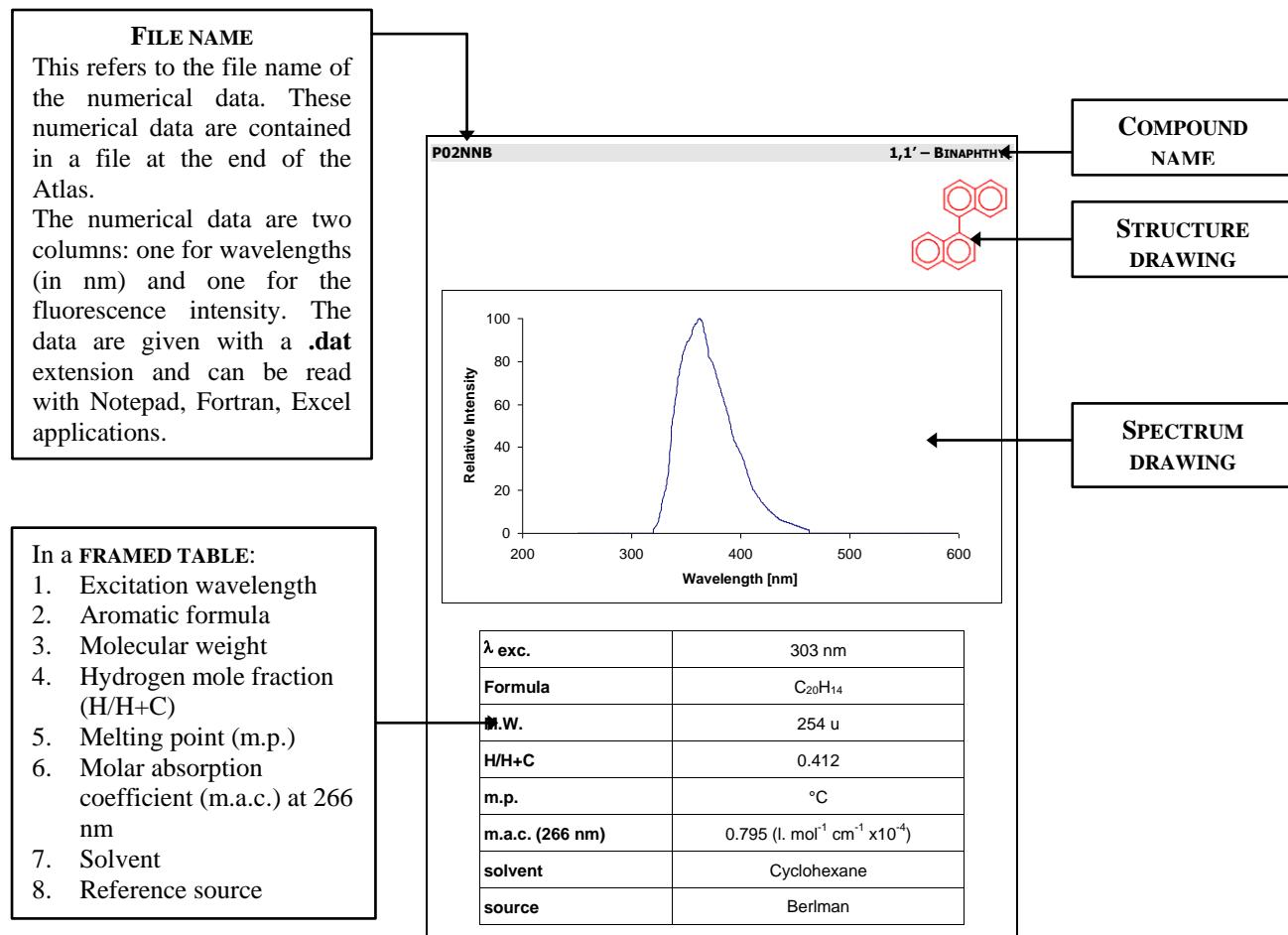
This Atlas (vol.1) contains the fluorescence spectra of pure aromatic hydrocarbons (only carbon and hydrogen, without non-hydrocarbon substituted or heterocycle compounds) as taken from literature data. In their original form, spectra were represented in different ways, so that they were not directly comparable. From that our decision to build this Atlas. The graphical spectra, taken from different sources, have been digitized with the help of a digital table, then normalized to the maximum, recalculated and interpolated in 0.5 nm wavelength intervals. The results have then been redrawn with the help of the Excel software. The authors are aware that this collection of data is not exhaustive of all the aromatic compounds and all data available in literature. It is the authors desire to continuously upgrade the Atlas at the best of their knowledge by adding new spectra as they appear in the open literature. If readers are aware of data not reported in this Atlas or wish to publish their personal data in this Atlas, please send an e-mail to zizak@tempe.mi.cnr.it. Their work will be duly referenced. We are still collecting spectra of non pure aromatic hydrocarbons and they will be the subject of a future Atlas vol. 2.

In general the spectra refer to solution of pure aromatic compounds in a non-polar solvent (mostly cyclohexane) with some exceptions such as acetonitrile or gas phase. Users must be aware that aromatic compounds in different solutions or experimental conditions can exhibit shifts and modifications in their fluorescence spectrum. The data here presented must be therefore taken with caution. For a direct comparison of the shapes of the spectra of different compounds, all spectra presented in this Atlas are normalized to 100 and the wavelength axis is kept the same. In their present digital form, as given in the Data folder, spectra can be easily handled and combined. However, comparison of intensities requires knowledge of the absorption coefficients at the excitation wavelength, the quantum efficiency for fluorescence in the given experimental conditions and the spectral efficiency of the detection equipment.

References are reported so that readers can verify the original spectra. Most of data are taken from two reference books. One is the *Spectral Atlas of Polycyclic Aromatic Compounds*, Kluwer Academic Publishing Company, Dordrecht, Holland, vol. I, II and III edited by W.Karcher. The other is the most cited *Handbook of Fluorescence Spectra of Aromatic Molecules*, Academic Press, New York, 1971 by I.B. Berlman. Spectra published on papers are referenced with the usual numerical style. The CNPM-CNR reference refers to unpublished data taken by the authors during the 1991-1992 years. The experimental apparatus was very simple. Mild heating of individual aromatic compounds, produced vapors then carried by a nitrogen flow out of a short brass pipe in quiescent air. Fluorescence was induced by the fourth harmonic (266 nm) of a pulsed Nd:YAG laser just above the exit of the vapors duct. LIF spectra were collected with a quartz optics a low resolution spectrograph (JY UFS 200) equipped with a gated intensified multichannel diode array detector (Tracor Northern TN 6132). Careful tests were carried out to verify that no residues were left when changing from one compound to another. Most of the measurements resulted from the integration of 200 laser shots with a measuring time of less than one minute.

In the Atlas each page contains one fluorescence spectrum of a specific aromatic compound. Some compounds have different spectra as taken from different literature sources. Some compounds have also different names. In the Atlas the spectra are ordered according to their **file name** (see further on). In each page some useful additional information are also given.

Page structure:



NOTE

FILE NAMES are given in a somewhat arbitrary way, in a combination of up to 7 characters (letters and numbers) according to the following PAH file name criterion:

PAH FILE NAME CRITERION

P123456.dat

P PAH (stands for all aromatic structures)

12 firsts two characters stand for:

a) number of fused aromatic rings (ex: 01, 02, ..15)

b) BP means biphenyl structure

c) FL means a fluoranthene structure

3 a letter or a number to label a particular structure within the range defined by the first two characters. In the particular case c) a number to indicate the total number of aromatic rings

456 combination of letters and numbers to indicate a different structure and/or solvent and/or excitation wavelength. In general:

M indicates a methylated PAH

The last letter can indicate a different solvent (A = acetonitrile, G = gas phase)

The last number indicates a different data file.

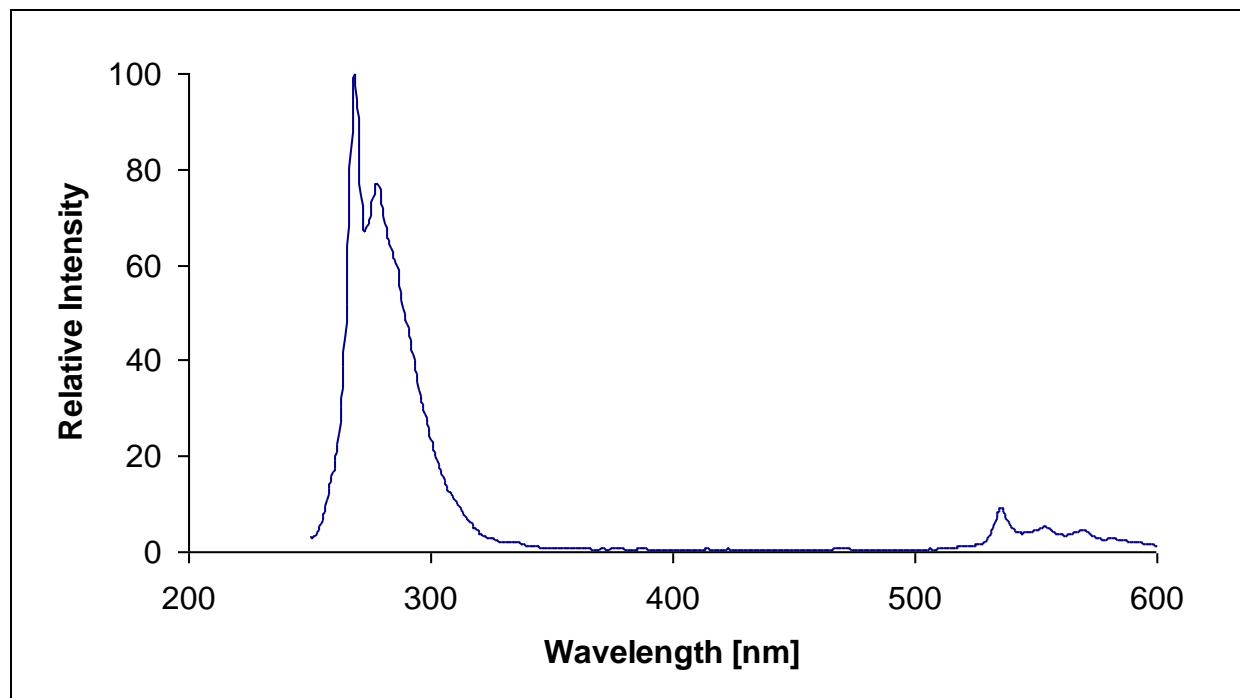
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List of compounds

P01B0G

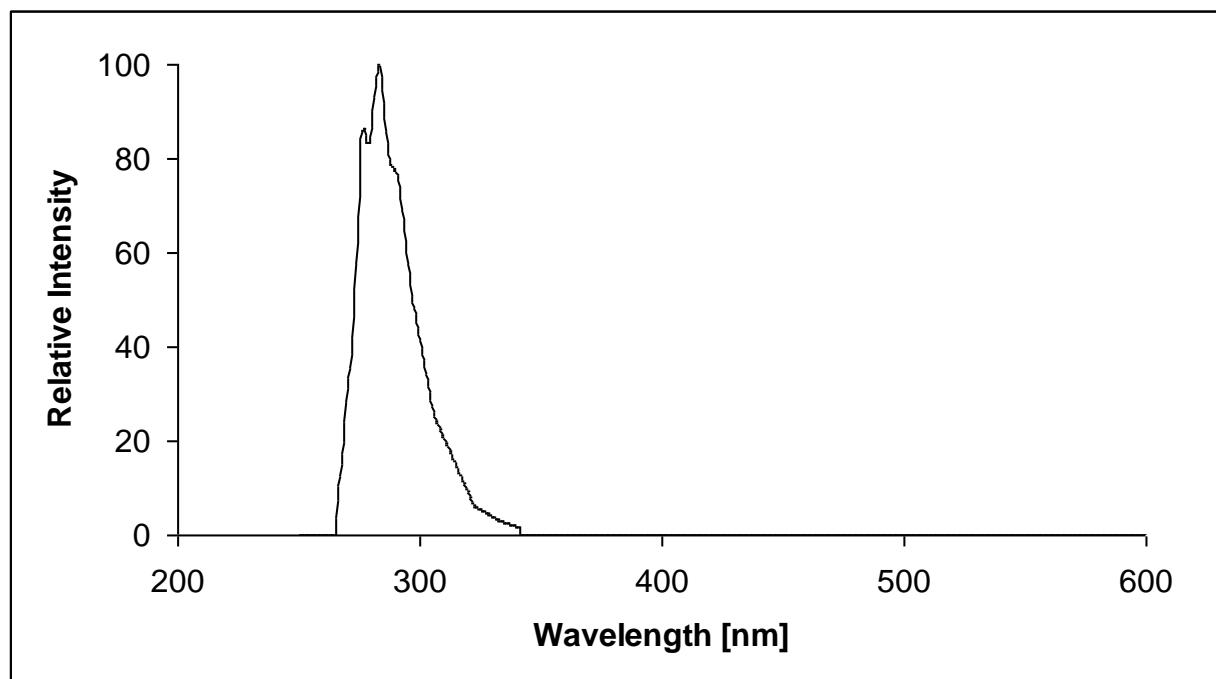
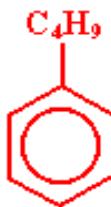
Benzene



λ exc.	266 nm
Formula	C_6H_6
M.W.	78 u
H/H+C	0.5
m.p.	°C
m.a.c. (266 nm)	(l. $mol^{-1} cm^{-1} \times 10^{-4}$)
solvent	Gas phase 300 K
source	CNPM-CNR

P01BB

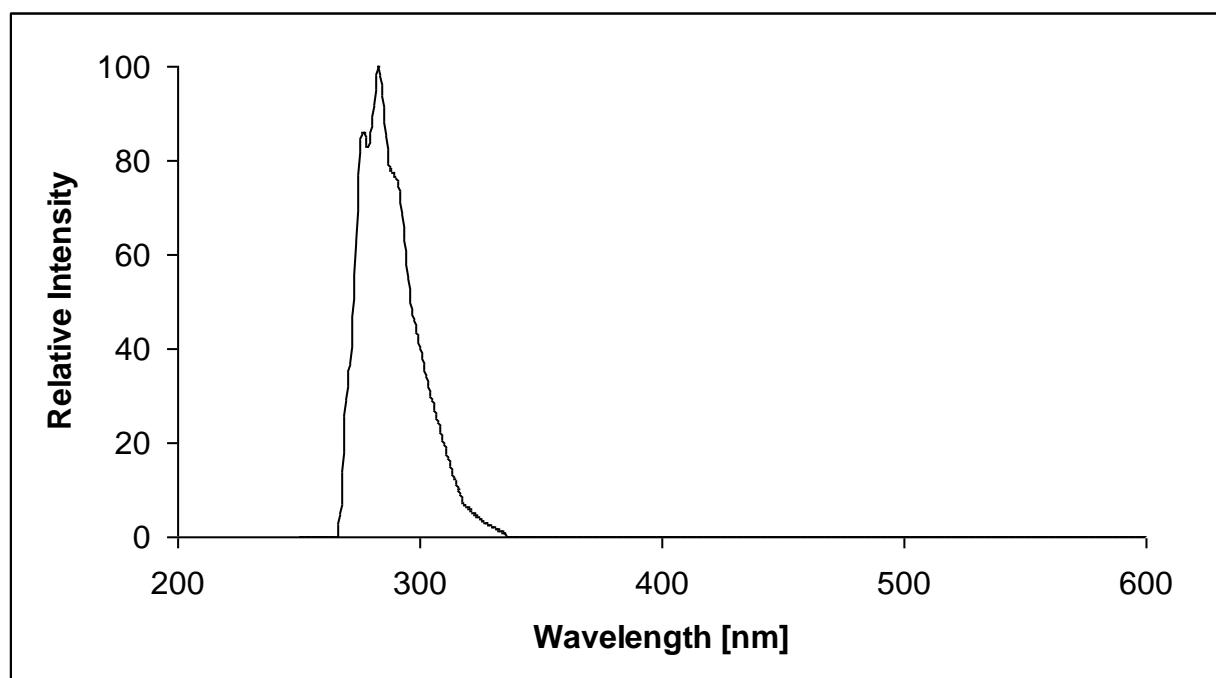
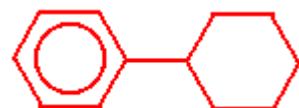
Sec-Butylbenzene



λ exc.	265 nm
Formula	$\text{C}_{10}\text{H}_{14}$
M.W.	134 u
H/H+C	0.583
m.p.	°C
m.a.c. (266 nm)	0.0111 ($\text{l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4}$)
solvent	Cyclohexane
source	Berlman

P01BC

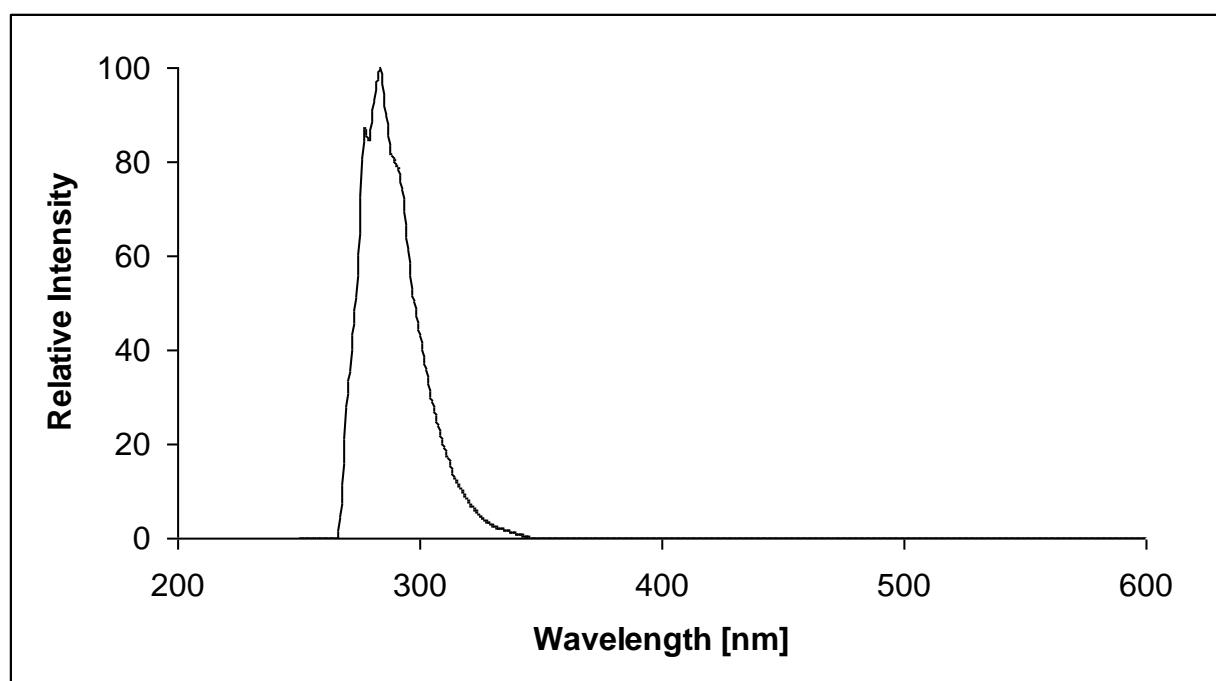
Phenylcyclohexane



λ exc.	265 nm
Formula	$C_{12}H_{16}$
M.W.	160 u
H/H+C	0.571
m.p.	°C
m.a.c. (266 nm)	$0.0135 \text{ (l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4})$
solvent	Cyclohexane
source	Berlman

P01BE

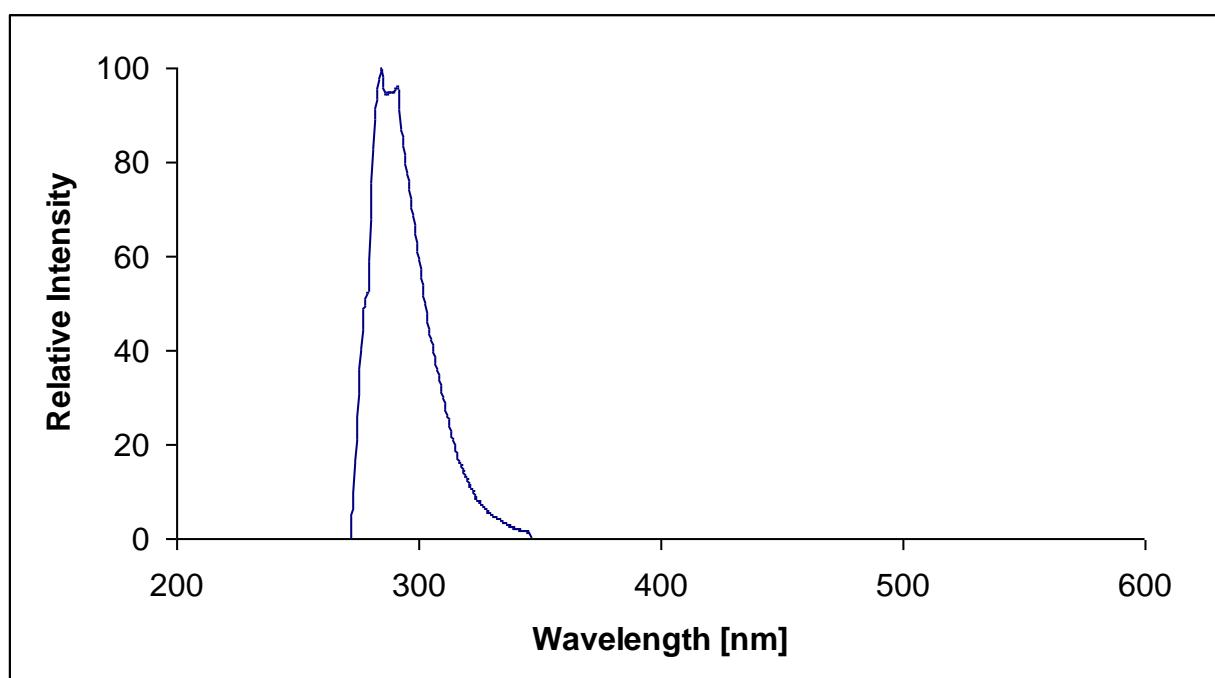
Ethylbenzene



λ exc.	265 nm
Formula	C_8H_{10}
M.W.	106 u
H/H+C	0.556
m.p.	°C
m.a.c. (266 nm)	0.016 ($l \cdot mol^{-1} cm^{-1} \times 10^{-4}$)
solvent	Cyclohexane
source	Berlman

P01BEM

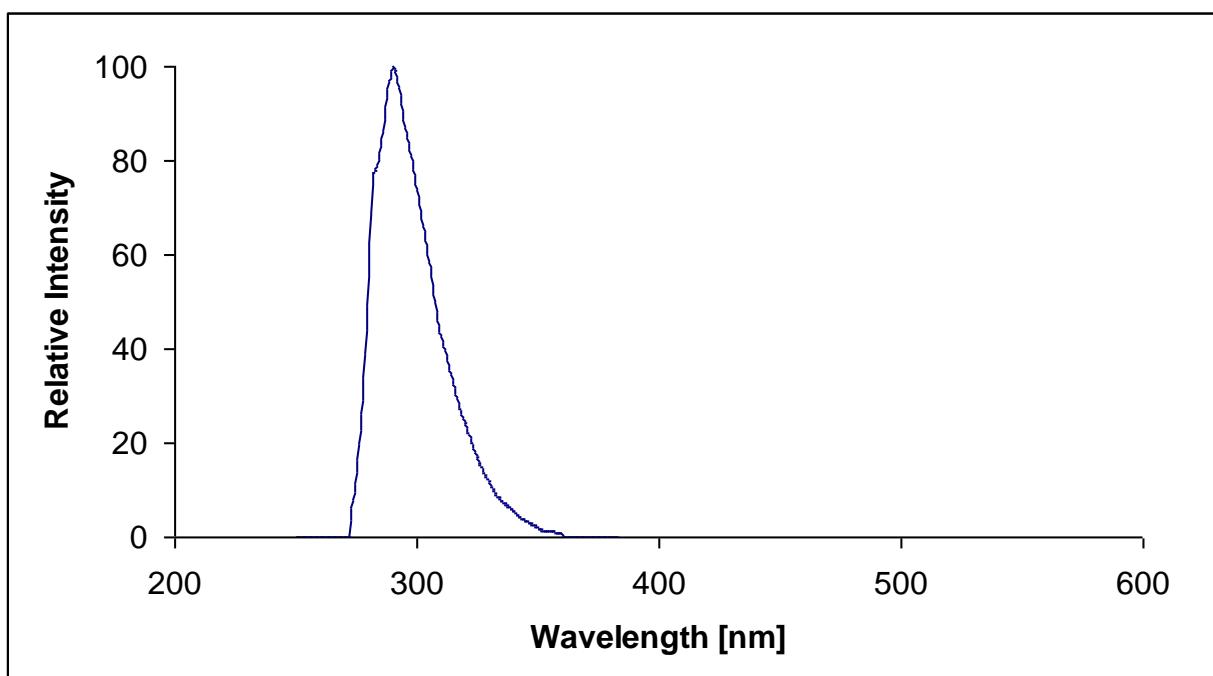
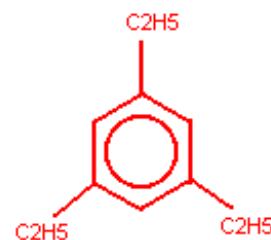
1 – Methyl-4-Ethylbenzene



λ exc.	265 nm
Formula	C_9H_{12}
M.W.	120 u
H/H+C	0.571
m.p.	°C
m.a.c. (266 nm)	$0.0425 (\text{l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4})$
solvent	Cyclohexane
source	Berlman

P01BES

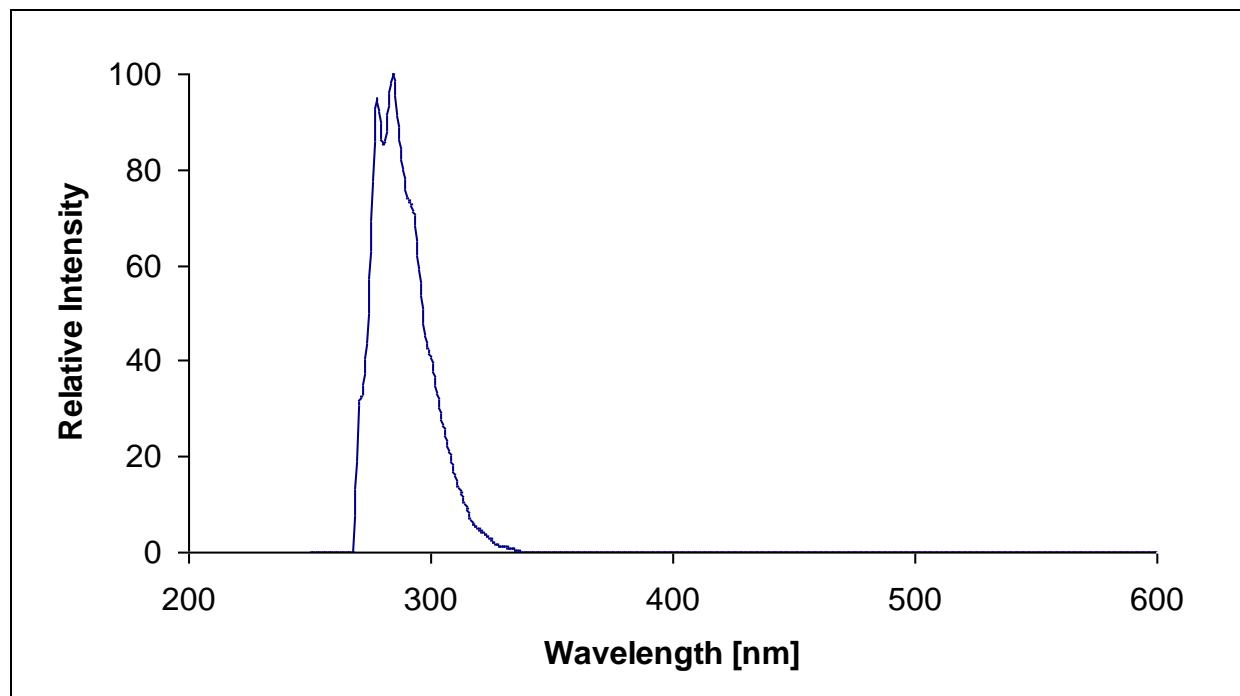
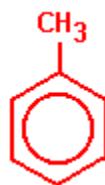
1,3,5 – Triethylbenzene



λ exc.	265 nm
Formula	C ₁₂ H ₁₈
M.W.	162 u
H/H+C	0.600
m.p.	°C
m.a.c. (266 nm)	0.0187 (l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	Cyclohexane
source	Berlman

P01BM

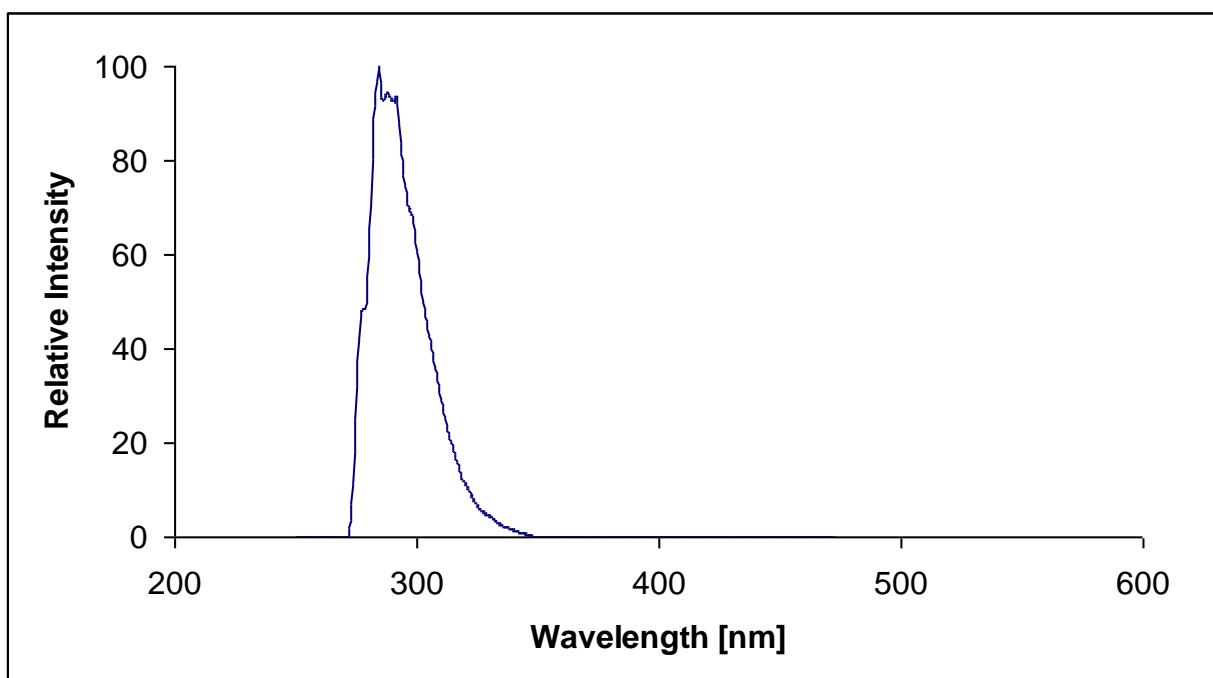
Toluene



λ exc.	265 nm
Formula	C_7H_8
M.W.	92 u
H/H+C	0.533
m.p.	°C
m.a.c. (266 nm)	$0.014 (\text{l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4})$
solvent	Cyclohexane
source	Berlman

P01BMJ

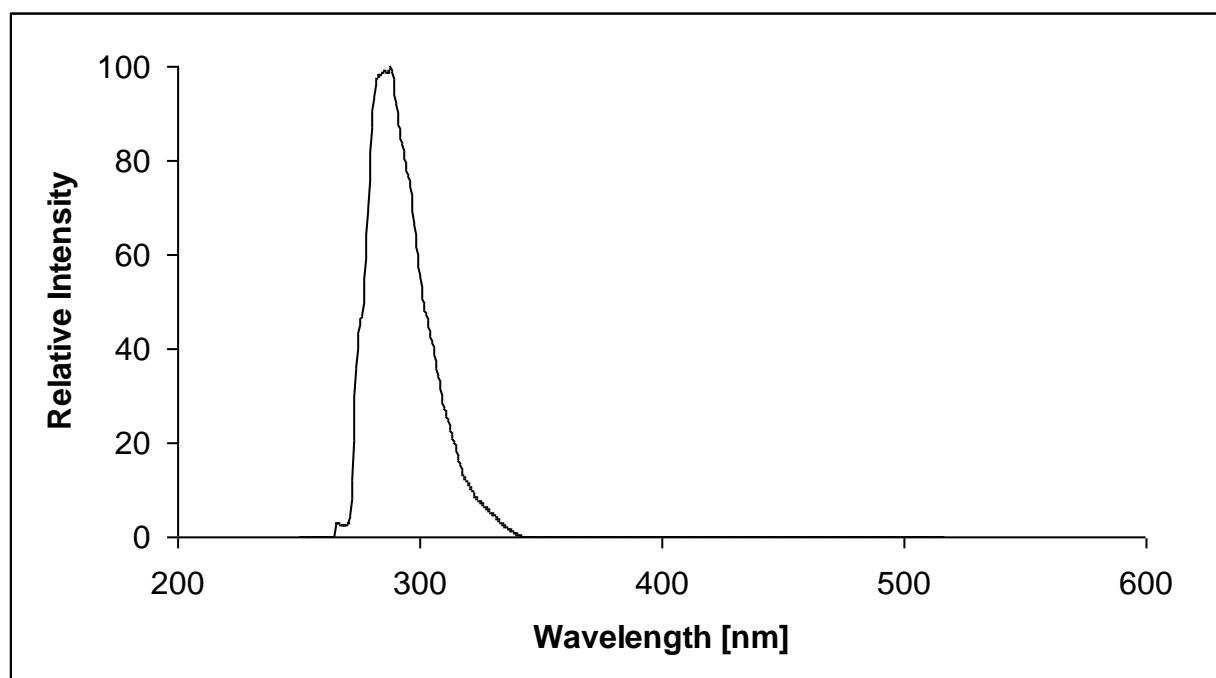
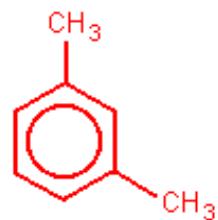
P-xylene



λ exc.	265 nm
Formula	C ₈ H ₁₀
M.W.	106 u
H/H+C	0.556
m.p.	°C
m.a.c. (266 nm)	0.0515 (l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	Cyclohexane
source	Berlman

P01BMK

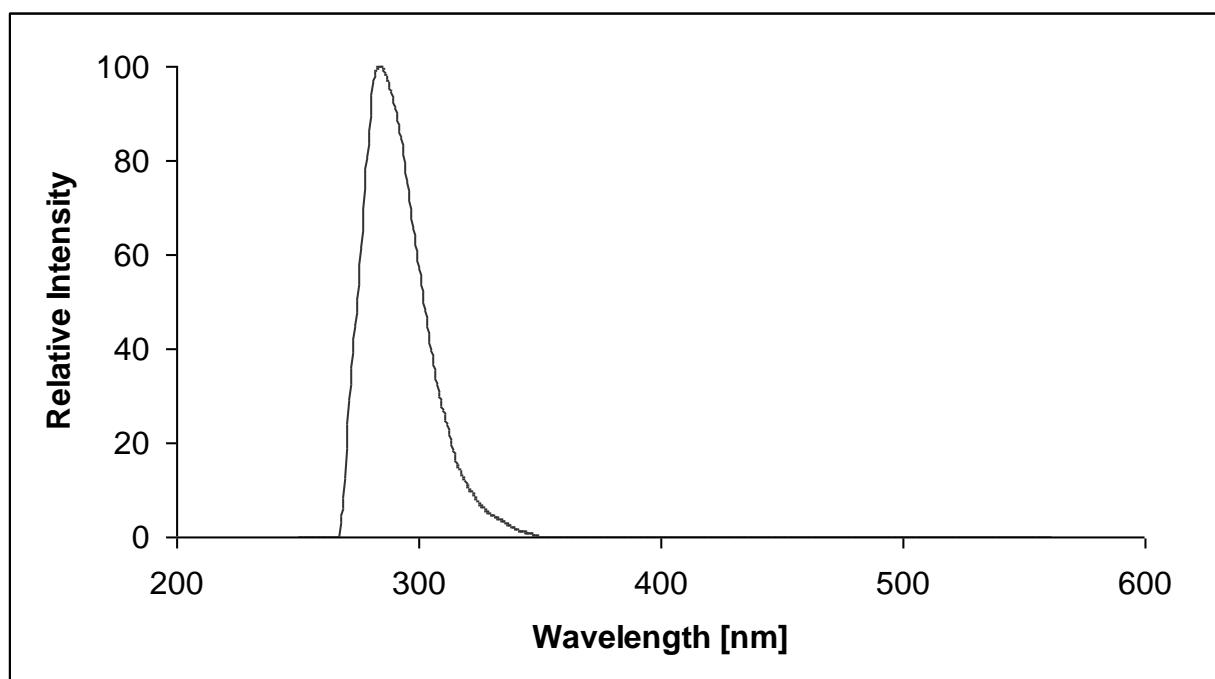
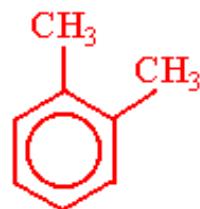
M-xylene



λ exc.	265 nm
Formula	C ₈ H ₁₀
M.W.	106 u
H/H+C	0.556
m.p.	°C
m.a.c. (266 nm)	0.027 (l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	Cyclohexane
source	Berlman

P01BML

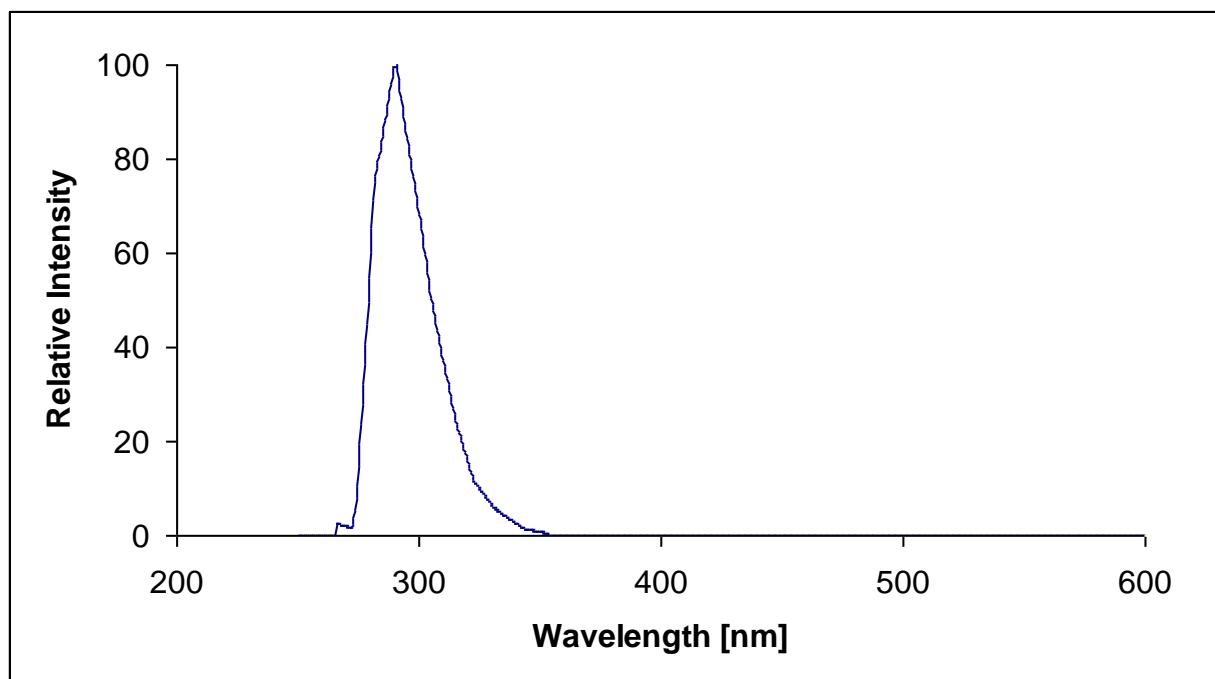
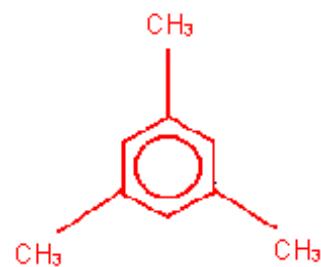
O-xylene



λ exc.	265 nm
Formula	C_8H_{10}
M.W.	106 u
H/H+C	0.556
m.p.	°C
m.a.c. (266 nm)	$0.0245 (\text{l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4})$
solvent	Cyclohexane
source	Berlman

P01BMS

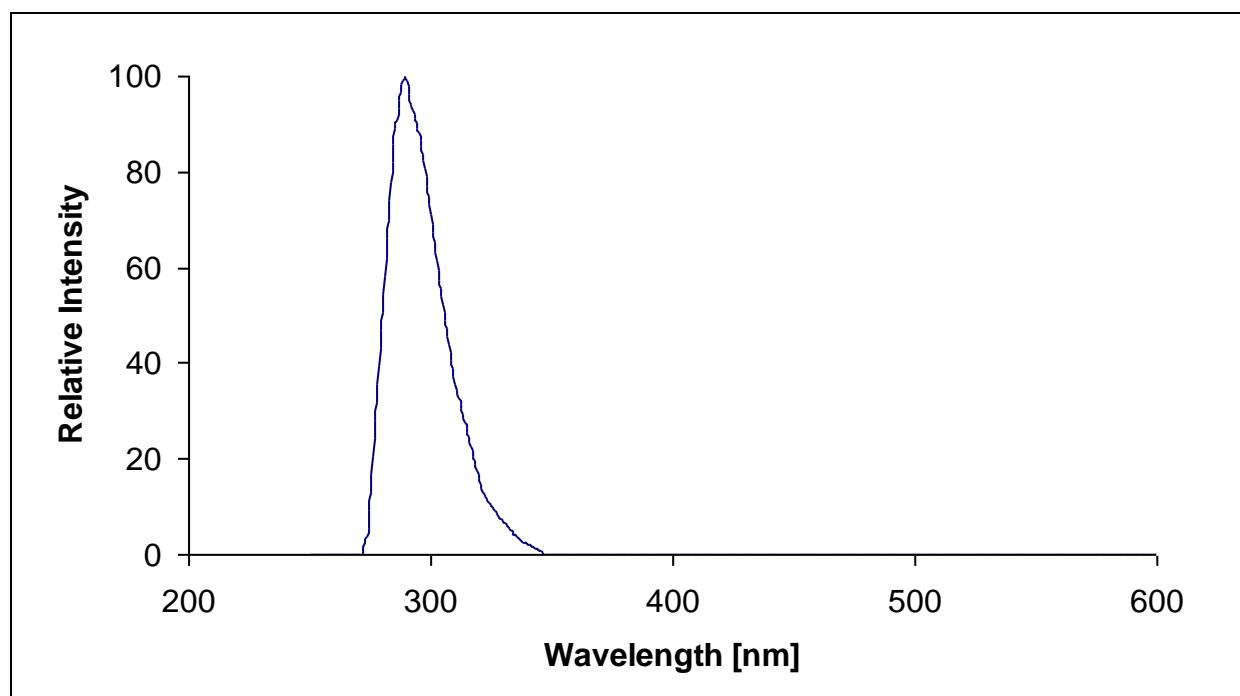
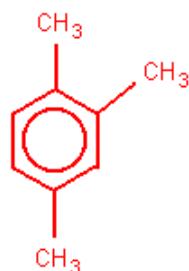
1,3,5 – Trimethylbenzene
Mesitylene



λ exc.	265 nm
Formula	C ₉ H ₁₂
M.W.	120 u
H/H+C	0.571
m.p.	°C
m.a.c. (266 nm)	0.023 (l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	Cyclohexane
source	Berlman

P01BMT

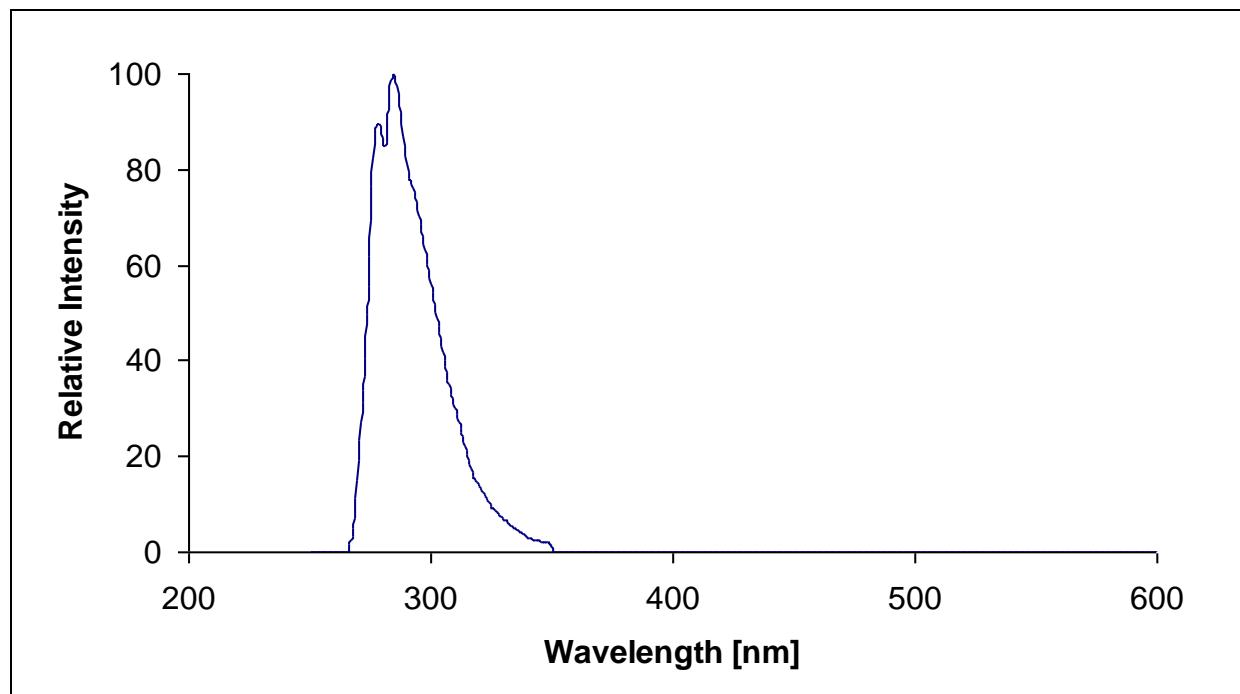
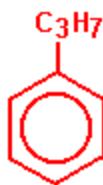
1,2,4 – Trimethylbenzene



λ exc.	265 nm
Formula	C ₉ H ₁₂
M.W.	120 u
H/H+C	0.571
m.p.	°C
m.a.c. (266 nm)	0.0565 (l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	Cyclohexane
source	Berlman

P01BR

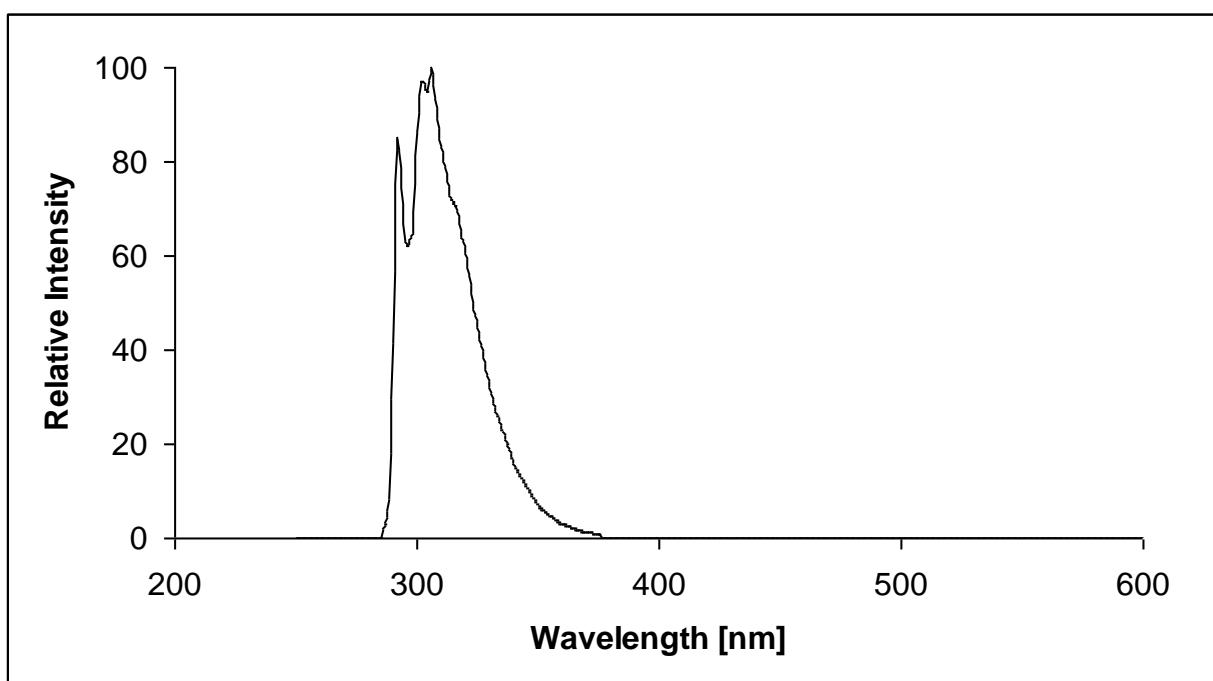
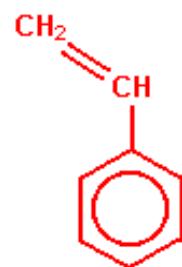
Prophylbenzene



λ exc.	265 nm
Formula	C_9H_{12}
M.W.	120 u
H/H+C	0.571
m.p.	°C
m.a.c. (266 nm)	$0.0155 \text{ (l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4})$
solvent	Cyclohexane
source	Berlman

P01BS

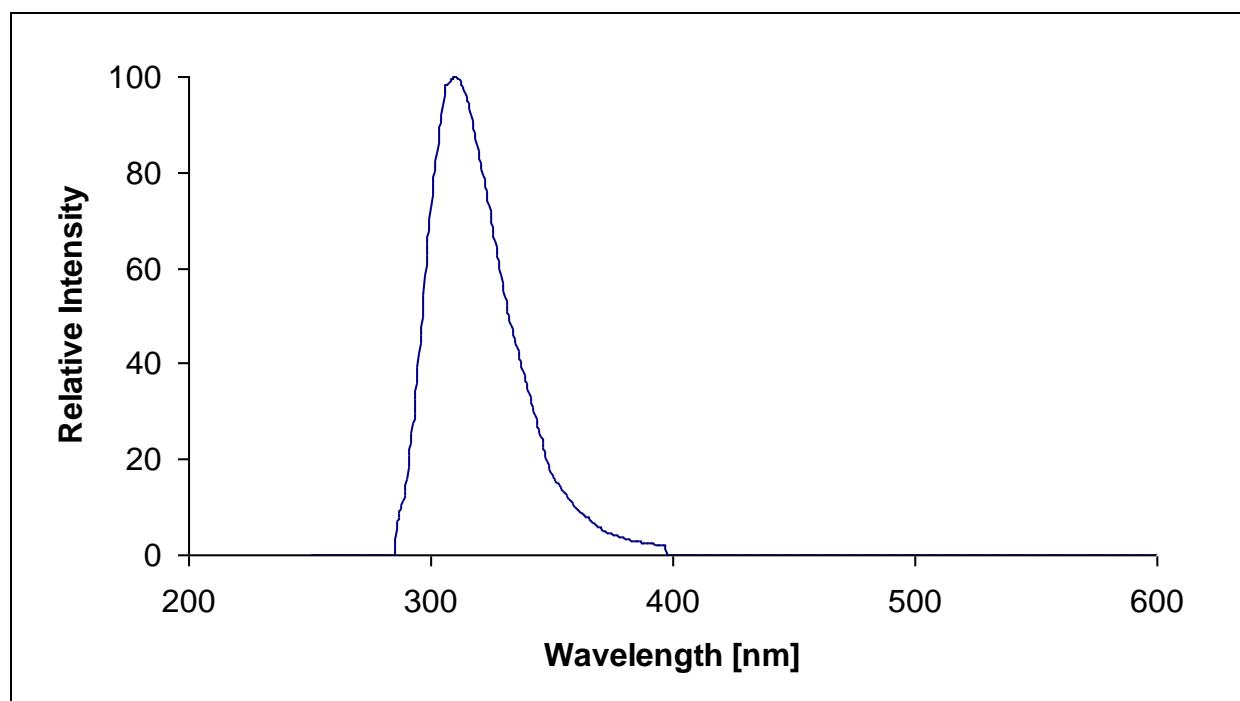
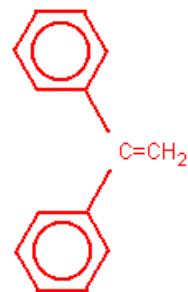
Styrene



λ exc.	253.5 nm
Formula	C ₈ H ₈
M.W.	104 u
H/H+C	0.500
m.p.	°C
m.a.c. (266 nm)	(l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	Cyclohexane
source	Berlman

P01DPE

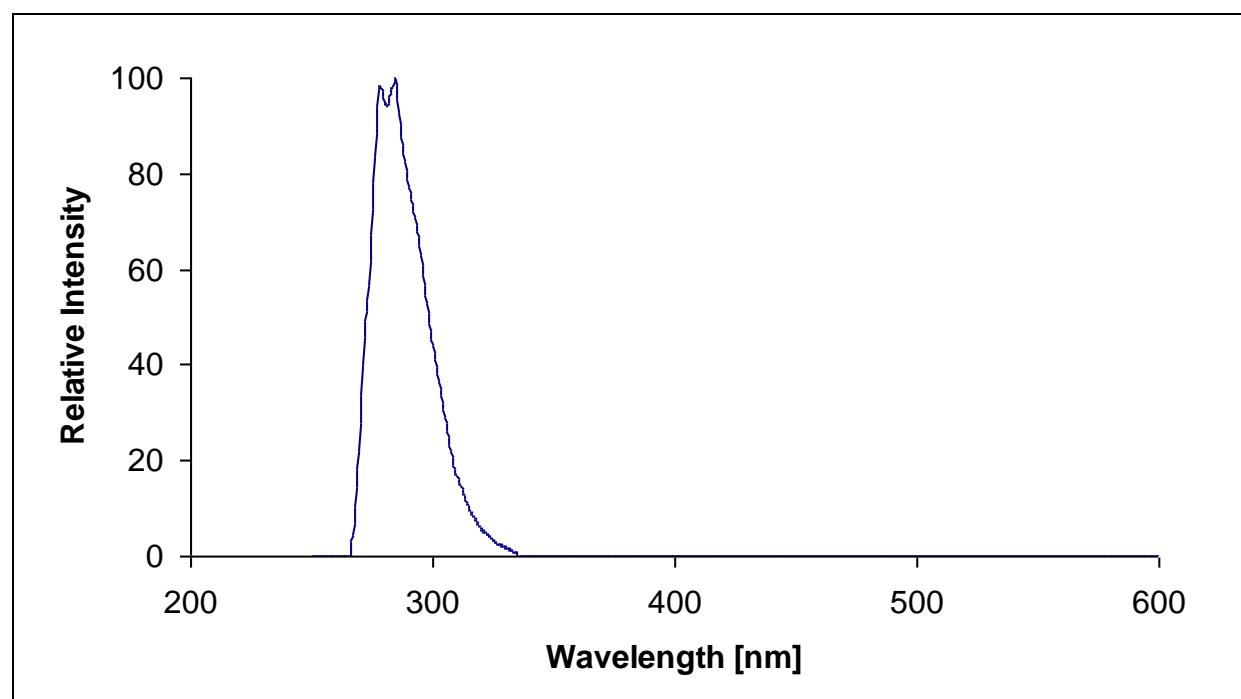
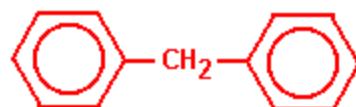
1,1 – Diphenylethylene



λ exc.	265 nm
Formula	$\text{C}_{14}\text{H}_{12}$
M.W.	180 u
H/H+C	0.462
m.p.	°C
m.a.c. (266 nm)	$0.600 (\text{l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4})$
solvent	Cyclohexane
source	Berlman

P01DPM

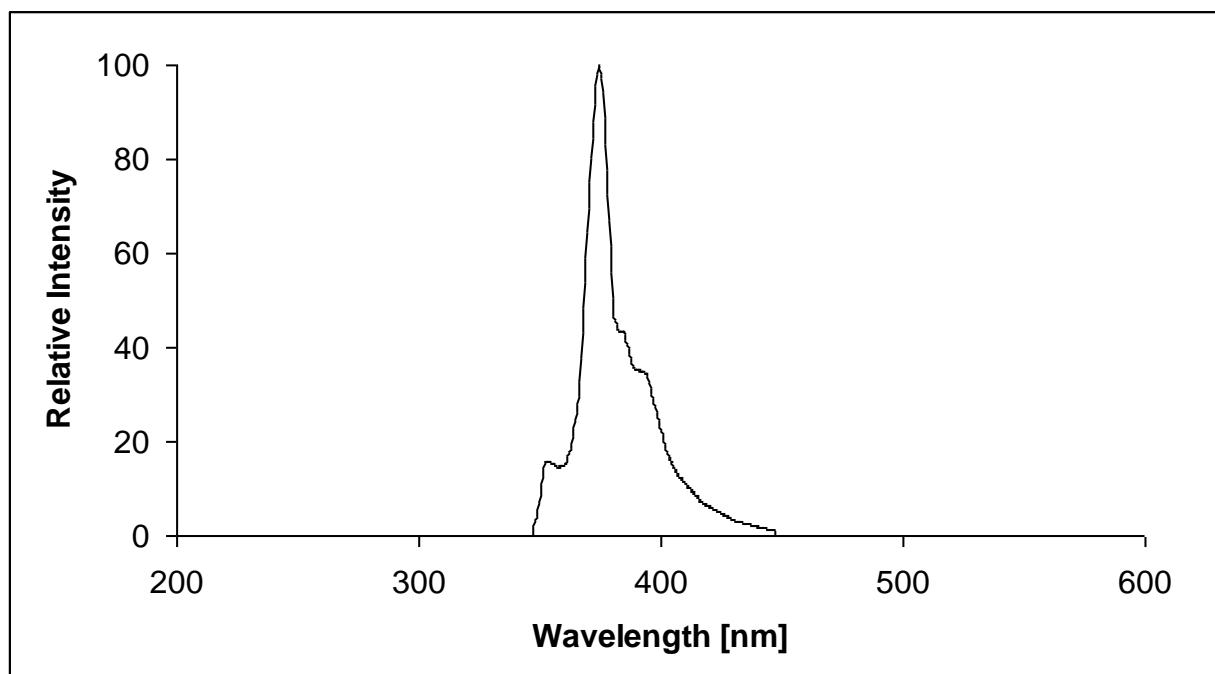
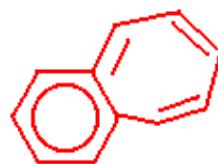
Diphenylmethane



λ exc.	265 nm
Formula	$\text{C}_{13}\text{H}_{12}$
M.W.	168 u
H/H+C	0.480
m.p.	°C
m.a.c. (266 nm)	$0.0315 (\text{l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4})$
solvent	Cyclohexane
source	Berlman

P01Z

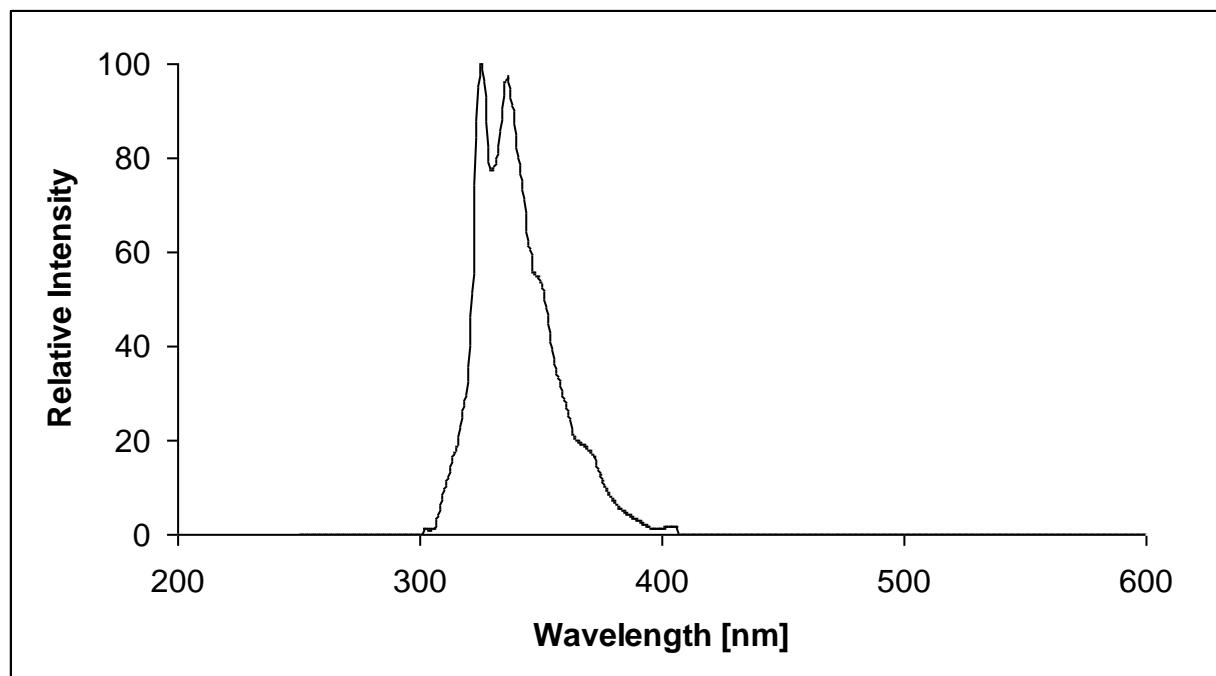
Azulene



λ exc.	313 nm
Formula	$C_{10}H_8$
M.W.	128 u
H/H+C	0.444
m.p.	°C
m.a.c. (266 nm)	(l. $mol^{-1} cm^{-1} \times 10^{-4}$)
solvent	Cyclohexane
source	Berlman

P02N01

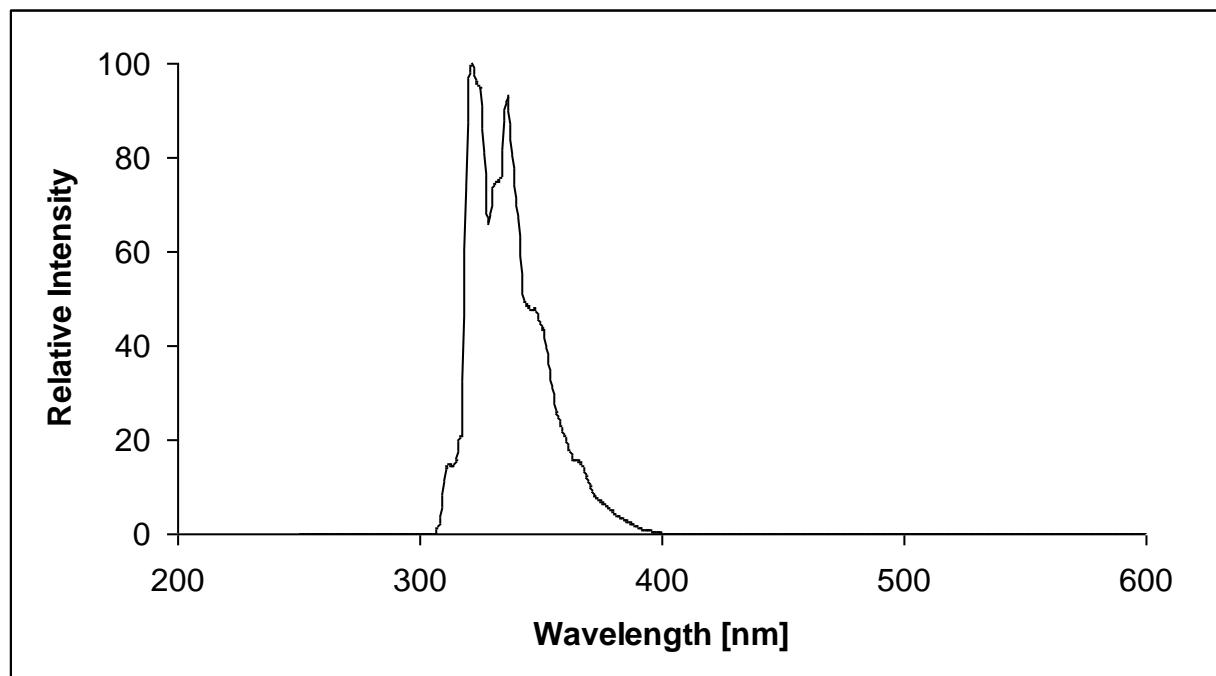
Naphthalene(1)



λ exc.	275 nm
Formula	$C_{10}H_8$
M.W.	128 u
H/H+C	0.444
m.p.	81°C
m.a.c. (266 nm)	0.62 ($l \cdot mol^{-1} cm^{-1} \times 10^{-4}$)
solvent	Cyclohexane
source	Atlas pg. 56-II

P02N02

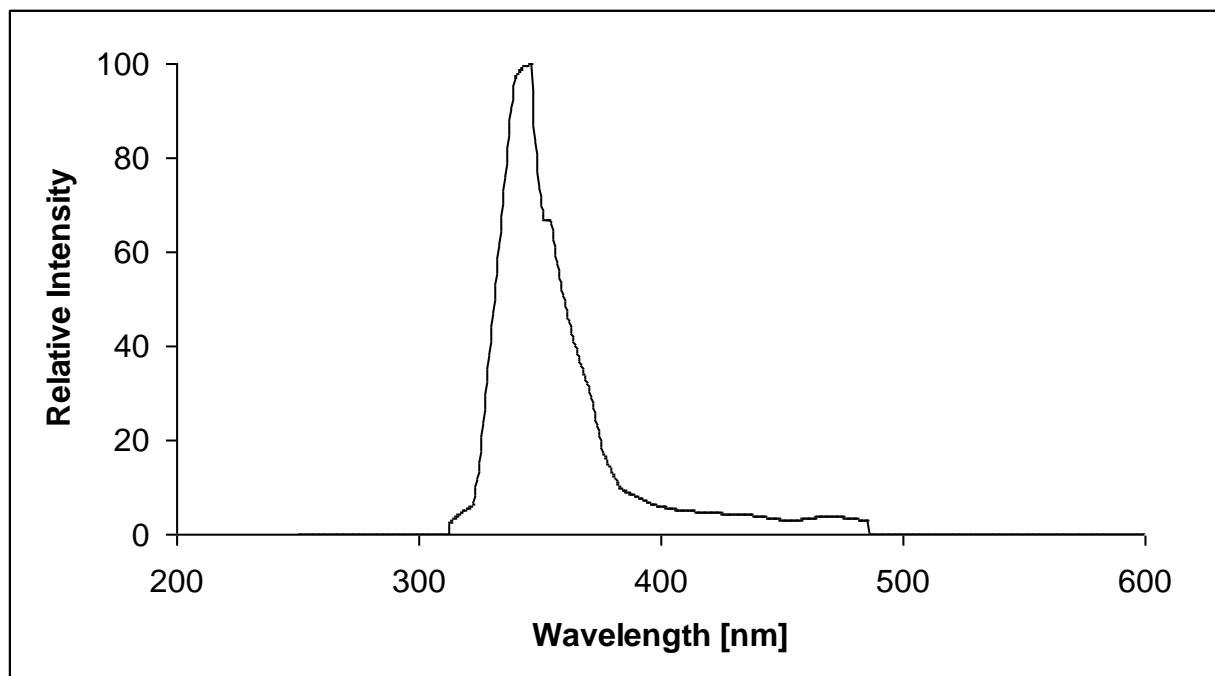
Naphthalene(2)



λ exc.	265 nm
Formula	$C_{10}H_8$
M.W.	128 u
H/H+C	0.444
m.p.	°C
m.a.c. (266 nm)	$0.535 \text{ (l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4})$
solvent	Cyclohexane
source	Berlman

P02N0A

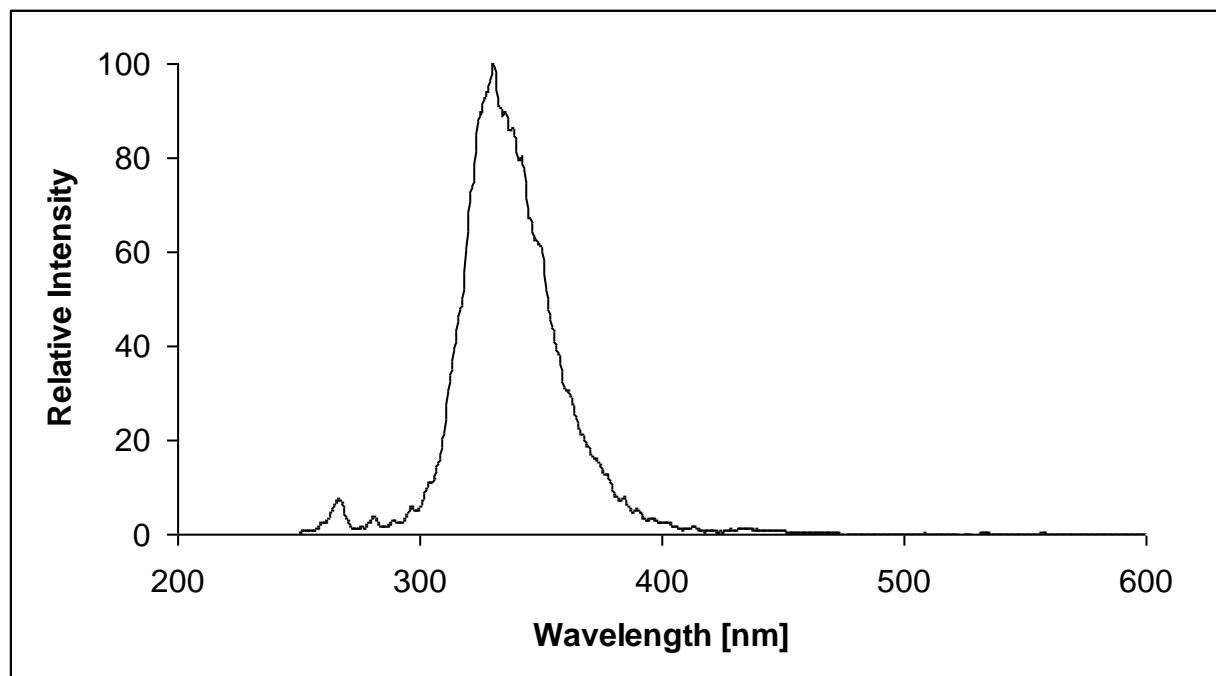
Naphthalene(3)



λ exc.	220 nm
Formula	$C_{10}H_8$
M.W.	128 u
H/H+C	0.444
m.p.	81°C
m.a.c. (266 nm)	(l. $mol^{-1} cm^{-1} \times 10^{-4}$)
solvent	Acetonitrile
source	[10]

P02N0G

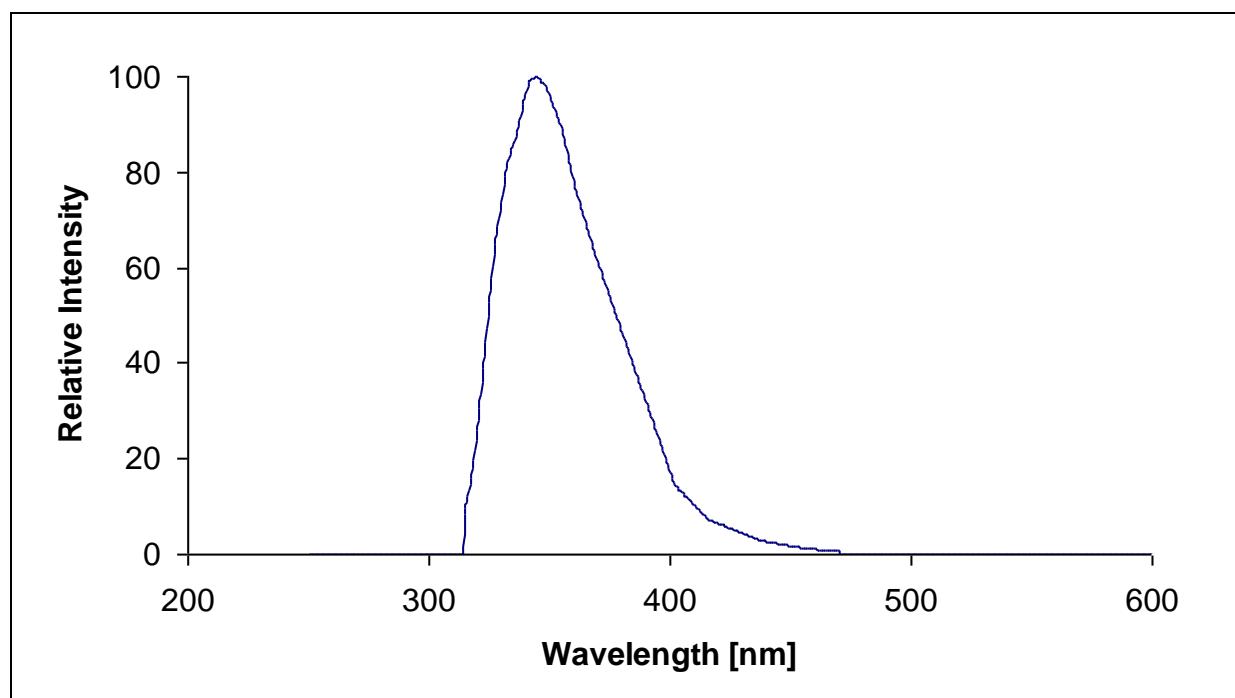
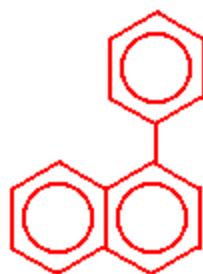
Naphthalene(4)



λ exc.	266 nm
Formula	$C_{10}H_8$
M.W.	128 u
H/H+C	0.444
m.p.	81°C
m.a.c. (266 nm)	0.62 ($l \cdot mol^{-1} cm^{-1} \times 10^{-4}$)
solvent	Gas phase 400 K
source	CNPM

P02NBA

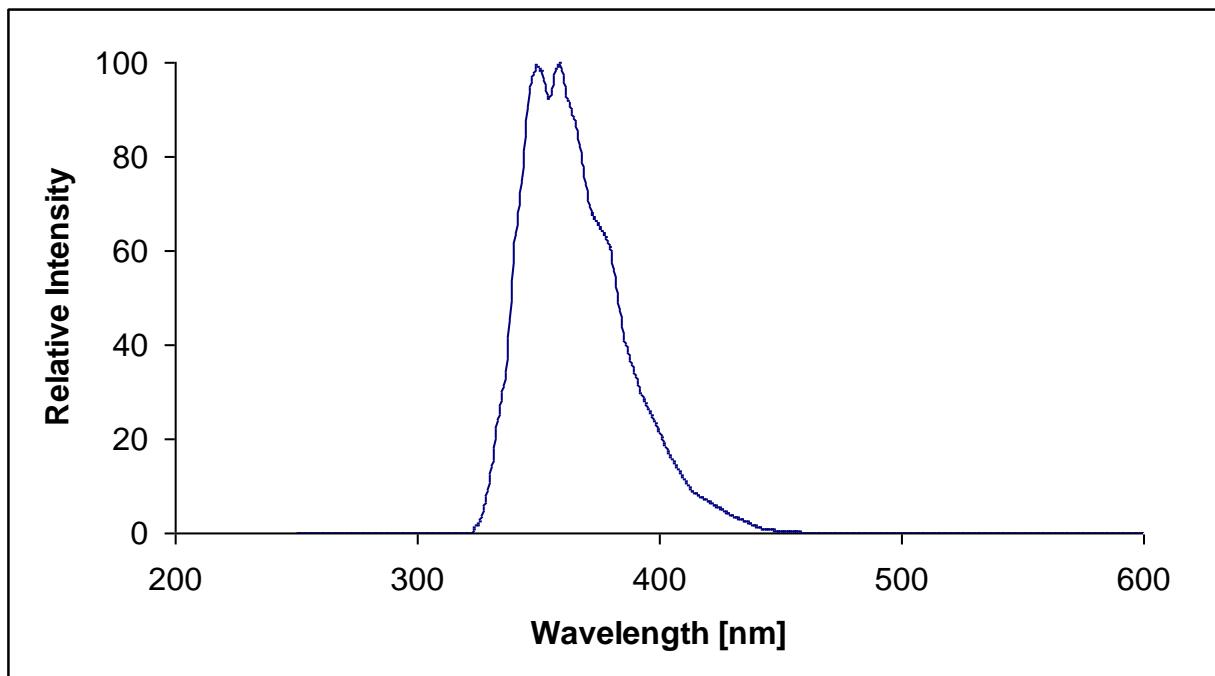
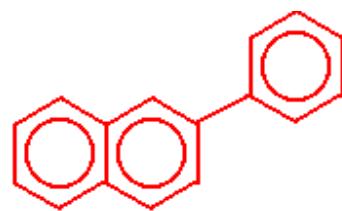
1-Phenylnaphthalene



λ exc.	303 nm
Formula	$C_{16}H_{12}$
M.W.	204 u
H/H+C	0.429
m.p.	°C
m.a.c. (266 nm)	$0.58 \text{ (l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4})$
solvent	Cyclohexane
source	Berlman

P02NBB

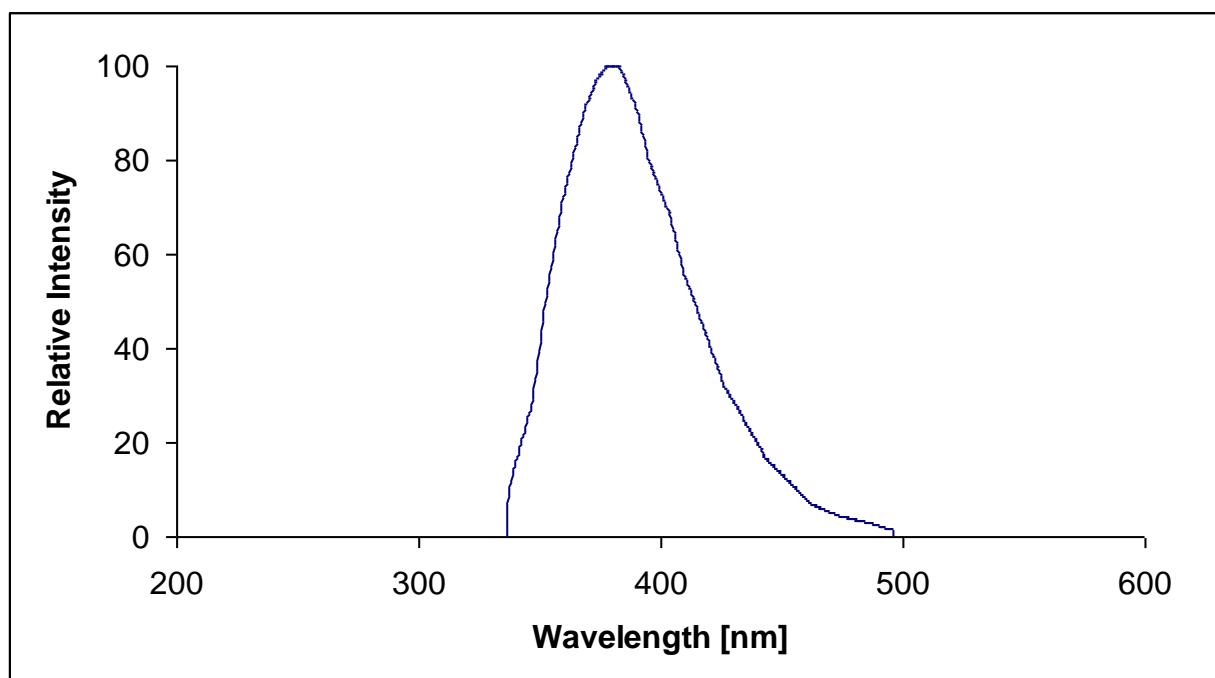
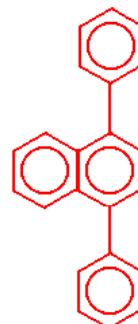
2 – Phenylnaphthalene



λ exc.	304 nm
Formula	C ₁₆ H ₁₂
M.W.	204 u
H/H+C	0.429
m.p.	°C
m.a.c. (266 nm)	11 (l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	Cyclohexane
source	Berlman

P02NBJ

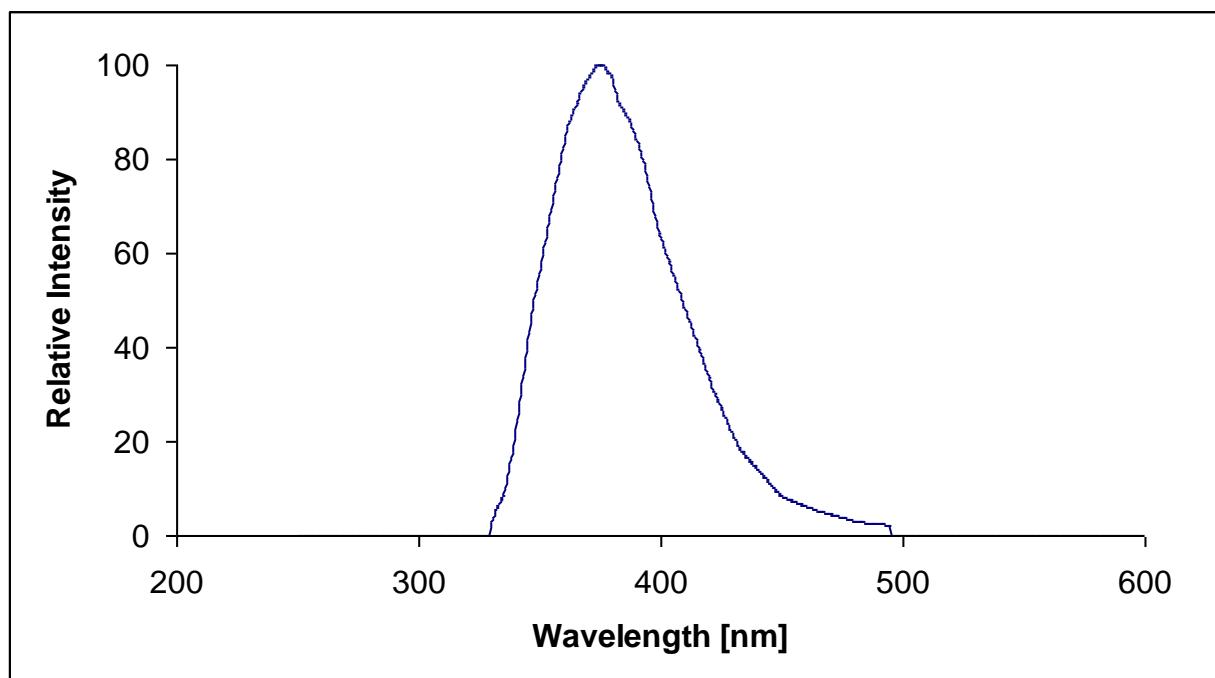
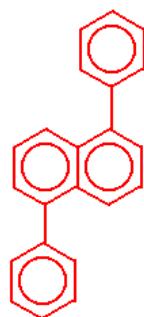
1,4 – Diphenylnaphthalene



λ exc.	313 nm
Formula	$C_{22}H_{16}$
M.W.	280 u
H/H+C	0.421
m.p.	°C
m.a.c. (266 nm)	$0.450 \text{ (l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4})$
solvent	Cyclohexane
source	Berlman

P02NBK

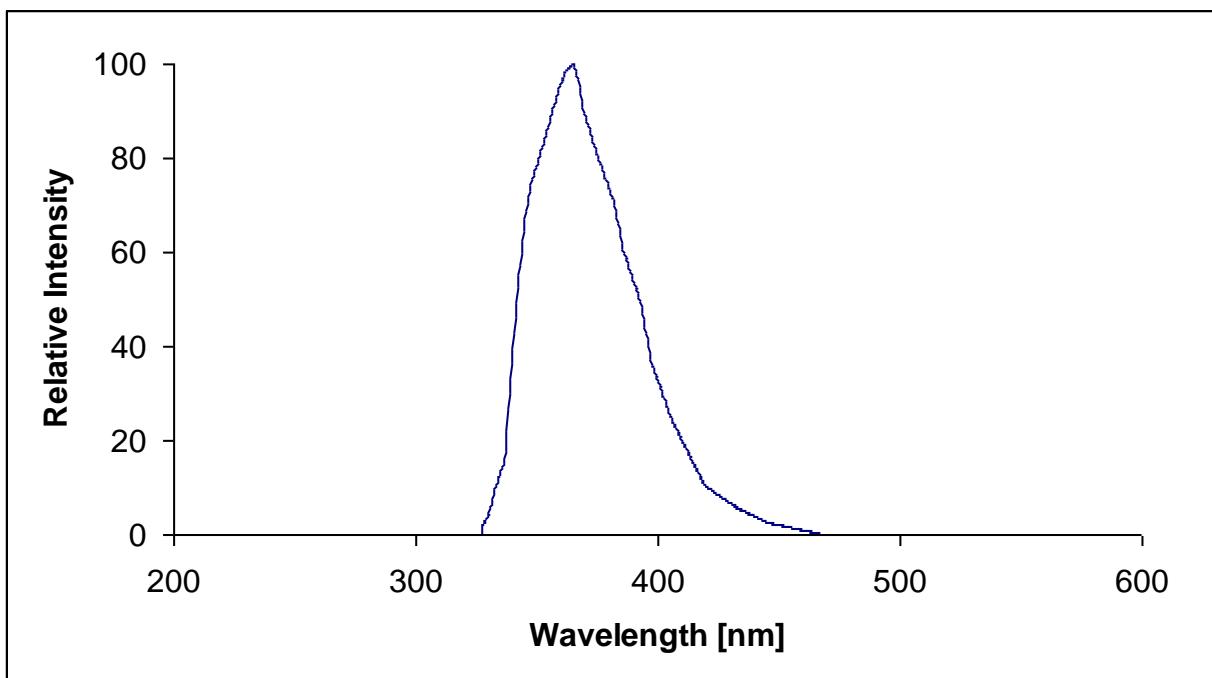
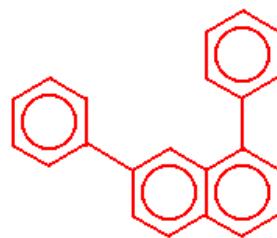
1,5 – Diphenylnaphthalene



λ exc.	304 nm
Formula	$C_{22}H_{16}$
M.W.	280 u
H/H+C	0.421
m.p.	°C
m.a.c. (266 nm)	0.4 ($l \cdot mol^{-1} cm^{-1} \times 10^{-4}$)
solvent	Cyclohexane
source	Berlman

P02NBL

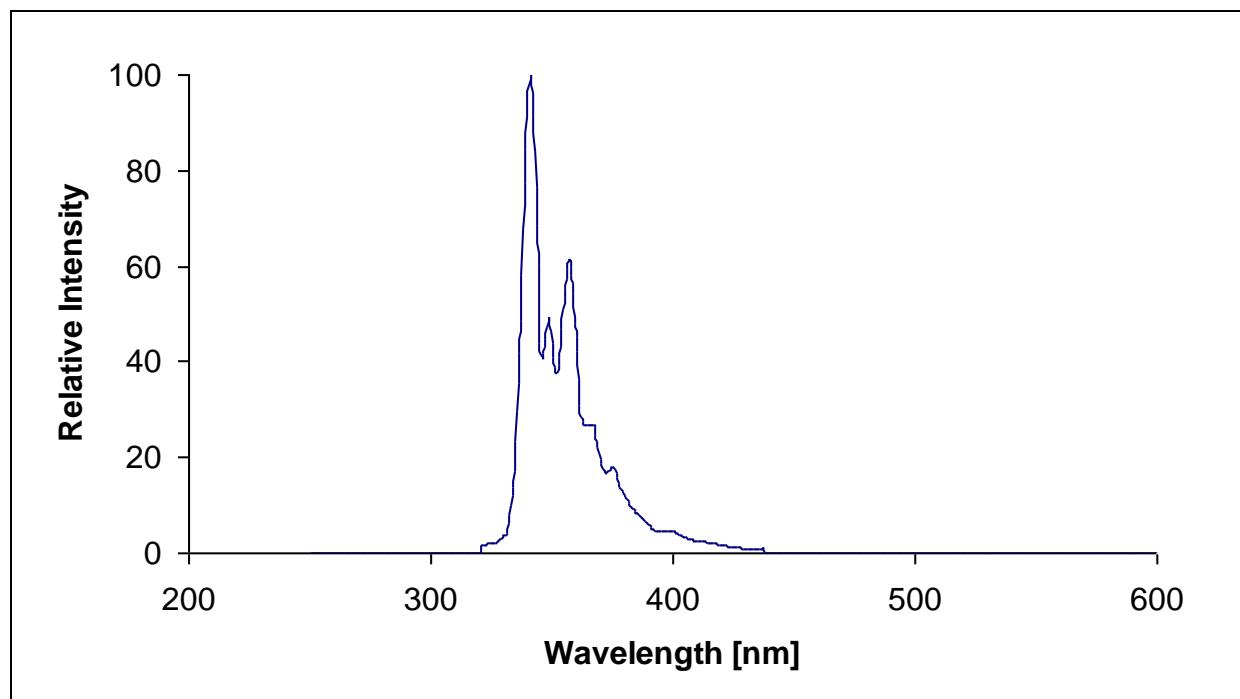
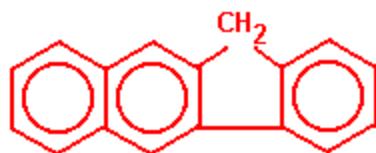
1,7 – Diphenylnaphthalene



λ exc.	303 nm
Formula	C ₂₂ H ₁₆
M.W.	280 u
H/H+C	0.421
m.p.	°C
m.a.c. (266 nm)	(l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	Cyclohexane
source	Berlman

P02NFA

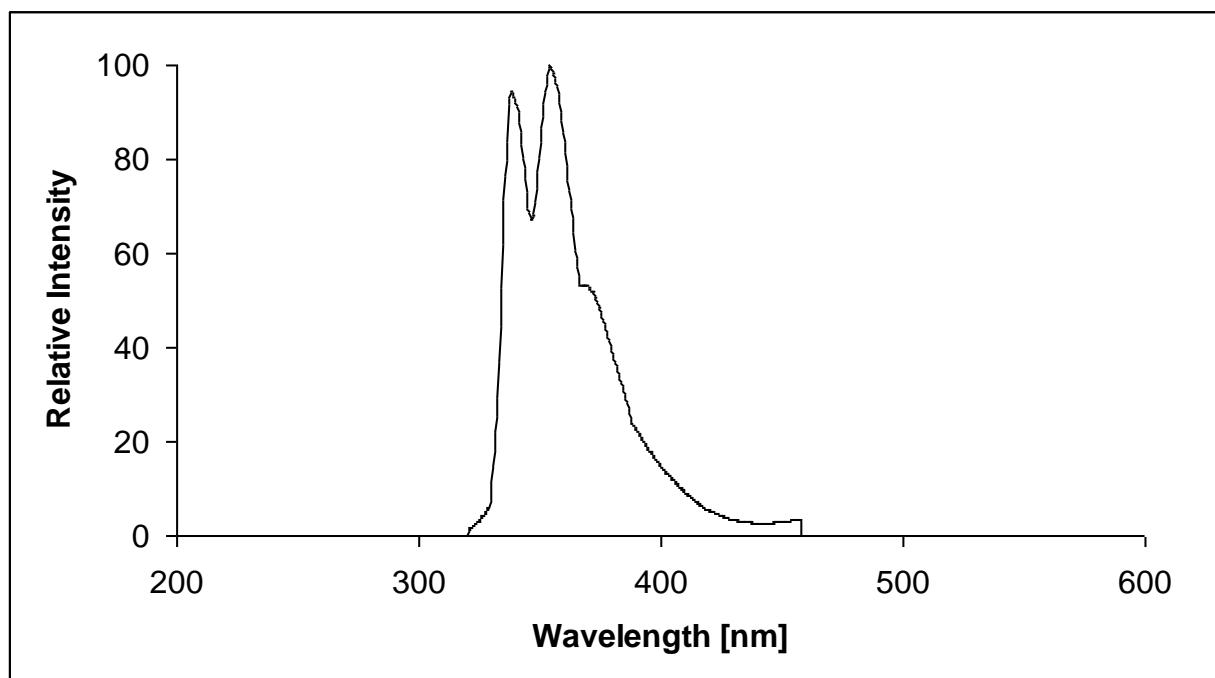
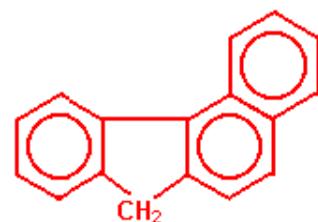
11H-Benzo(b)fluorene



λ exc.	304 nm
Formula	$\text{C}_{17}\text{H}_{12}$
M.W.	216 u
H/H+C	0.414
m.p.	213.5 °C
m.a.c. (266 nm)	4.65 ($\text{l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4}$)
solvent	Cyclohexane
source	Atlas pag. 176-II

P02NFB

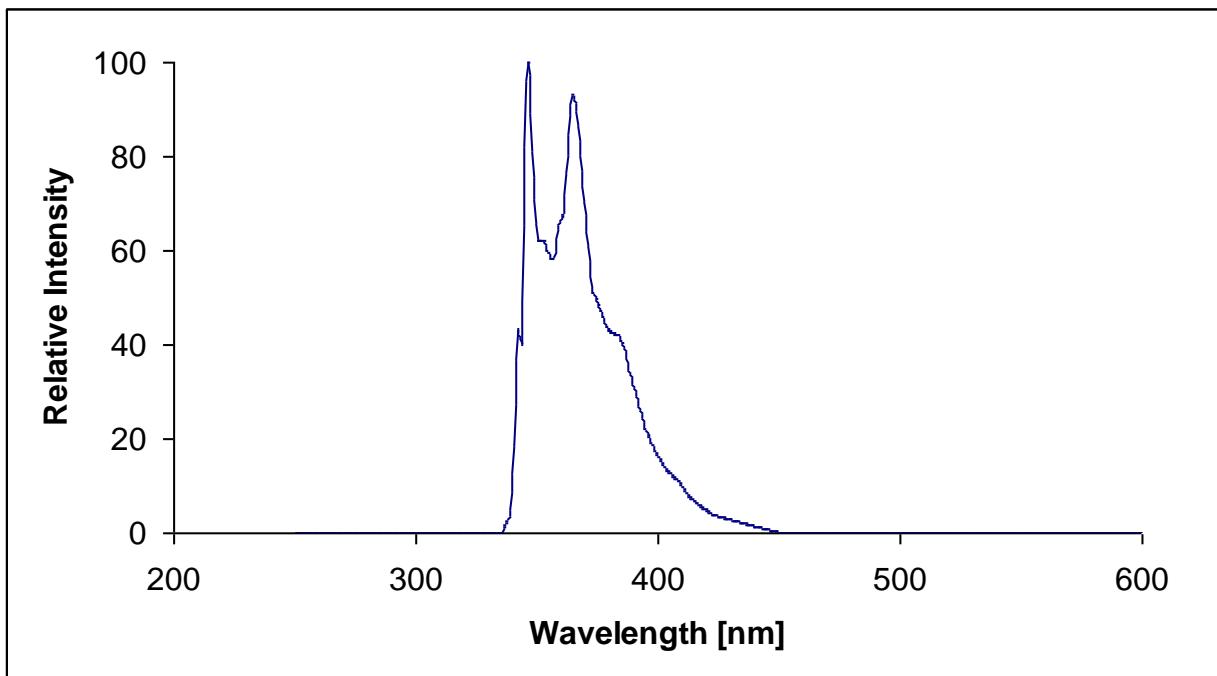
7H – Benzo(c)fluorene



λ exc.	311 nm
Formula	$\text{C}_{17}\text{H}_{12}$
M.W.	216 u
H/H+C	0.414
m.p.	126.5°C
m.a.c. (266 nm)	$0.65 (\text{l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4})$
solvent	Cyclohexane
source	Atlas pg. 194-II

P02NFC

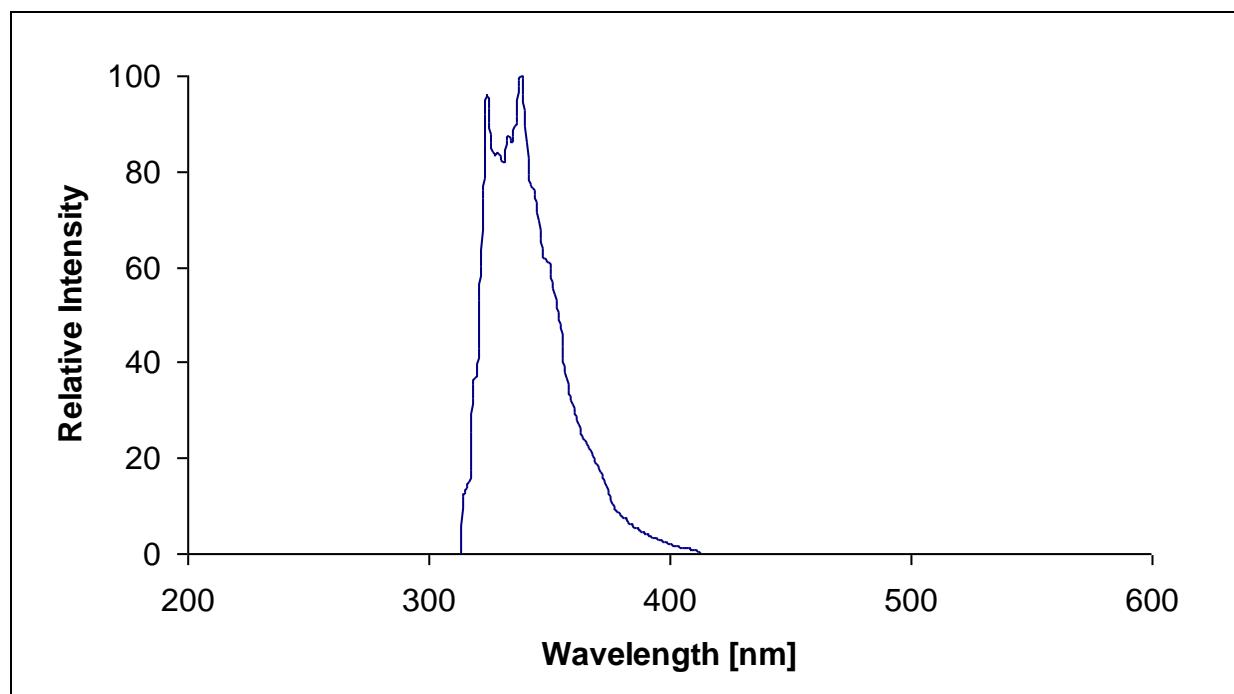
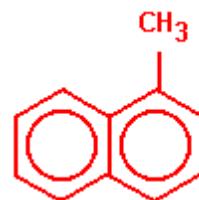
1,2 – Benzofluorene



λ exc.	313 nm
Formula	$C_{17}H_{12}$
M.W.	216 u
H/H+C	0.414
m.p.	°C
m.a.c. (266 nm)	(l. $mol^{-1} cm^{-1} \times 10^{-4}$)
solvent	Cyclohexane
source	Berlman

P02NMA

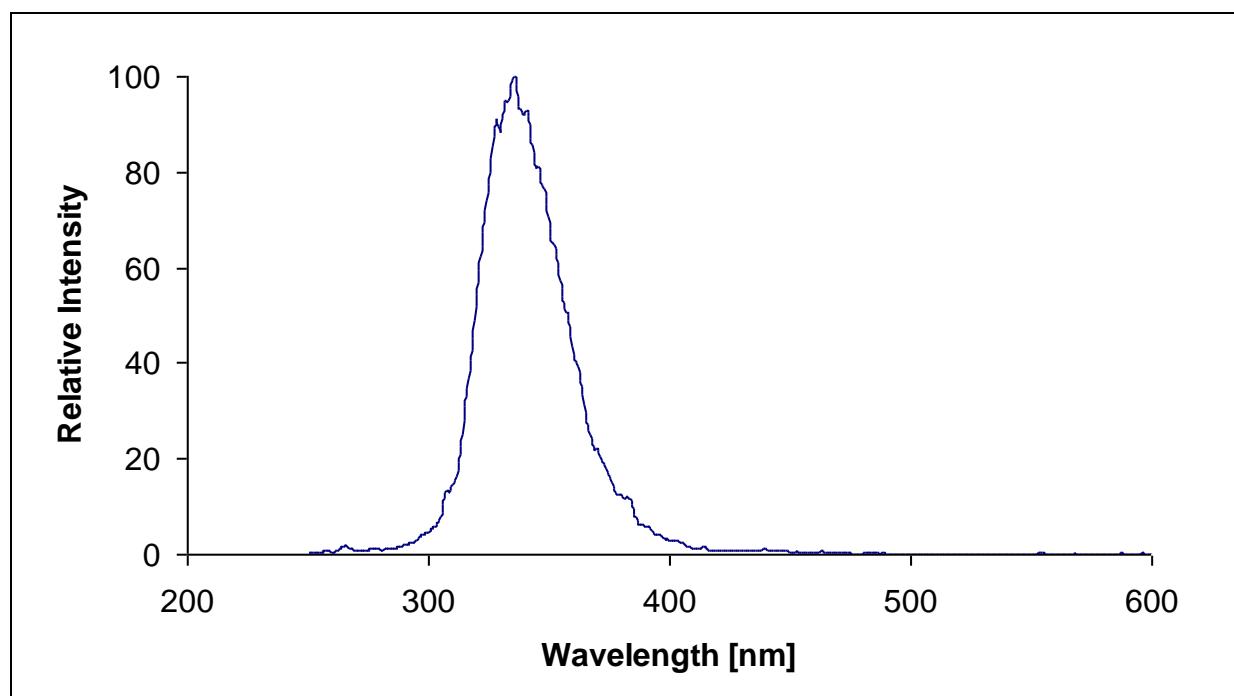
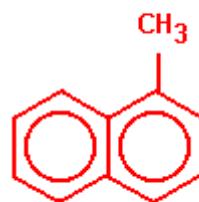
1-Methylnaphthalene (1)



λ exc.	265 nm
Formula	$\text{C}_{11}\text{H}_{10}$
M.W.	142 u
H/H+C	0.476
m.p.	°C
m.a.c. (266 nm)	0.4 ($\text{l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4}$)
solvent	Cyclohexane
source	Berlman

P02NMAG

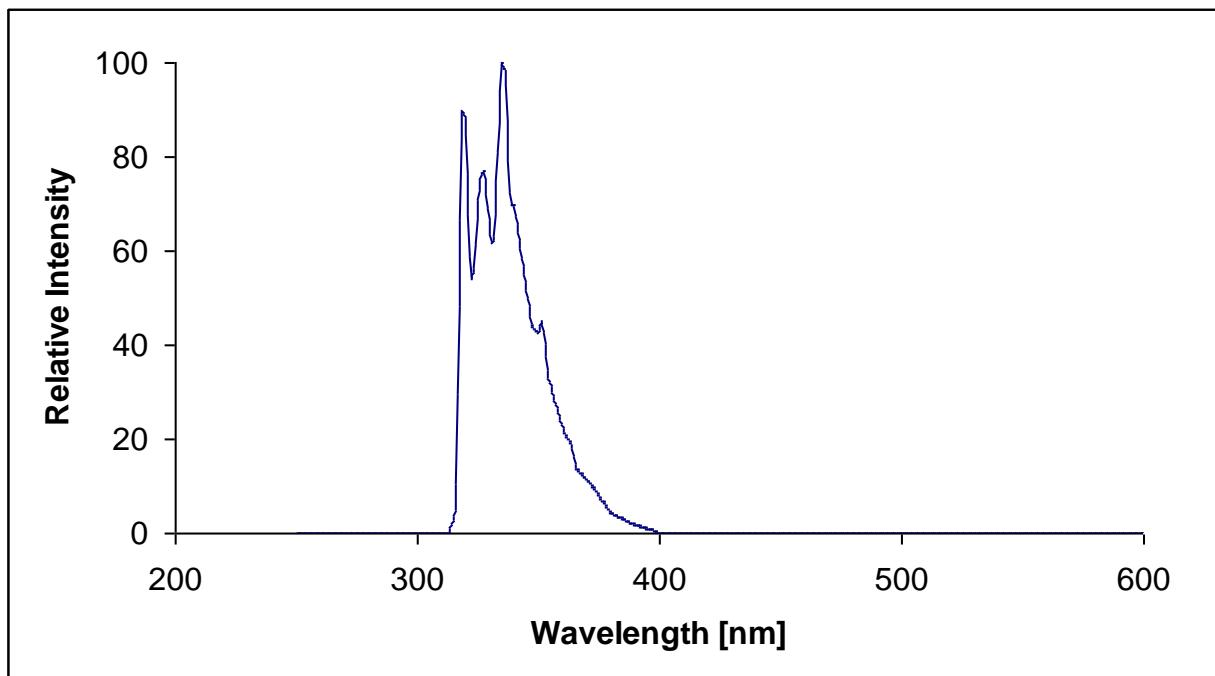
1-Methylnaphthalene (2)



λ exc.	266 nm
Formula	C ₁₁ H ₁₀
M.W.	142 u
H/H+C	0.476
m.p.	°C
m.a.c. (266 nm)	0.4 (l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	Gas phase 400 K
source	CNPM

P02NMB

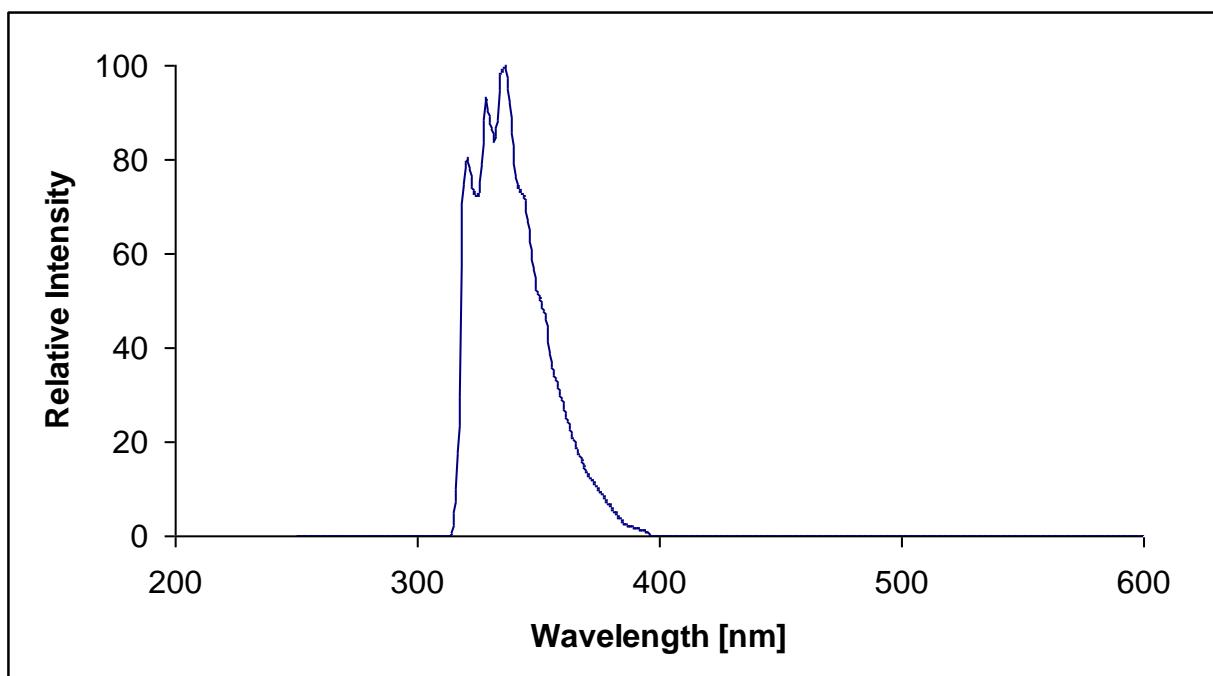
2 – Methylnaphthalene



λ exc.	265 nm
Formula	C ₁₁ H ₁₀
M.W.	142 u
H/H+C	0.476
m.p.	°C
m.a.c. (266 nm)	0.47 (l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	Cyclohexane
source	Berlman

P02NMJ

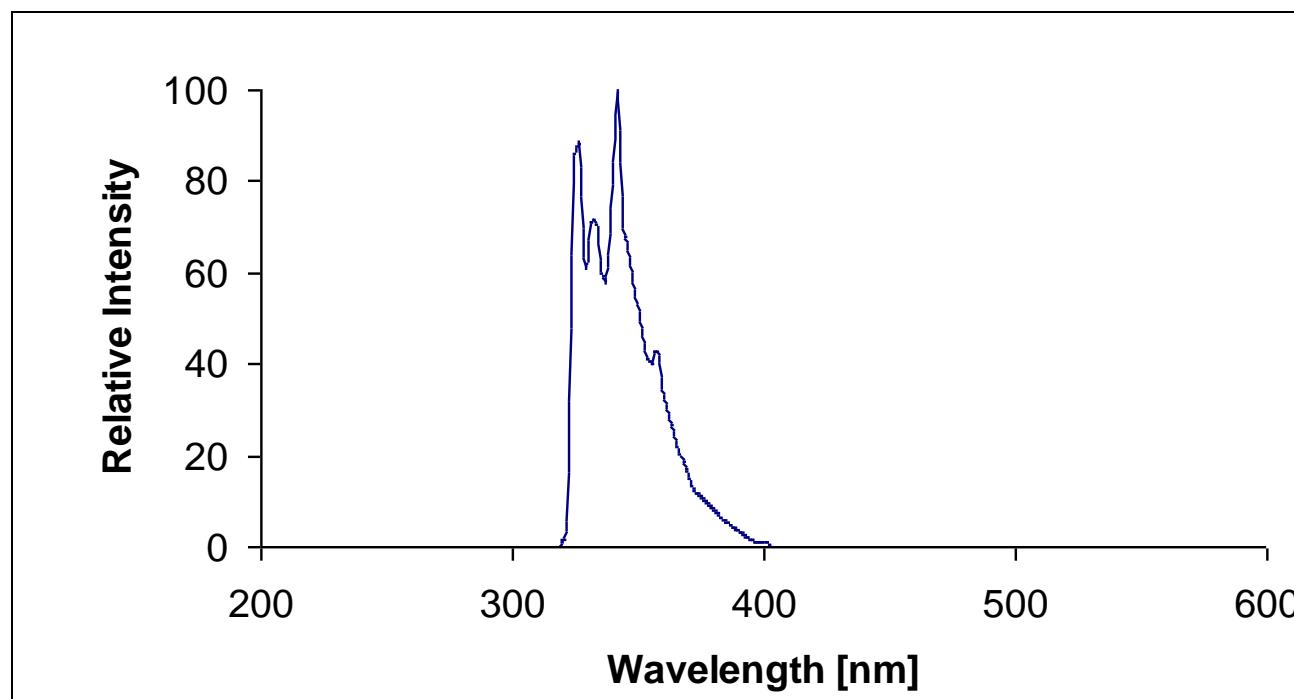
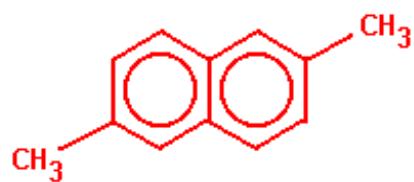
2,3 – dimethylnaphthalene



λ exc.	265 nm
Formula	$\text{C}_{12}\text{H}_{12}$
M.W.	156 u
H/H+C	0.500
m.p.	°C
m.a.c. (266 nm)	$0.475 (\text{l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4})$
solvent	Cyclohexane
source	Berlman

P02NMK

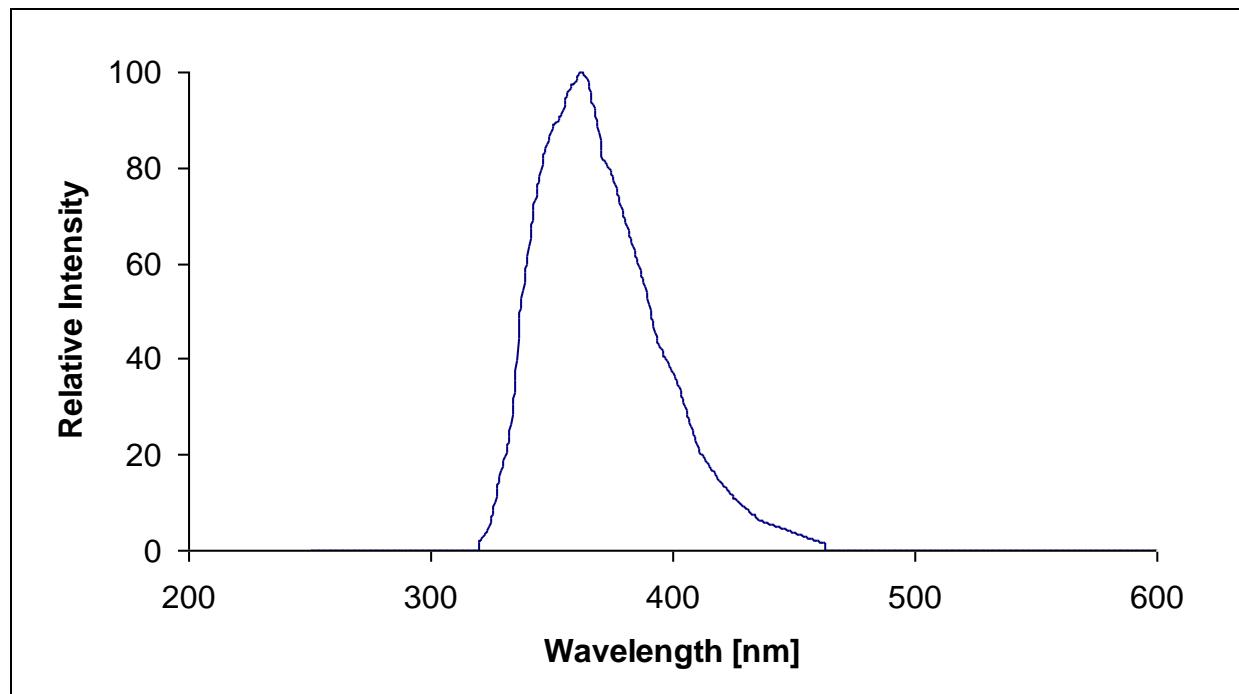
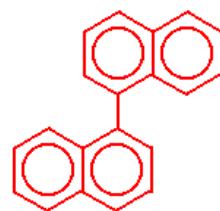
2,6 – Dimethylnaphthalene



λ exc.	265 nm
Formula	C ₁₂ H ₁₂
M.W.	156 u
H/H+C	0.500
m.p.	°C
m.a.c. (266 nm)	0.515 (l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	Cyclohexane
source	Berlman

P02NNB

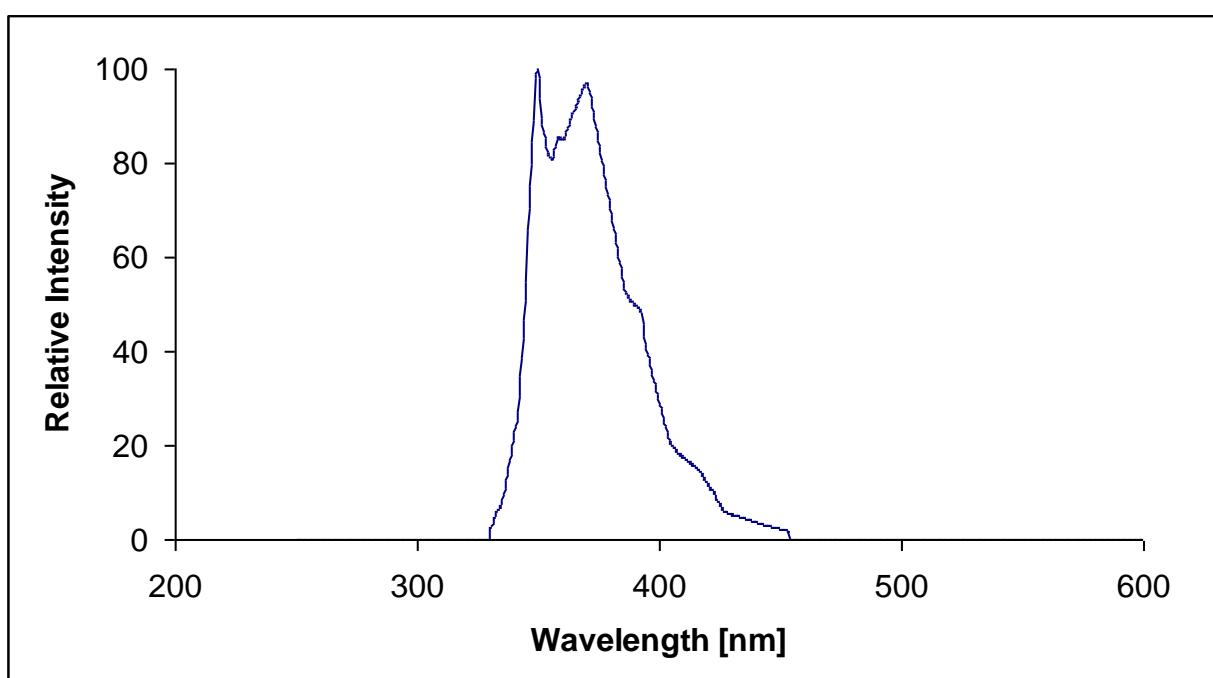
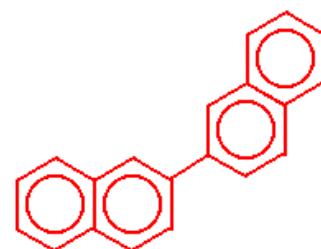
1,1' – Binaphthyl



λ exc.	303 nm
Formula	C ₂₀ H ₁₄
M.W.	254 u
H/H+C	0.412
m.p.	°C
m.a.c. (266 nm)	0.795 (l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	Cyclohexane
source	Berlman

P02NNC

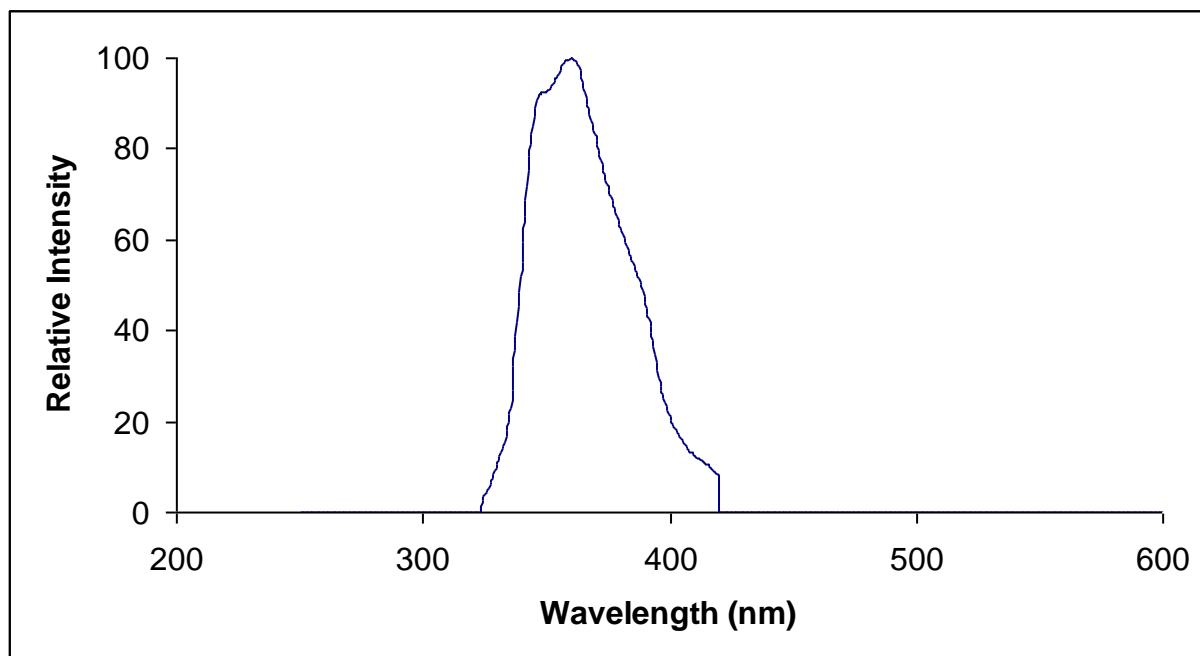
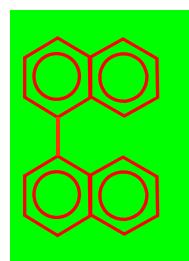
2,2' – Binaphthyl



λ exc.	265 nm
Formula	$C_{20}H_{14}$
M.W.	254 u
H/H+C	0.412
m.p.	°C
m.a.c. (266 nm)	$3.65 \text{ (l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4})$
solvent	Cyclohexane
source	Berlman

P02NND

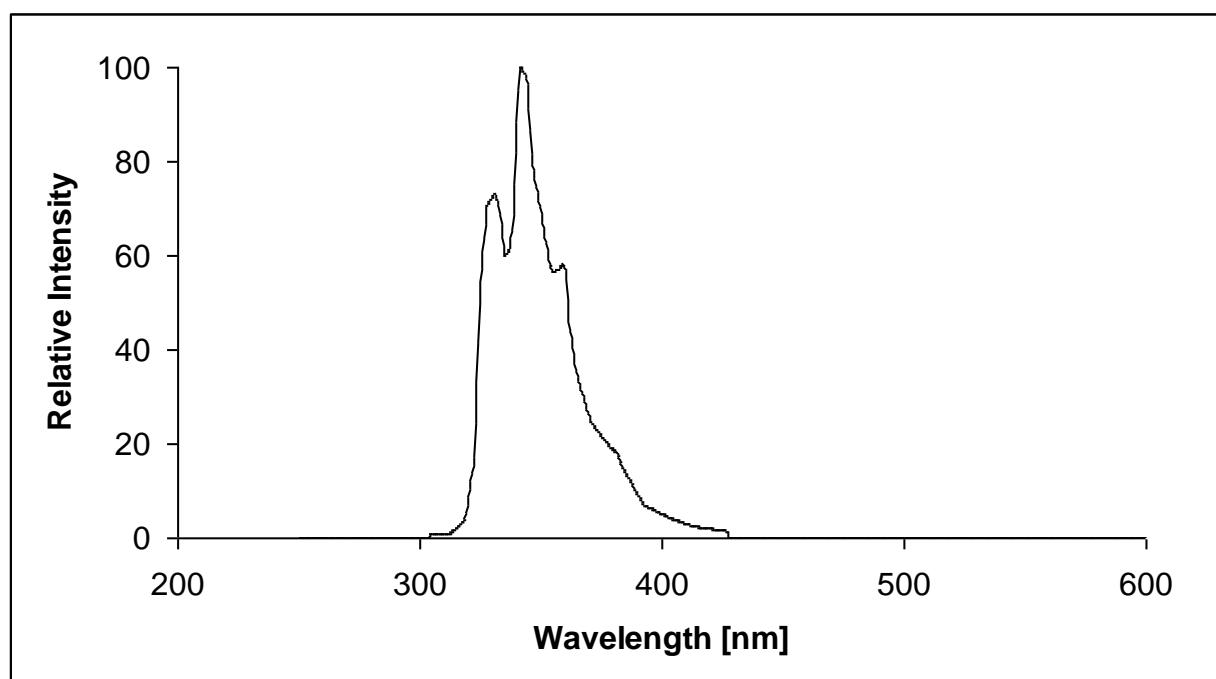
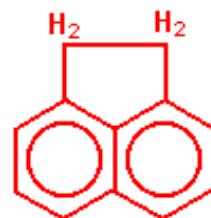
1,1' Binaphthalene



λ exc.	300 nm
Formula	$C_{20}H_{14}$
M.W.	254 u
H/H+C	0.412
m.p.	°C
m.a.c. (266 nm)	(l. $mol^{-1} cm^{-1} \times 10^{-4}$)
solvent	n-hexadecane
source	[15]

P02P1

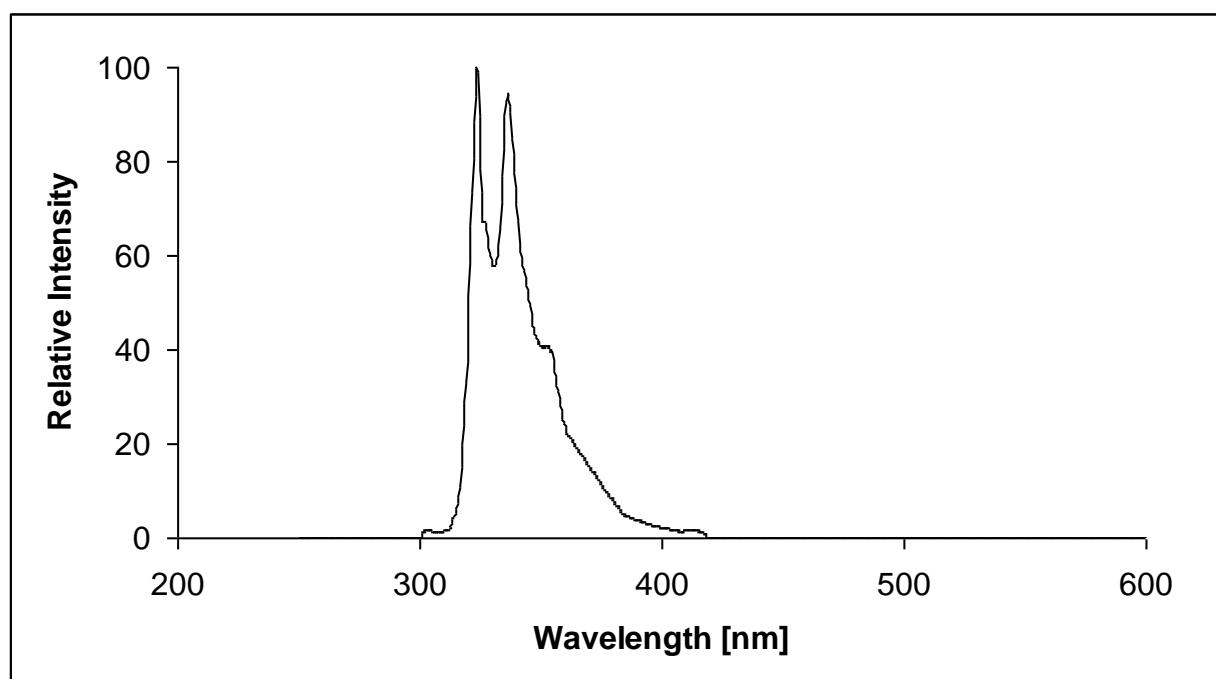
Acenaphthene (1)



λ exc.	266 nm
Formula	$C_{12}H_{10}$
M.W.	154 u
H/H+C	0.455
m.p.	95°C
m.a.c. (266 nm)	(l. $mol^{-1} cm^{-1} \times 10^{-4}$)
solvent	Cyclohexane
source	[11]

P02P2

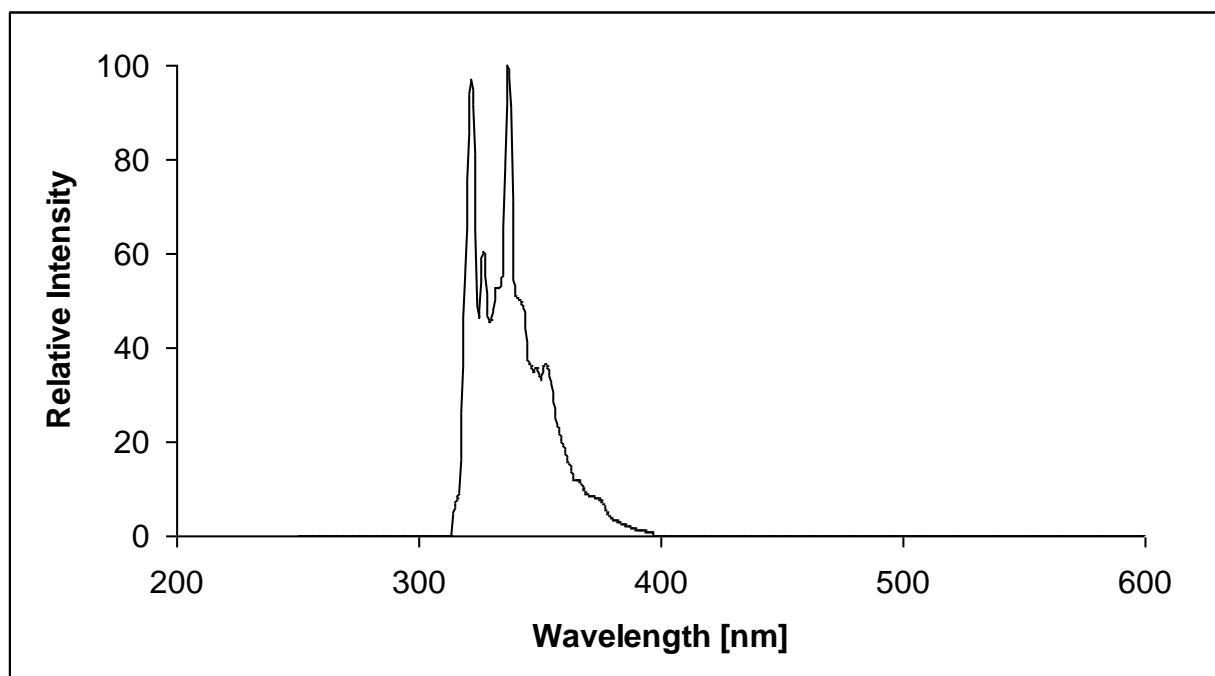
Acenaphthene (2)



λ exc.	289 nm
Formula	C ₁₂ H ₁₀
M.W.	154 u
H/H+C	0.455
m.p.	95°C
m.a.c. (266 nm)	0.5 (l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	Cyclohexane
source	Atlas pg. 90-II

P02P3

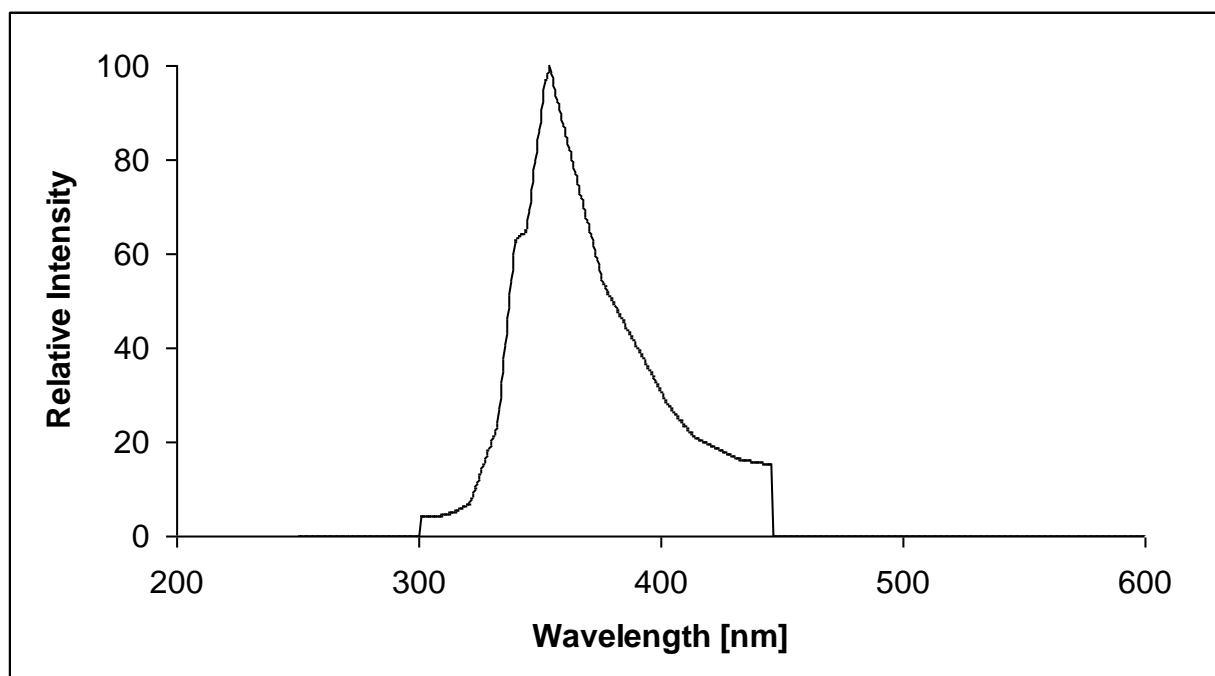
Acenaphthene (3)



λ exc.	303 nm
Formula	C ₁₂ H ₁₀
M.W.	154 u
H/H+C	0.455
m.p.	°C
m.a.c. (266 nm)	0.36 (l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	Cyclohexane
source	Berlman

P02T1

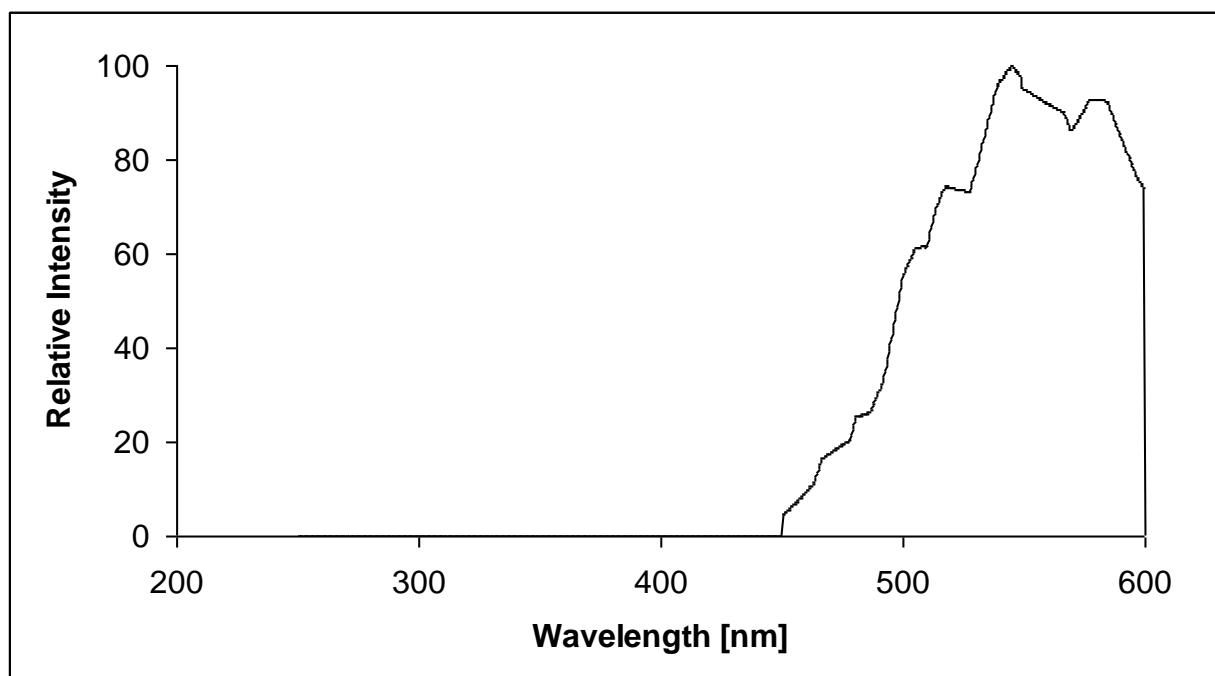
Acenaphthylene (1)



λ exc.	266 nm
Formula	$C_{12}H_8$
M.W.	152 u
H/H+C	0.400
m.p.	92-93°C
m.a.c. (266 nm)	(l. $mol^{-1} cm^{-1} \times 10^{-4}$)
solvent	Cyclohexane
source	[11]

P02T2

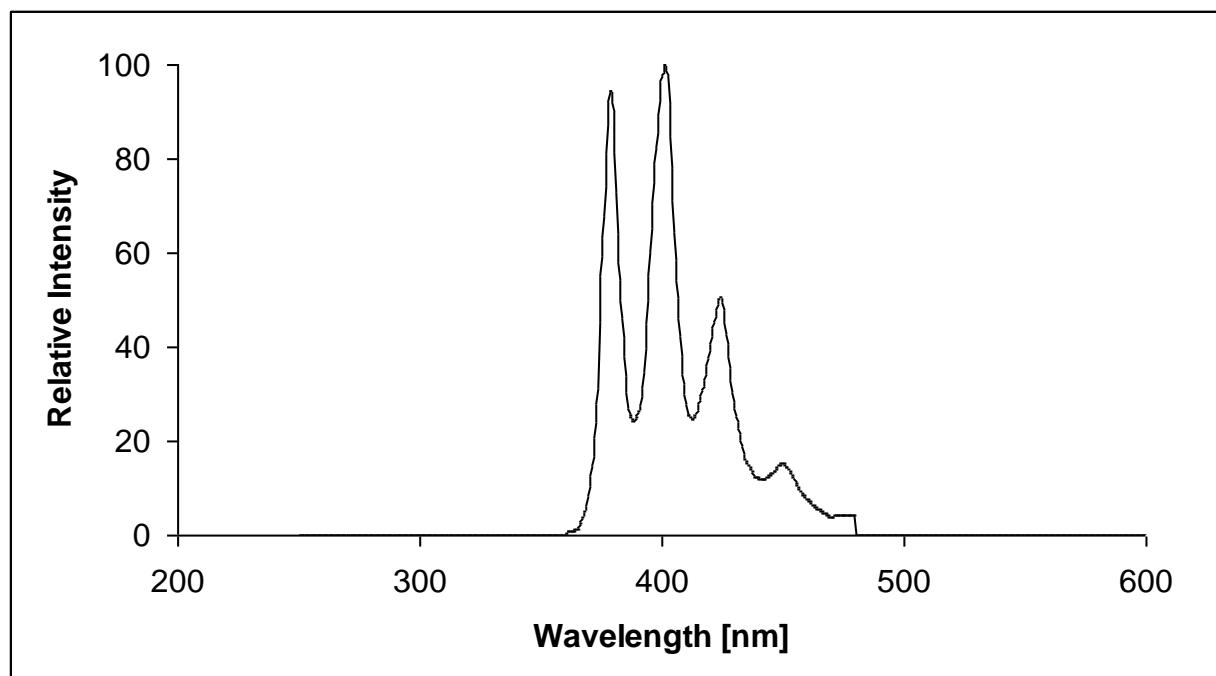
Acenaphthylene (2)



λ exc.	355 nm
Formula	$C_{12}H_8$
M.W.	152 u
H/H+C	0.400
m.p.	92-93°C
m.a.c. (266 nm)	(l. $mol^{-1} cm^{-1} \times 10^{-4}$)
solvent	Cyclohexane
source	[11]

P03A0

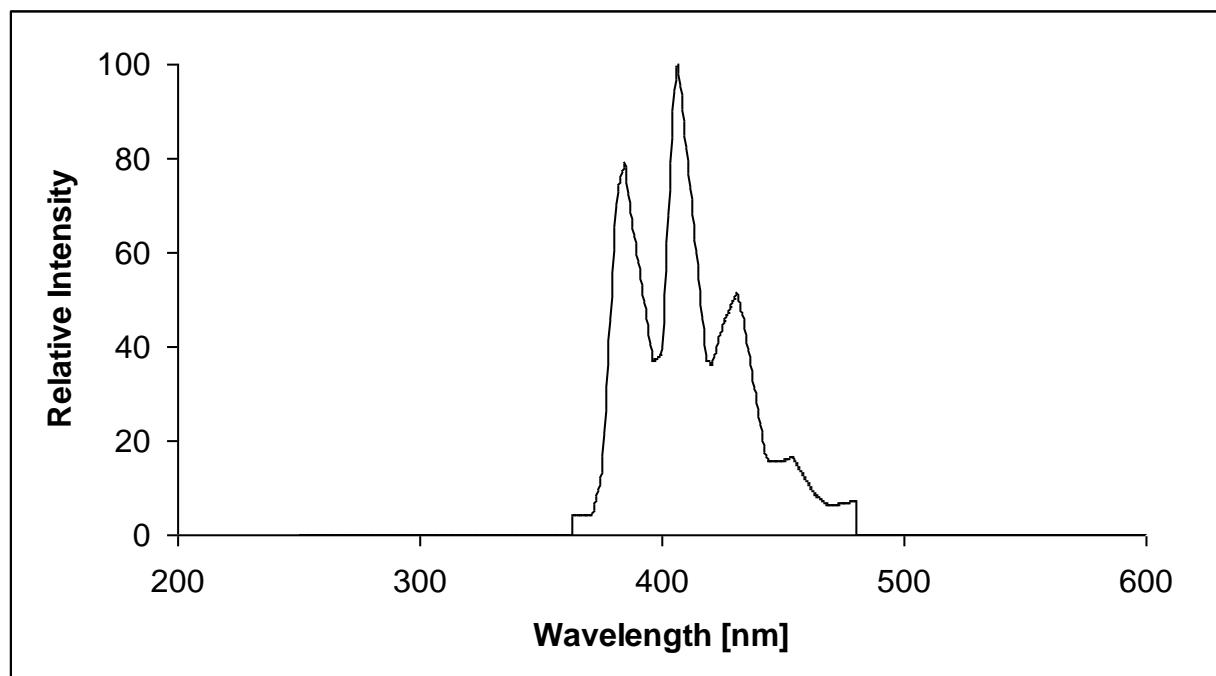
Anthracene (1)



λ exc.	252 nm
Formula	C ₁₄ H ₁₀
M.W.	178 u
H/H+C	0.417
m.p.	216.4°C
m.a.c. (266 nm)	0.2 (l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	Cyclohexane
source	Atlas pg. 32-I

P03A0A

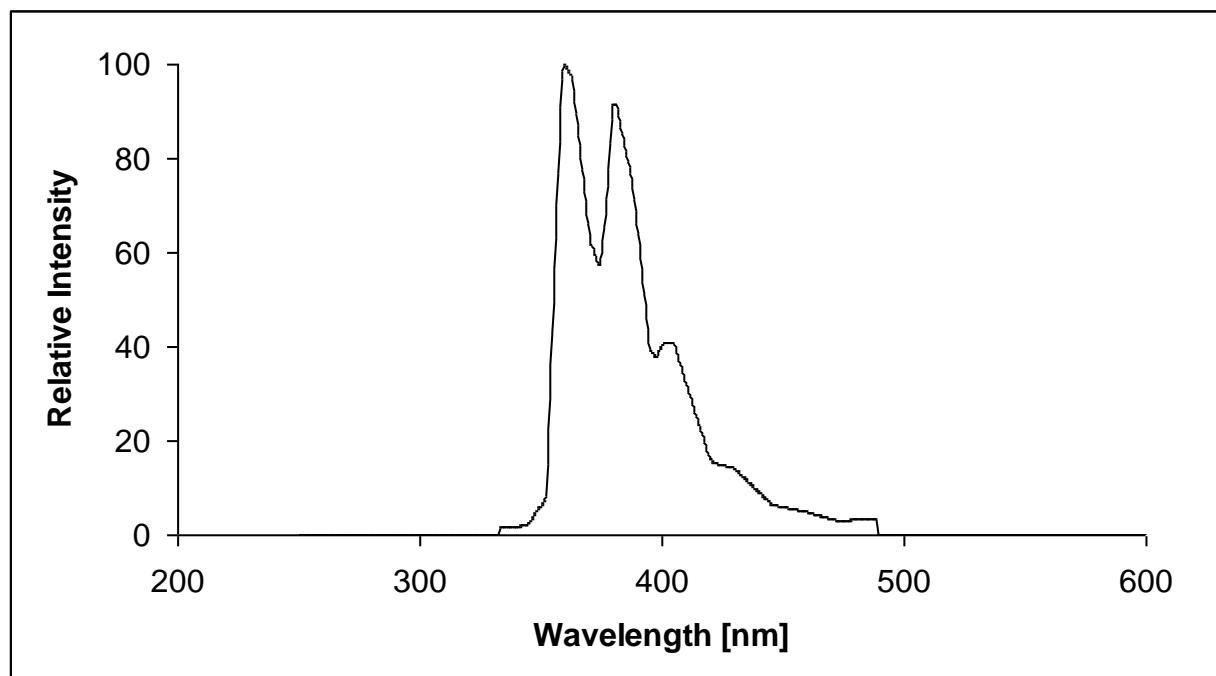
Anthracene (2)



λ exc.	220 nm
Formula	$C_{14}H_{10}$
M.W.	178 u
H/H+C	0.417
m.p.	216.4°C
m.a.c. (266 nm)	(l. $mol^{-1} cm^{-1} \times 10^{-4}$)
solvent	Acetonitrile
source	[10]

P03A0G

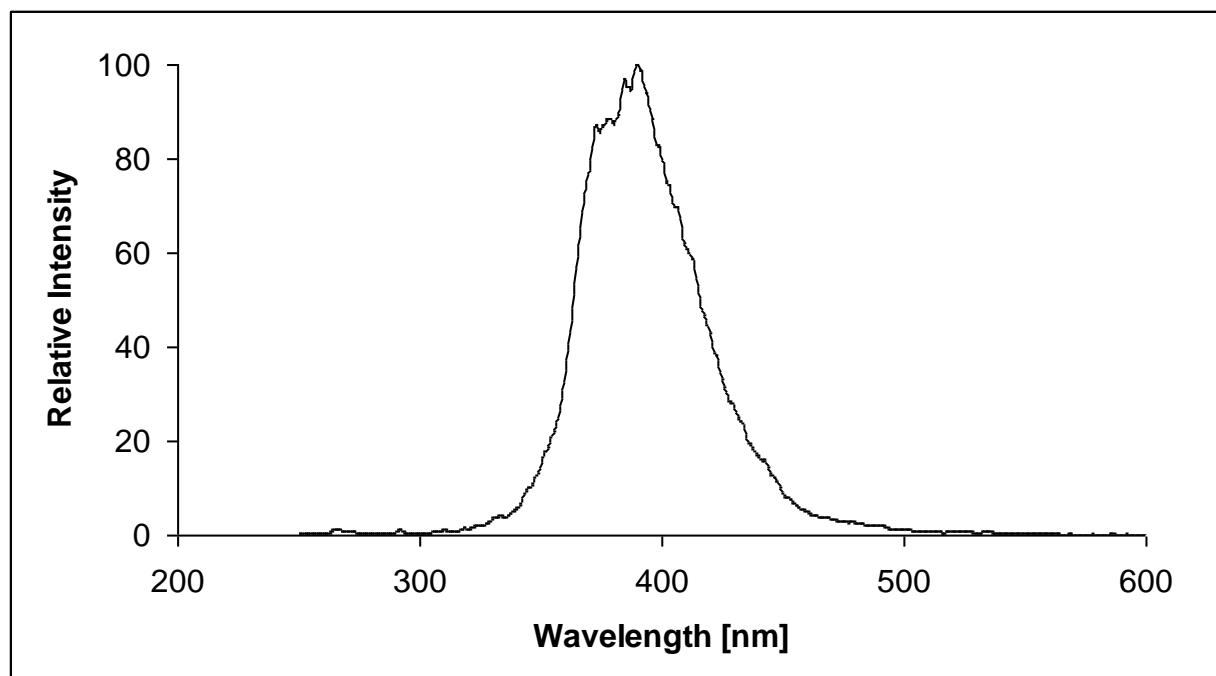
Anthracene (3)



λ exc.	337.1 nm
Formula	C ₁₄ H ₁₀
M.W.	178 u
H/H+C	0.417
m.p.	216.4°C
m.a.c. (266 nm)	(l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	Gas phase 298-423 K
source	[2]

P03A0G2

Anthracene (4)

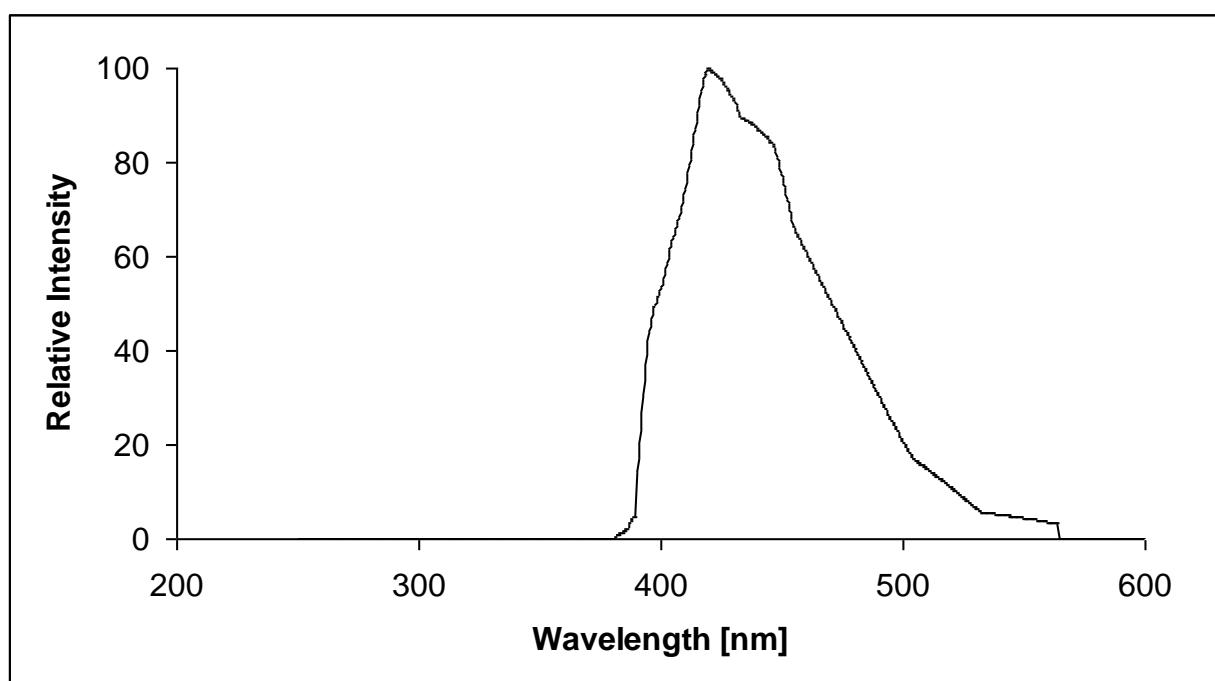
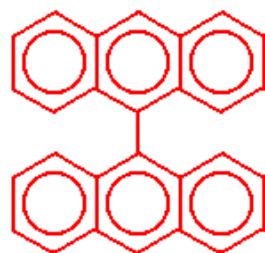


λ exc.	266 nm
Formula	$C_{14}H_{10}$
M.W.	178 u
H/H+C	0.417
m.p.	216.4°C
m.a.c. (266 nm)	(l. $mol^{-1} cm^{-1} \times 10^{-4}$)
solvent	Gas phase
source	CNPM

P03AAB

9,9' – Biantharanyl(1)

9,9' - Bianthracene

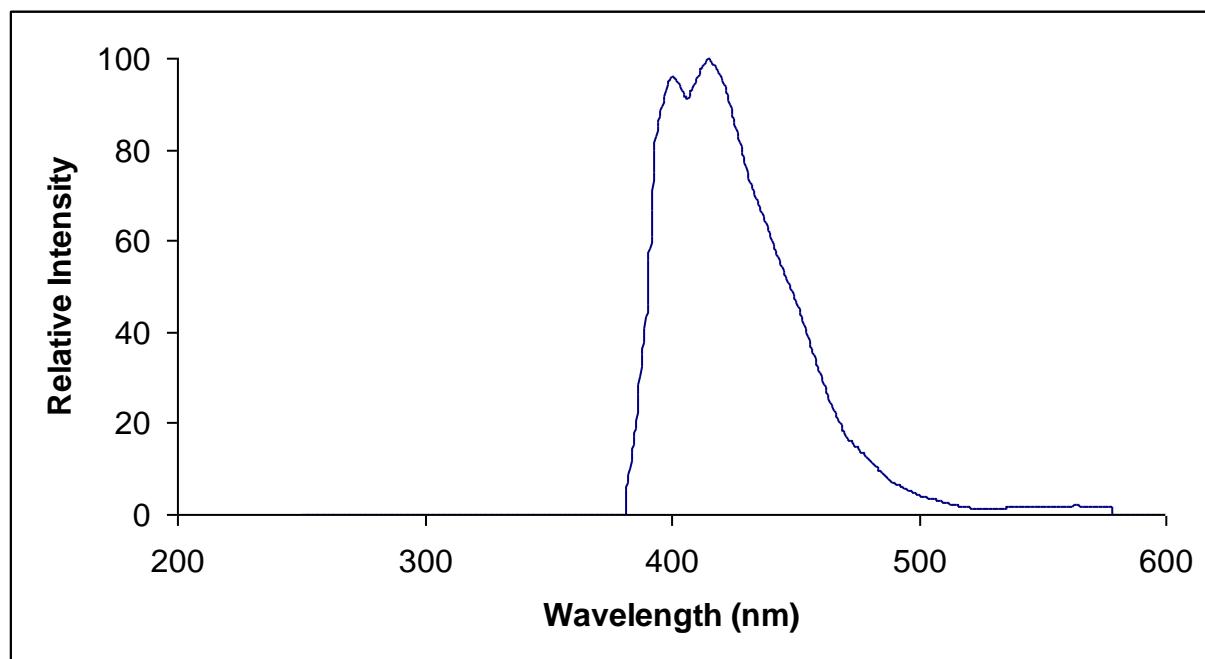
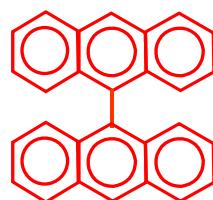


λ exc.	365 nm
Formula	$C_{28}H_{18}$
M.W.	354 u
H/H+C	0.391
m.p.	°C
m.a.c. (266 nm)	(l. $mol^{-1} cm^{-1} \times 10^{-4}$)
solvent	Benzene
source	Berlman

P03AABC

9,9' – Biantharanyl (2)

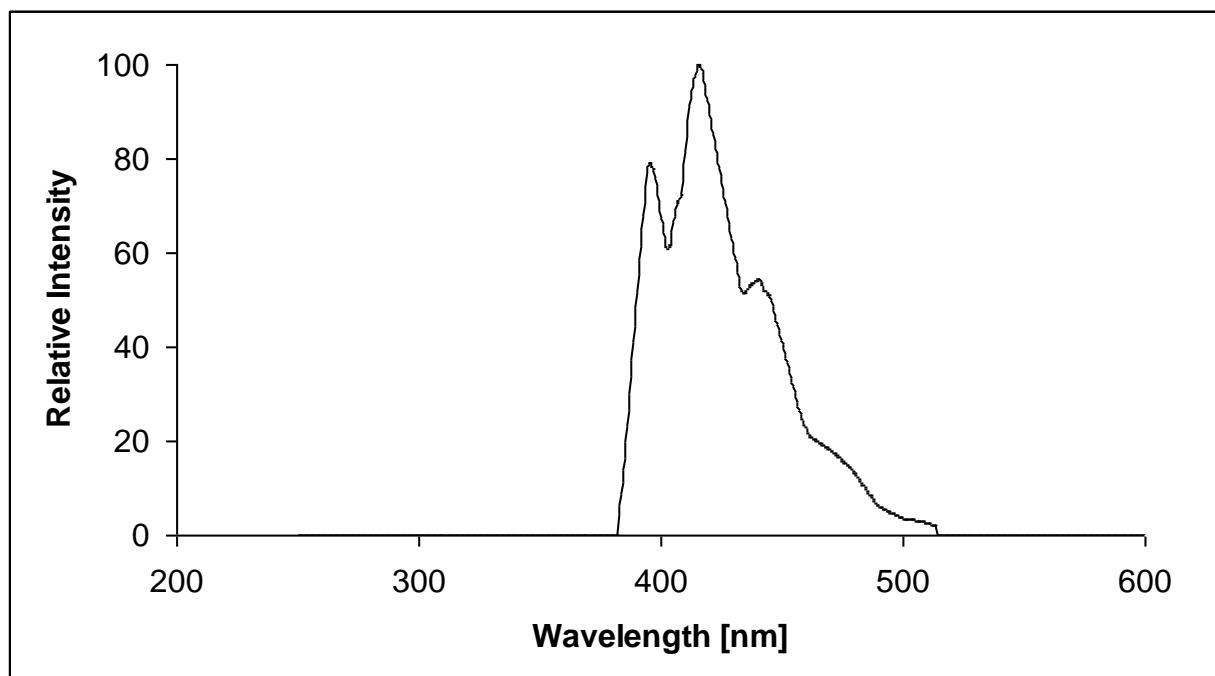
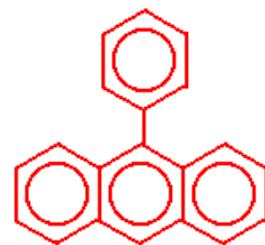
9,9' - Bianthracene



λ exc.	366 nm
Formula	$C_{28}H_{18}$
M.W.	354 u
H/H+C	0.391
m.p.	°C
m.a.c. (266 nm)	(l. $mol^{-1} cm^{-1} \times 10^{-4}$)
solvent	n-hexadecane
source	[15]

P03ABA

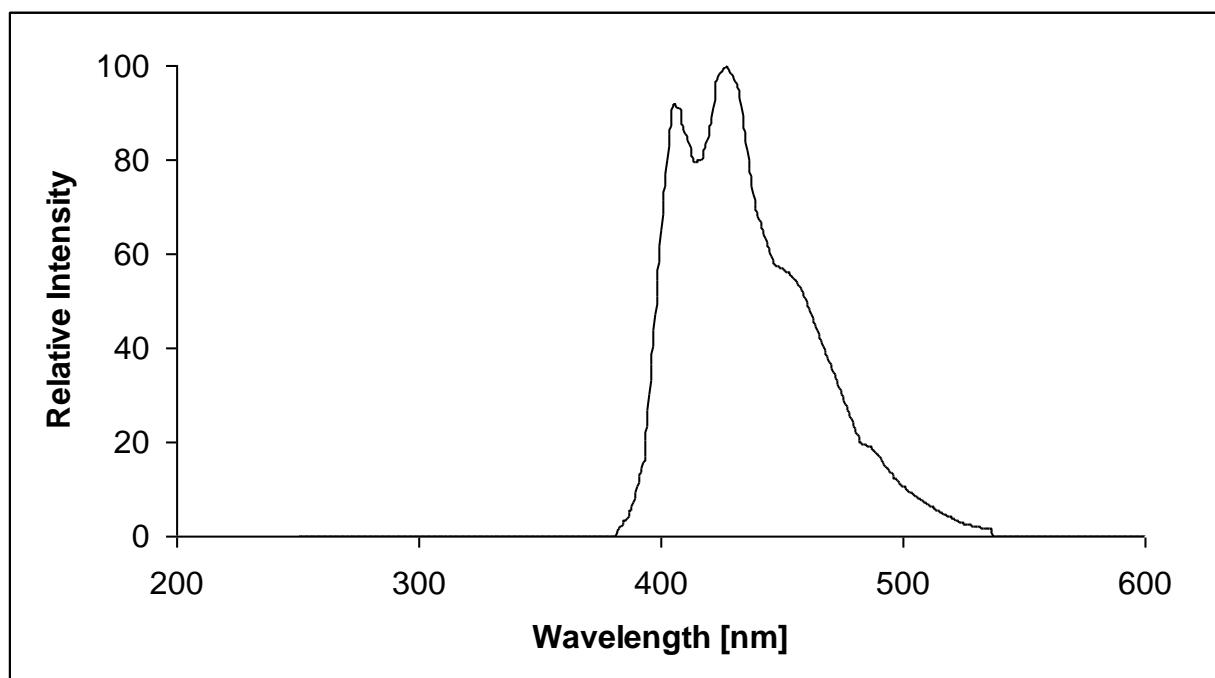
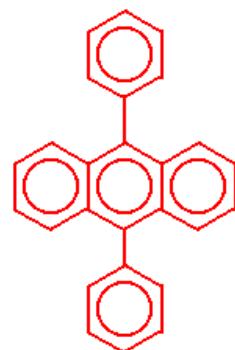
9 – Phenylanthracene



λ exc.	365 nm
Formula	$C_{20}H_{14}$
M.W.	254 u
H/H+C	0.412
m.p.	°C
m.a.c. (266 nm)	(l. $mol^{-1} cm^{-1} \times 10^{-4}$)
solvent	Cyclohexane
source	Berlman

P03ABJ

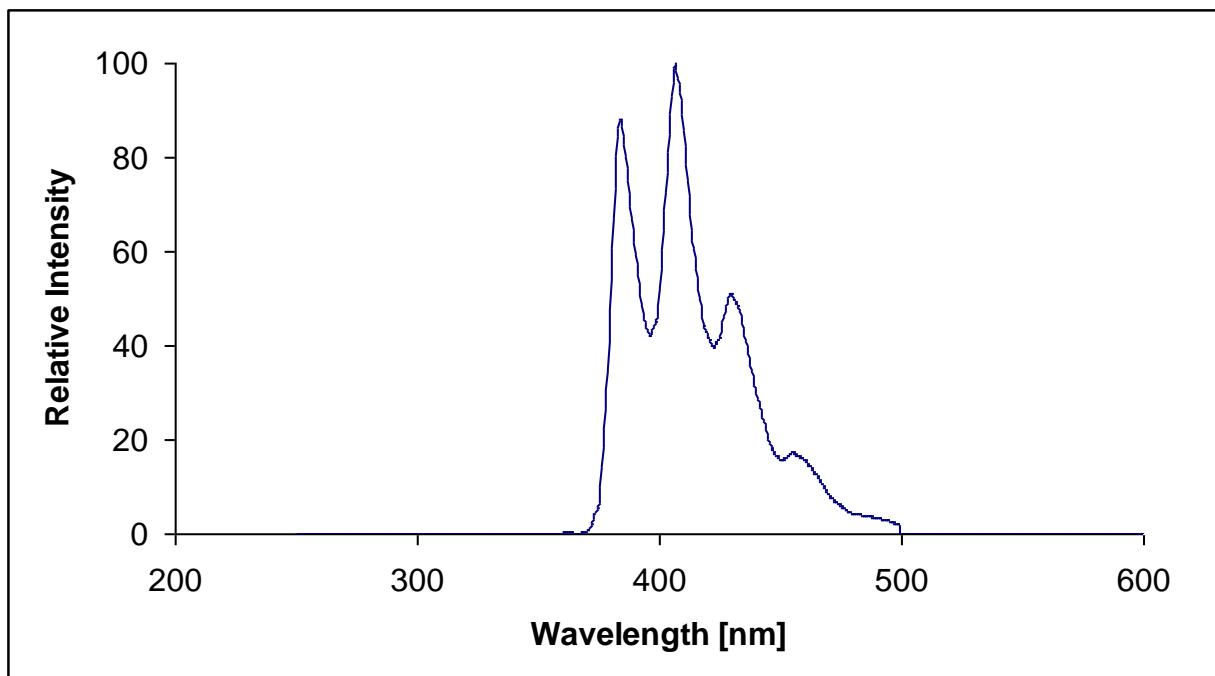
9,10 – Diphenylanthracene



λ exc.	365 nm
Formula	$C_{26}H_{18}$
M.W.	330 u
H/H+C	0.409
m.p.	°C
m.a.c. (266 nm)	(l. $mol^{-1} cm^{-1} \times 10^{-4}$)
solvent	Cyclohexane
source	Berlman

P03AMA

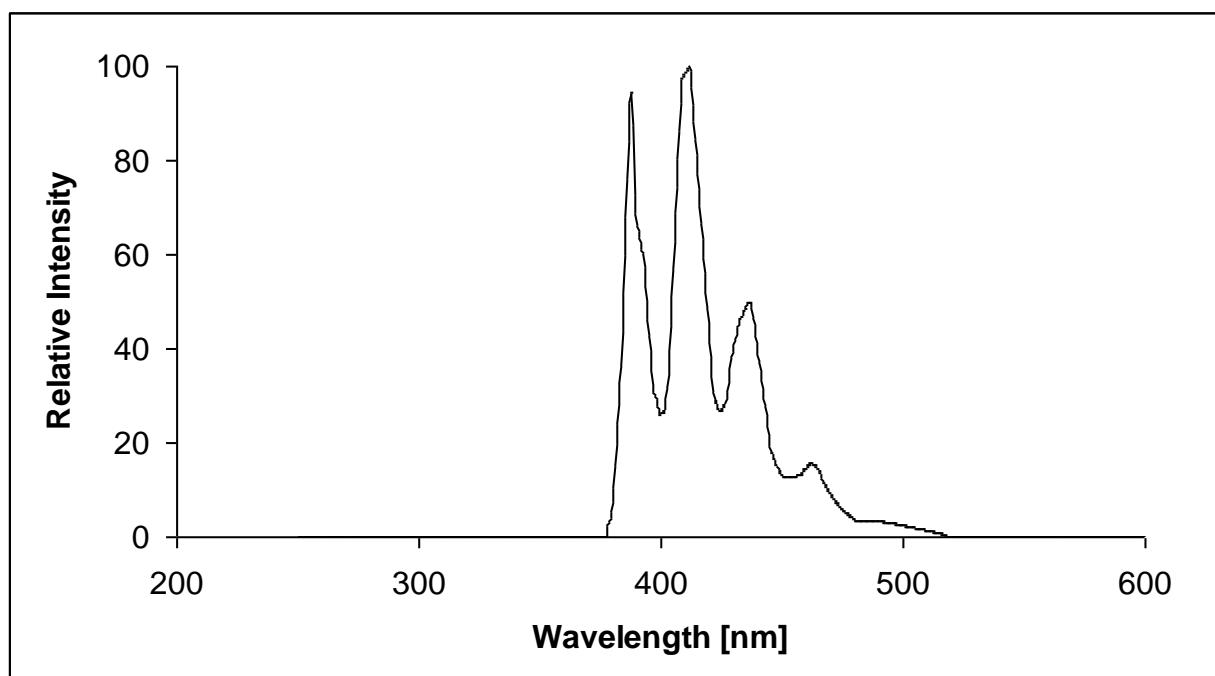
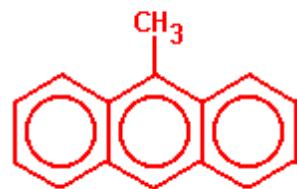
2 – Methylanthracene



λ exc.	255.5 nm
Formula	$\text{C}_{15}\text{H}_{12}$
M.W.	192 u
H/H+C	0.444
m.p.	203°C
m.a.c. (266 nm)	$3.5 (\text{l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4})$
solvent	Cyclohexane
source	Atlas pg. 140-II

P03AMB1

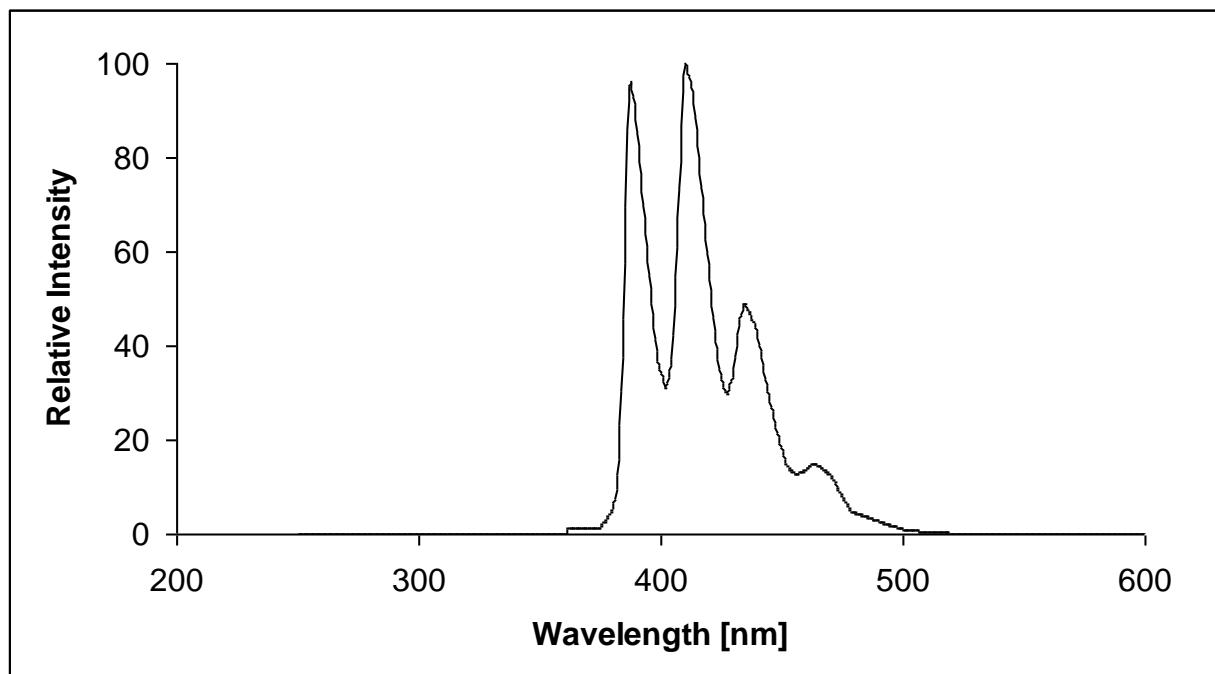
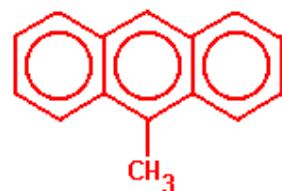
9 – Methylanthracene(1)



λ exc.	253.7 nm
Formula	C ₁₅ H ₁₂
M.W.	192 u
H/H+C	0.444
m.p.	°C
m.a.c. (266 nm)	(l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	Cyclohexane
source	Berlman

P03AMB2

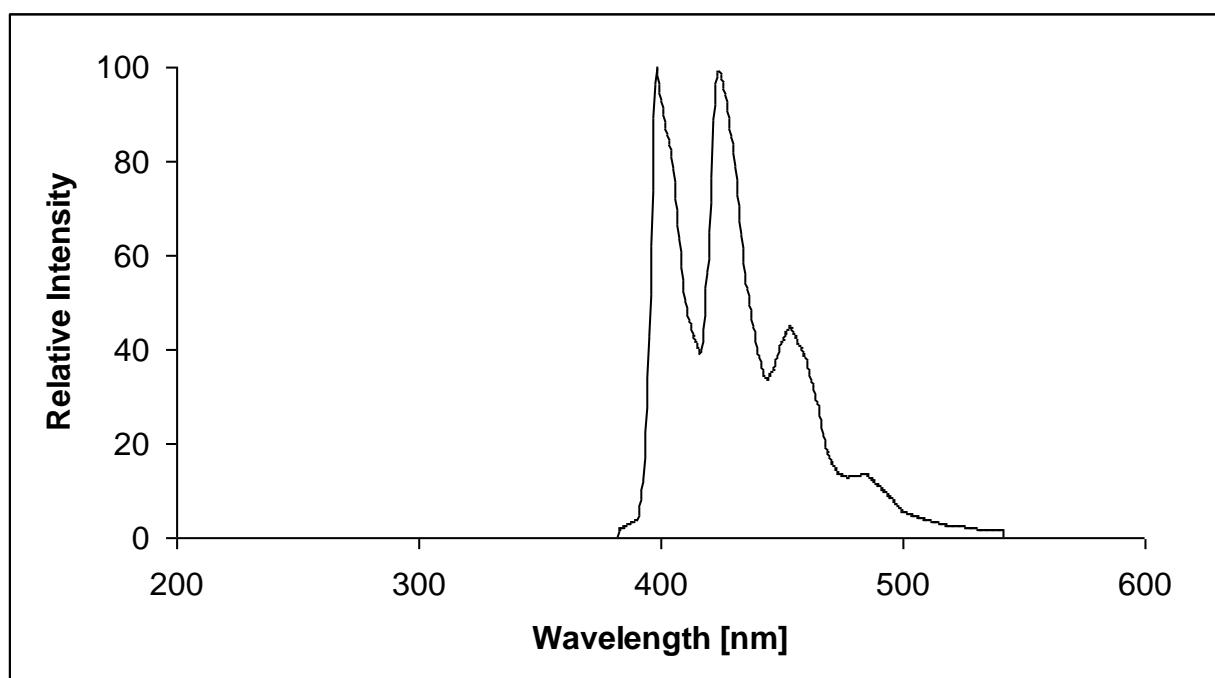
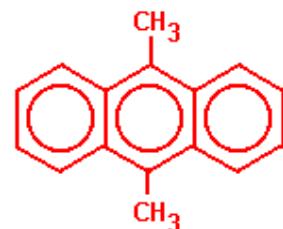
9 – Methylanthracene(2)



λ exc.	257 nm
Formula	$\text{C}_{15}\text{H}_{12}$
M.W.	192 u
H/H+C	0.444
m.p.	79°C
m.a.c. (266 nm)	1.5 ($\text{l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4}$)
solvent	Cyclohexane
source	Atlas pg. 158-II

P03AMJ

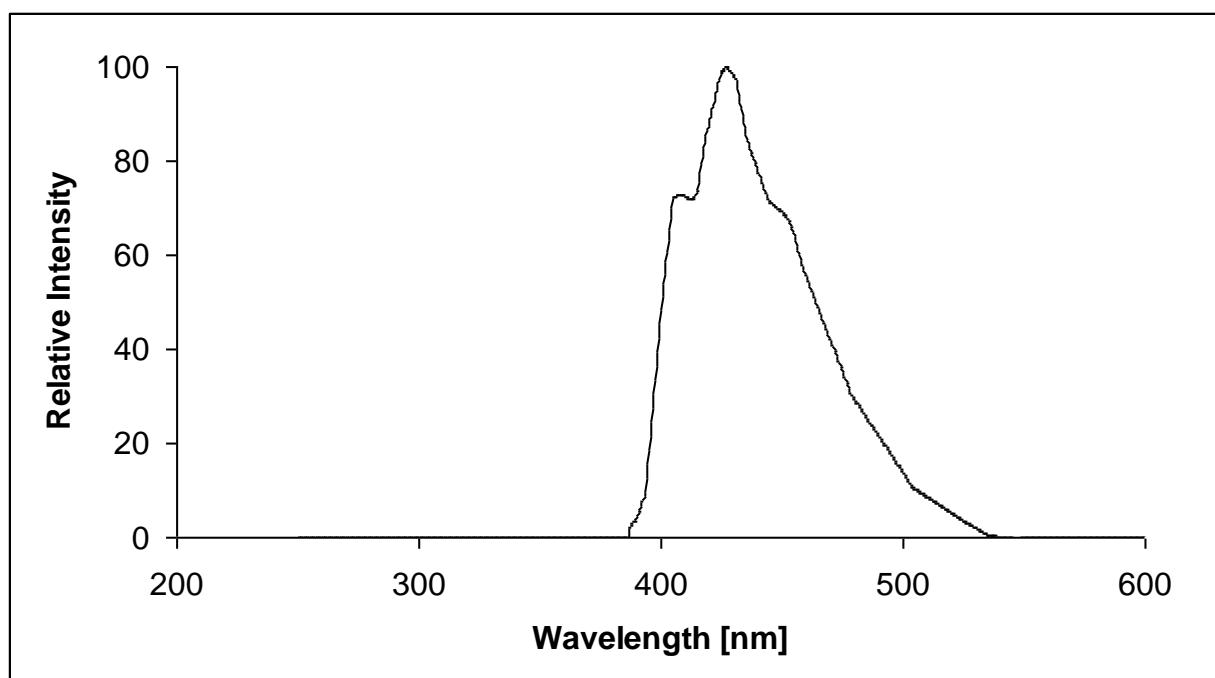
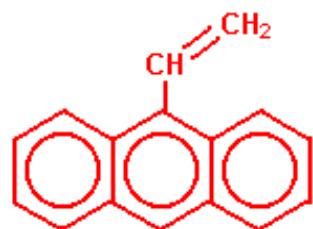
9,10 – Dimethylanthracene



λ exc.	377.5 nm
Formula	$\text{C}_{16}\text{H}_{14}$
M.W.	206 u
H/H+C	0.467
m.p.	183-184°C
m.a.c. (266 nm)	$9.1 (\text{l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4})$
solvent	Cyclohexane
source	Atlas pg. 822-III

P03AV

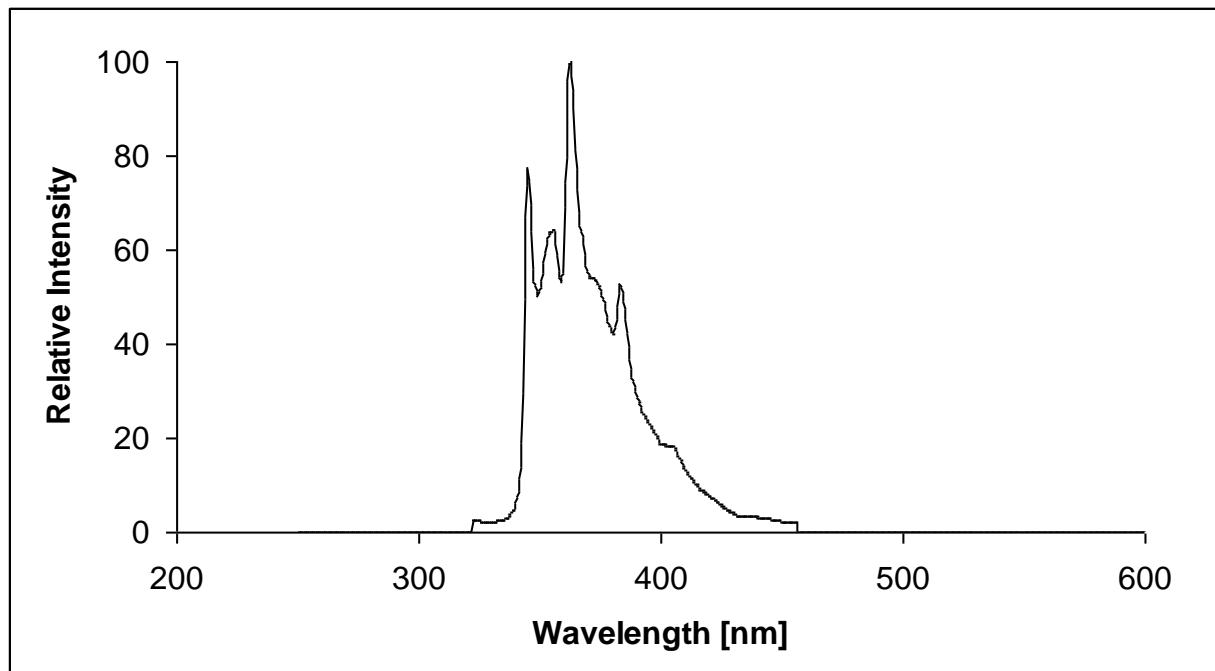
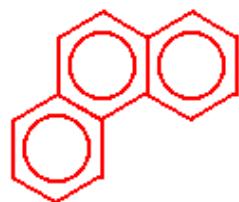
9 – Vinylanthracene



λ exc.	365 nm
Formula	C ₁₆ H ₁₂
M.W.	204 u
H/H+C	0.429
m.p.	°C
m.a.c. (266 nm)	(l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	Cyclohexane
source	Berlman

P03P0

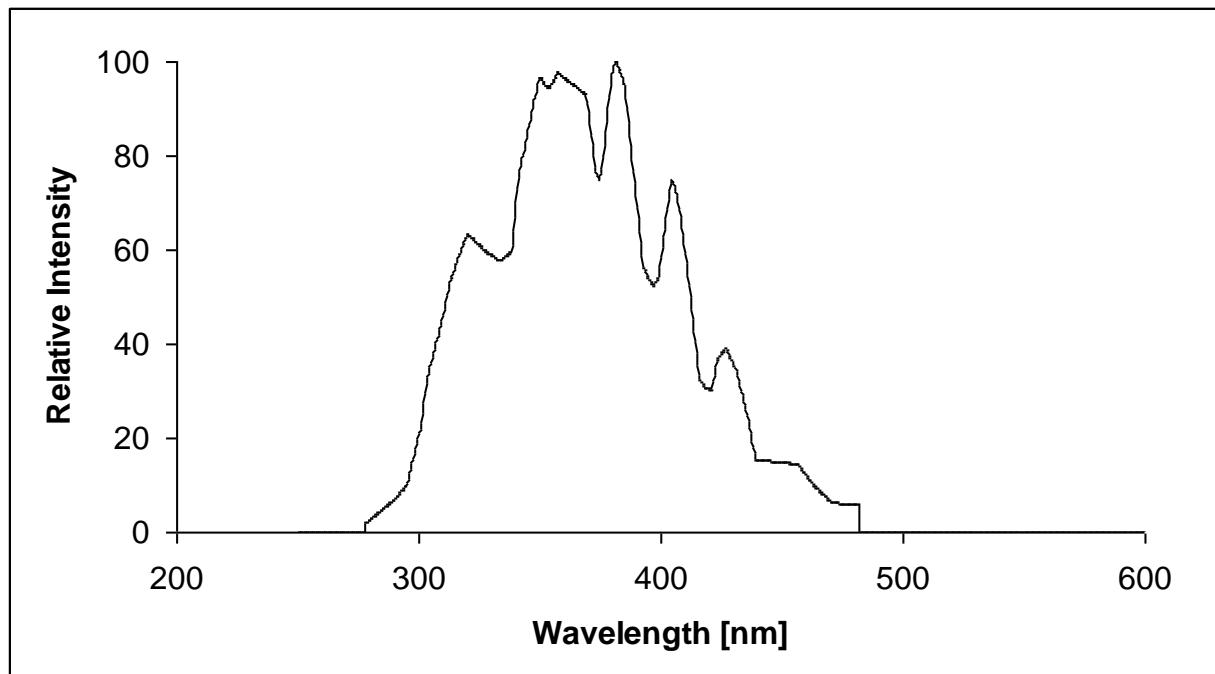
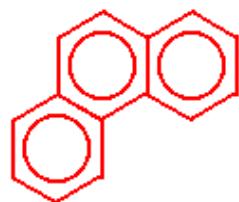
Phenanthrene (1)



λ exc.	252 nm
Formula	$C_{14}H_{10}$
M.W.	178 u
H/H+C	0.417
m.p.	100.5°C
m.a.c. (266 nm)	$4.8 \text{ (l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4})$
solvent	Cyclohexane
source	Atlas pg. 54-I

P03P0A

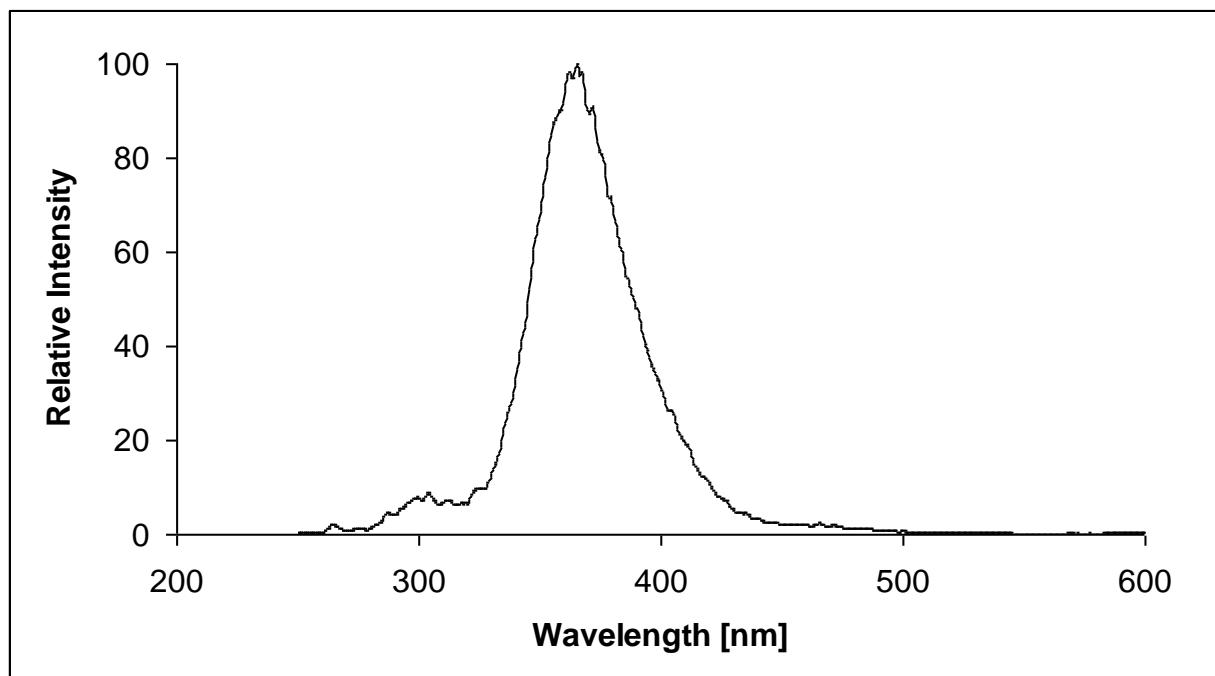
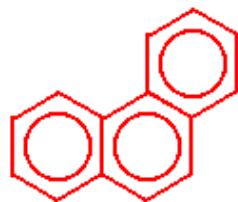
Phenanthrene (2)



λ exc.	220 nm
Formula	$C_{14}H_{10}$
M.W.	178 u
H/H+C	0.417
m.p.	100.5°C
m.a.c. (266 nm)	($l \cdot mol^{-1} cm^{-1} \times 10^{-4}$)
solvent	Acetonitrile
source	[10]

P03P0G

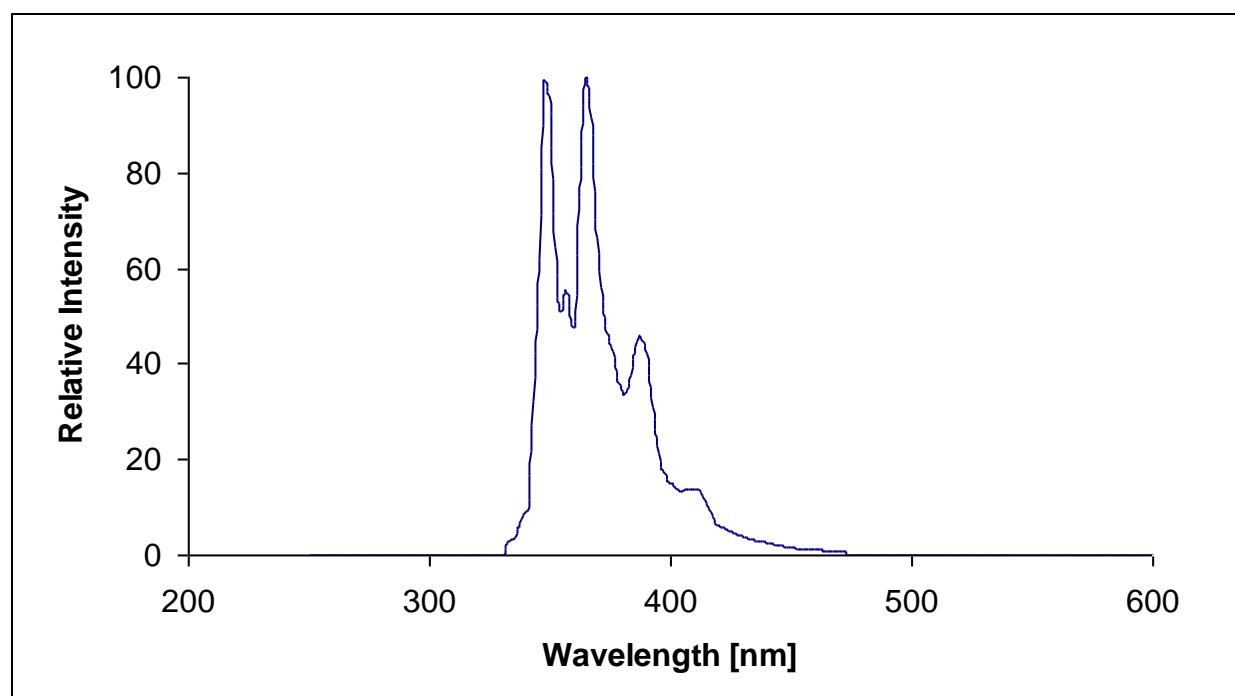
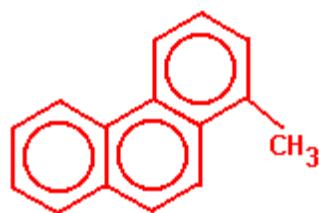
Phenanthrene (3)



λ exc.	266 nm
Formula	$C_{14}H_{10}$
M.W.	178 u
H/H+C	0.417
m.p.	100.5°C
m.a.c. (266 nm)	4.8 ($l \cdot mol^{-1} cm^{-1} \times 10^{-4}$)
solvent	Gas phase
source	CNPM

P03PMA

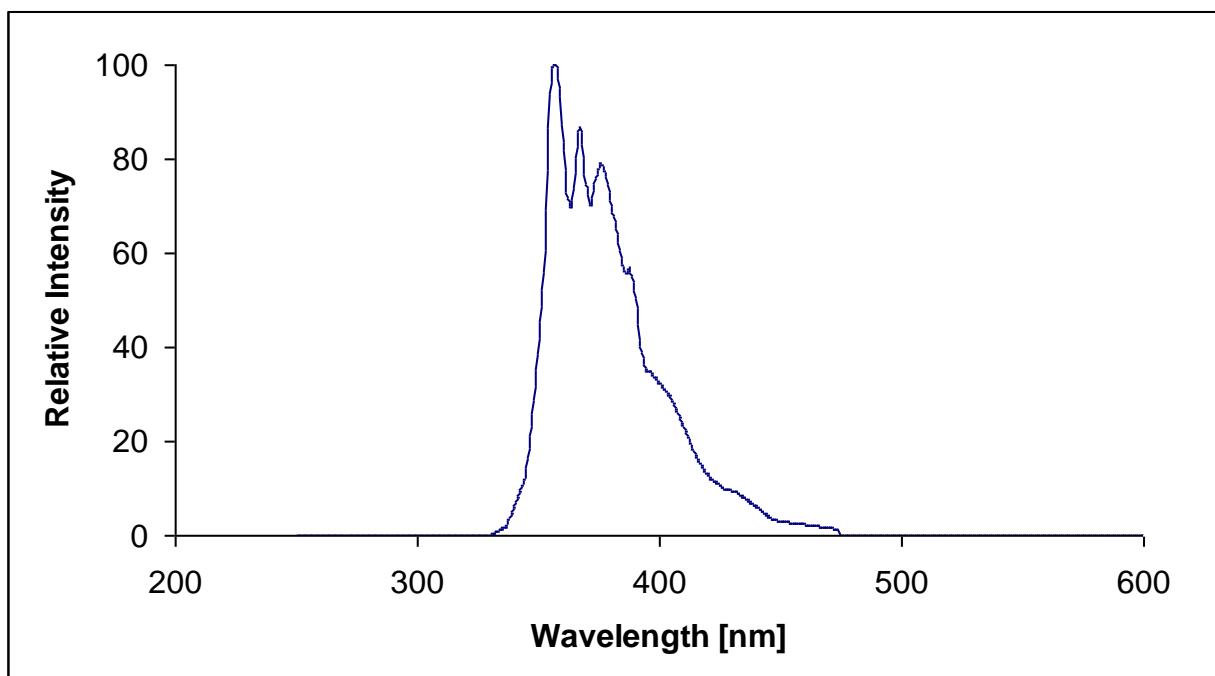
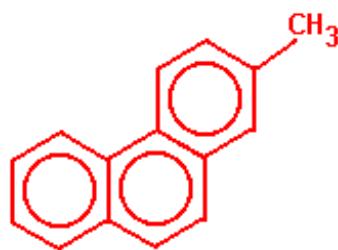
1-Methylphenanthrene



λ exc.	300 nm
Formula	C ₁₅ H ₁₂
M.W.	192 u
H/H+C	0.444
m.p.	123 °C
m.a.c. (266 nm)	0.5 (l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	Cyclohexane
source	Atlas pag 704-III

P03PMB

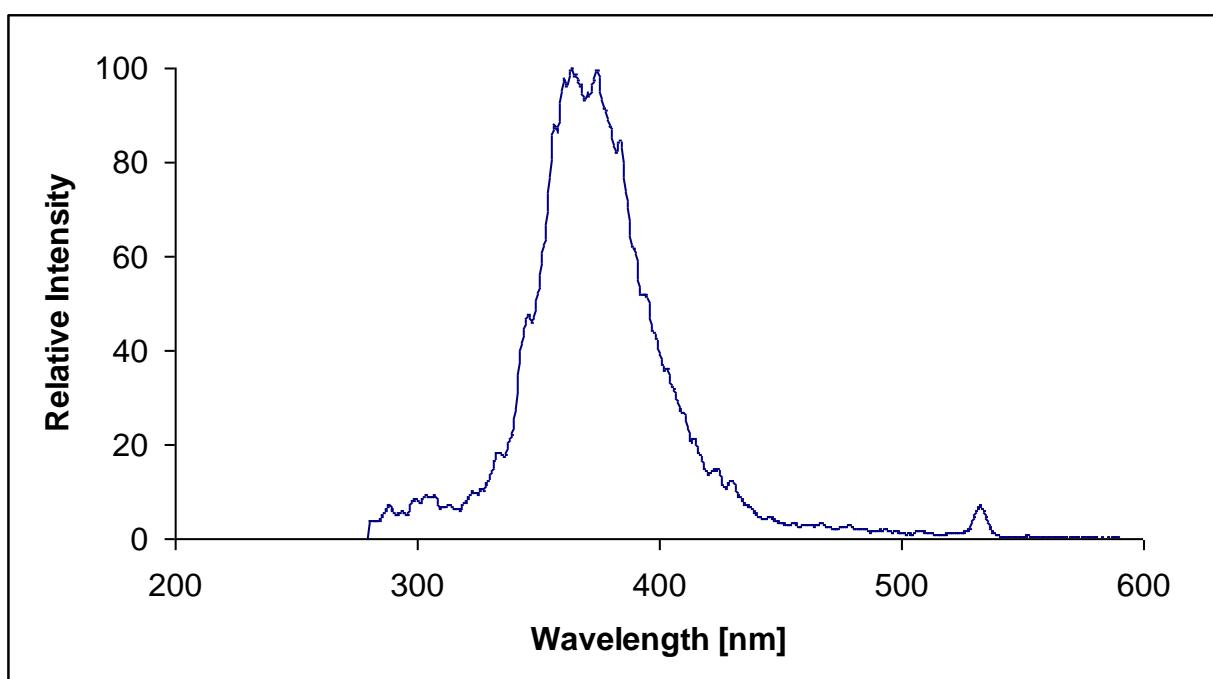
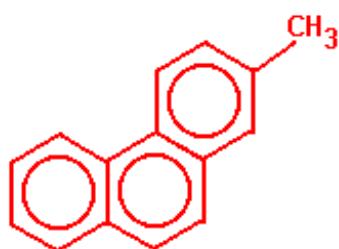
2 – Methylphenanthrene (1)



λ exc.	253.5 nm
Formula	$C_{15}H_{12}$
M.W.	192 u
H/H+C	0.444
m.p.	58-59°C
m.a.c. (266 nm)	$1.7 \text{ (l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4})$
solvent	Cyclohexane
source	Atlas pg. 724-III

P03PMBG

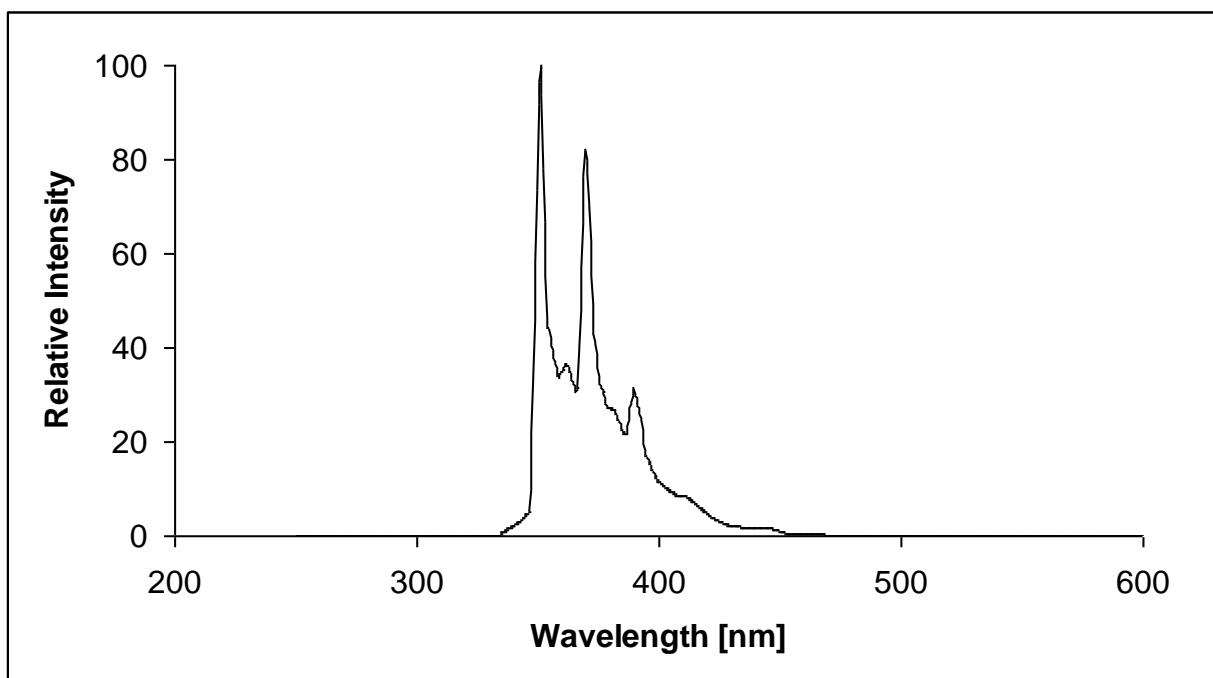
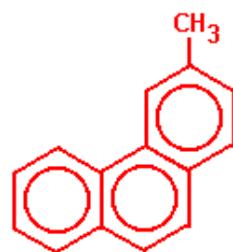
2 – Methylphenanthrene (1)



λ exc.	266 nm
Formula	$C_{15}H_{12}$
M.W.	192 u
H/H+C	0.444
m.p.	58-59°C
m.a.c. (266 nm)	$1.7 \text{ (l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4})$
solvent	Gas phase 330°K
source	CNPM

P03PMC

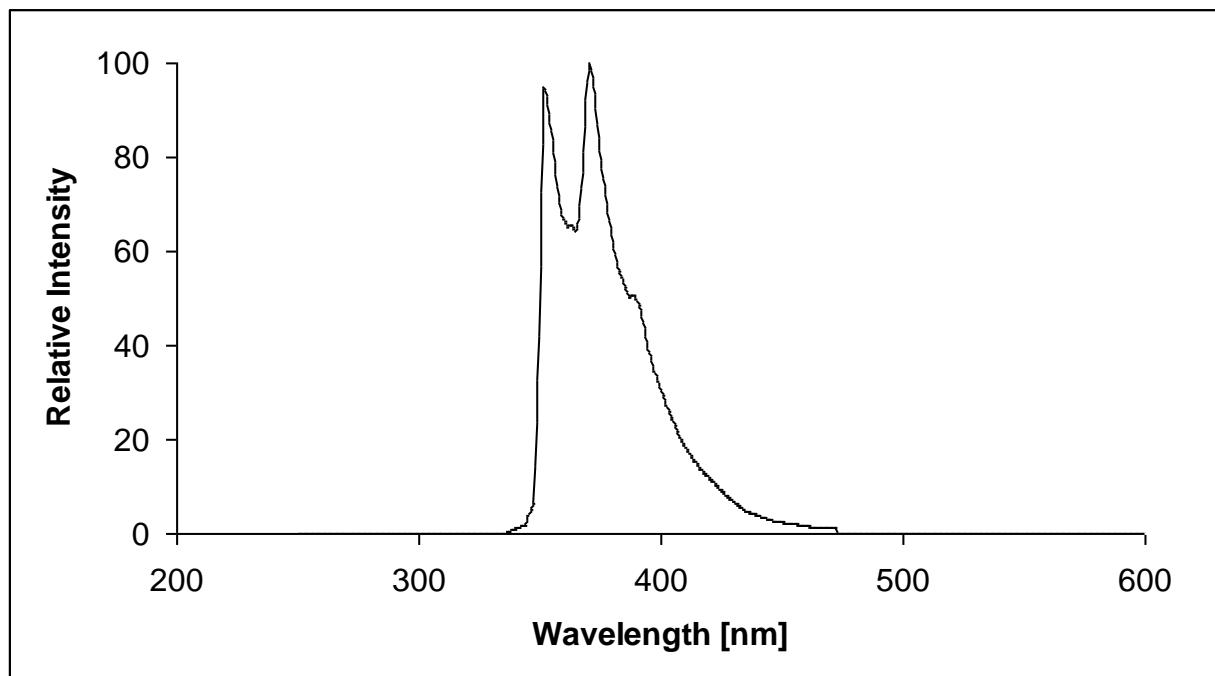
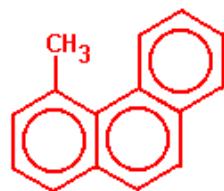
3 – Methylphenanthrene



λ exc.	253 nm
Formula	$C_{15}H_{12}$
M.W.	192 u
H/H+C	0.444
m.p.	65°C
m.a.c. (266 nm)	$1.4 \text{ (l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4})$
solvent	Cyclohexane
source	Atlas pg. 744-III

P03PMD

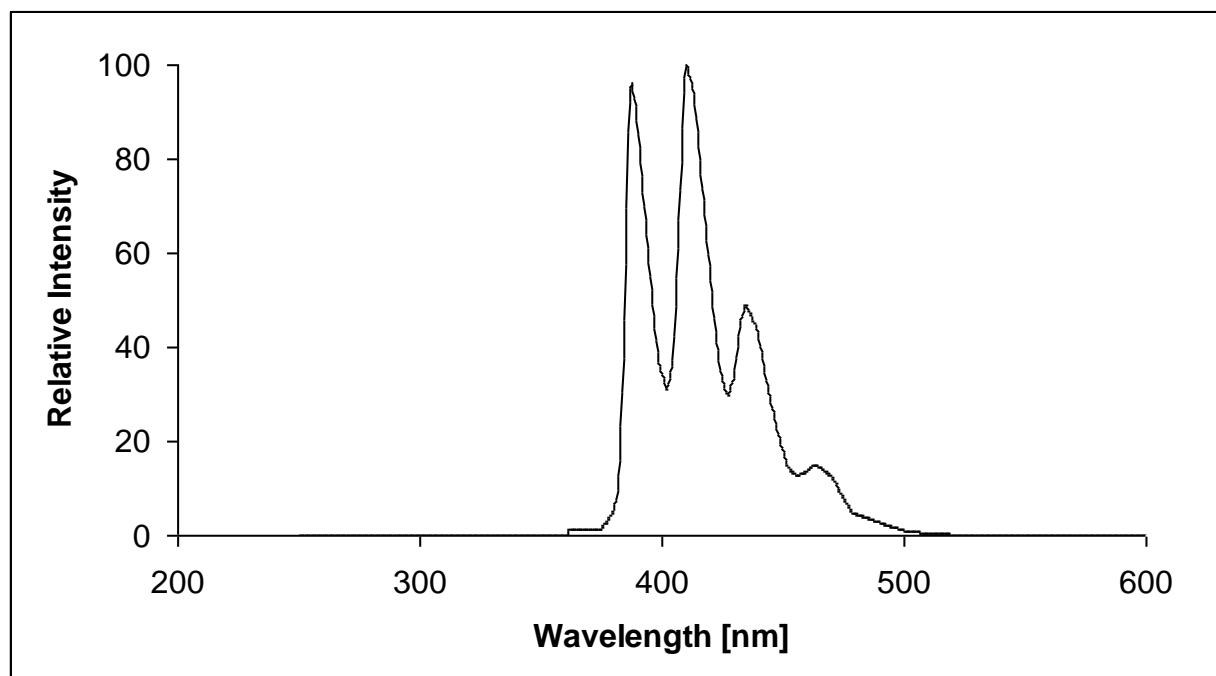
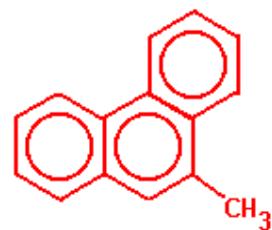
4 – Methylphenanthrene



λ exc.	251.5 nm
Formula	C ₁₅ H ₁₂
M.W.	192 u
H/H+C	0.444
m.p.	52-53°C
m.a.c. (266 nm)	1.2 (l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	Cyclohexane
source	Atlas pg. 765-III

P03PME

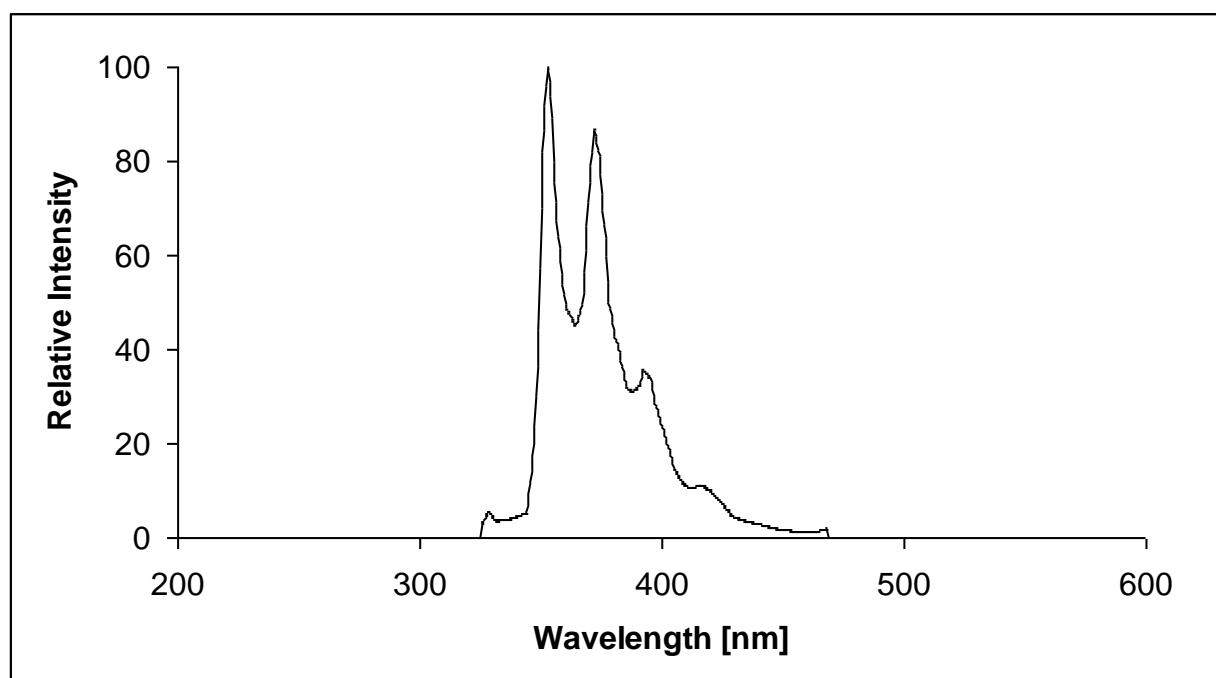
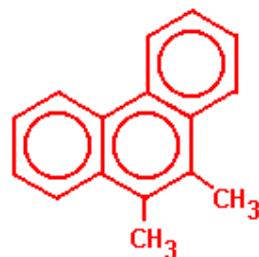
9 – Methylphenanthrene



λ exc.	235.5 nm
Formula	$\text{C}_{15}\text{H}_{12}$
M.W.	192 u
H/H+C	0.444
m.p.	91-92 °C
m.a.c. (266 nm)	$1.65 (\text{l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4})$
solvent	Cyclohexane
source	Atlas pg. 784-III

P03PMJ

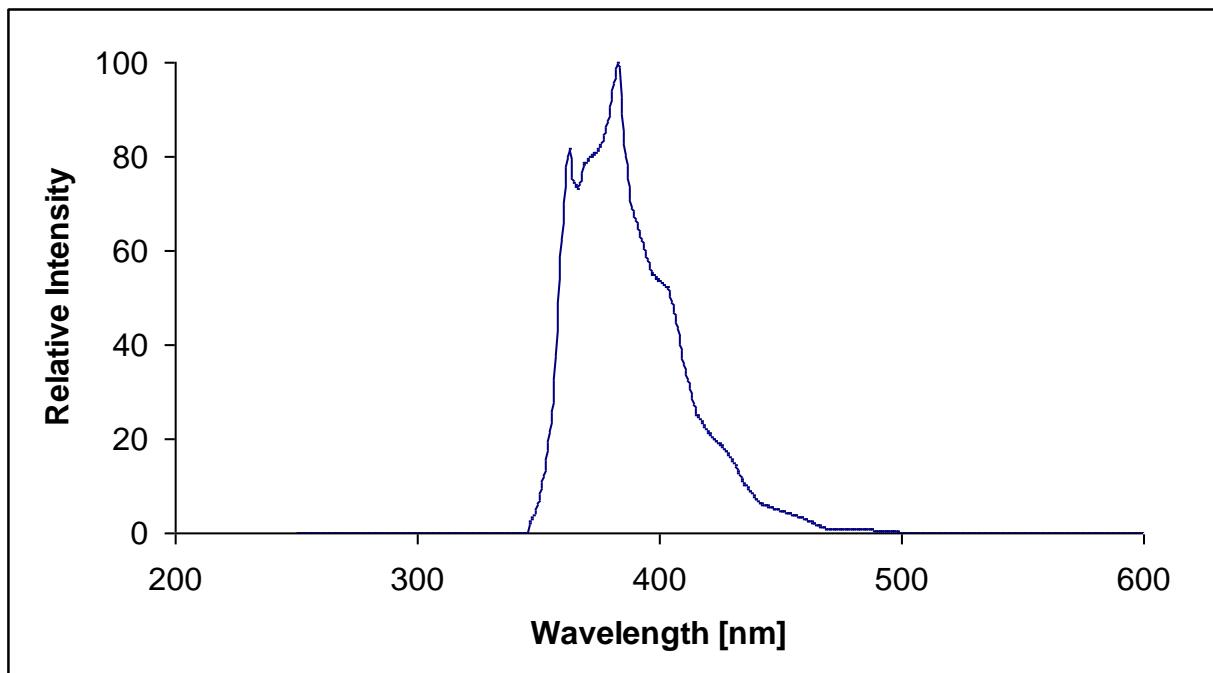
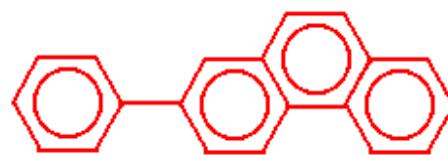
9,10 – Dimethylphenanthrene



λ exc.	255.5 nm
Formula	C ₁₆ H ₁₄
M.W.	206 u
H/H+C	0.467
m.p.	114-115°C
m.a.c. (266 nm)	1.75 (l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	Cyclohexane
source	Atlas pg. 804-III

P03PPB

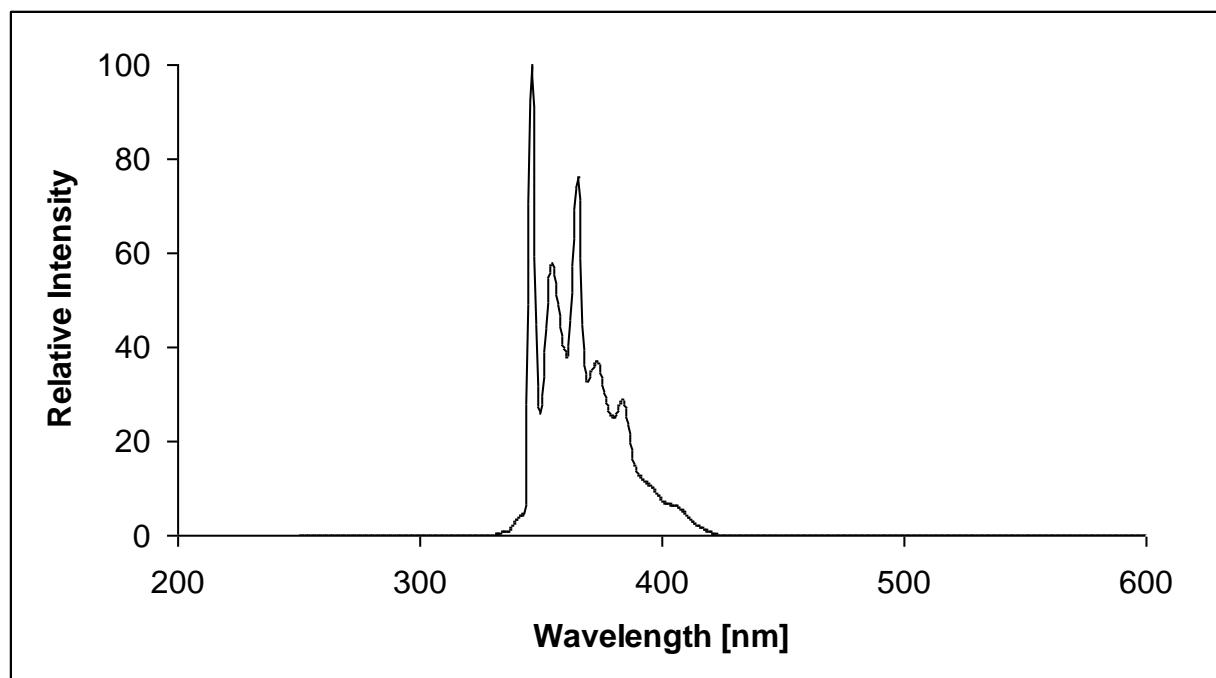
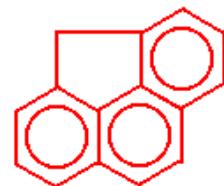
2 – Phenylphenanthrene



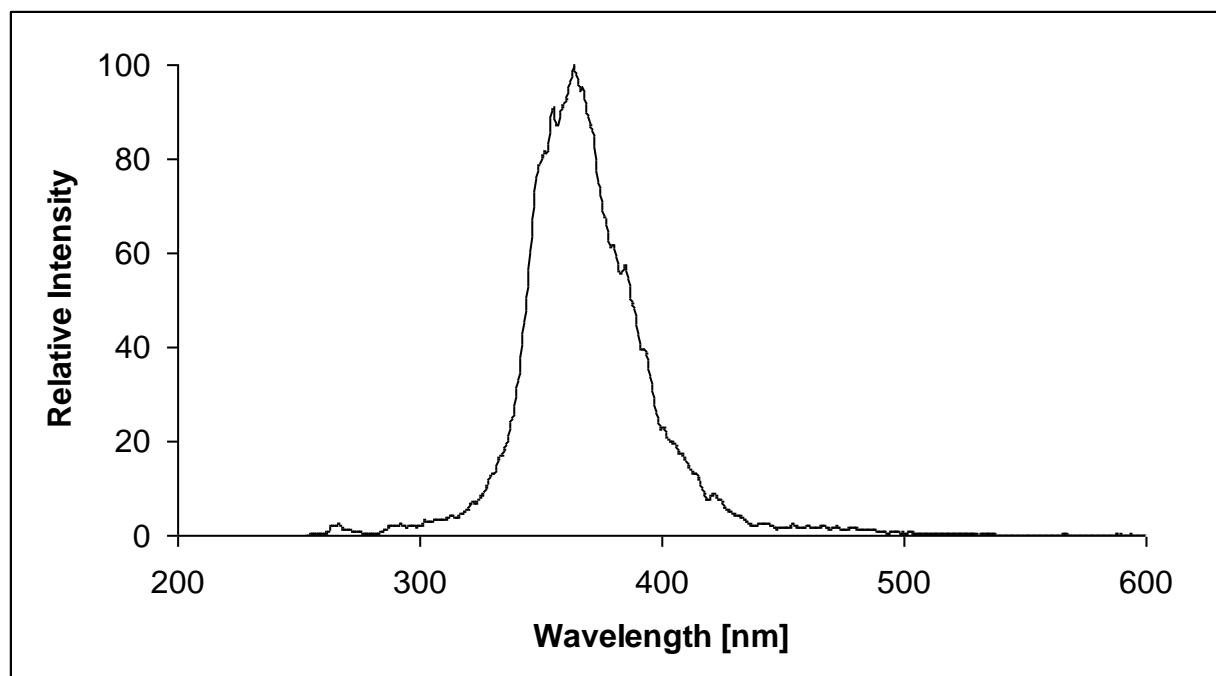
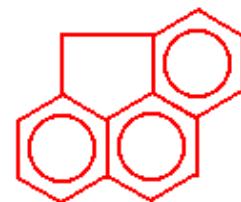
λ exc.	265 nm
Formula	$C_{20}H_{14}$
M.W.	254 u
H/H+C	0.412
m.p.	°C
m.a.c. (266 nm)	$0.76 \text{ (l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4})$
solvent	Cyclohexane
source	Berlman

P03PQ1

4,5 – Methylenephenanthrene (1)
4h-Cyclopenta(def)phenanthrene



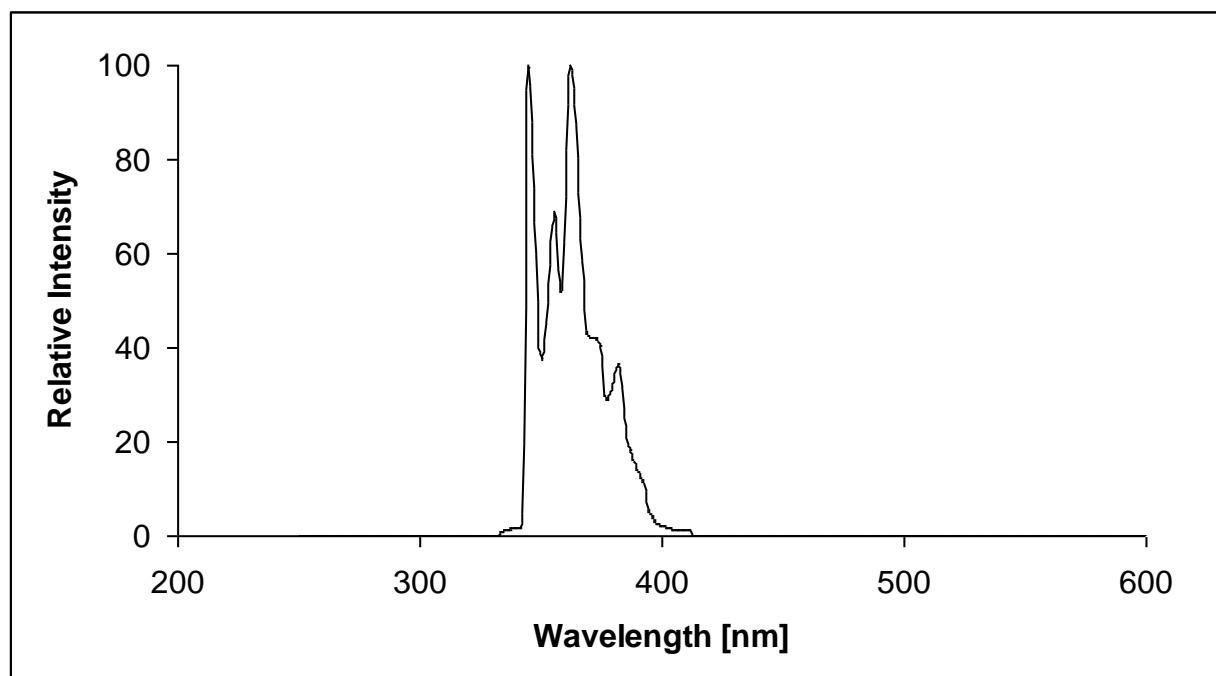
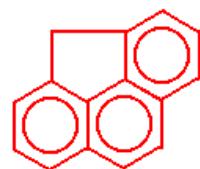
λ exc.	287.5 nm
Formula	$C_{15}H_{10}$
M.W.	190 u
H/H+C	0.400
m.p.	114-115°C
m.a.c. (266 nm)	$1.6 \text{ (l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4})$
solvent	Cyclohexane
source	Atlas pg. 684-III

P03PQ1G**4,5 – Methylenephenanthrene (2)**
4h-Cyclopenta(def)phenanthrene

λ exc.	266 nm
Formula	C ₁₅ H ₁₀
M.W.	190 u
H/H+C	0.400
m.p.	114-115°C
m.a.c. (266 nm)	1.6 (l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	Gas phase
source	CNPM

P03PQ2

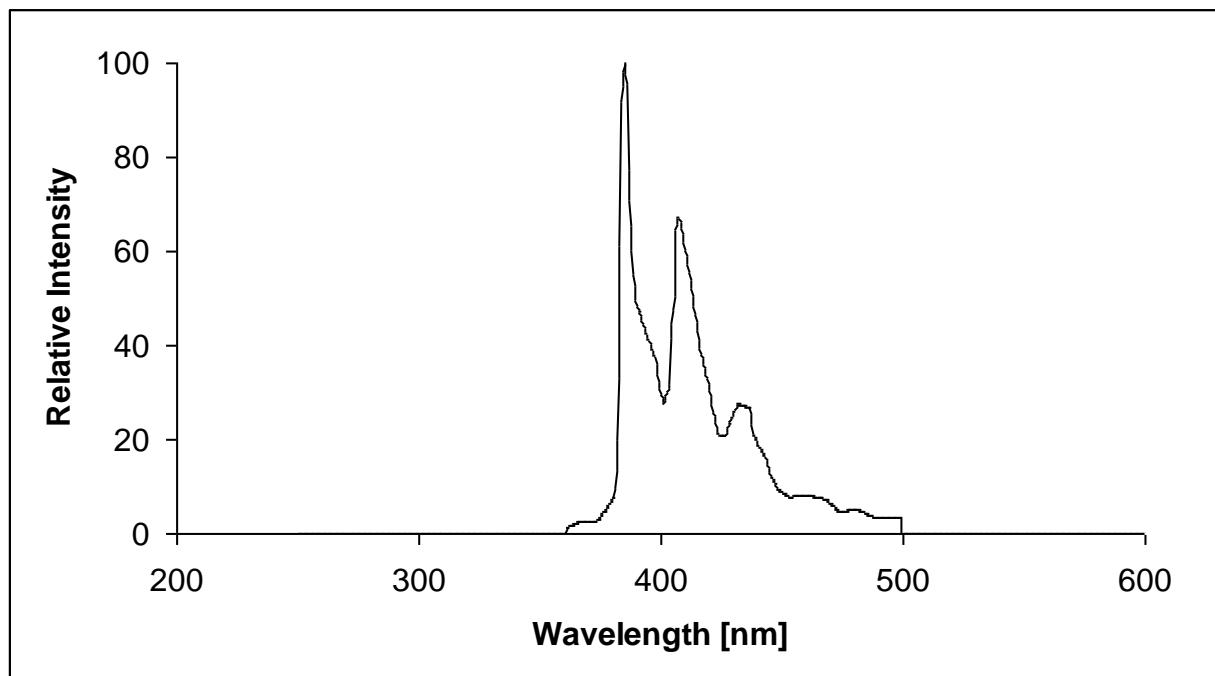
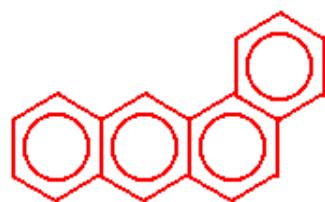
4,5 – Methylenephenanthrene (3)
4h-Cyclopenta(def)phenanthrene



λ exc.	320 nm
Formula	C ₁₅ H ₁₀
M.W.	190 u
H/H+C	0.400
m.p.	°C
m.a.c. (266 nm)	1.65 (l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	n-hexadecane
source	[9]

P04A0

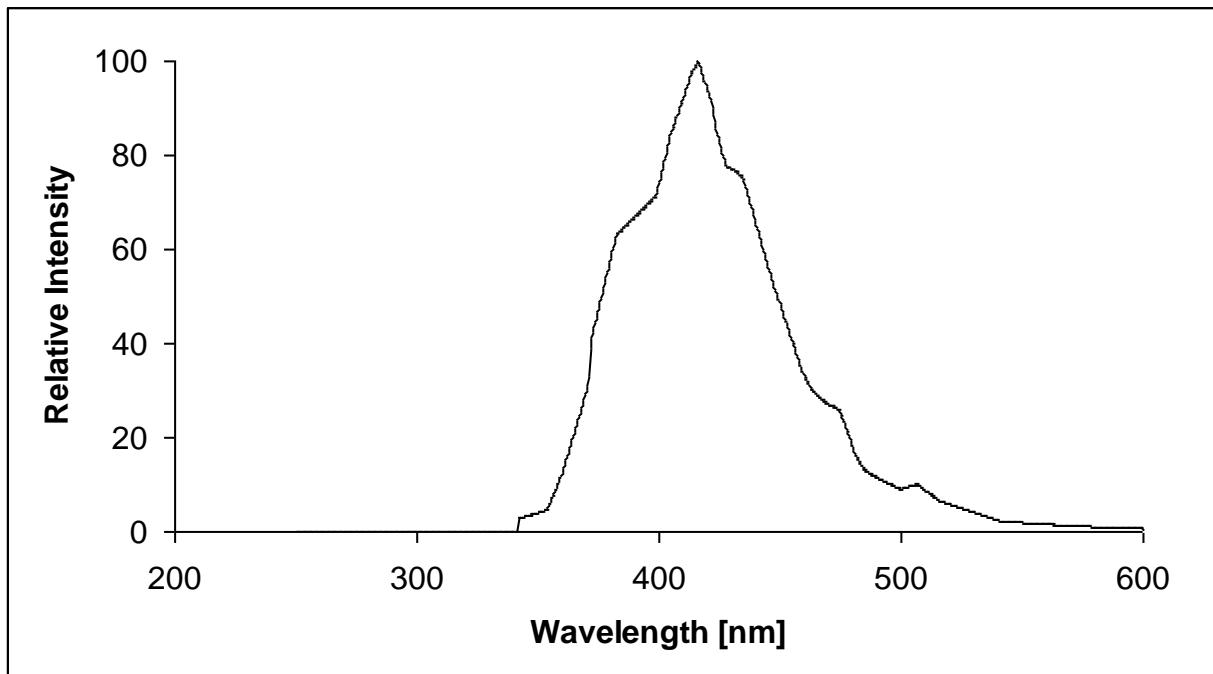
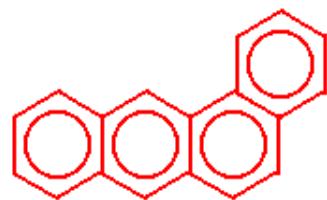
Benz(a)anthracene (1)



λ exc.	289 nm
Formula	$C_{18}H_{12}$
M.W.	228 u
H/H+C	0.400
m.p.	160.7°C
m.a.c. (266 nm)	$3.2 \text{ (l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4})$
solvent	Cyclohexane
source	Atlas pg. 206-I

P04A0G

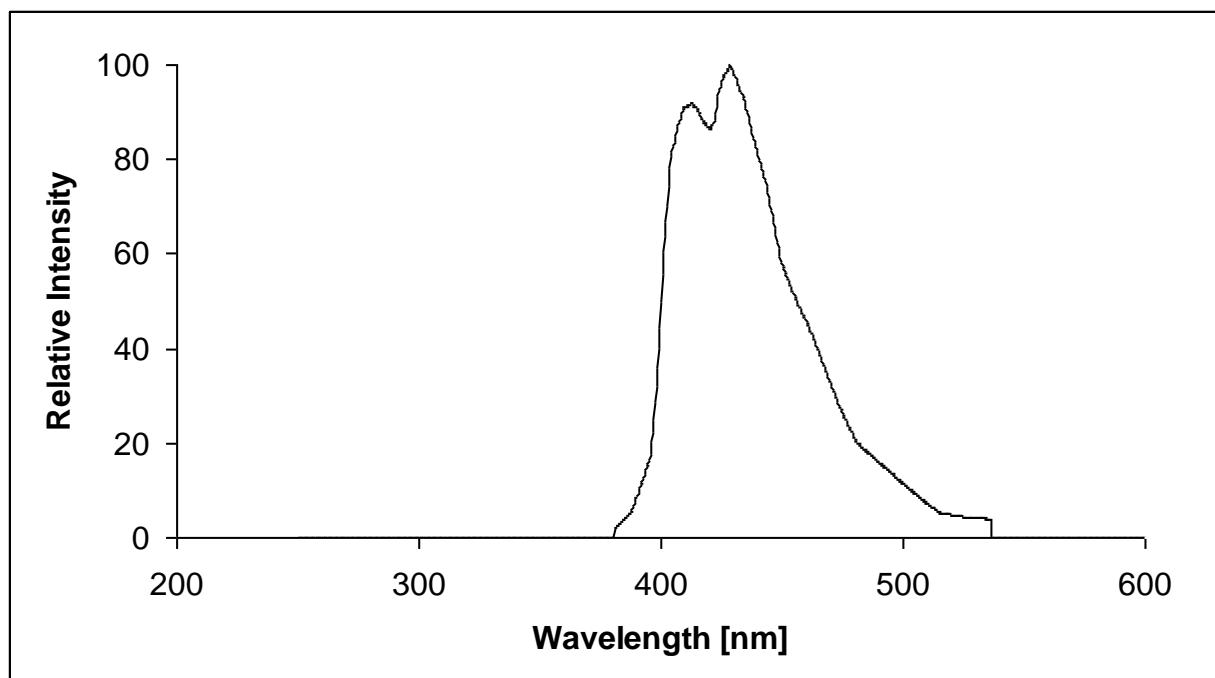
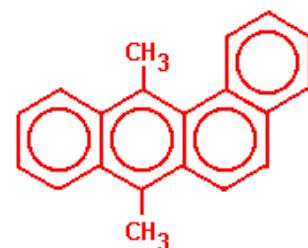
Benz(a)anthracene (2)



λ exc.	337 nm
Formula	C ₁₈ H ₁₂
M.W.	228 u
H/H+C	0.400
m.p.	160.7°C
m.a.c. (266 nm)	(l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	Gas phase 711 K
source	[1]

P04AM0

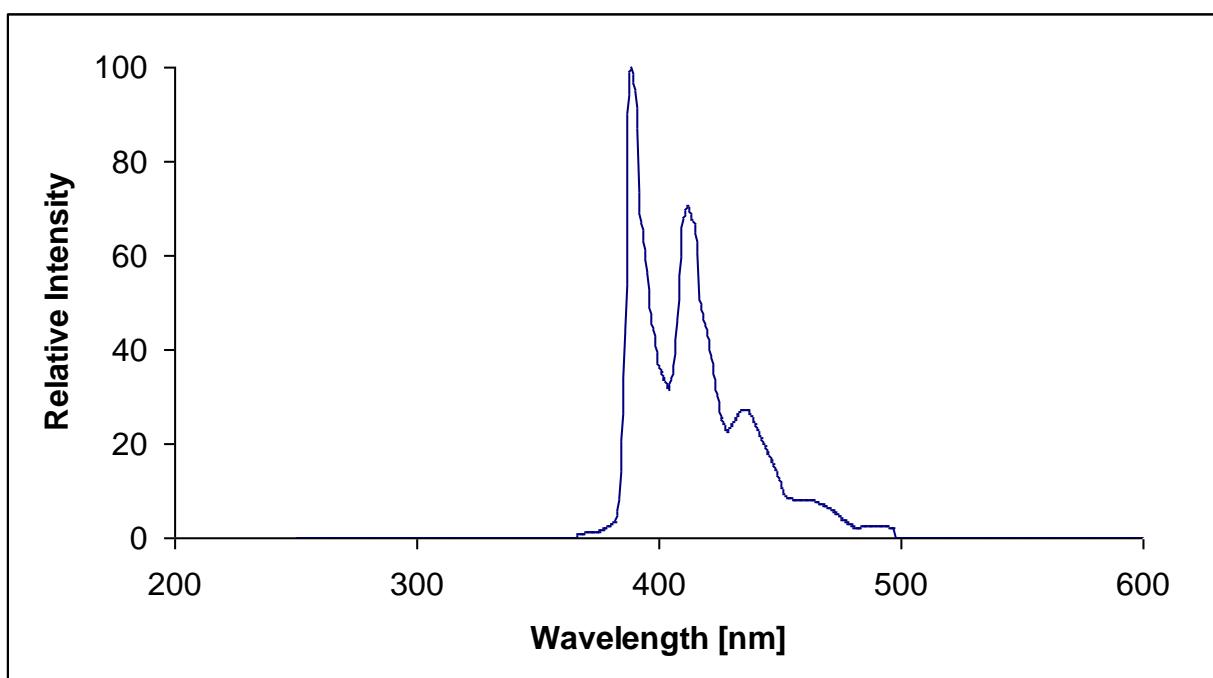
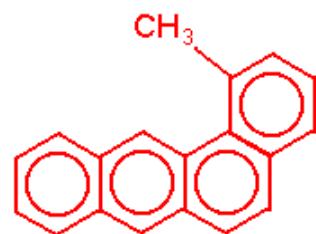
7,12 – Dimethylbenz(a)anthracene



λ exc.	298 nm
Formula	C ₂₀ H ₁₆
M.W.	256 u
H/H+C	0.444
m.p.	122°C
m.a.c. (266 nm)	2.8 (l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	Cyclohexane
source	Atlas pg. 502-II

P04AMA1

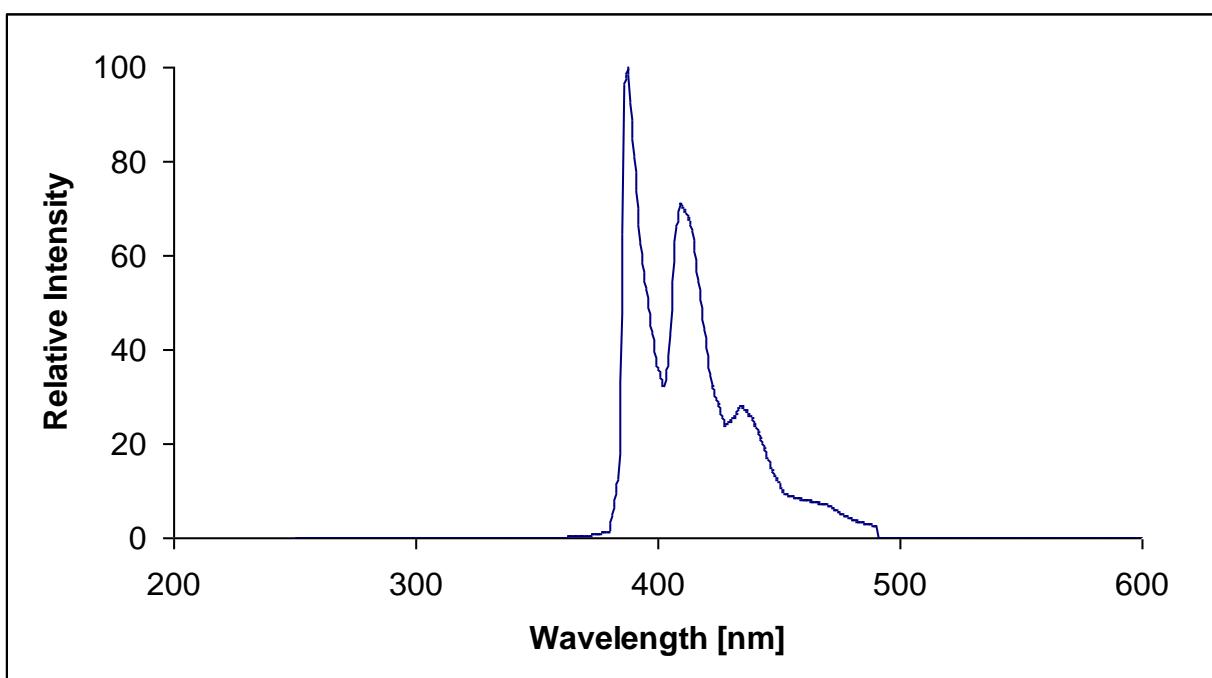
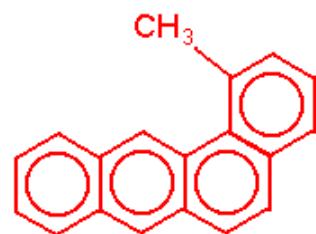
1 – Methylbenz(a)anthracene(1)



λ exc.	279 nm
Formula	C ₉ H ₁₄
M.W.	240 u
H/H+C	0.424
m.p.	138.6°C
m.a.c. (266 nm)	3.6 (l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	Cyclohexane
source	Atlas pag. 320 -I

P04AMA2

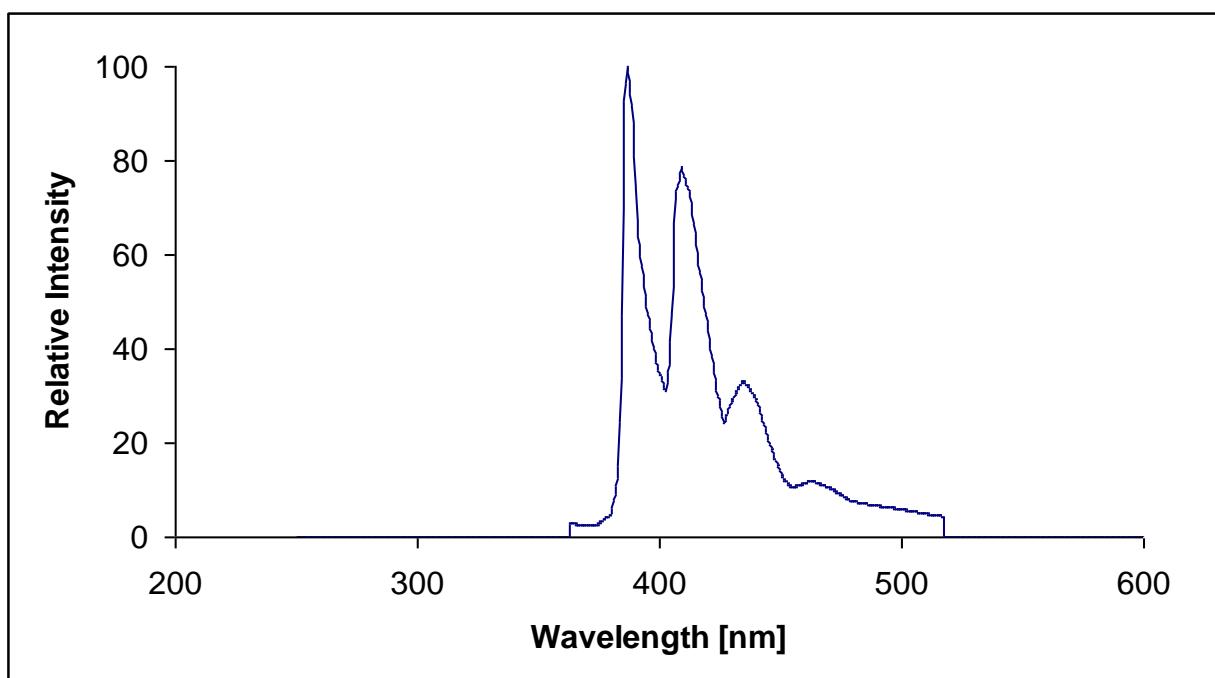
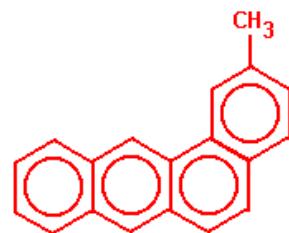
1 – Methylbenz(a)anthracene(2)



λ exc.	279 nm
Formula	C ₁₉ H ₁₄
M.W.	242 u
H/H+C	0.424
m.p.	138.6°C
m.a.c. (266 nm)	3.6 (l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	Cyclohexane
source	Atlas pag. 280 -II

P04AMB

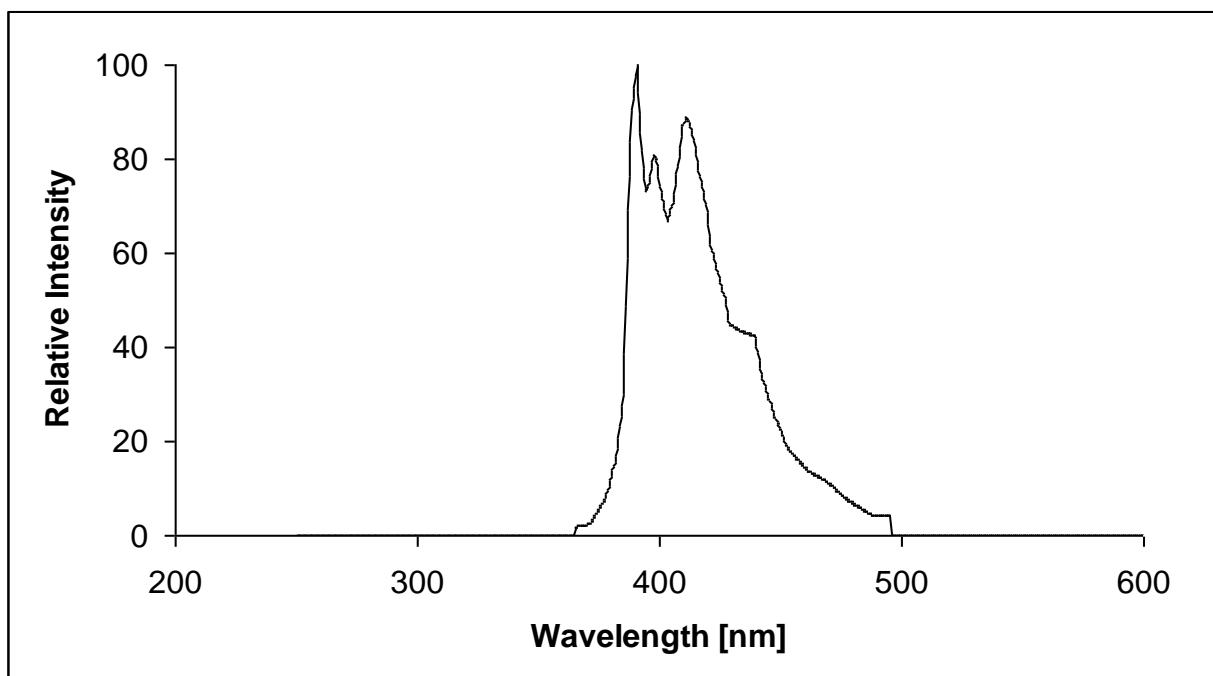
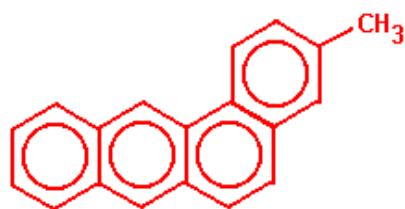
2 – Methylbenz(a)anthracene



λ exc.	291 nm
Formula	C ₁₉ H ₁₄
M.W.	242 u
H/H+C	0.424
m.p.	146-147°C
m.a.c. (266 nm)	3.16 (l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	Cyclohexane
source	Atlas pg. 298-II

P04AMC

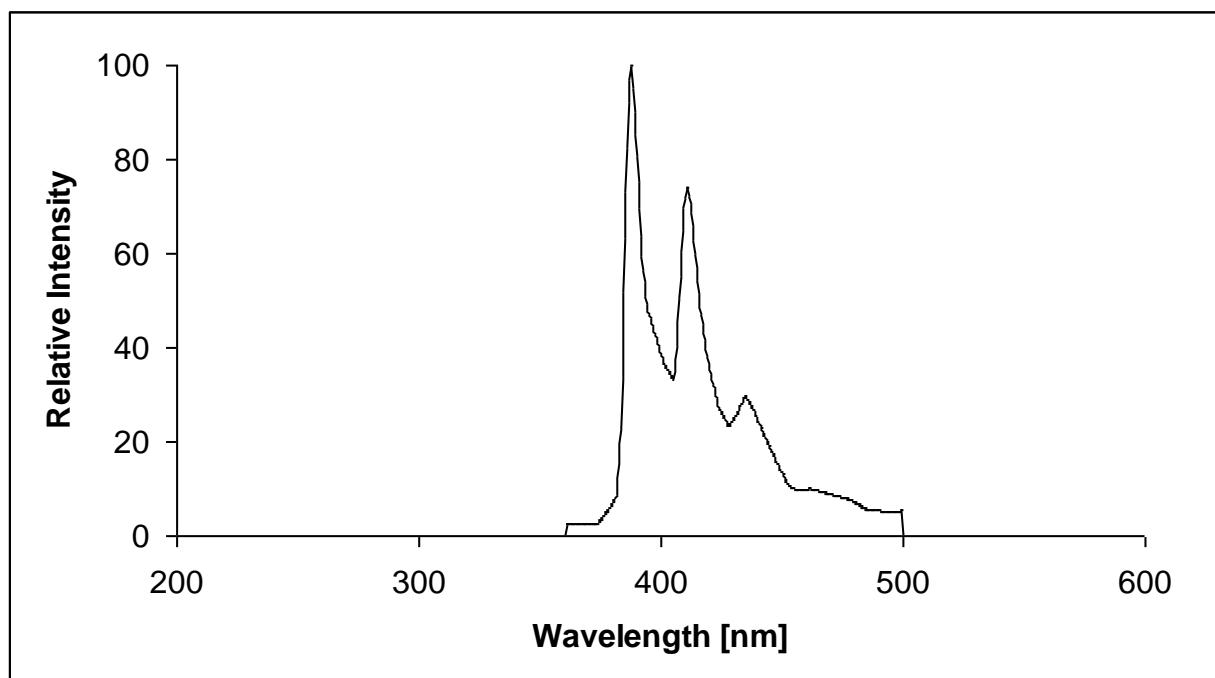
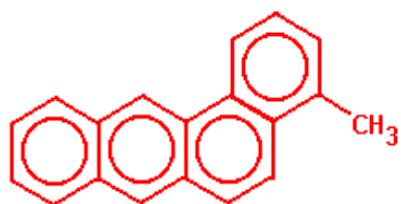
3 – Methylbenz(a)anthracene



λ exc.	290 nm
Formula	$\text{C}_{19}\text{H}_{14}$
M.W.	242 u
H/H+C	0.424
m.p.	159-161°C
m.a.c. (266 nm)	$2.5 (\text{l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4})$
solvent	Cyclohexane
source	Atlas pg. 306-II

P04AMD

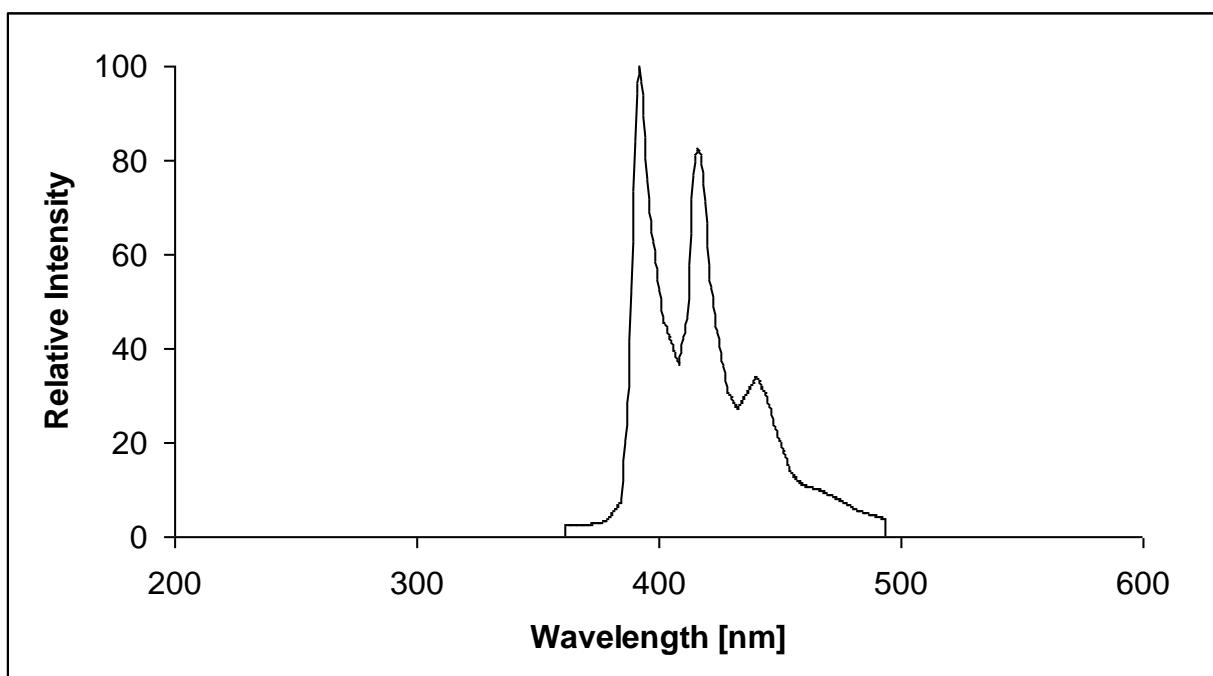
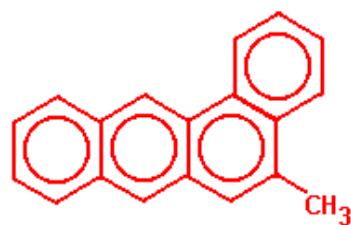
4 – Methylbenz(a)anthracene



λ exc.	294 nm
Formula	$C_{19}H_{14}$
M.W.	242 u
H/H+C	0.424
m.p.	193°C
m.a.c. (266 nm)	$2.9 \text{ (l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4})$
solvent	Cyclohexane
source	Atlas pg. 320-II

P04AME

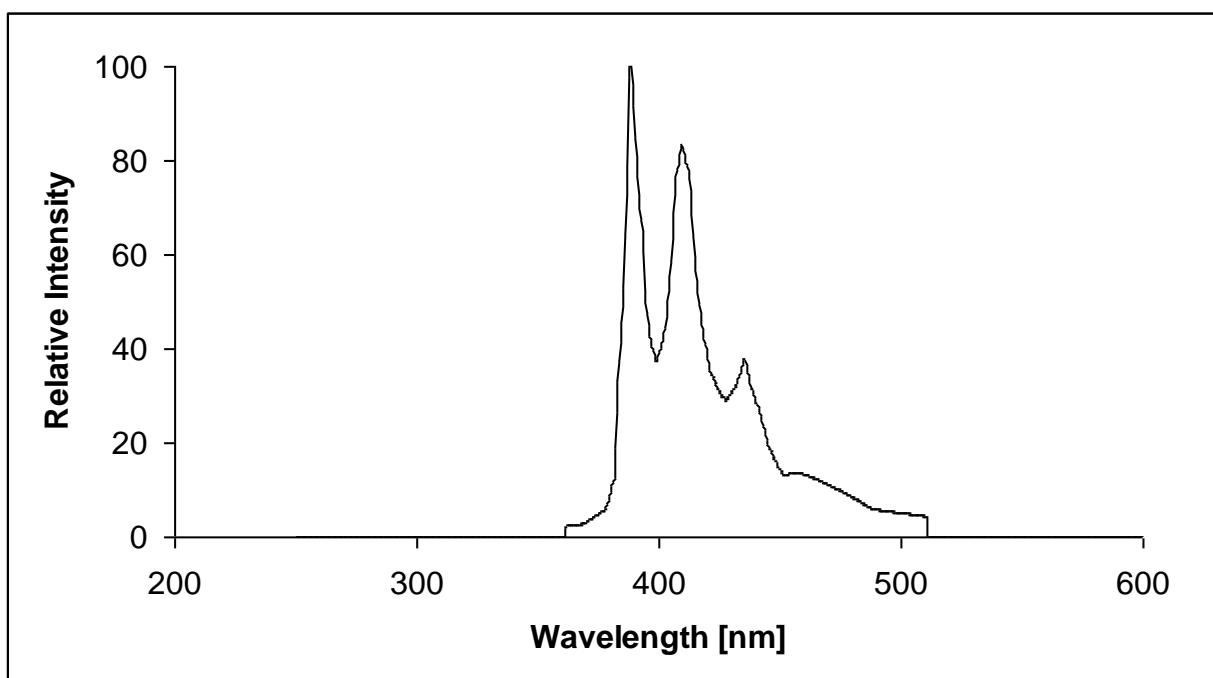
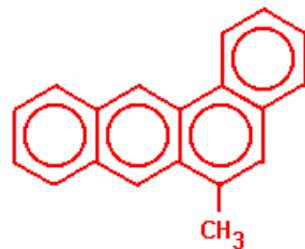
5 – Methylbenz(a)anthracene



λ exc.	291 nm
Formula	$\text{C}_{19}\text{H}_{14}$
M.W.	242 u
H/H+C	0.424
m.p.	154°C
m.a.c. (266 nm)	$2.5 (\text{l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4})$
solvent	Cyclohexane
source	Atlas pg. 334-II

P04AMF

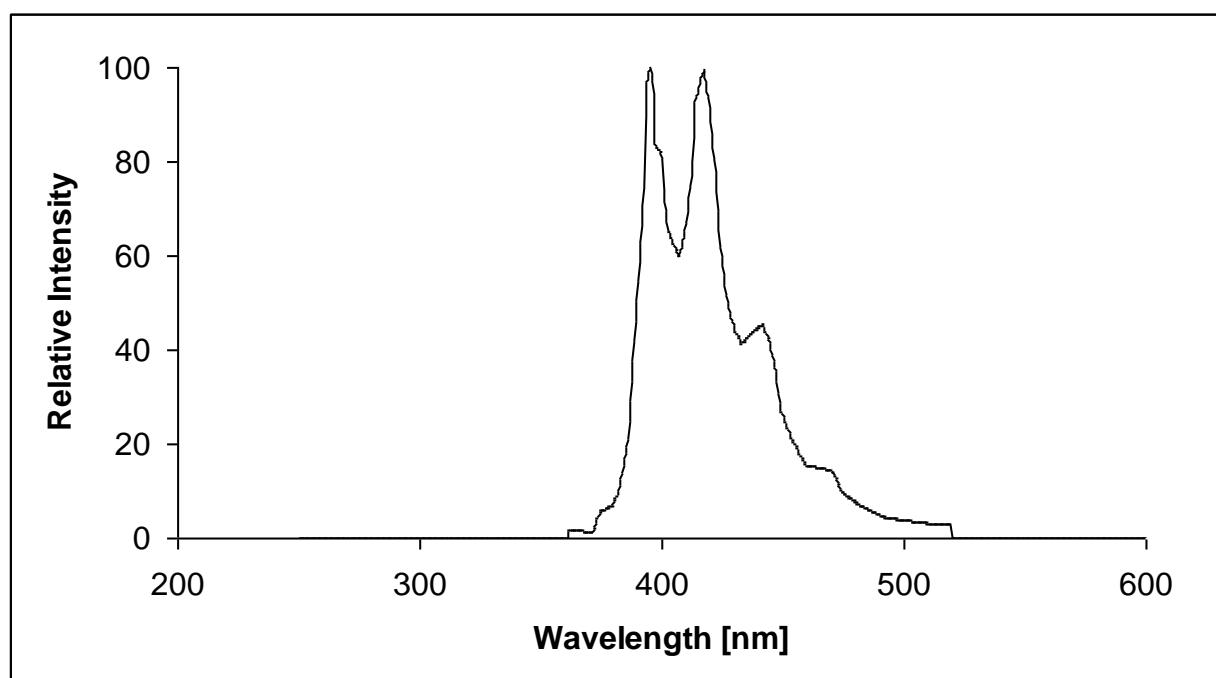
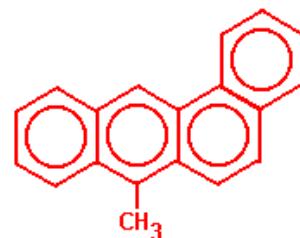
6 – Methylbenz(a)anthracene



λ exc.	290 nm
Formula	$\text{C}_{19}\text{H}_{14}$
M.W.	242 u
H/H+C	0.424
m.p.	125°C
m.a.c. (266 nm)	7 ($\text{l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4}$)
solvent	Cyclohexane
source	Atlas pg. 348-II

P04AMG

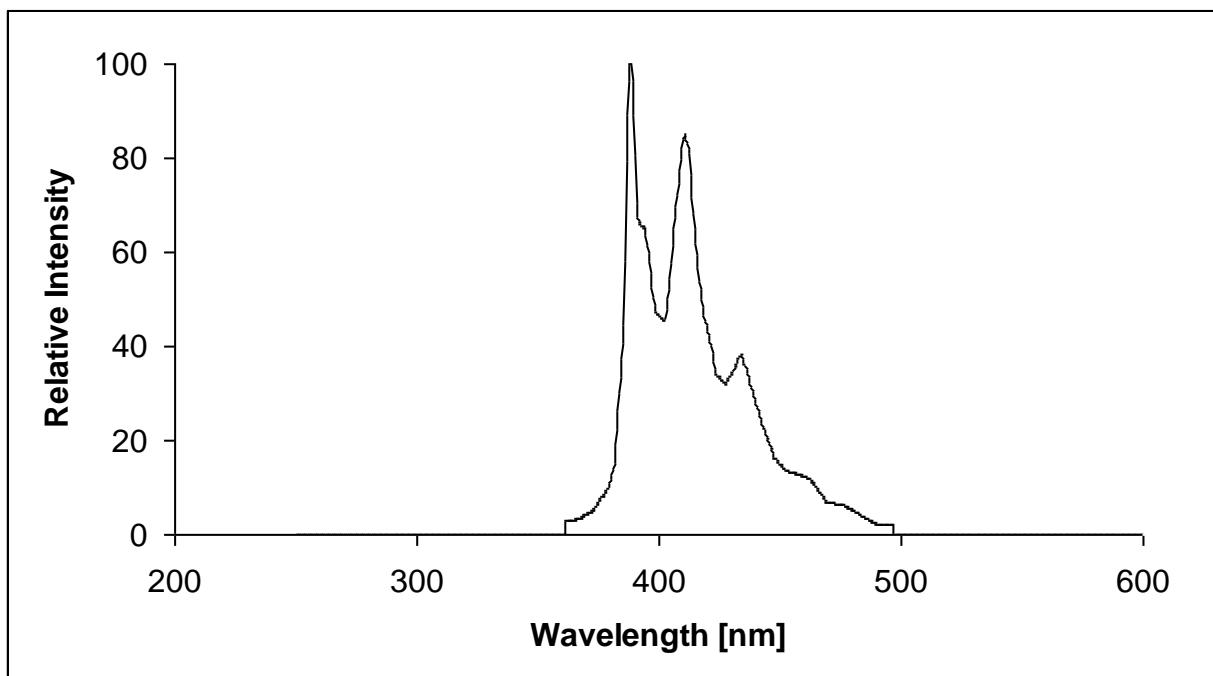
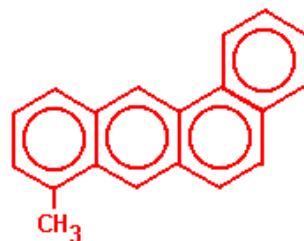
7 – Methylbenz(a)anthracene



λ exc.	293 nm
Formula	$\text{C}_{19}\text{H}_{14}$
M.W.	242 u
H/H+C	0.424
m.p.	134°C
m.a.c. (266 nm)	$3.9 (\text{l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4})$
solvent	Cyclohexane
source	Atlas pg. 362-II

P04AMH

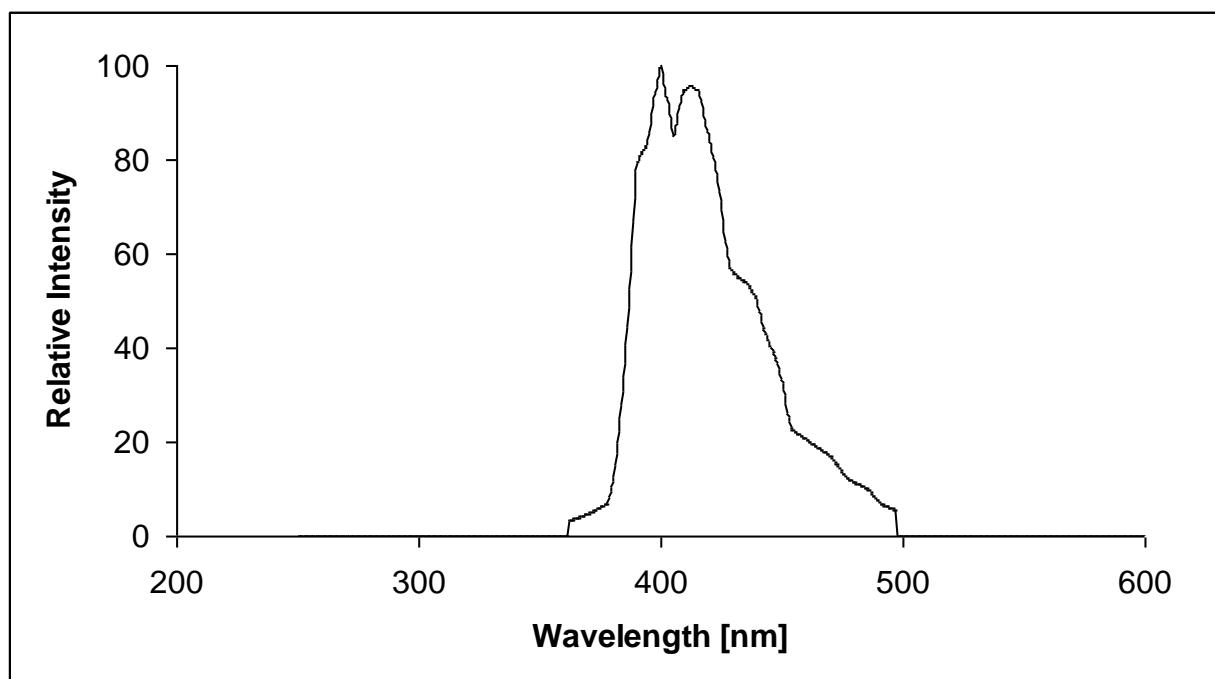
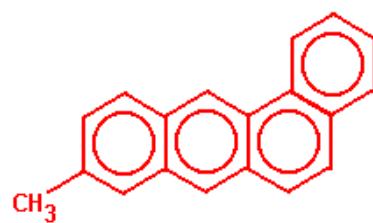
8 - Methylbenz(a)anthracene



λ exc.	291 nm
Formula	$\text{C}_{19}\text{H}_{14}$
M.W.	242 u
H/H+C	0.424
m.p.	158°C
m.a.c. (266 nm)	$3.3 (\text{l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4})$
solvent	Cyclohexane
source	Atlas pg. 374-II

P04AMI

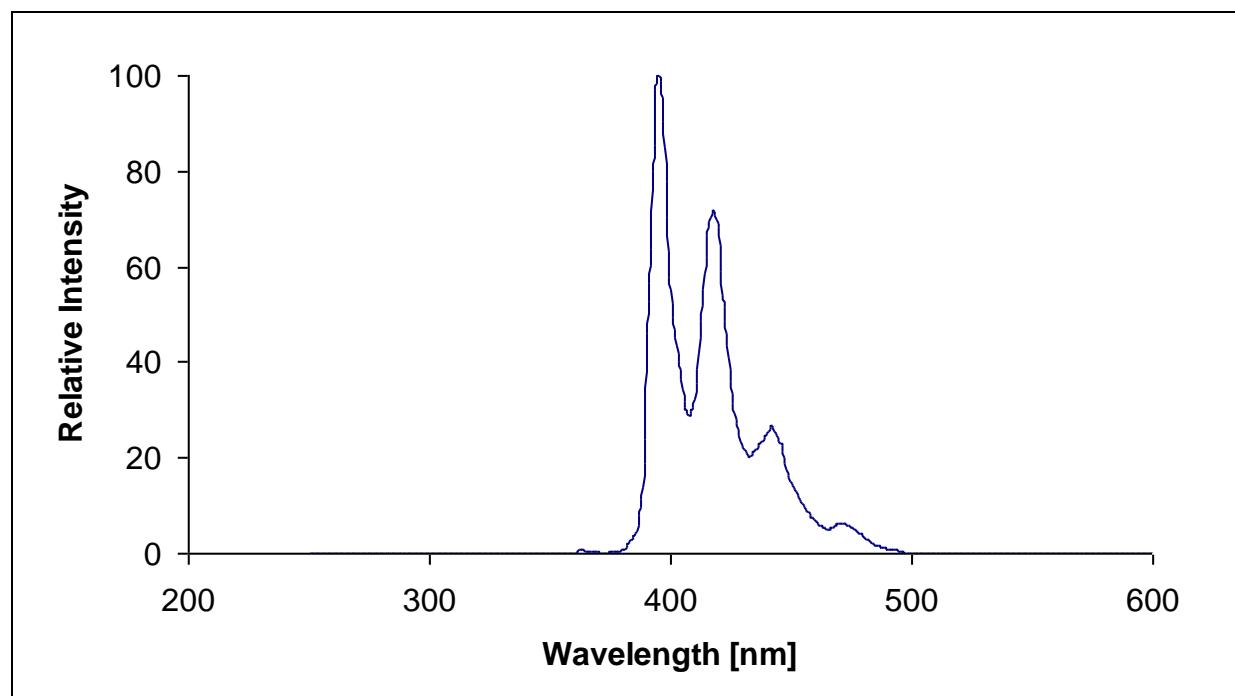
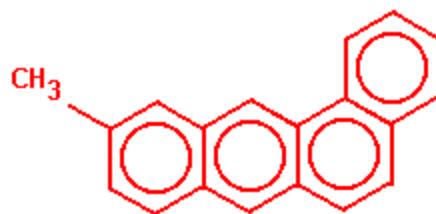
9 – Methylbenz(a)anthracene



λ exc.	290 nm
Formula	$\text{C}_{19}\text{H}_{14}$
M.W.	242 u
H/H+C	0.424
m.p.	151°C
m.a.c. (266 nm)	$3.65 (\text{l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4})$
solvent	Cyclohexane
source	Atlas pg. 390-II

P04AMJ

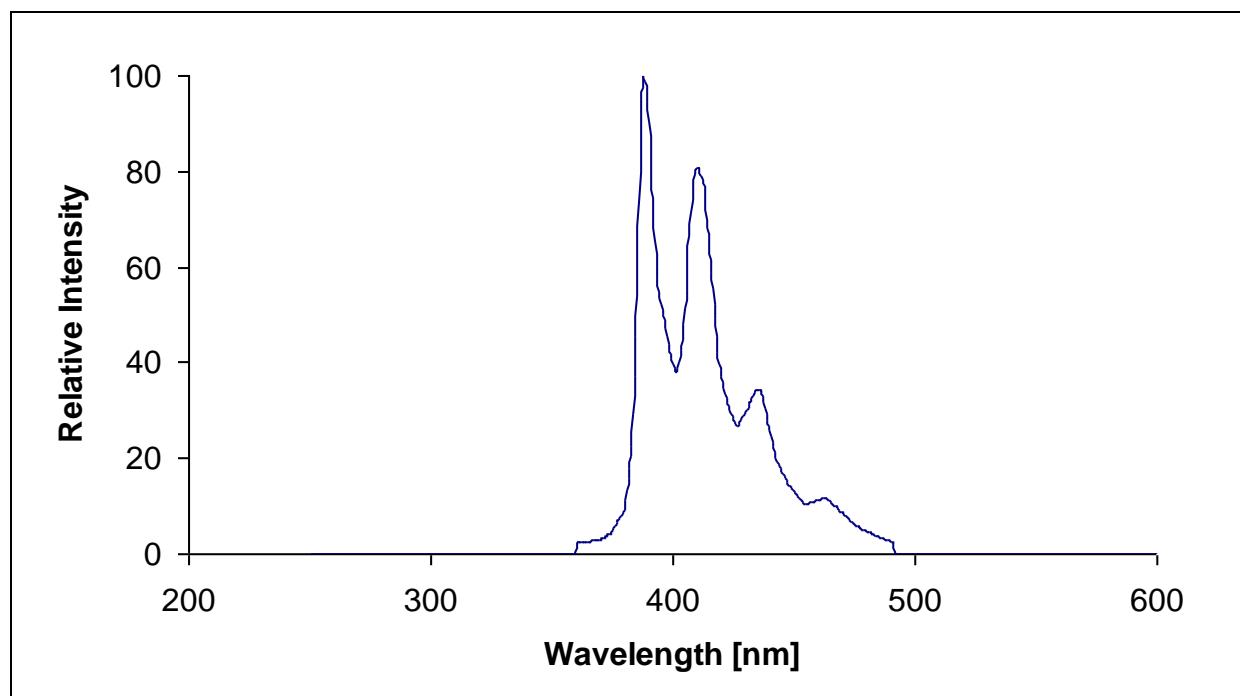
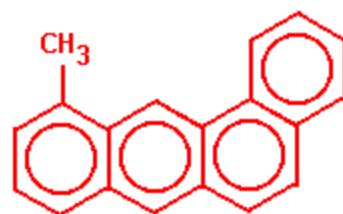
10-Methylbenz(a)anthracene



λ exc.	290.5 nm
Formula	$\text{C}_{19}\text{H}_{14}$
M.W.	242 u
H/H+C	0.424
m.p.	139 °C
m.a.c. (266 nm)	$2.1 (\text{l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4})$
solvent	Cyclohexane
source	Atlas pag. 404-II

P04AMK

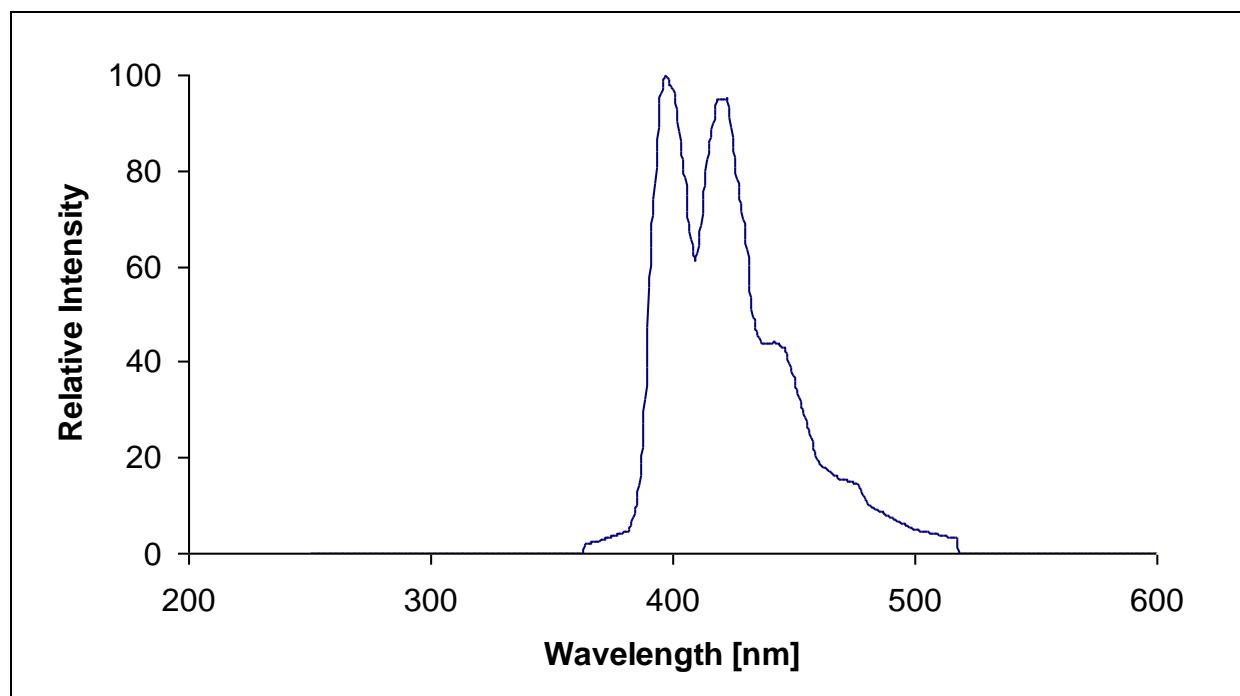
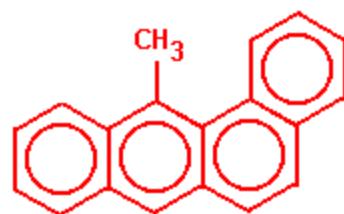
11-Methylbenz(a)anthracene



λ exc.	292.5 nm
Formula	$\text{C}_{19}\text{H}_{14}$
M.W.	242 u
H/H+C	0.424
m.p.	117.5 °C
m.a.c. (266 nm)	$4.2 (\text{l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4})$
solvent	Cyclohexane
source	Atlas pag. 418-II

P04AML

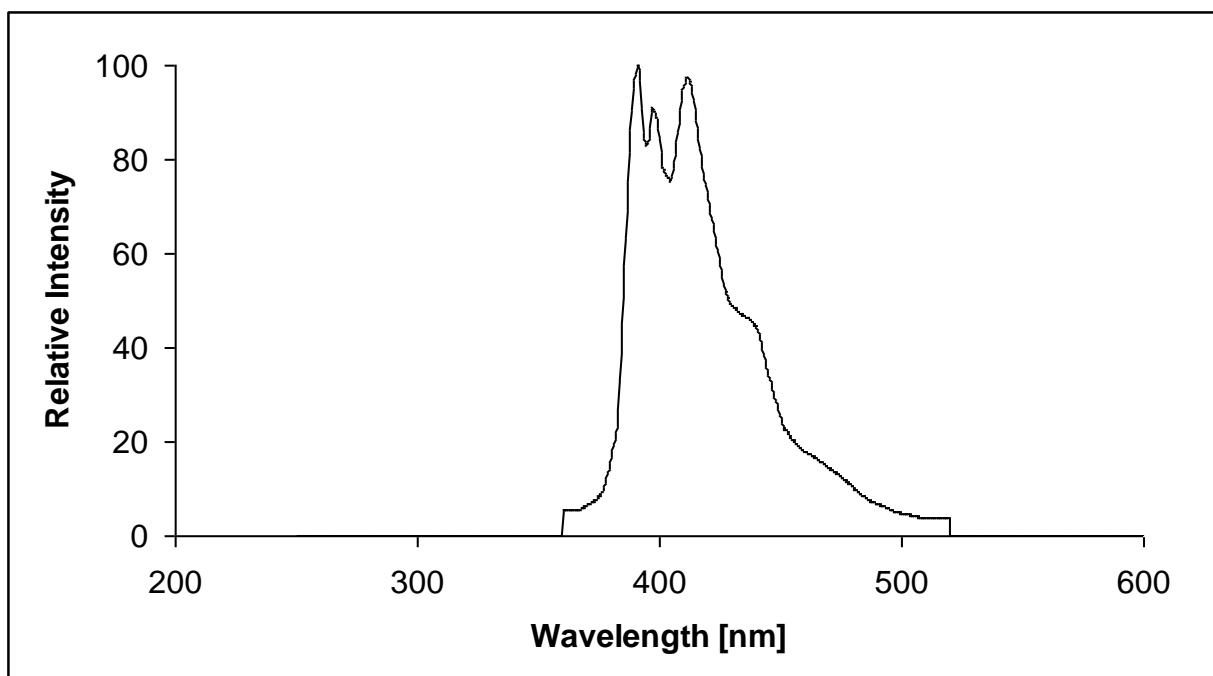
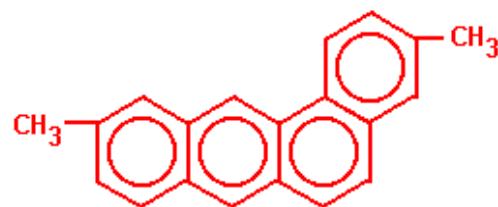
12-Methylbenz(a)anthracene



λ exc.	292 nm
Formula	C ₁₉ H ₁₄
M.W.	242 u
H/H+C	0.424
m.p.	138 °C
m.a.c. (266 nm)	3.15(l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	Cyclohexane
source	ATLAS pag.338-I (432-II)

P04AMM

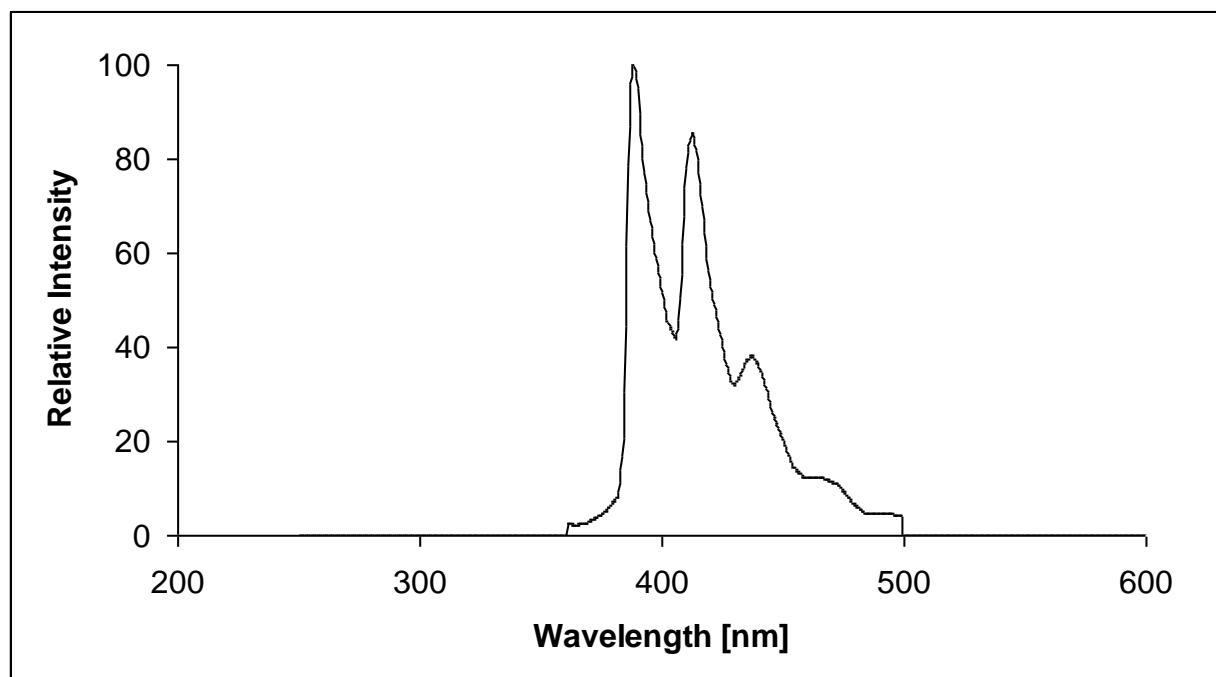
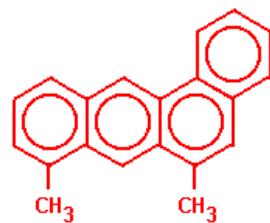
3,9 – Dimethylbenz(a)anthracene



λ exc.	292 nm
Formula	$\text{C}_{20}\text{H}_{16}$
M.W.	256 u
H/H+C	0.444
m.p.	186°C
m.a.c. (266 nm)	2 ($\text{l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4}$)
solvent	Cyclohexane
source	Atlas pg. 466-II

P04AMN

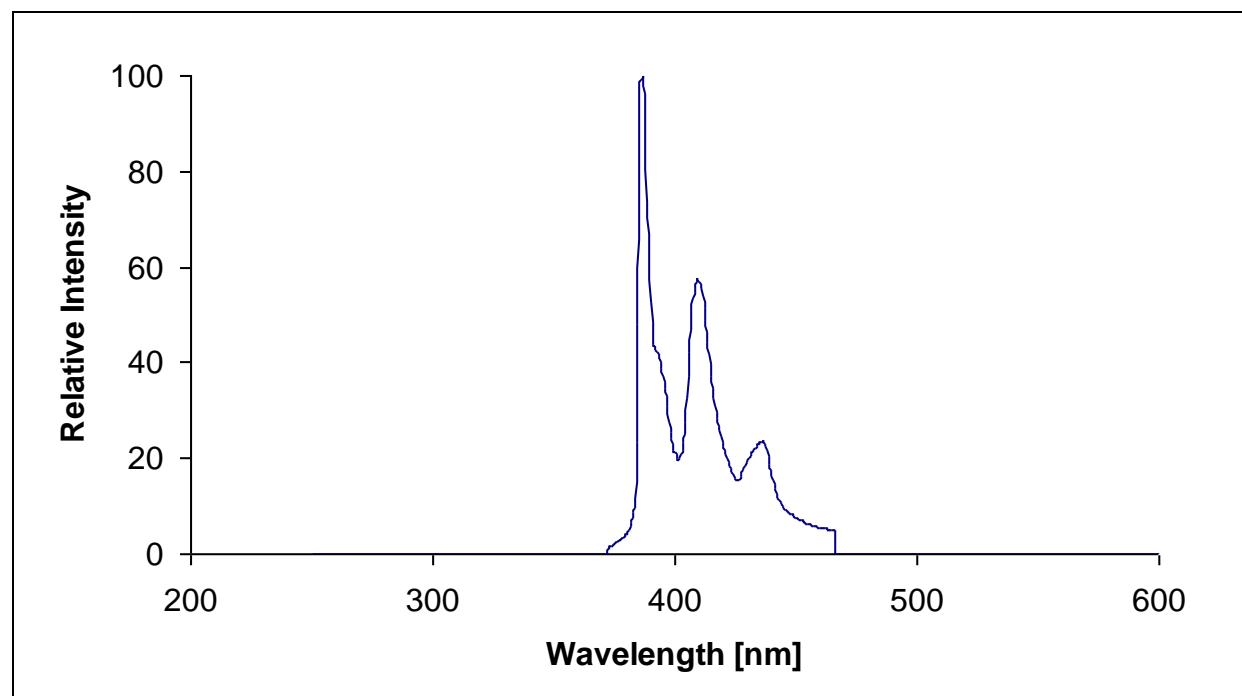
6,8 – Dimethylbenz(a)anthracene



λ exc.	291.5 nm
Formula	$\text{C}_{20}\text{H}_{16}$
M.W.	256 u
H/H+C	0.444
m.p.	143°C
m.a.c. (266 nm)	$4.4 (\text{l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4})$
solvent	Cyclohexane
source	Atlas pg. 484-II

P04AQ

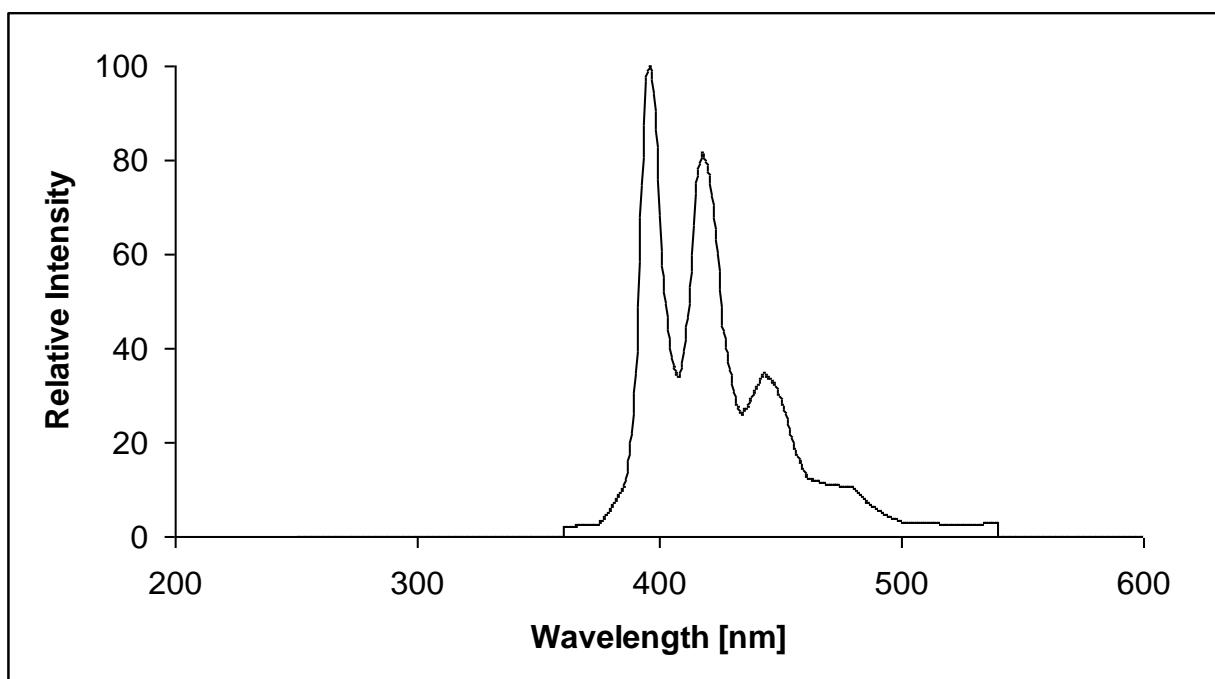
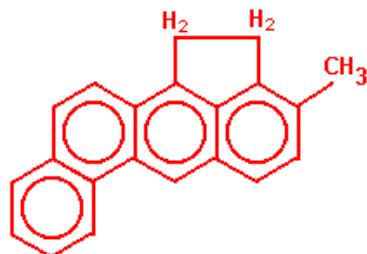
11H-Benzo(bc)aceanthrylene



λ exc.	350 nm
Formula	C ₁₉ H ₁₂
M.W.	240 u
H/H+C	0.387
m.p.	°C
m.a.c. (266 nm)	(l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	n-hexadecane
source	[9]

P04AW

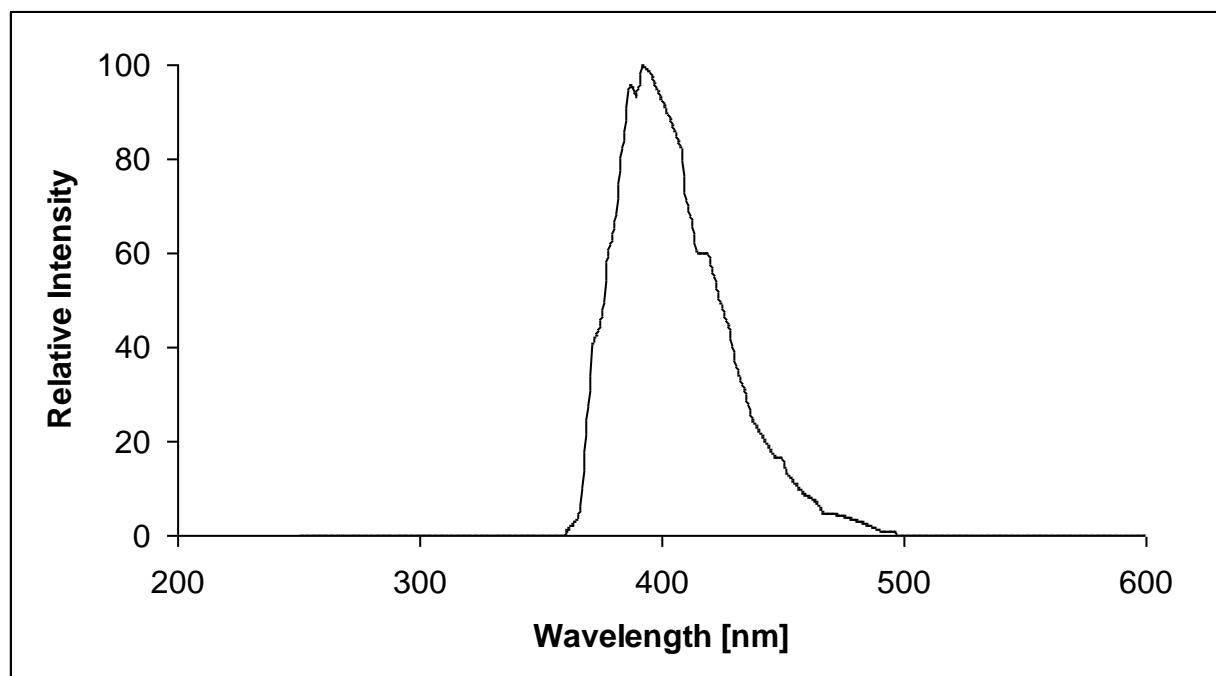
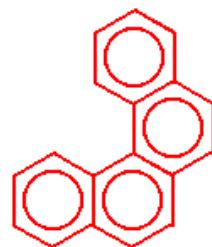
3 – Methyl-1,2-dihydrobenz(j)aceanthrylene
3 – Methylchcolanthrene



λ exc.	297 nm
Formula	$C_{21}H_{16}$
M.W.	268 u
H/H+C	0.432
m.p.	179°C
m.a.c. (266 nm)	$2.65 \text{ (l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4})$
solvent	Cyclohexane
source	Atlas pg. 554-II

P04B

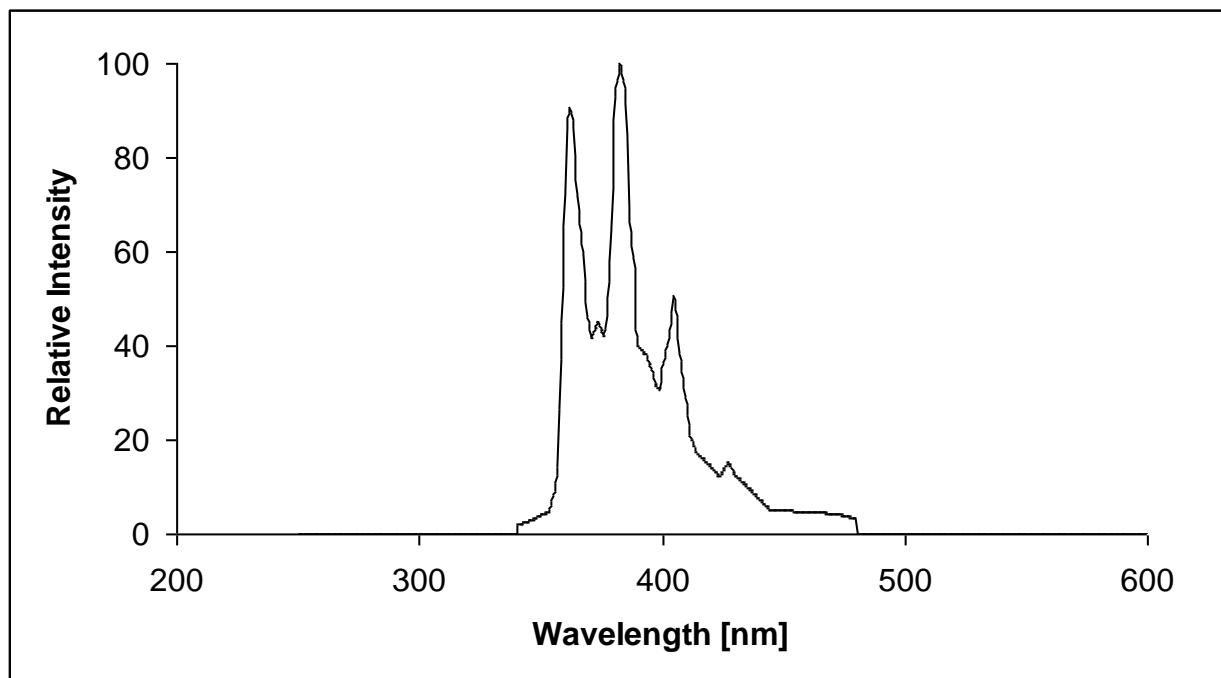
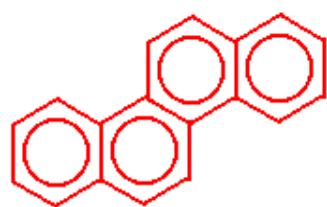
Benzo(c)phenanthrene



λ exc.	300 nm
Formula	$C_{18}H_{12}$
M.W.	228 u
H/H+C	0.400
m.p.	66.1°C
m.a.c. (266 nm)	$3.2 \text{ (l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4})$
solvent	Cyclohexane
source	Atlas pg. 226-I

P04C

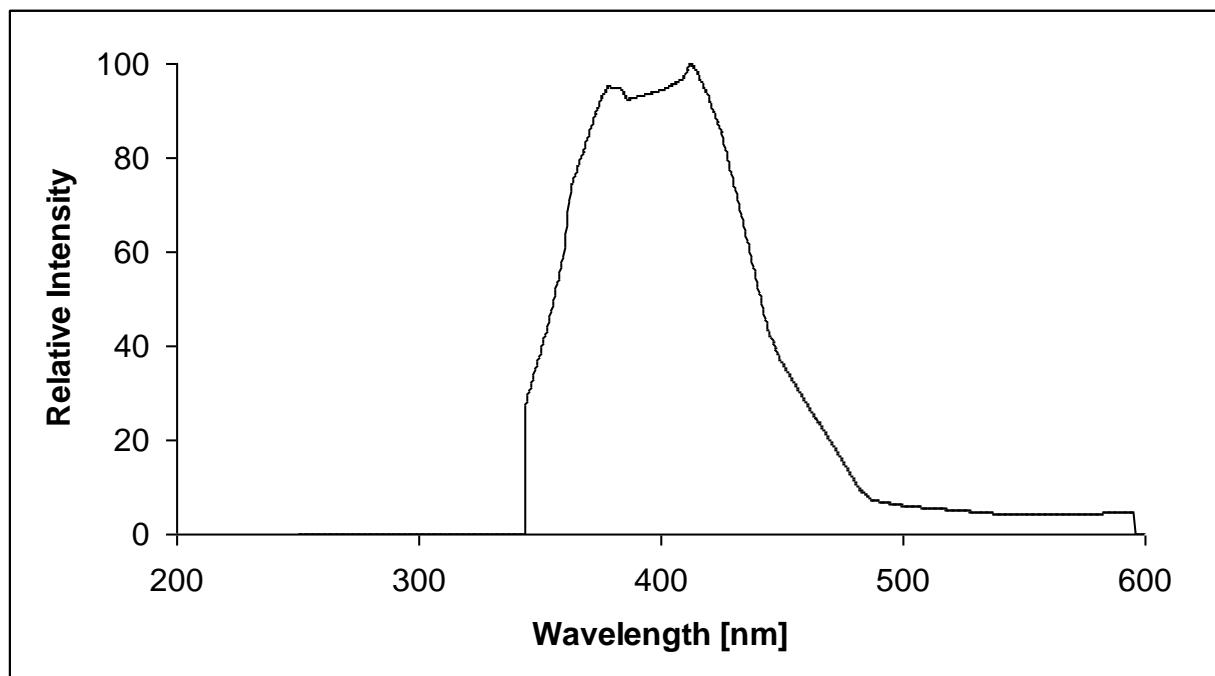
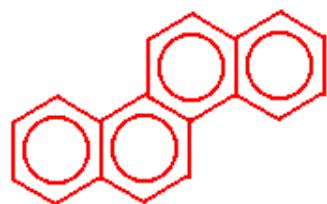
Chrysene (1)



λ exc.	269 nm
Formula	C ₁₈ H ₁₂
M.W.	228 u
H/H+C	0.400
m.p.	253.5°C
m.a.c. (266 nm)	9.6 (l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	Cyclohexane
source	Atlas pg. 246-I

P04CG

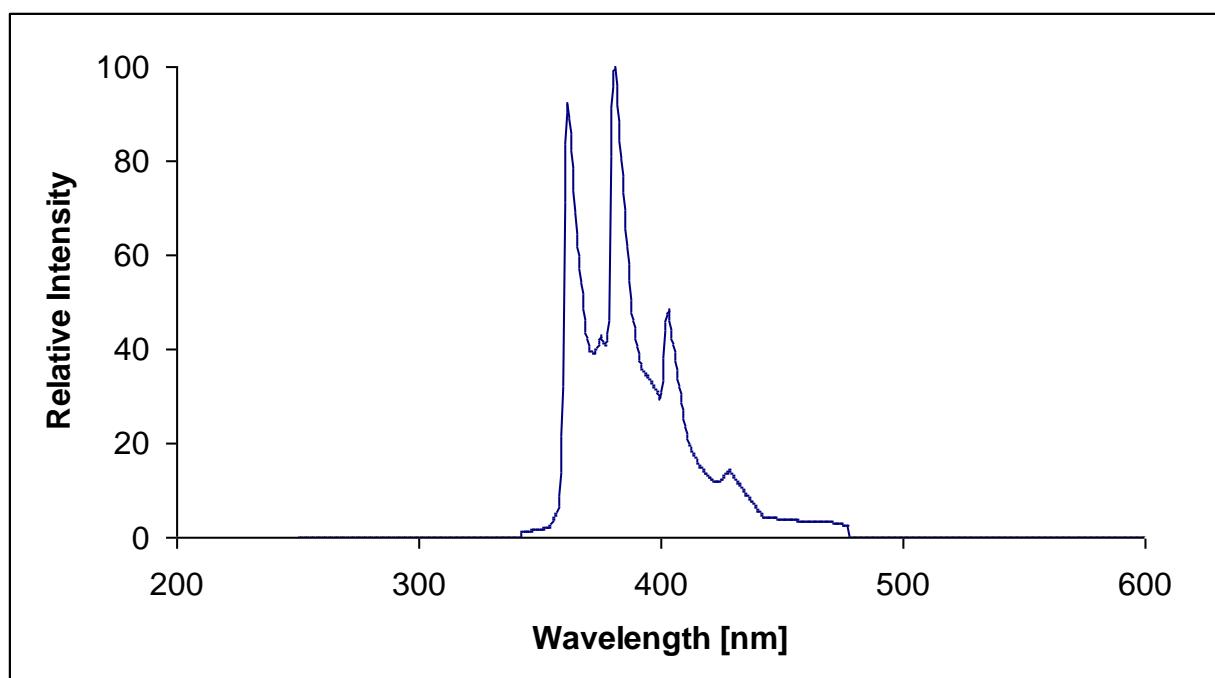
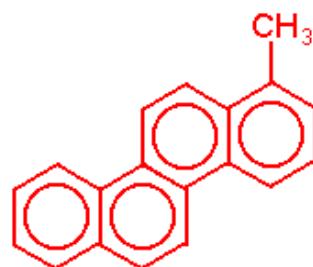
Chrysene (2)



λ exc.	337 nm
Formula	C ₁₈ H ₁₂
M.W.	228 u
H/H+C	0.400
m.p.	253.5°C
m.a.c. (266 nm)	(l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	Gas phase 714 K
source	[1]

P04CMA

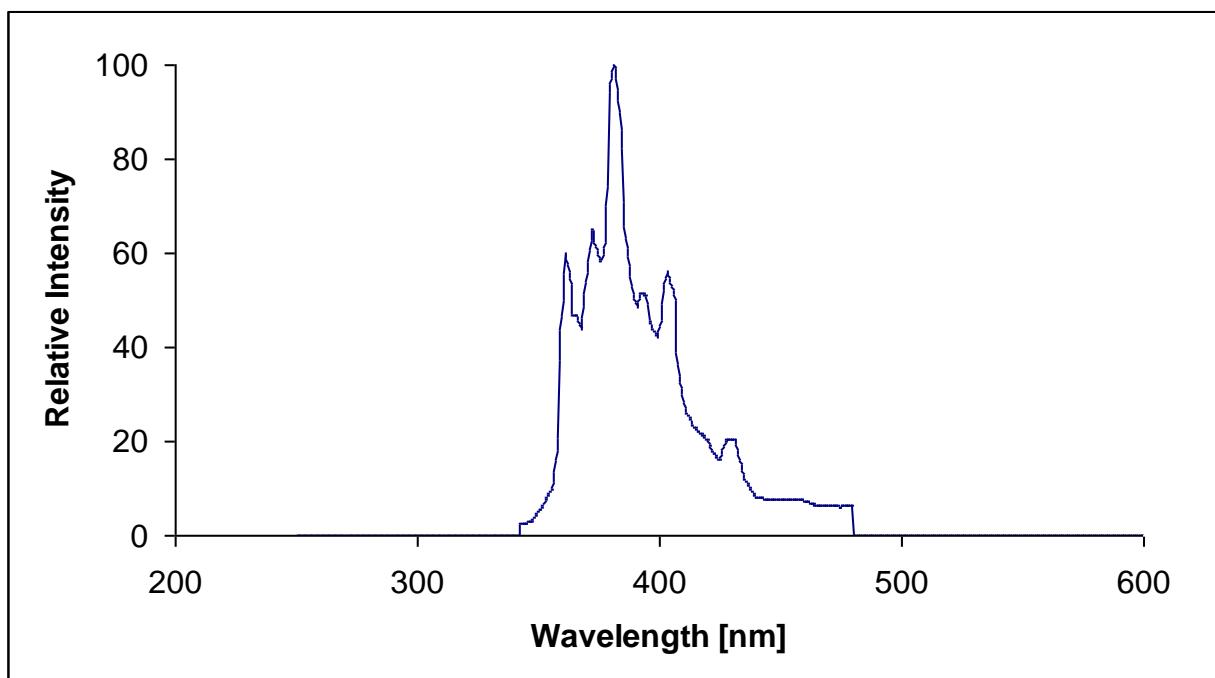
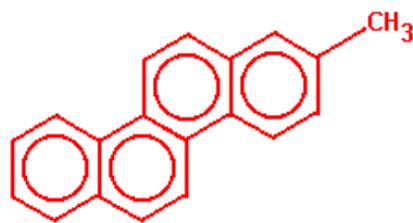
1 – Methylchrysene



λ exc.	271 nm
Formula	C ₁₉ H ₁₄
M.W.	242 u
H/H+C	0.424
m.p.	254.4°C
m.a.c. (266 nm)	7.4 (l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	Cyclohexane
source	Atlas pag. 354 -I

P04CMB

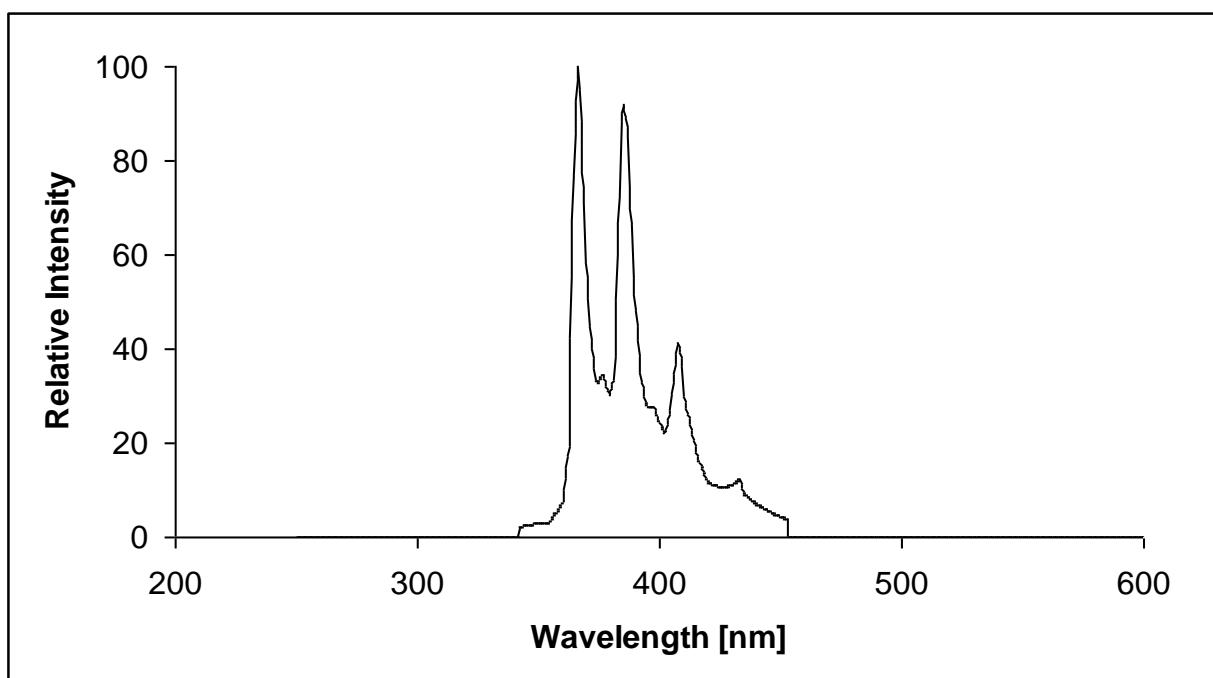
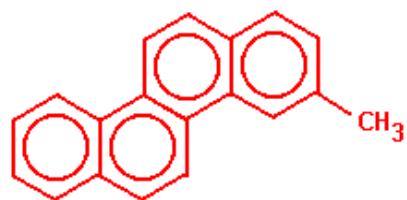
2 – Methylchrysene



λ exc.	217 nm
Formula	C ₁₉ H ₁₄
M.W.	242 u
H/H+C	0.424
m.p.	230.2°C
m.a.c. (266 nm)	7 (l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	Cyclohexane
source	Atlas pg. 372-I

P04CMC

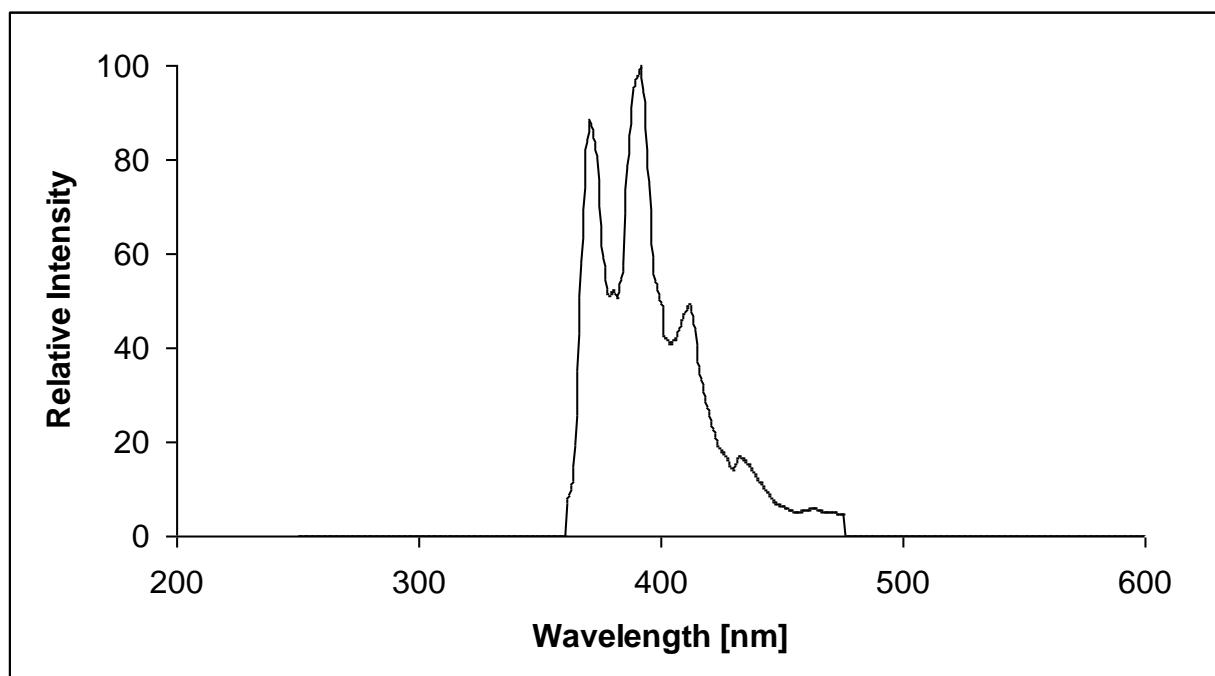
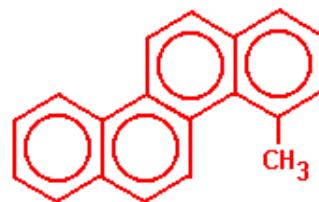
3 – Methylchrysene



λ exc.	271 nm
Formula	C ₁₉ H ₁₄
M.W.	242 u
H/H+C	0.424
m.p.	171.9°C
m.a.c. (266 nm)	7.1 (l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	Cyclohexane
source	Atlas pg. 390-I

P04CMD

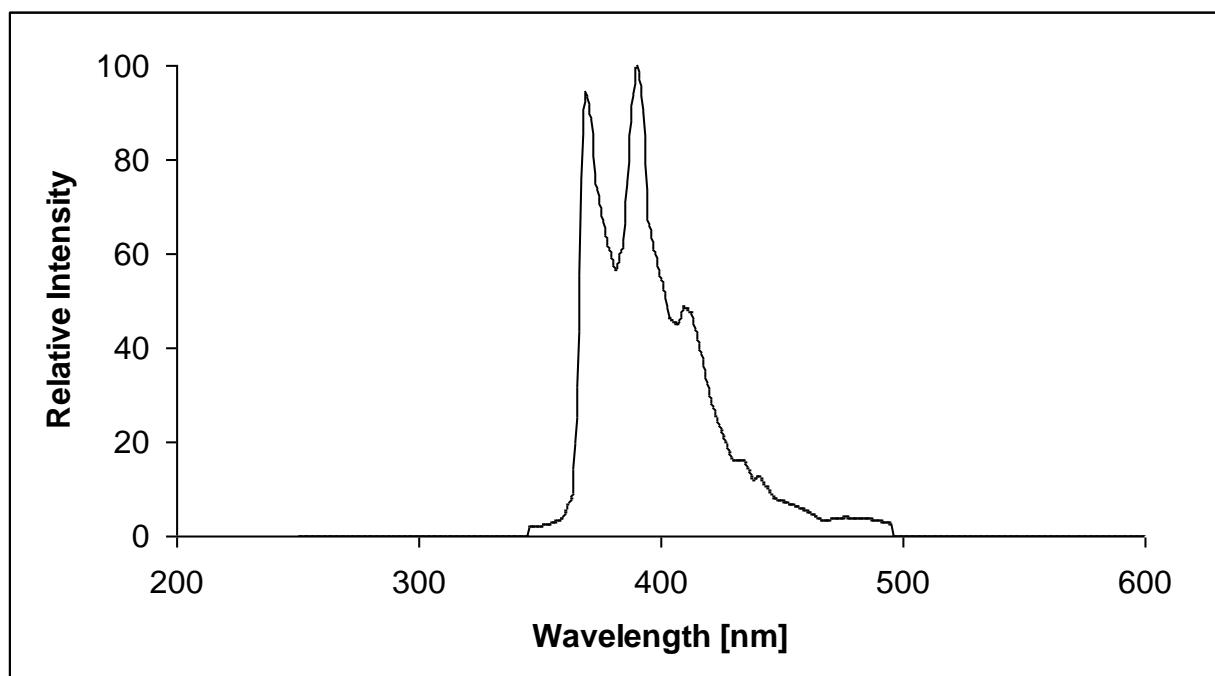
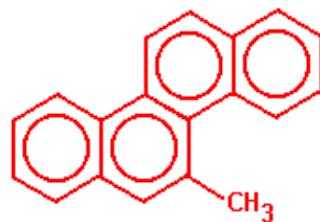
4 – Methylchrysene



λ exc.	272 nm
Formula	C ₁₉ H ₁₄
M.W.	242 u
H/H+C	0.424
m.p.	150.6°C
m.a.c. (266 nm)	7.5 (l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	Cyclohexane
source	Atlas pg. 408-I

P04CME

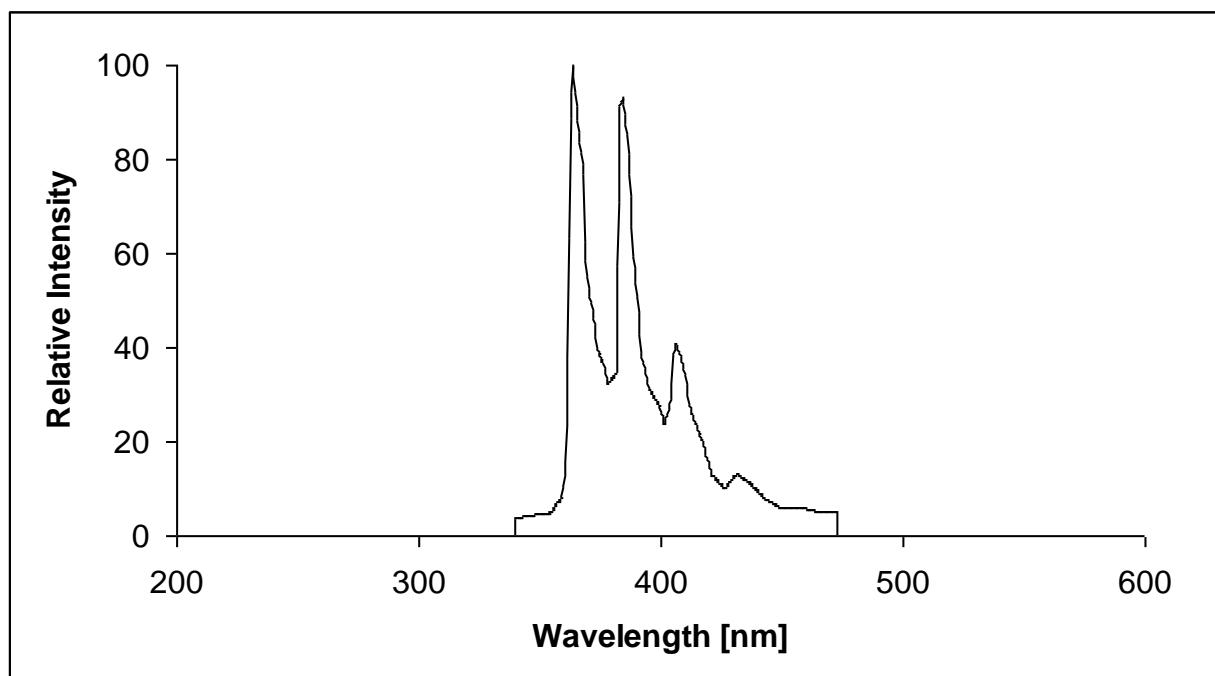
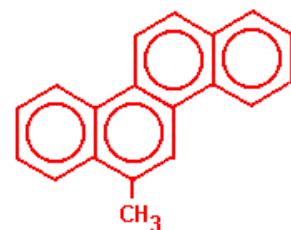
5 – Methylchrysene



λ exc.	271 nm
Formula	C ₁₉ H ₁₄
M.W.	242 u
H/H+C	0.424
m.p.	117.1°C
m.a.c. (266 nm)	7.7 (l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	Cyclohexane
source	Atlas pg. 426-I

P04CMF

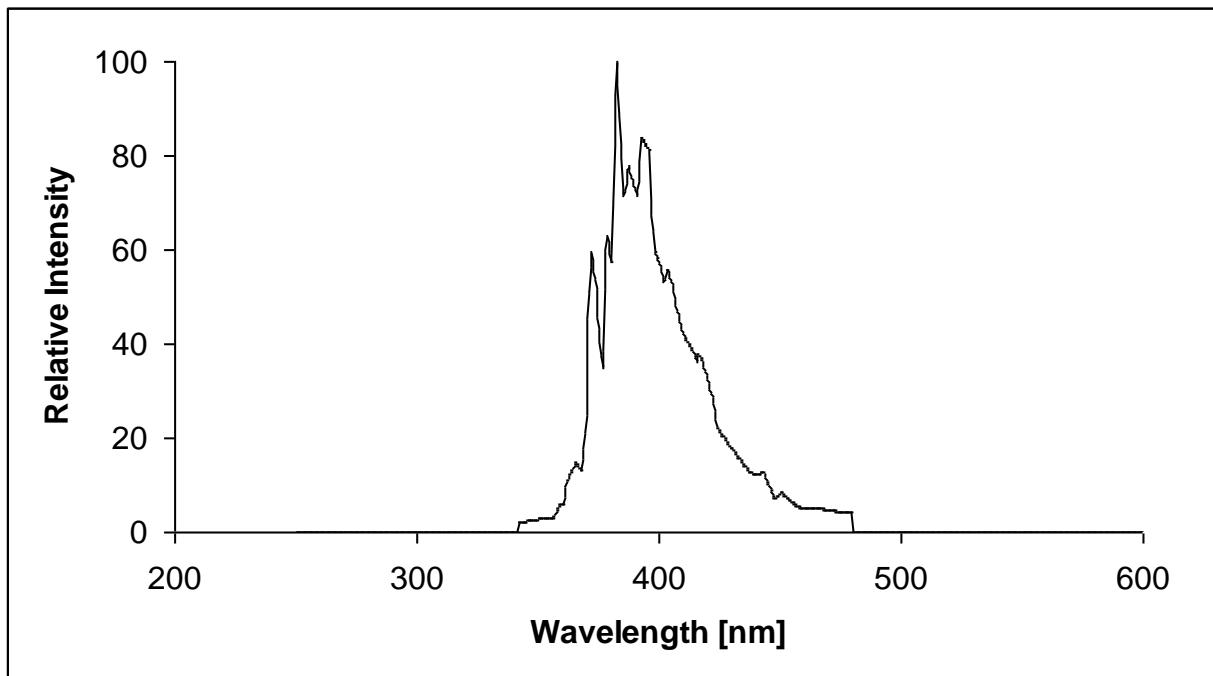
6 – Methylchrysene



λ exc.	270 nm
Formula	$\text{C}_{19}\text{H}_{14}$
M.W.	242 u
H/H+C	0.424
m.p.	160-161°C
m.a.c. (266 nm)	$6.95 (\text{l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4})$
solvent	Cyclohexane
source	Atlas pg. 444-I

P04P0

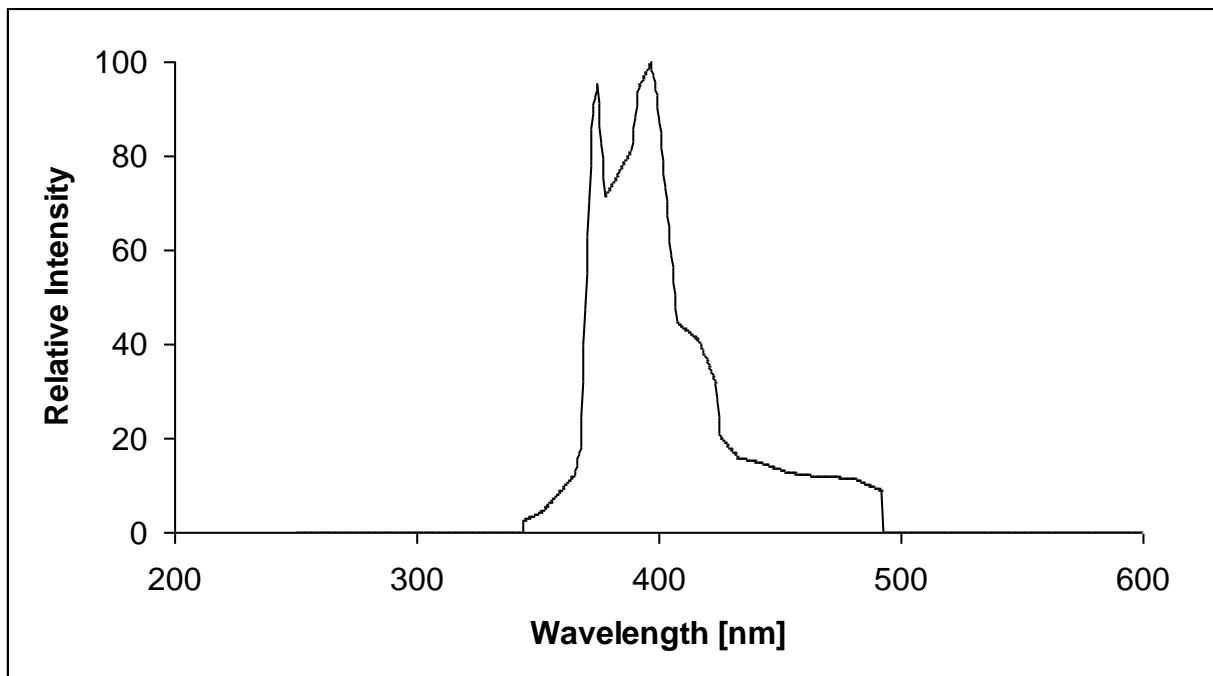
Pyrene (1)



λ exc.	335.5 nm
Formula	C ₁₆ H ₁₀
M.W.	202 u
H/H+C	0.385
m.p.	150.4°C
m.a.c. (266 nm)	1.8 (l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	Cyclohexane
source	Atlas pg. 94-I

P04P0A

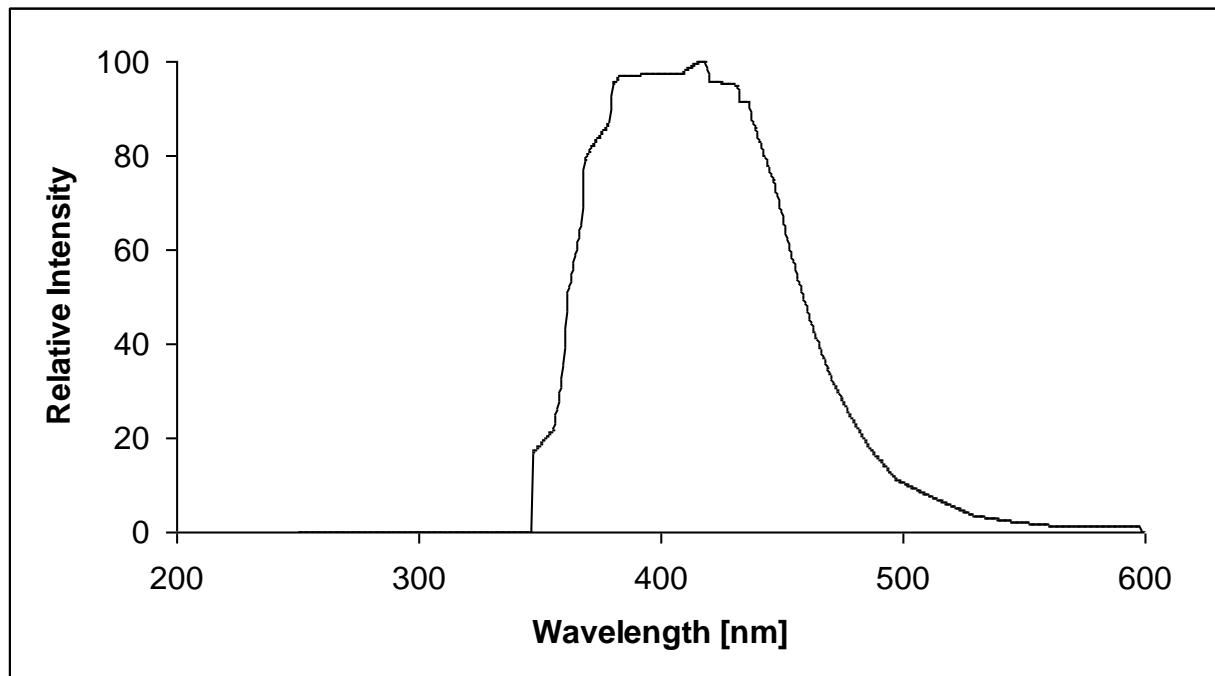
Pyrene (2)



λ exc.	220 nm
Formula	$C_{16}H_{10}$
M.W.	202 u
H/H+C	0.385
m.p.	150.4°C
m.a.c. (266 nm)	(l. $mol^{-1} cm^{-1} \times 10^{-4}$)
solvent	Acetonitrile
source	[10]

P04P0G1

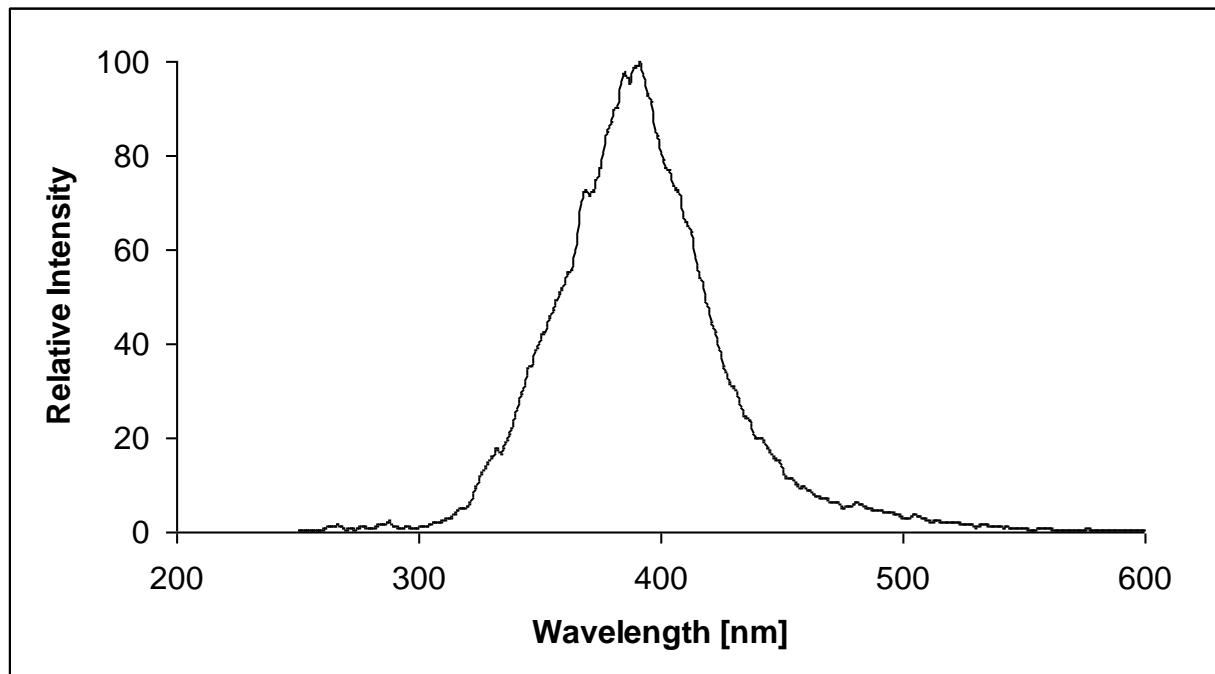
Pyrene (3)



λ exc.	337 nm
Formula	$C_{16}H_{10}$
M.W.	202 u
H/H+C	0.385
m.p.	150.4°C
m.a.c. (266 nm)	(l. $mol^{-1} cm^{-1} \times 10^{-4}$)
solvent	Gas phase 667 K
source	[1]

P04P0G2

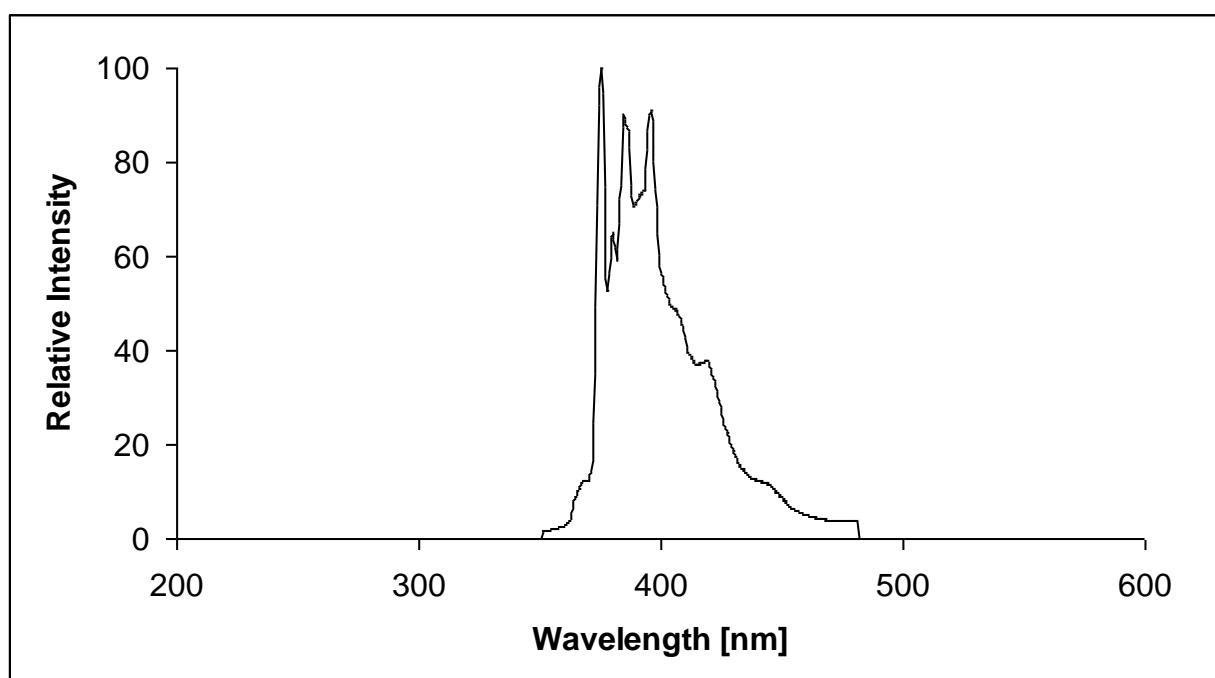
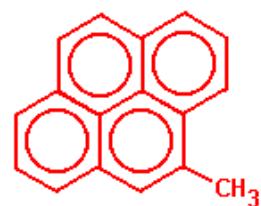
Pyrene (4)



λ exc.	266 nm
Formula	$C_{16}H_{10}$
M.W.	202 u
H/H+C	0.385
m.p.	150.4°C
m.a.c. (266 nm)	(l. $mol^{-1} cm^{-1} \times 10^{-4}$)
solvent	Gas phase
source	CNPM

P04PMA

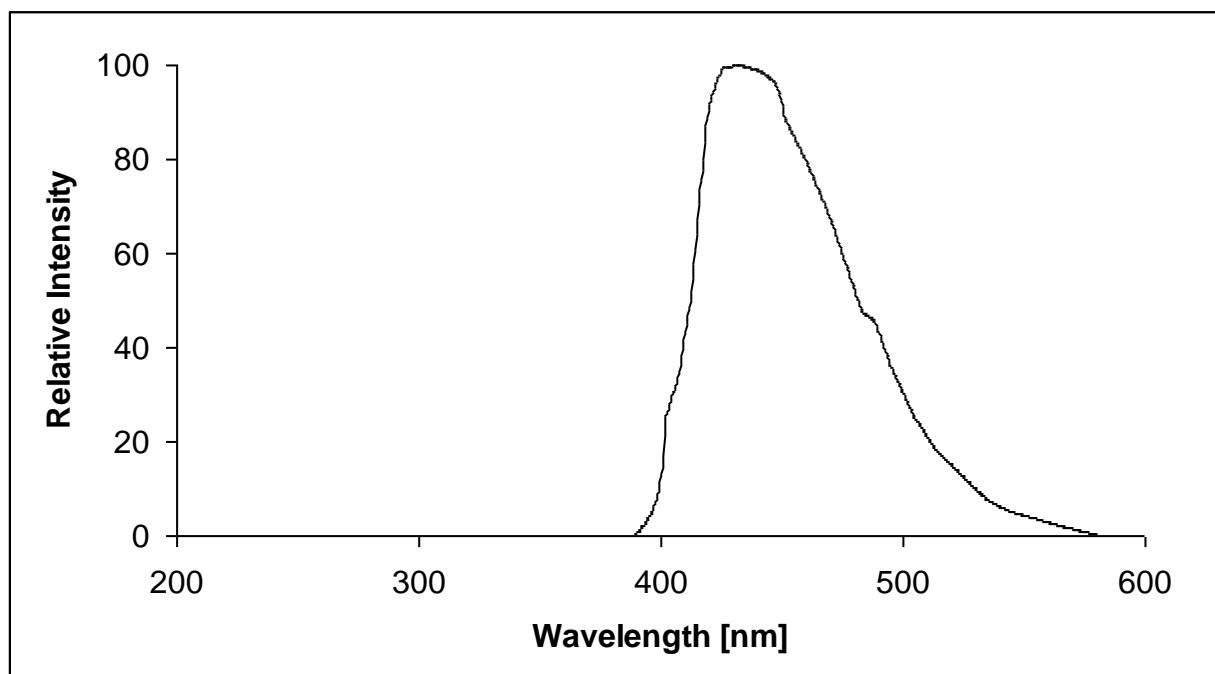
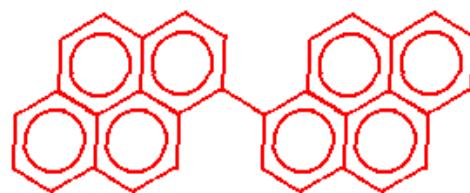
4 – Methylpyrene



λ exc.	337.5 nm
Formula	$\text{C}_{17}\text{H}_{12}$
M.W.	216 u
H/H+C	0.414
m.p.	146°C
m.a.c. (266 nm)	$1.25 (\text{l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4})$
solvent	Cyclohexane
source	Atlas pg. 914-III

P04PP

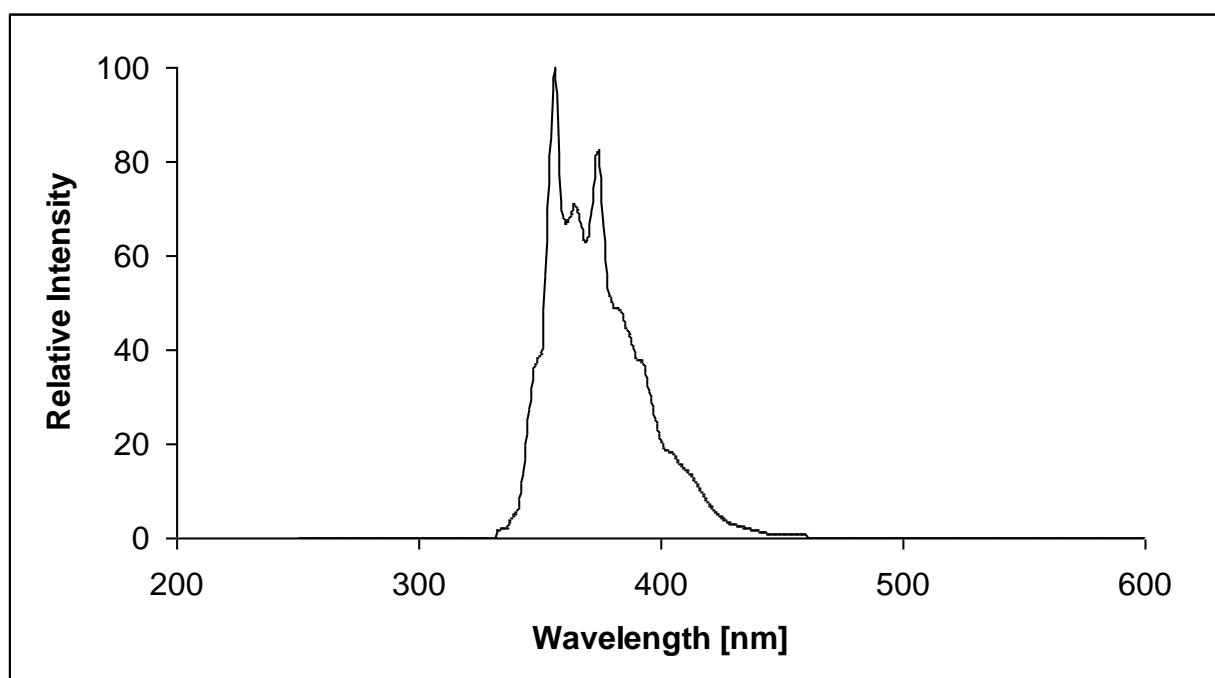
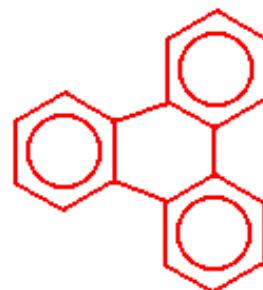
Biphenyl



λ exc.	365 nm
Formula	$C_{32}H_{18}$
M.W.	402 u
H/H+C	0.360
m.p.	°C
m.a.c. (266 nm)	(l. $mol^{-1} cm^{-1} \times 10^{-4}$)
solvent	Cyclohexane
source	Berlman

P04R

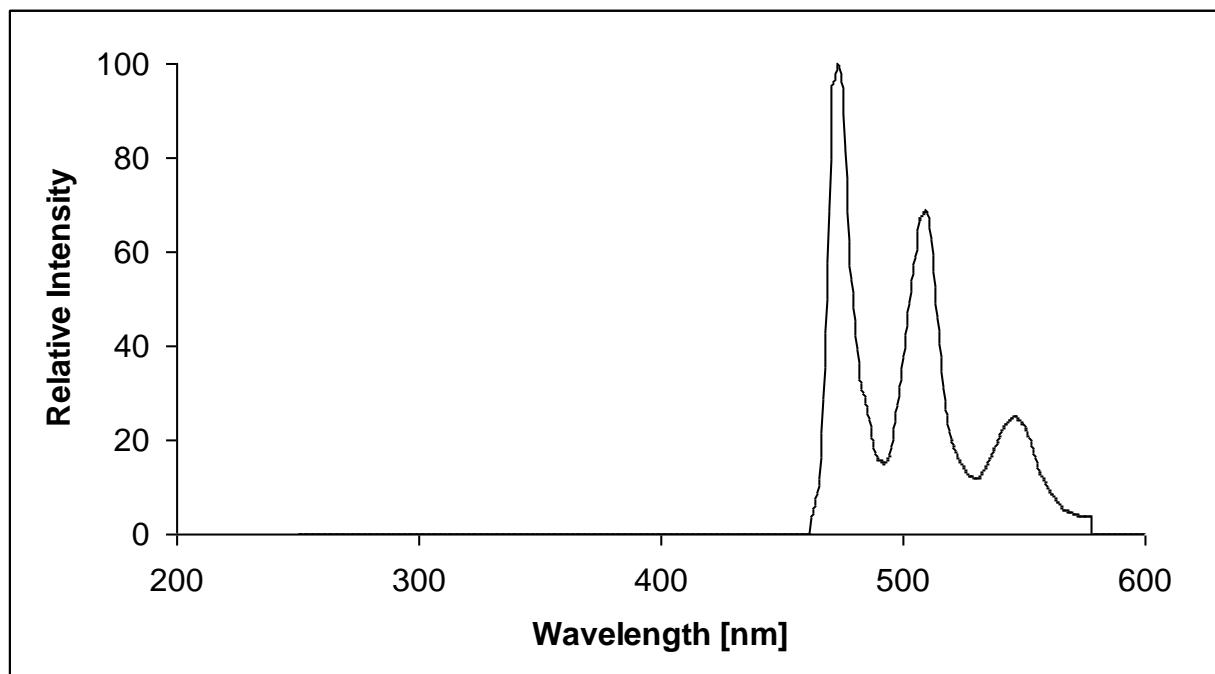
Triphenylene



λ exc.	258 nm
Formula	$C_{18}H_{12}$
M.W.	228 u
H/H+C	0.400
m.p.	199°C
m.a.c. (266 nm)	$1.95 \text{ (l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4})$
solvent	Cyclohexane
source	Atlas pg. 212-II

P04T1

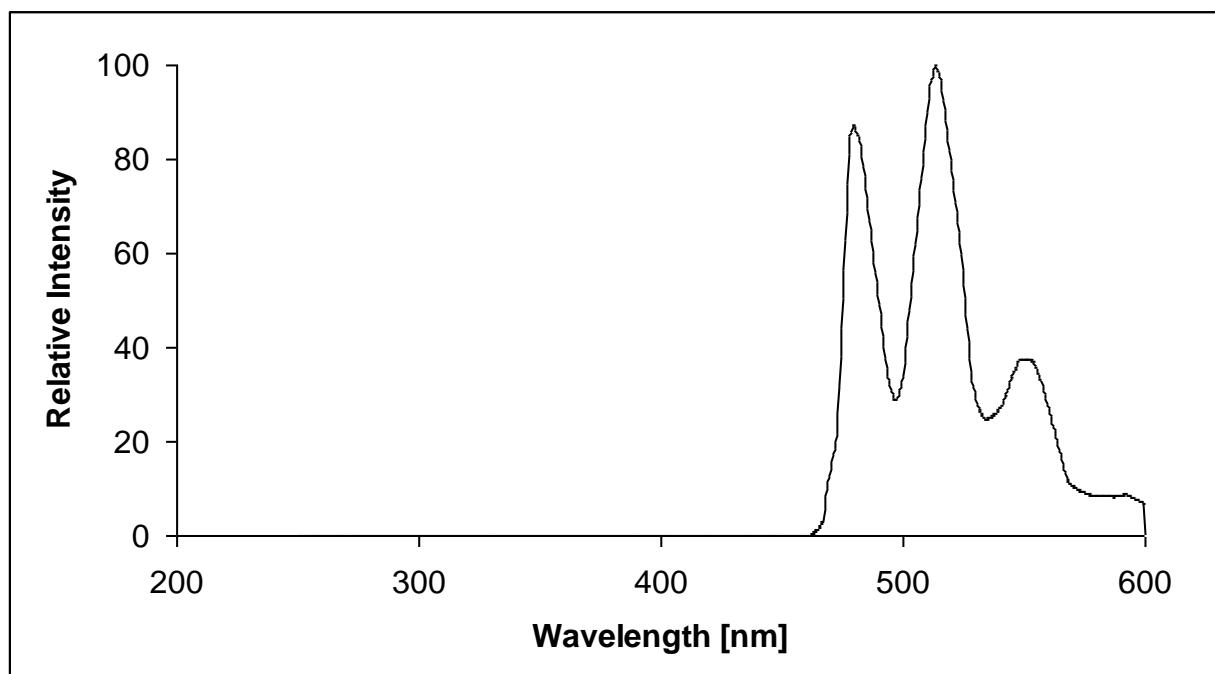
Tetracene(1)
Naphthacene



λ exc.	294 nm
Formula	$C_{18}H_{12}$
M.W.	228 u
H/H+C	0.400
m.p.	357°C
m.a.c. (266 nm)	$7.22 \text{ (l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4})$
solvent	Cyclohexane
source	Atlas pg. 230-II

P04T2

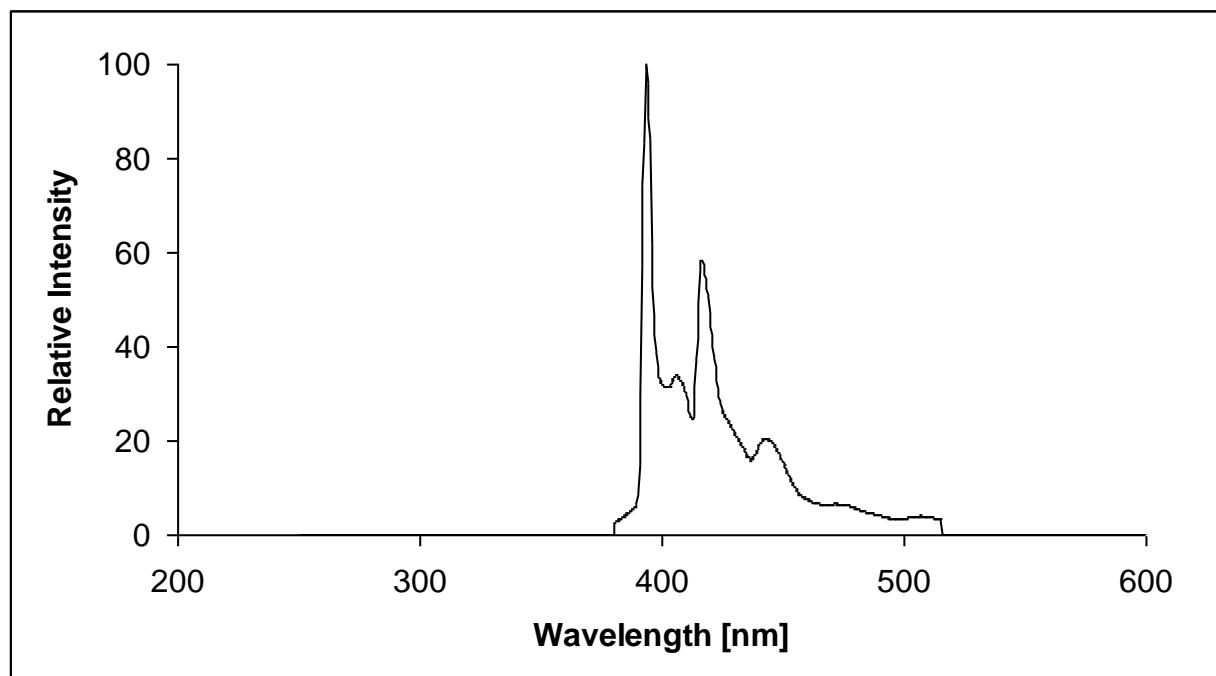
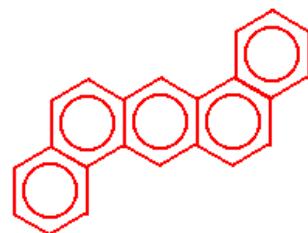
Tetracene(2)
Naphthacene



λ exc.	303 nm
Formula	$C_{18}H_{12}$
M.W.	228 u
H/H+C	0.400
m.p.	°C
m.a.c. (266 nm)	(l. $mol^{-1} cm^{-1} \times 10^{-4}$)
solvent	Benzene
source	Berlman

P05B1

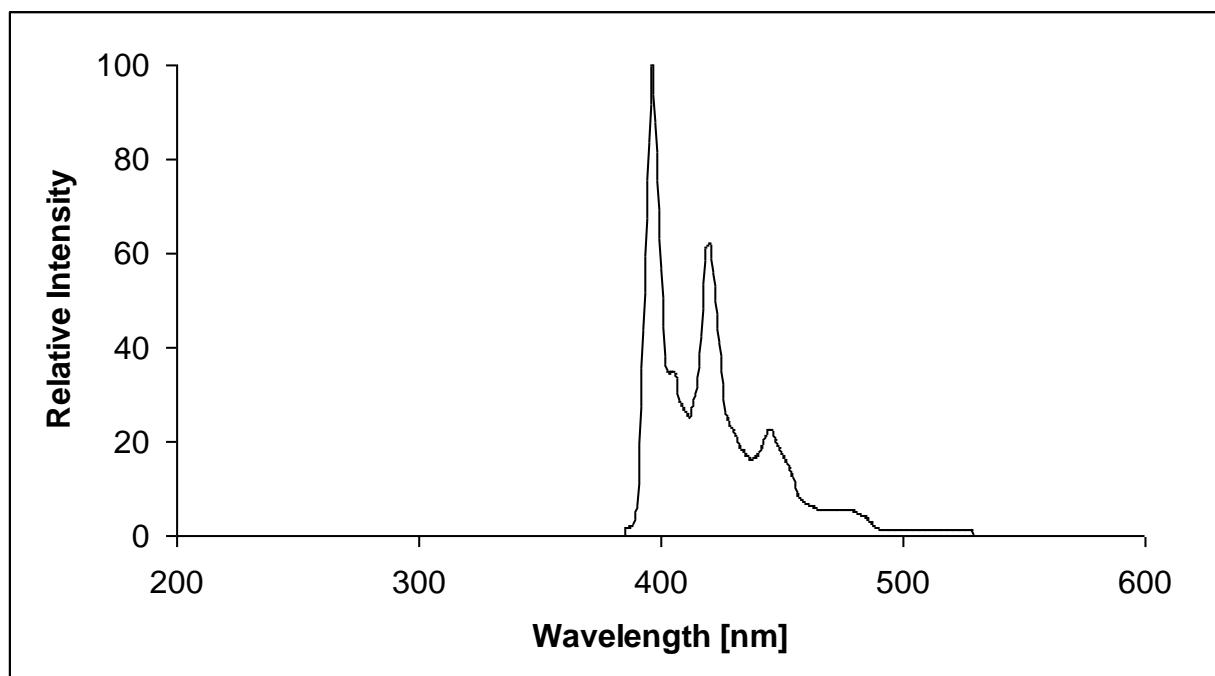
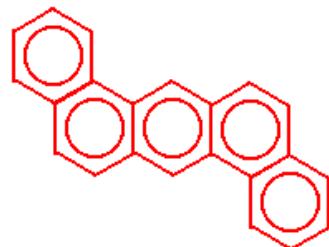
Dibenz(a,h)anthracene (1)



λ exc.	298 nm
Formula	$C_{22}H_{14}$
M.W.	278 u
H/H+C	0.389
m.p.	266.6°C
m.a.c. (266 nm)	$1.85 \text{ (l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4})$
solvent	Cyclohexane
source	Atlas pg. 726-I

P05B2

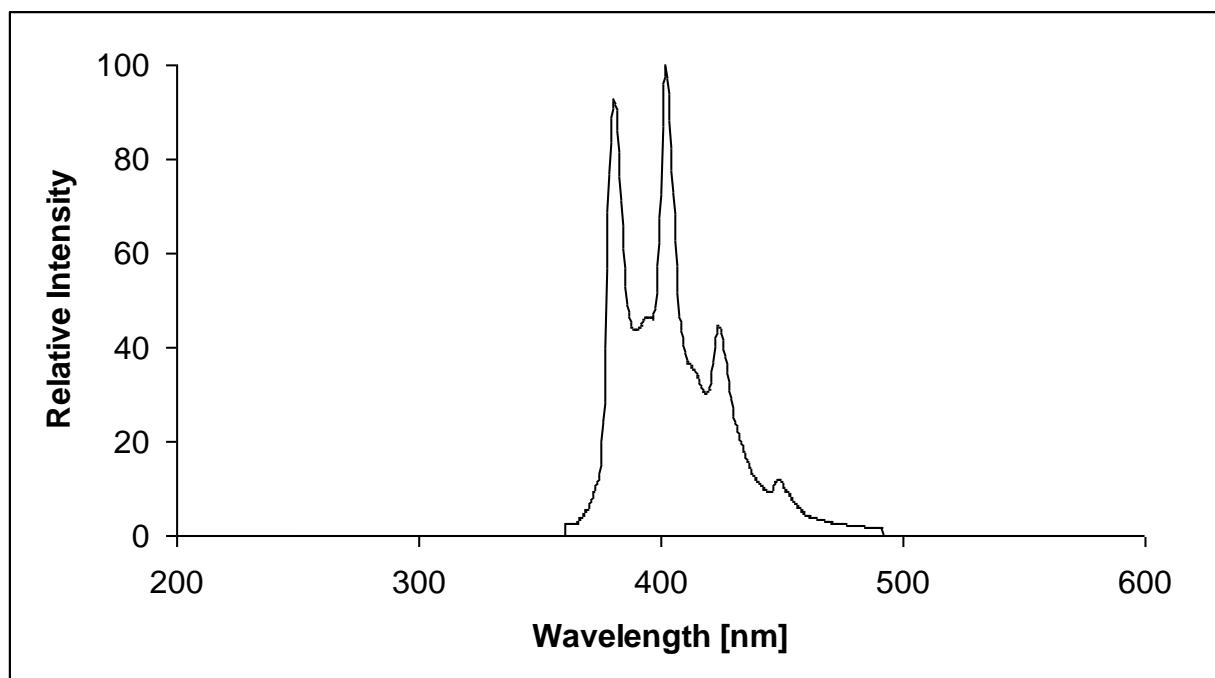
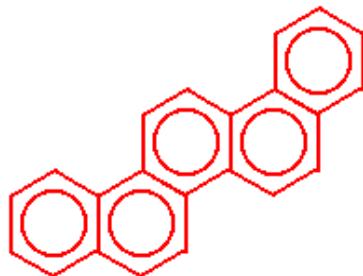
Dibenz(a,h)anthracene (2)



λ exc.	303 nm
Formula	$C_{22}H_{14}$
M.W.	278 u
H/H+C	0.389
m.p.	°C
m.a.c. (266 nm)	(l. $mol^{-1} cm^{-1} \times 10^{-4}$)
solvent	Benzene
source	Berlman

P05C

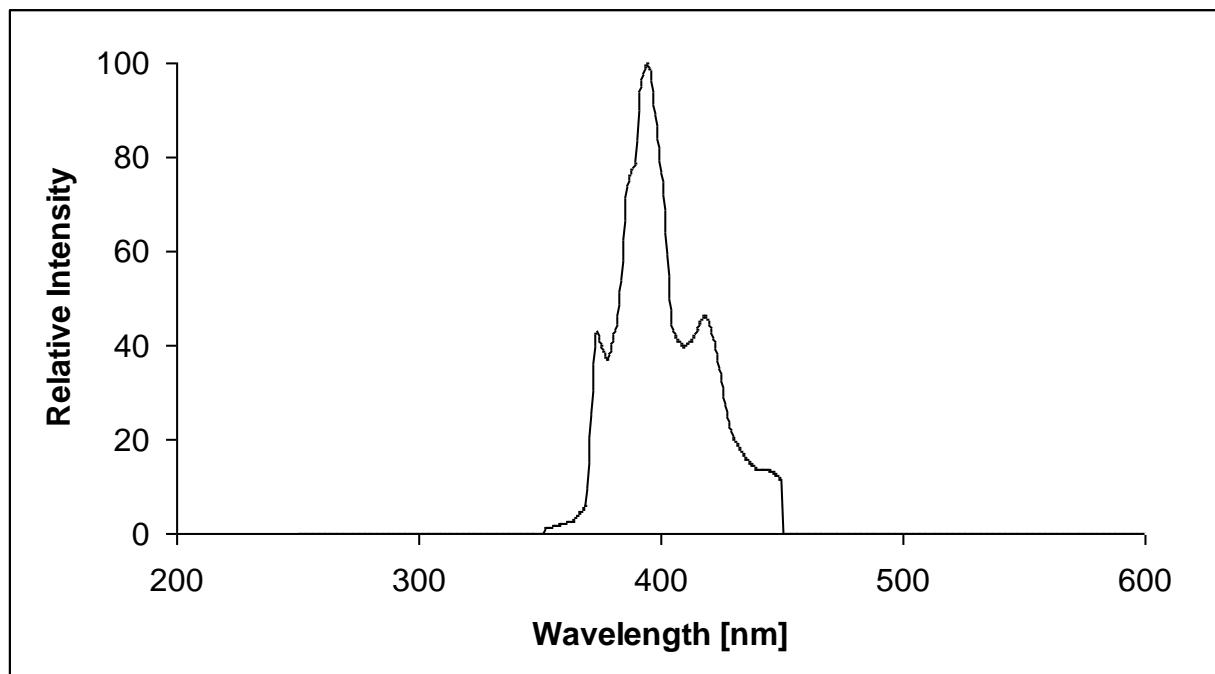
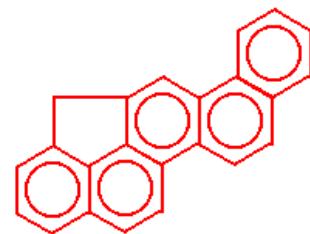
Picene



λ exc.	284 nm
Formula	C ₂₂ H ₁₄
M.W.	278 u
H/H+C	0.389
m.p.	364°C
m.a.c. (266 nm)	4.65 (l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	Cyclohexane
source	Atlas pg. 604-II

P05CC

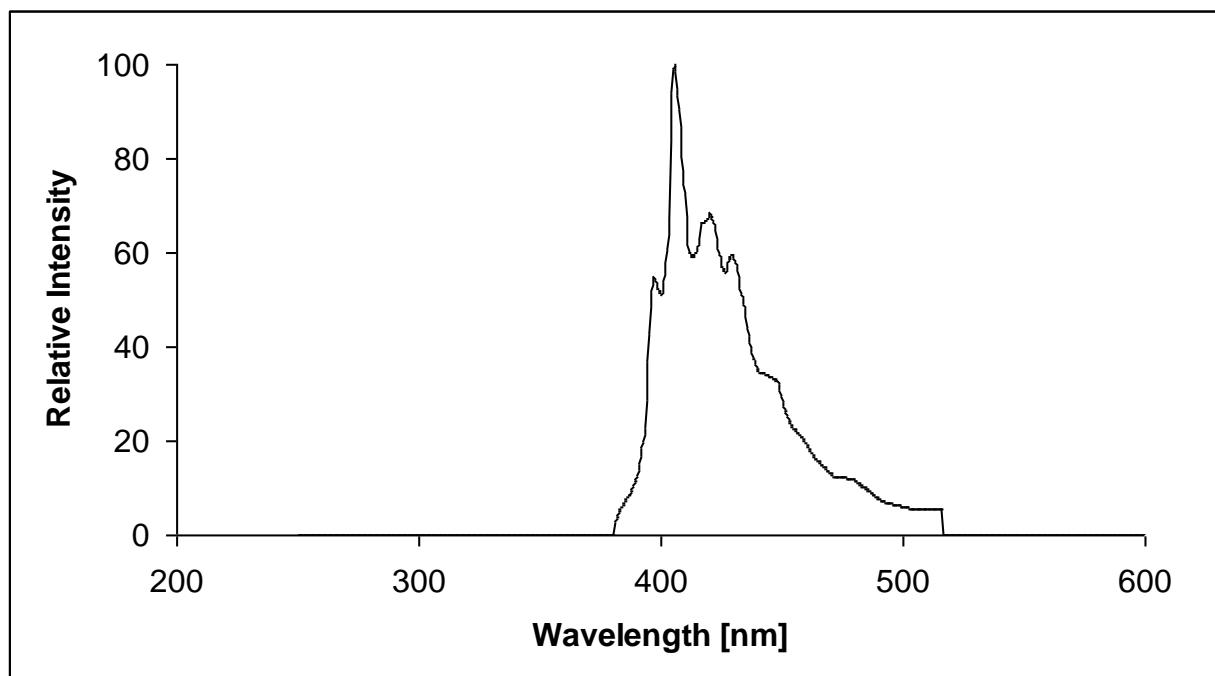
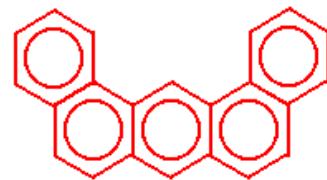
4H – Cyclopenta(pqr)picene



λ exc.	330 nm
Formula	C ₂₃ H ₁₄
M.W.	290 u
H/H+C	0.378
m.p.	°C
m.a.c. (266 nm)	(l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	n-hexadecane
source	[9]

P05D

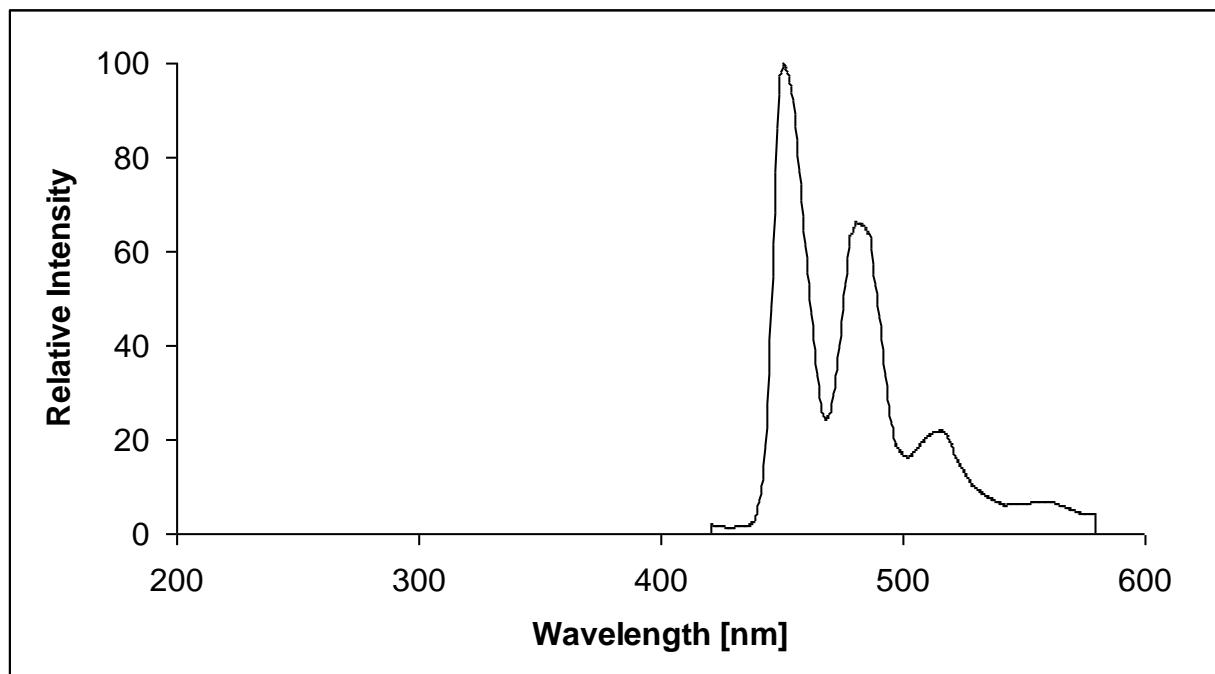
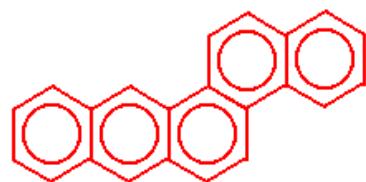
Dibenz(a,j)anthracene



λ exc.	298 nm
Formula	$C_{22}H_{14}$
M.W.	278 u
H/H+C	0.389
m.p.	197.3°C
m.a.c. (266 nm)	1.9 ($l \cdot mol^{-1} cm^{-1} \times 10^{-4}$)
solvent	Cyclohexane
source	Atlas pg. 744-I

P05E

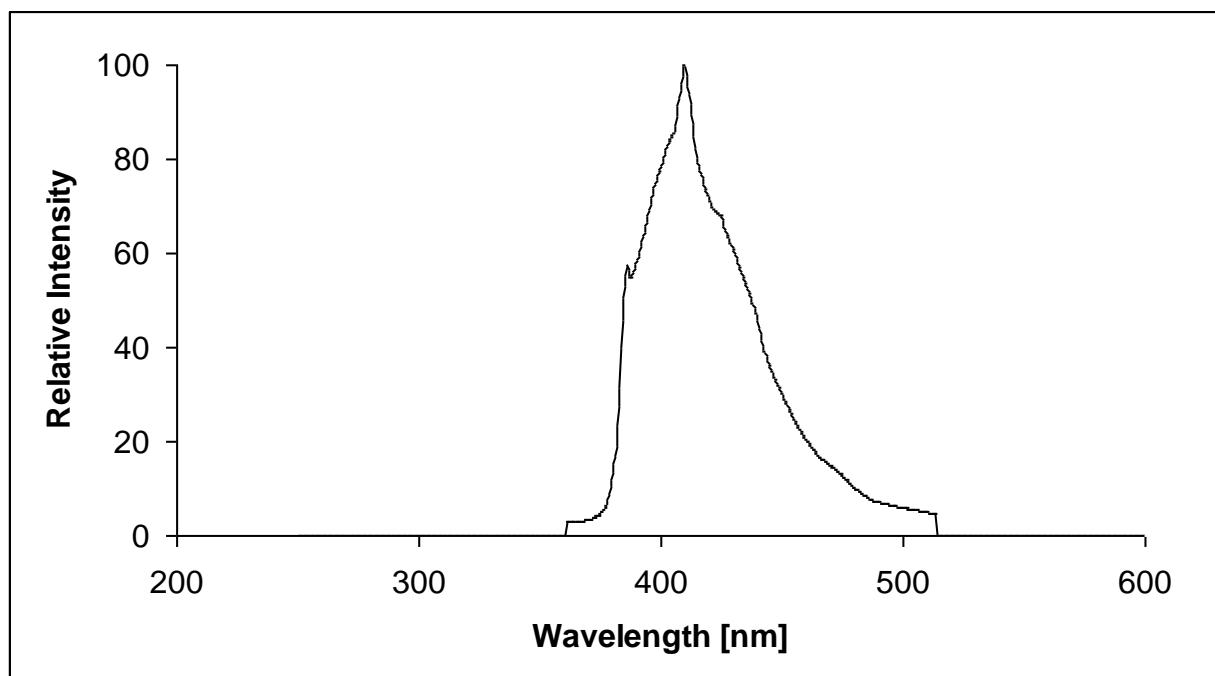
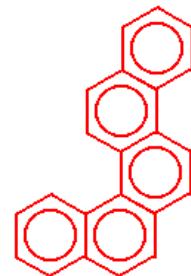
Benzo(b)chrysene



λ exc.	287 nm
Formula	$C_{22}H_{14}$
M.W.	278 u
H/H+C	0.389
m.p.	299.7°C
m.a.c. (266 nm)	$3.25 \text{ (l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4})$
solvent	Cyclohexane
source	Atlas pg. 670-I

P05F

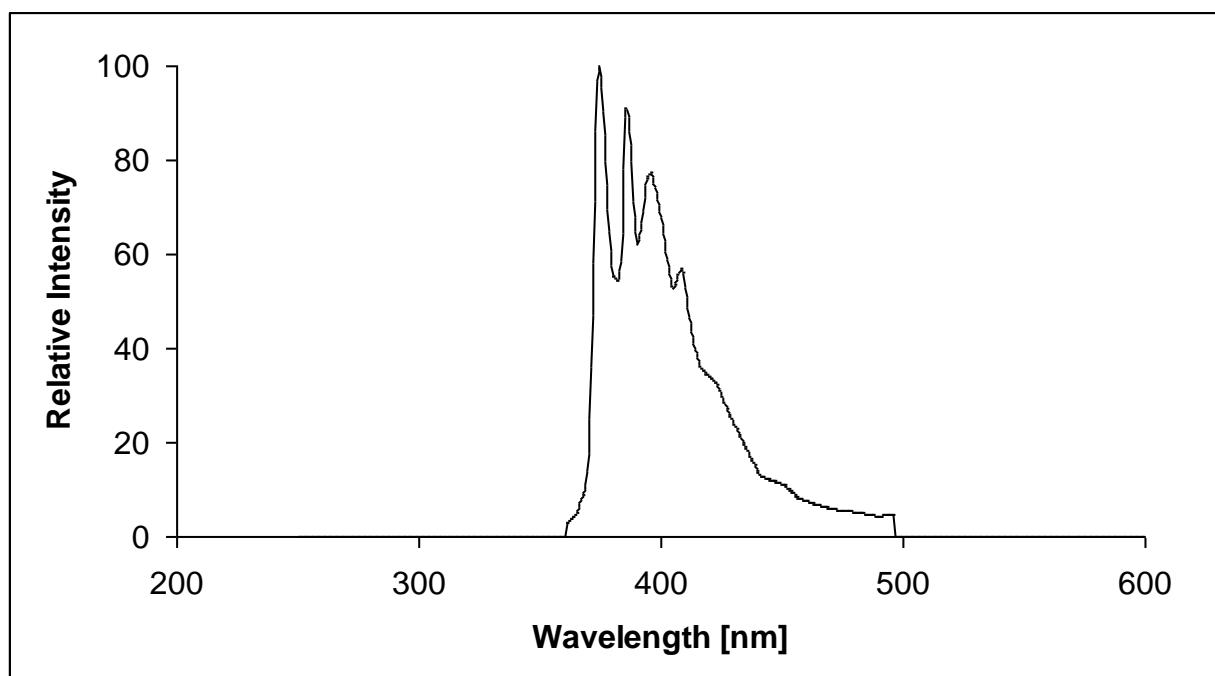
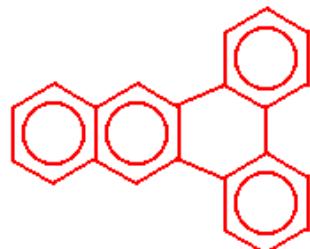
Benzo(c)chrysene



λ exc.	293 nm
Formula	C ₂₂ H ₁₄
M.W.	278 u
H/H+C	0.389
m.p.	126.1°C
m.a.c. (266 nm)	1.54 (l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	Cyclohexane
source	Atlas pg. 686-I

P05G

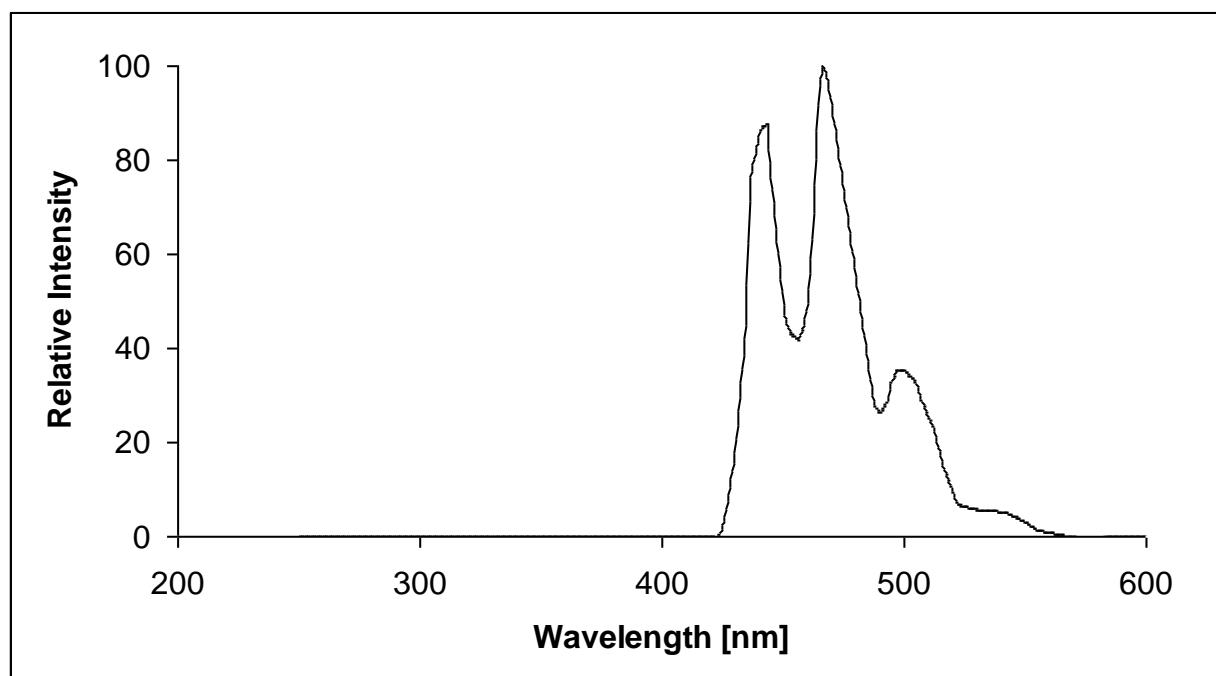
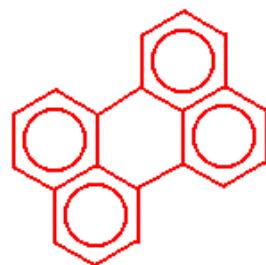
Benzo(b)triphenylene
Dibenz(ac)anthracene



λ exc.	287 nm
Formula	$C_{22}H_{14}$
M.W.	278 u
H/H+C	0.389
m.p.	205.6°C
m.a.c. (266 nm)	$5.26 \text{ (l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4})$
solvent	Cyclohexane
source	Atlas pg. 708-I

P05P01

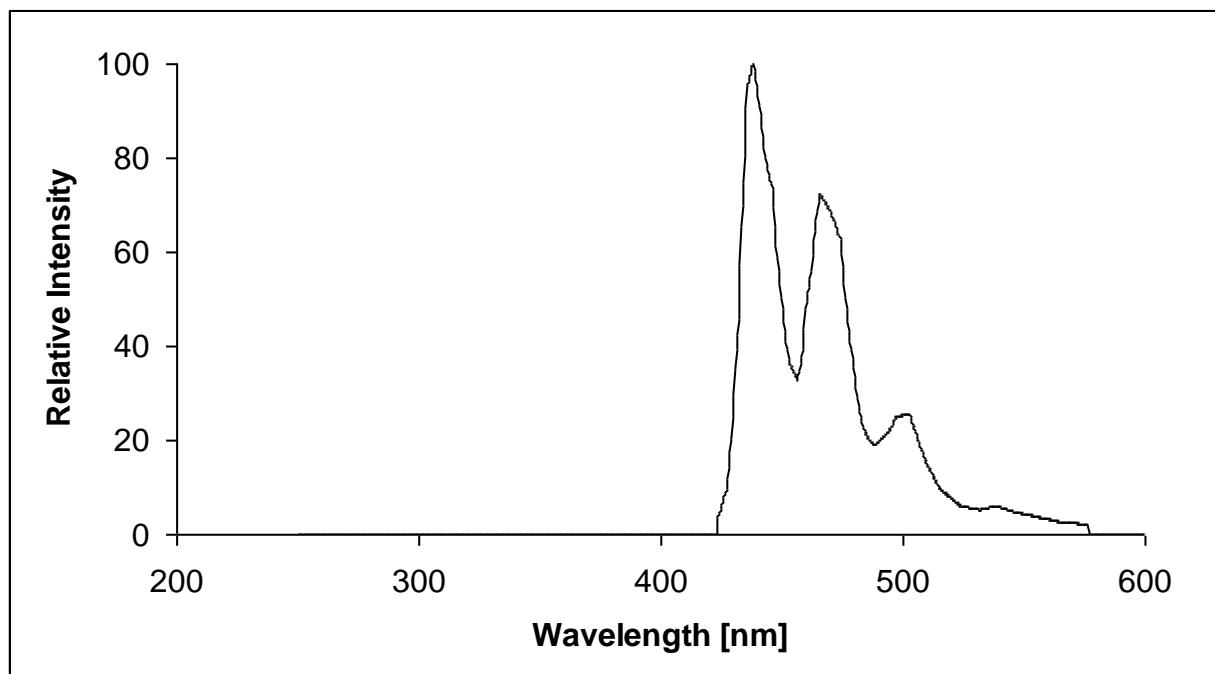
Perylene (1)



λ exc.	313 nm
Formula	$C_{20}H_{12}$
M.W.	252 u
H/H+C	0.375
m.p.	°C
m.a.c. (266 nm)	(l. $mol^{-1} cm^{-1} \times 10^{-4}$)
solvent	Cyclohexane
source	Berlman

P05P02

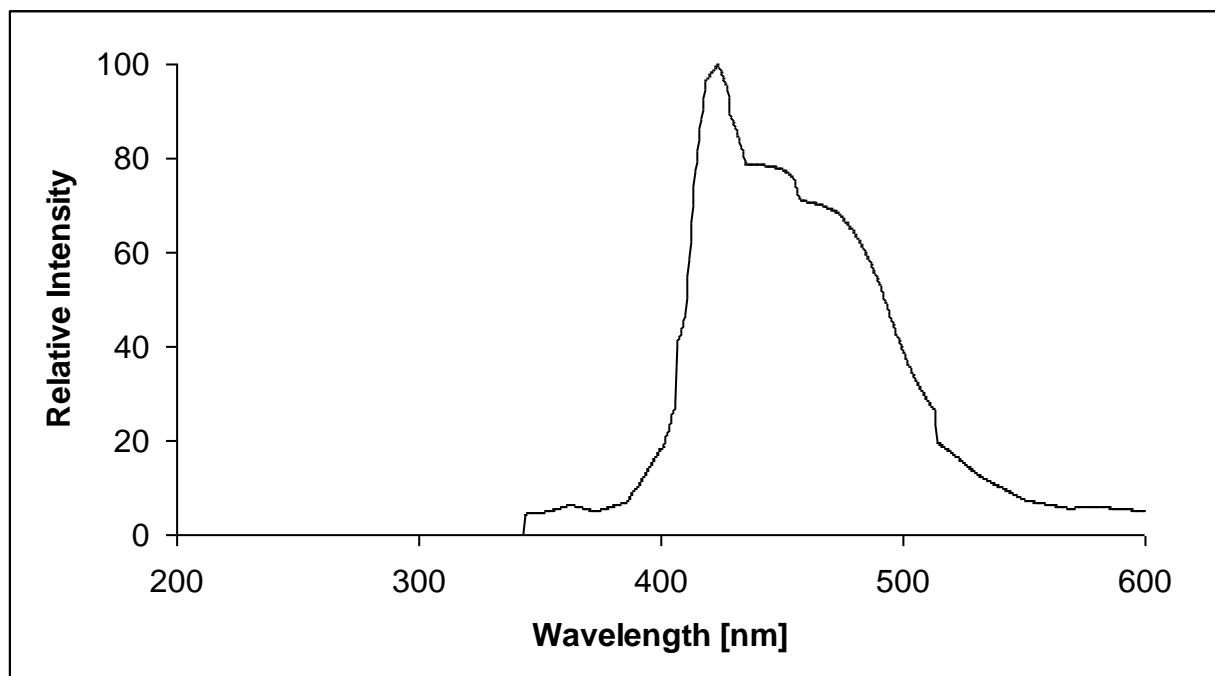
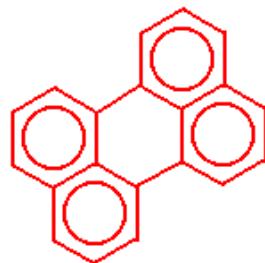
Perylene (2)



λ exc.	408 nm
Formula	$C_{20}H_{12}$
M.W.	252 u
H/H+C	0.375
m.p.	277.5°C
m.a.c. (266 nm)	$0.9 \text{ (l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4})$
solvent	Cyclohexane
source	Atlas pg. 448-II

P05P0G1

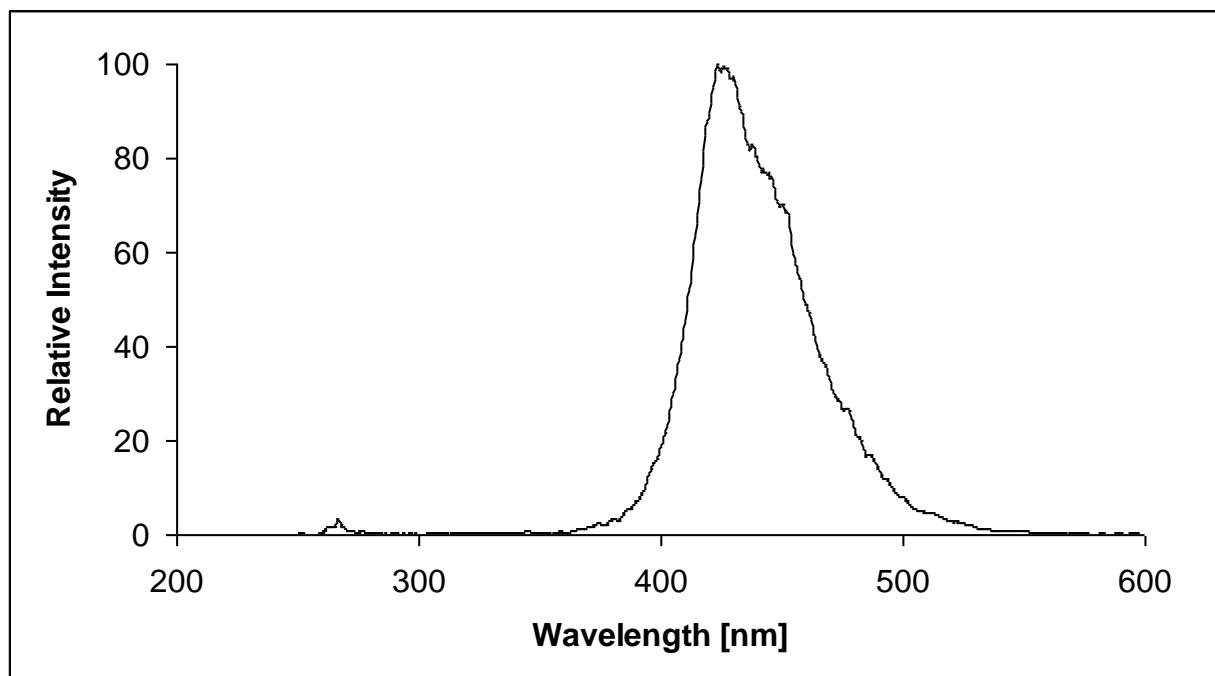
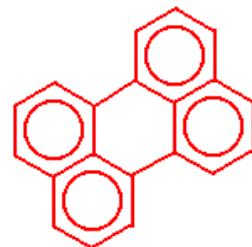
Perylene (3)



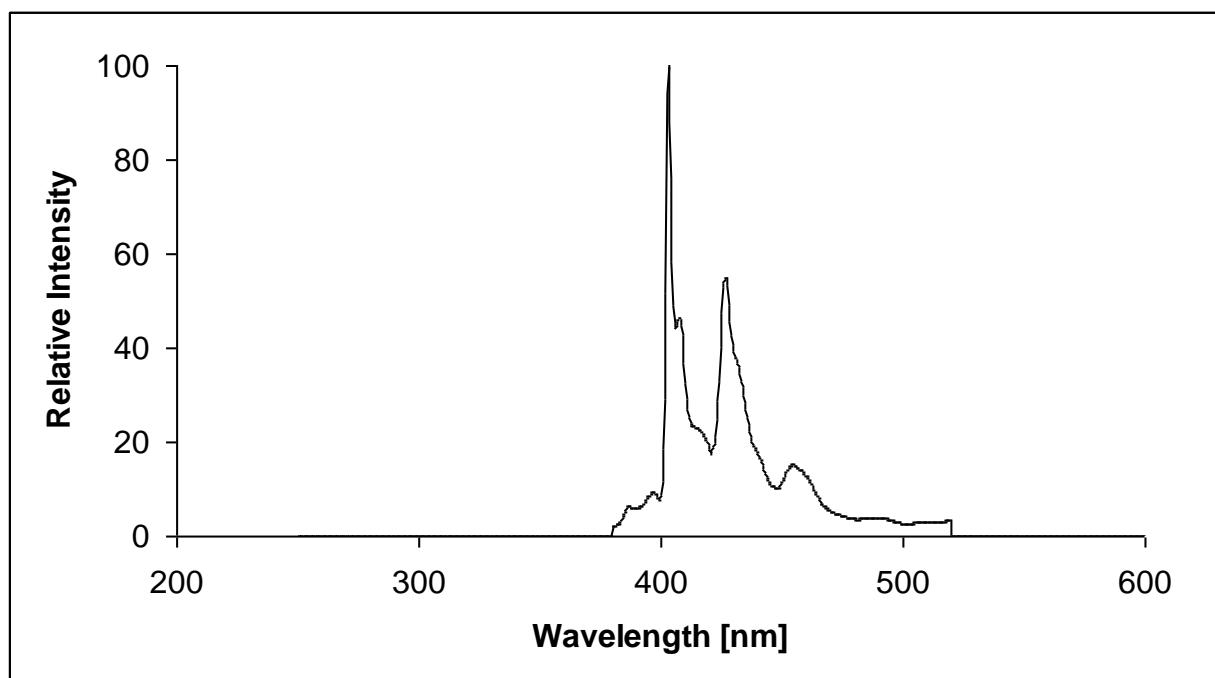
λ exc.	337 nm
Formula	$C_{20}H_{12}$
M.W.	252 u
H/H+C	0.375
m.p.	277.5°C
m.a.c. (266 nm)	($l \cdot mol^{-1} cm^{-1} \times 10^{-4}$)
solvent	Gas phase 770 K
source	[1]

P05P0G2

Perylene (4)



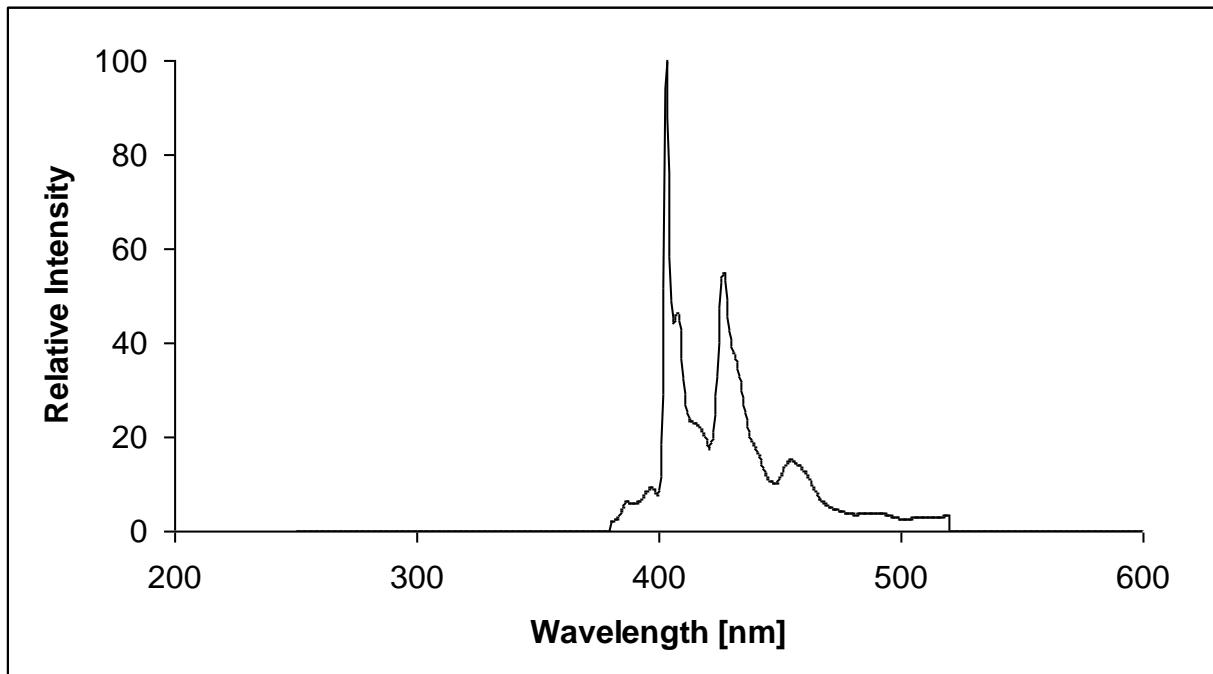
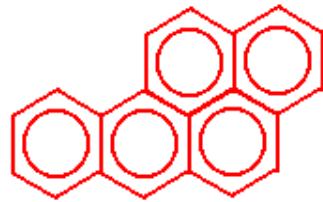
λ exc.	266 nm
Formula	C ₂₀ H ₁₂
M.W.	252 u
H/H+C	0.375
m.p.	277.5°C
m.a.c. (266 nm)	0.9 (l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	Gas phase 400 K
source	CNPM

P05R**Benzo(A)pyrene (1)**
Benzo(d, e, f)chrysene (1)

λ exc.	365 nm
Formula	C ₂₀ H ₁₂
M.W.	252 u
H/H+C	0.375
m.p.	178.1°C
m.a.c. (266 nm)	4.91 (l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	Cyclohexane
source	Atlas pg. 536-I

P05R

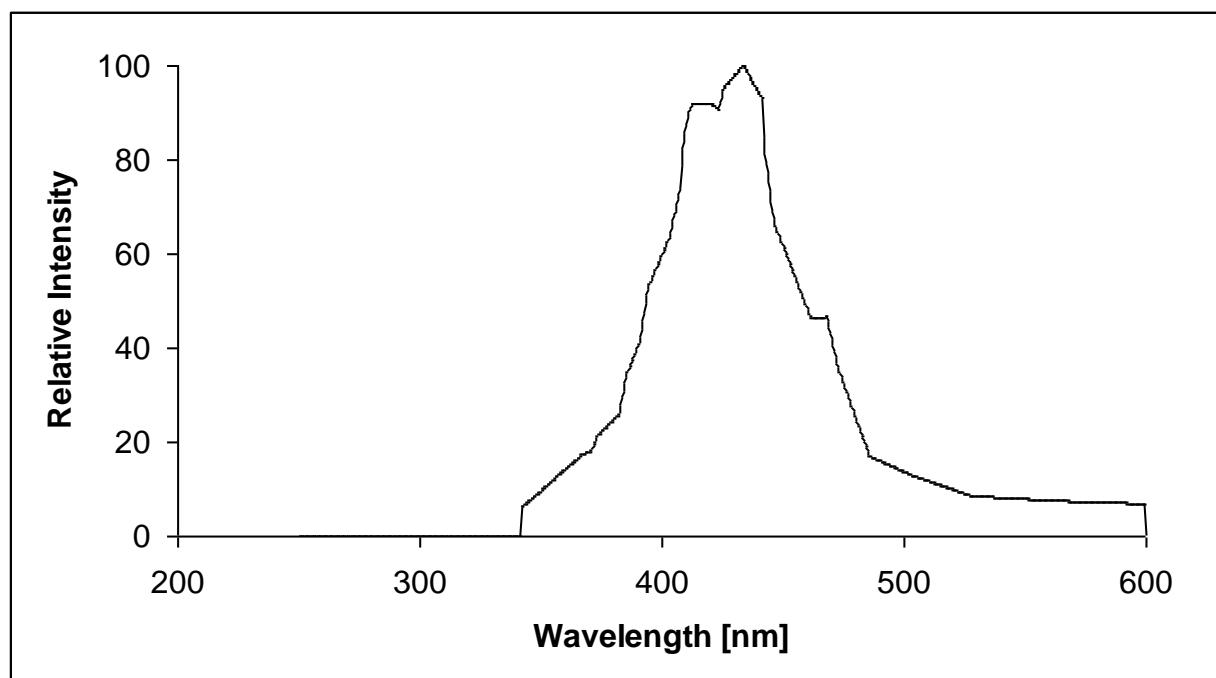
Benzo(def)chrysene (1)



λ exc.	365 nm
Formula	C ₂₀ H ₁₂
M.W.	252 u
H/H+C	0.375
m.p.	178.1°C
m.a.c. (266 nm)	4.91 (l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	Cyclohexane
source	Atlas pg. 536-I

P05RG

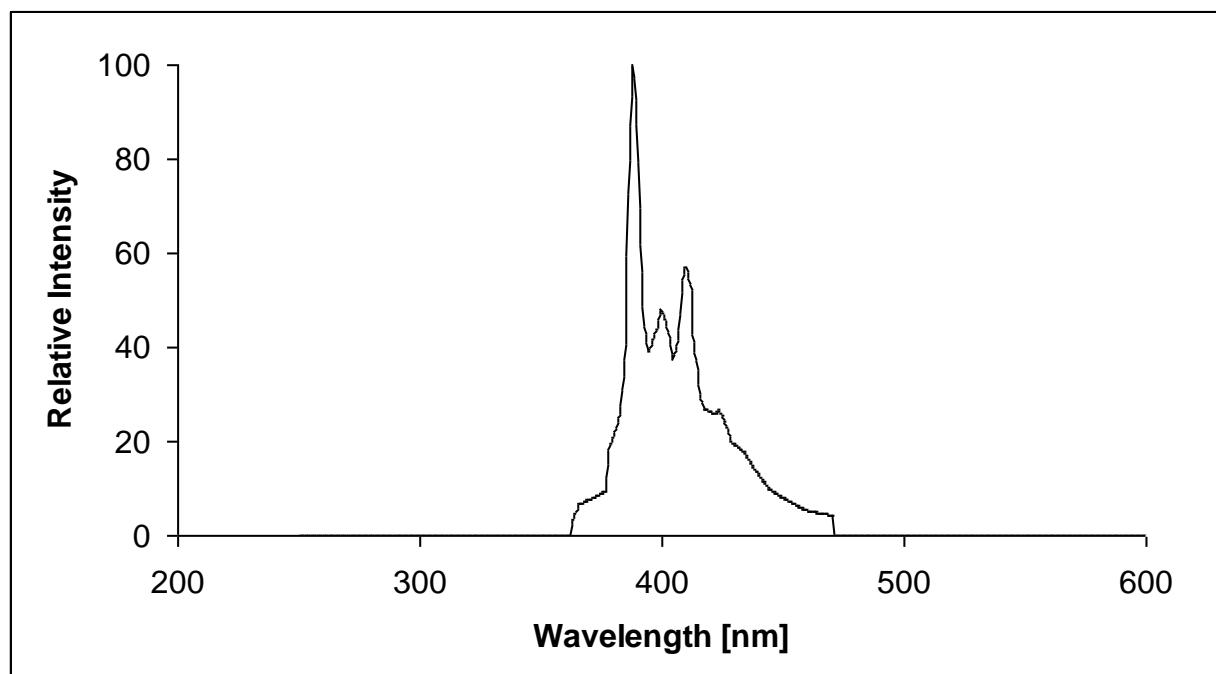
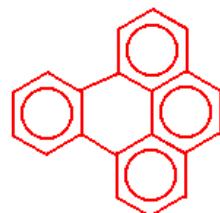
Benzo(a)pyrene (2)
Benzo(def)chrysene (2)



λ exc.	337 nm
Formula	C ₂₀ H ₁₂
M.W.	252 u
H/H+C	0.375
m.p.	178.7°C
m.a.c. (266 nm)	(l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	Gas phase 769 K
source	[1]

P05S

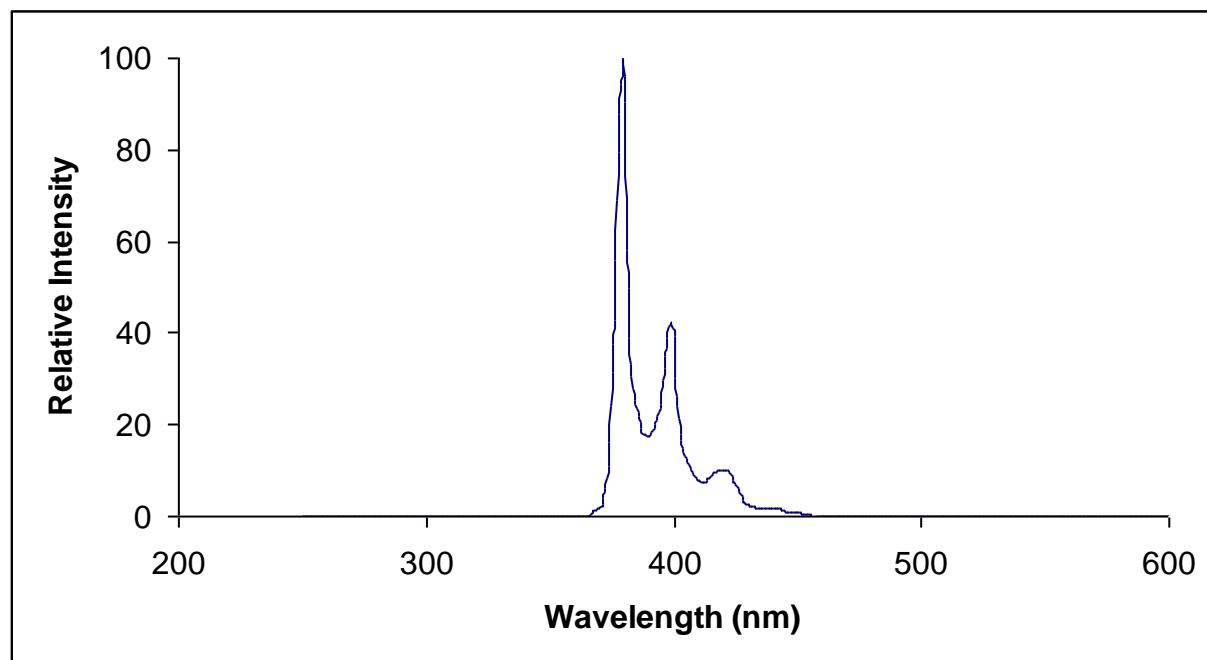
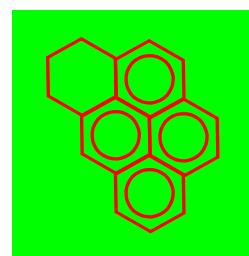
Benzo(e)pyrene



λ exc.	290 nm
Formula	C ₂₀ H ₁₂
M.W.	252 u
H/H+C	0.375
m.p.	178.7°C
m.a.c. (266 nm)	2.85 (l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	Cyclohexane
source	Atlas pg. 556-I

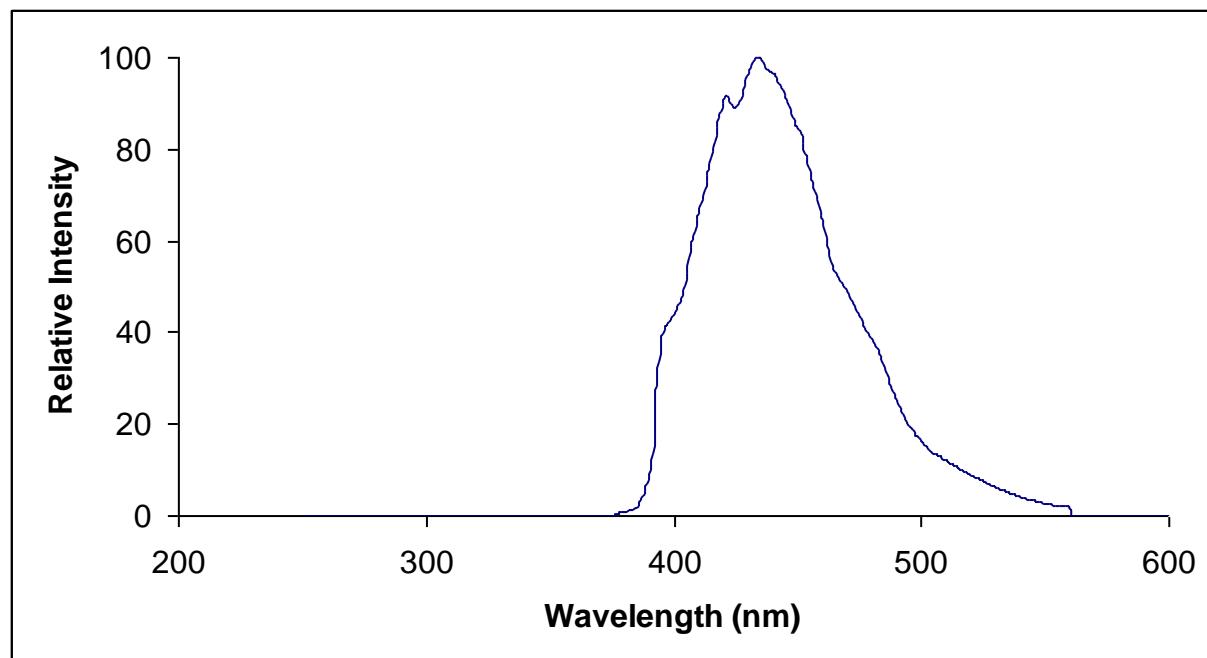
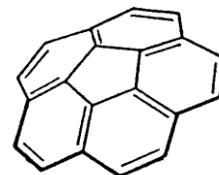
P05T

3,4,5-Trihydrobenzo[cd]pyrene



λ exc.	340 nm
Formula	$C_{19}H_{14}$
M.W.	242 u
H/H+C	0.424
m.p.	°C
m.a.c. (266 nm)	(l. $mol^{-1} cm^{-1} \times 10^{-4}$)
solvent	n-hexadecane
source	[14]

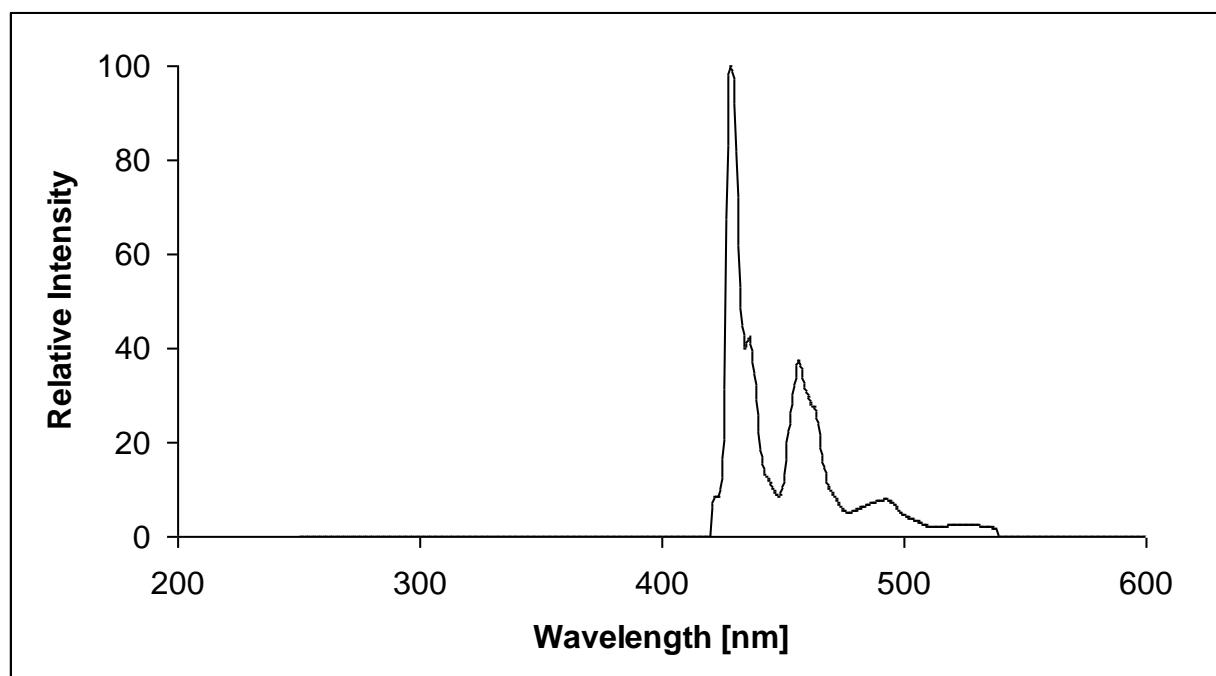
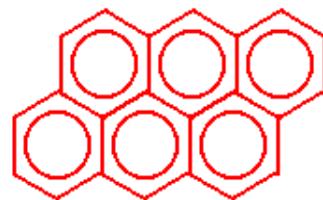
P05V

Corannulene
Dibenzo[ghi,mno]fluoranthene

λ exc.	290 nm
Formula	C ₂₀ H ₁₀
M.W.	250 u
H/H+C	0.333
m.p.	°C
m.a.c. (266 nm)	(l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	n-hexadecane
source	[13]

P06A1

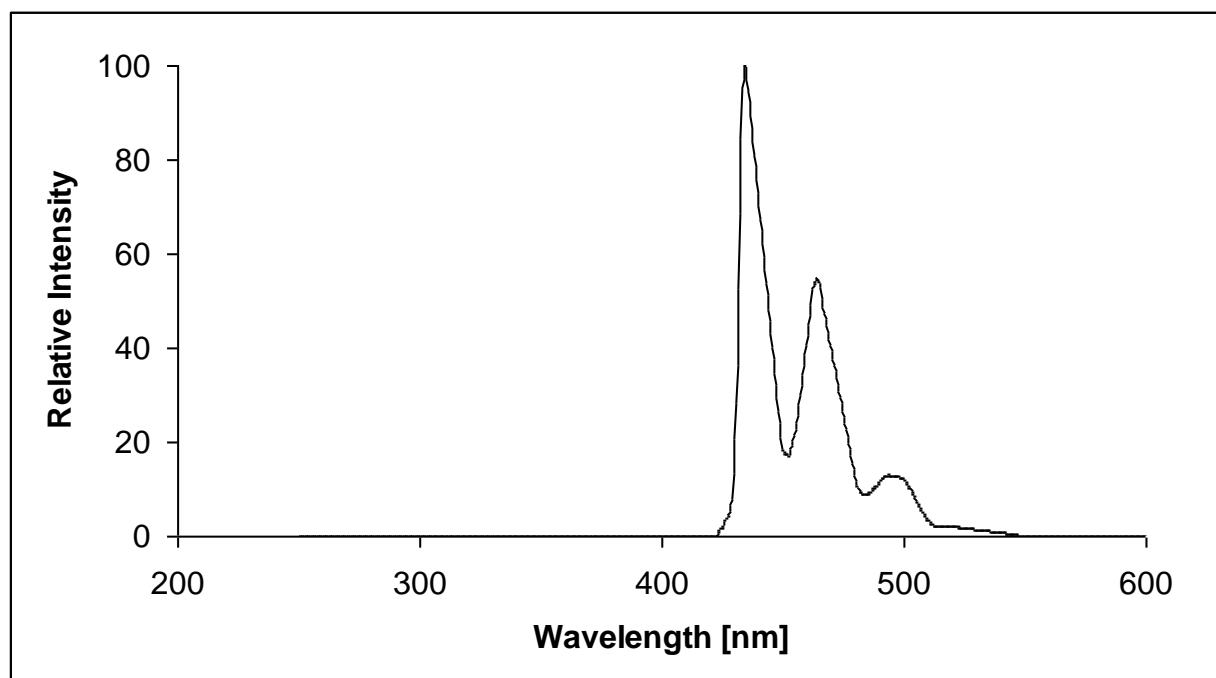
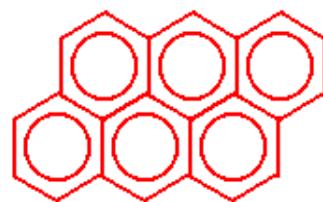
Anthanthrene (1)
Dibenzo(de,mno)chrysene



λ exc.	307 nm
Formula	$C_{22}H_{12}$
M.W.	276 u
H/H+C	0.353
m.p.	264°C
m.a.c. (266 nm)	0.31 ($l \cdot mol^{-1} cm^{-1} \times 10^{-4}$)
solvent	Cyclohexane
source	Atlas pg. 612-I

P06A2

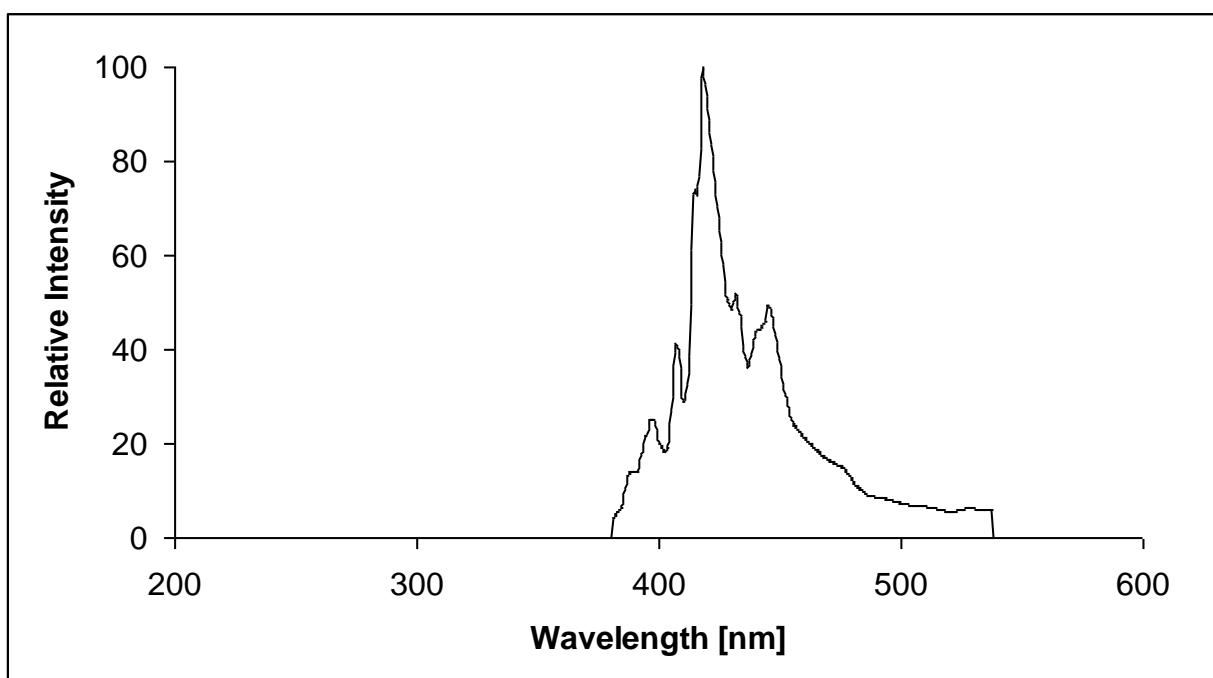
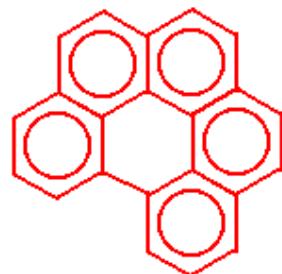
Anthanthrene (2)



λ exc.	435 nm
Formula	$C_{22}H_{12}$
M.W.	276 u
H/H+C	0.353
m.p.	°C
m.a.c. (266 nm)	(l. $mol^{-1} cm^{-1} \times 10^{-4}$)
solvent	Benzene
source	Berlman

P06B1

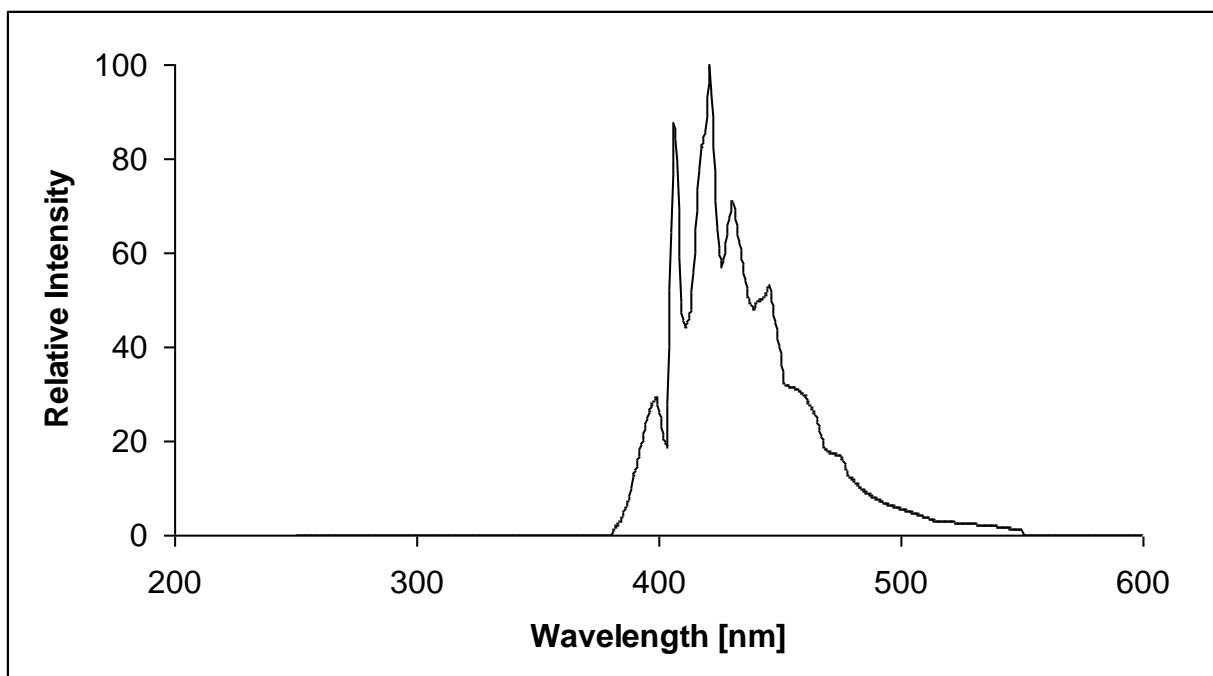
Benzo(ghi)perylene (1)



λ exc.	300 nm
Formula	$C_{22}H_{12}$
M.W.	276 u
H/H+C	0.353
m.p.	278.3°C
m.a.c. (266 nm)	$2.15 \text{ (l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4})$
solvent	Cyclohexane
source	Atlas pg. 630-I

P06B2

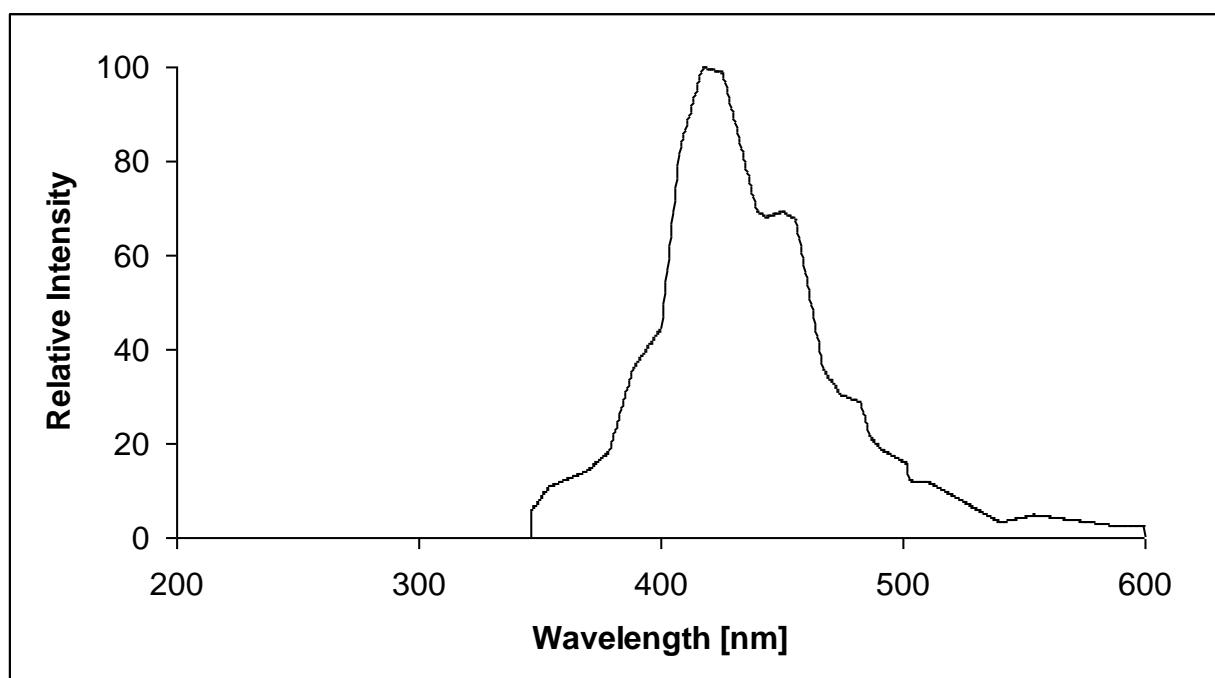
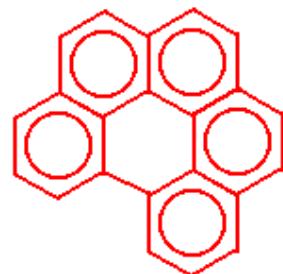
Benzo(ghi)perylene (2)



λ exc.	365 nm
Formula	C ₂₂ H ₁₂
M.W.	276 u
H/H+C	0.353
m.p.	°C
m.a.c. (266 nm)	(l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	Benzene
source	Berlman

P06BG

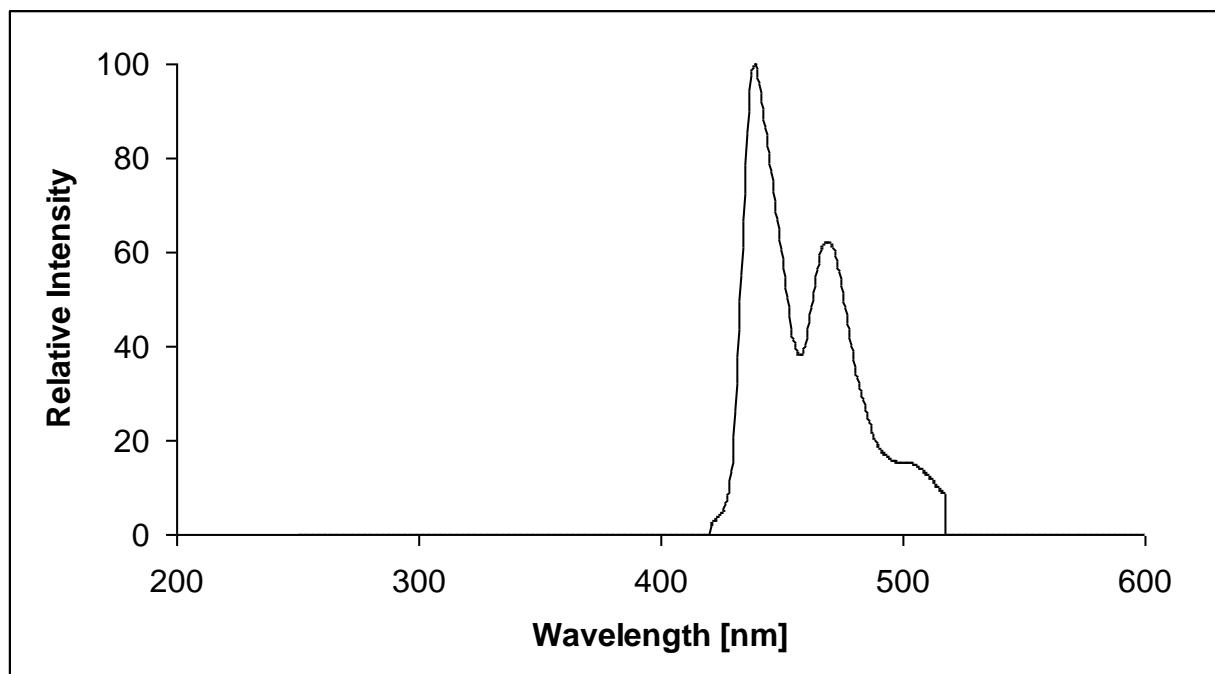
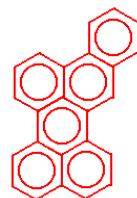
Benzo(ghi)perylene (3)



λ exc.	337 nm
Formula	$C_{22}H_{12}$
M.W.	276 u
H/H+C	0.353
m.p.	278.3°C
m.a.c. (266 nm)	($I \cdot mol^{-1} cm^{-1} \times 10^{-4}$)
solvent	Gas phase 815 K
source	[1]

P06C

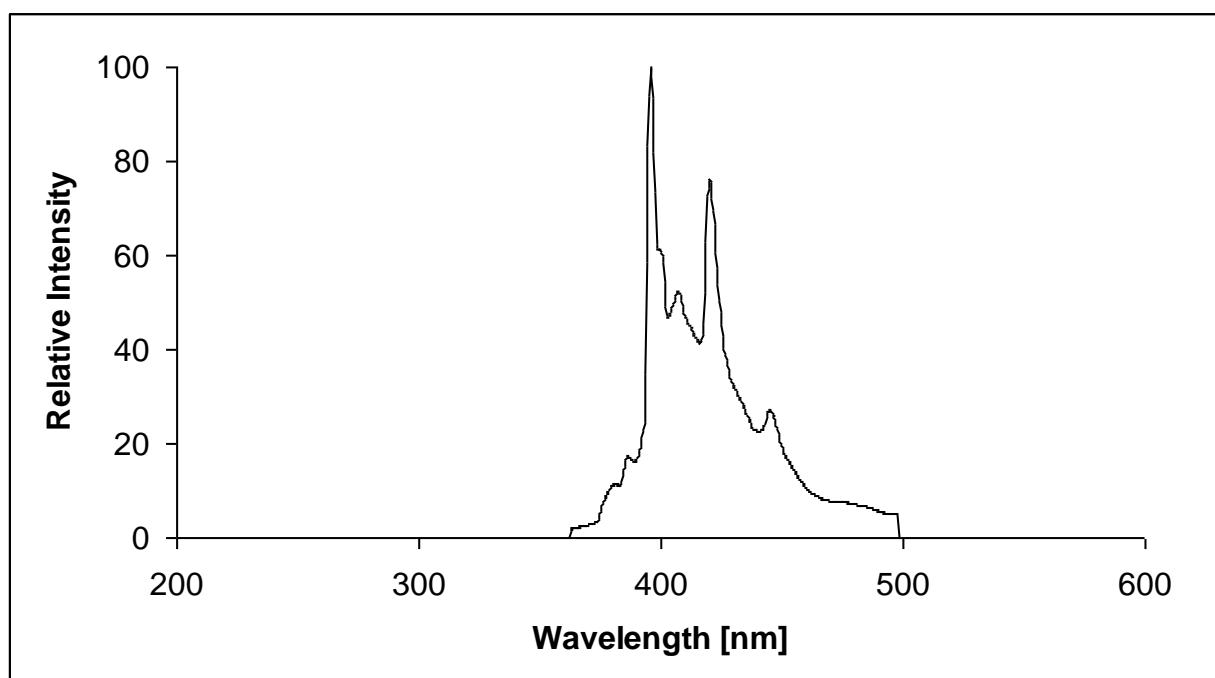
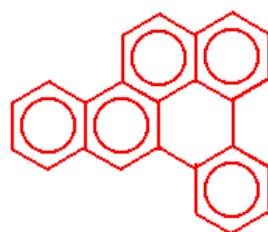
Benzo(b)perylene



λ exc.	312 nm
Formula	$C_{24}H_{14}$
M.W.	302 u
H/H+C	0.368
m.p.	°C
m.a.c. (266 nm)	(l. $mol^{-1} cm^{-1} \times 10^{-4}$)
solvent	n-hexadecane
source	[4]

P06D

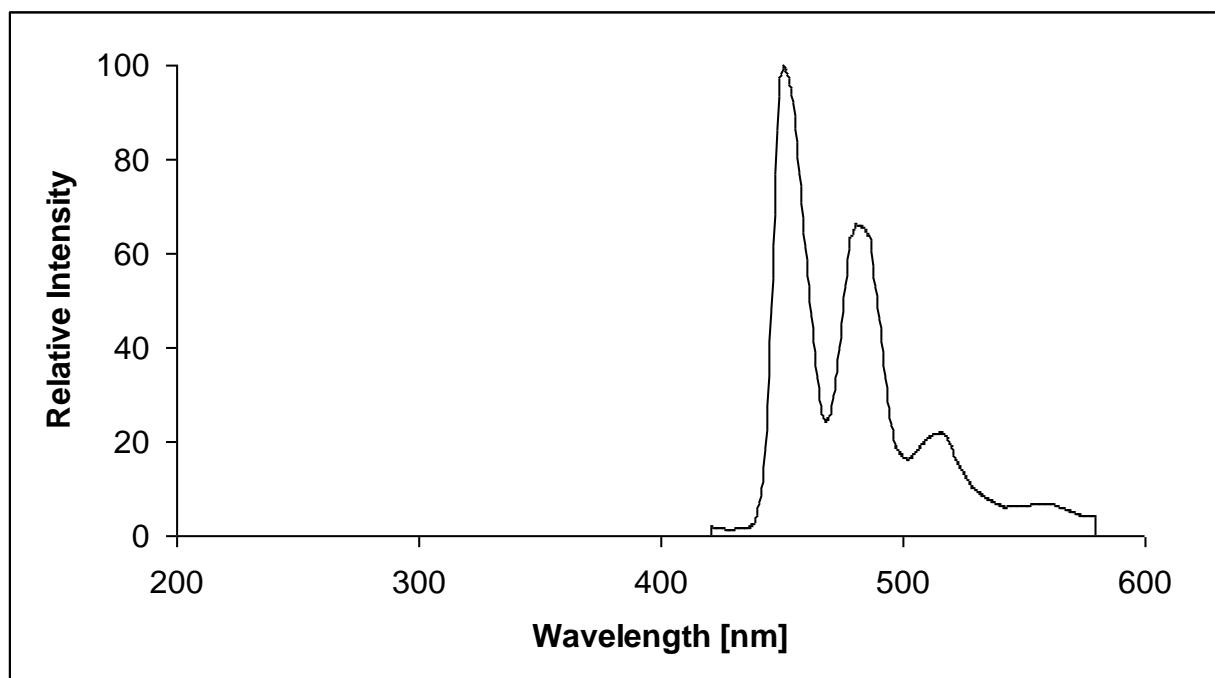
Naphtho(1,2,3,4-def)chrysene
Bibenzo(a,e)pyrene



λ exc.	304.5 nm
Formula	C ₂₄ H ₁₄
M.W.	302 u
H/H+C	0.368
m.p.	244.4°C
m.a.c. (266 nm)	3.6 (l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	Cyclohexane
source	Atlas pg. 764-I

P06E

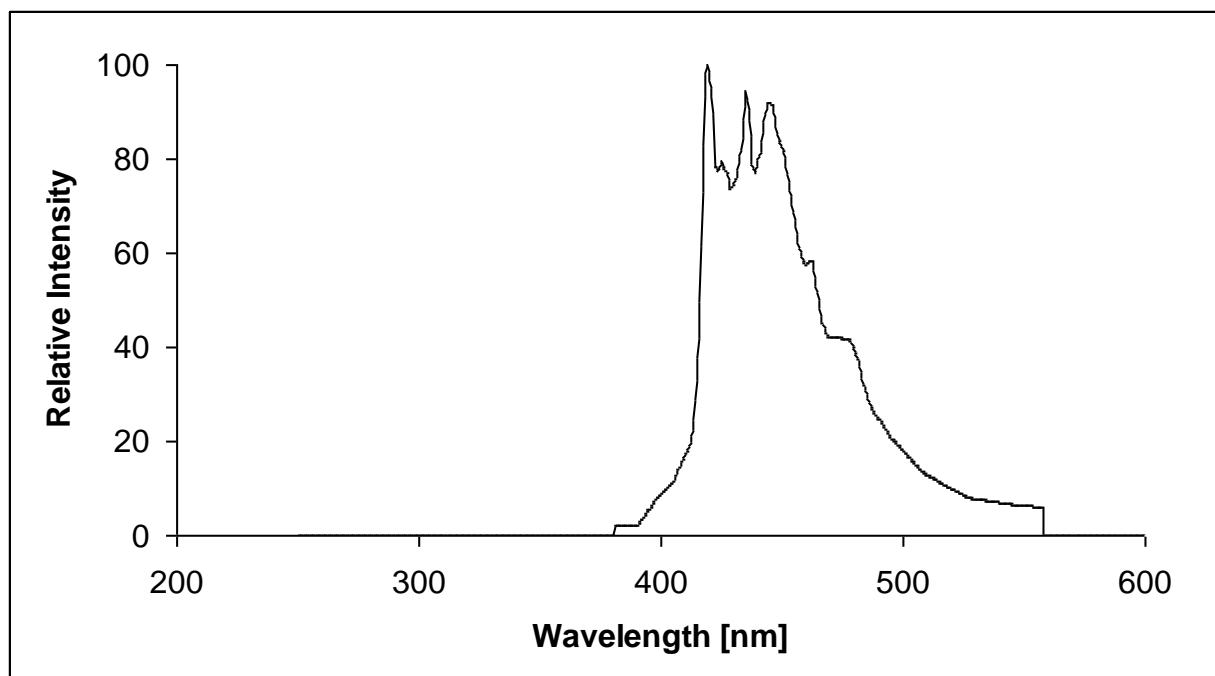
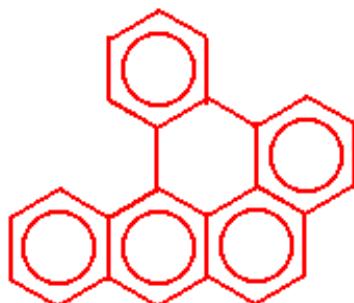
Dibenzo(b,def)chrysene
Dibenzo(a,h)pyrene



λ exc.	311 nm
Formula	$C_{24}H_{14}$
M.W.	302 u
H/H+C	0.368
m.p.	317°C
m.a.c. (266 nm)	$2.45 \text{ (l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4})$
solvent	Cyclohexane
source	Atlas pg. 776-II

P06F1

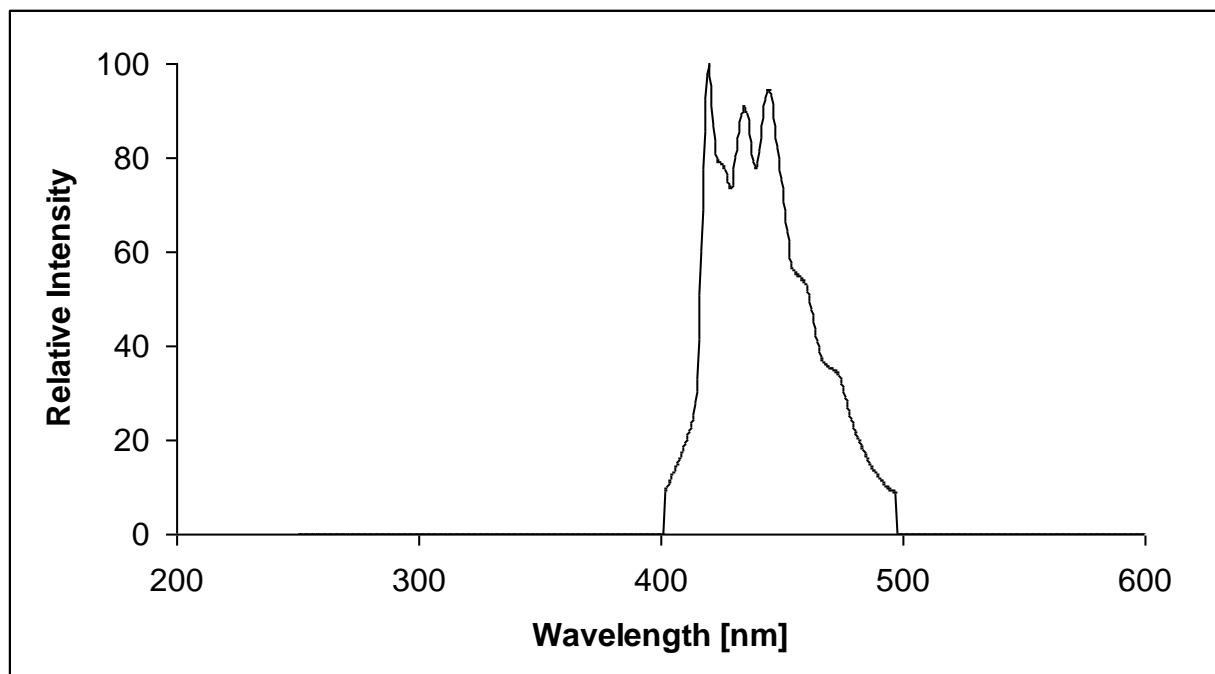
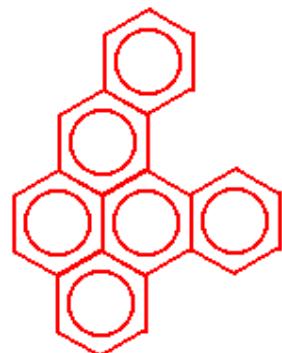
Dibenzo(def,p)chrysene(1)
Dibenzo(a,l)pyrene



λ exc.	317 nm
Formula	$C_{24}H_{14}$
M.W.	302 u
H/H+C	0.368
m.p.	162.4°C
m.a.c. (266 nm)	$3.1 \text{ (l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4})$
solvent	Cyclohexane
source	Atlas pg. 780-I

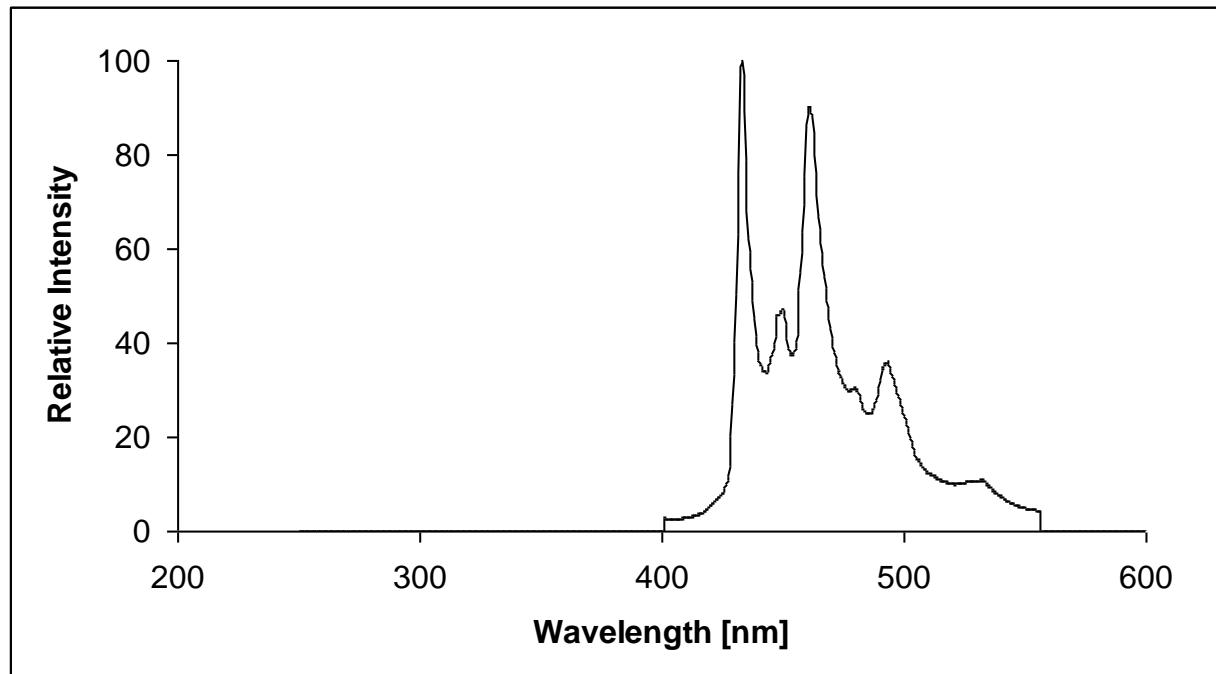
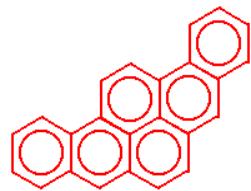
P06F2

Dibenzo(def,p)chrysene(2)



λ exc.	310 nm
Formula	C ₂₄ H ₁₄
M.W.	302 u
H/H+C	0.368
m.p.	°C
m.a.c. (266 nm)	(l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	n-hexadecane
source	[8]

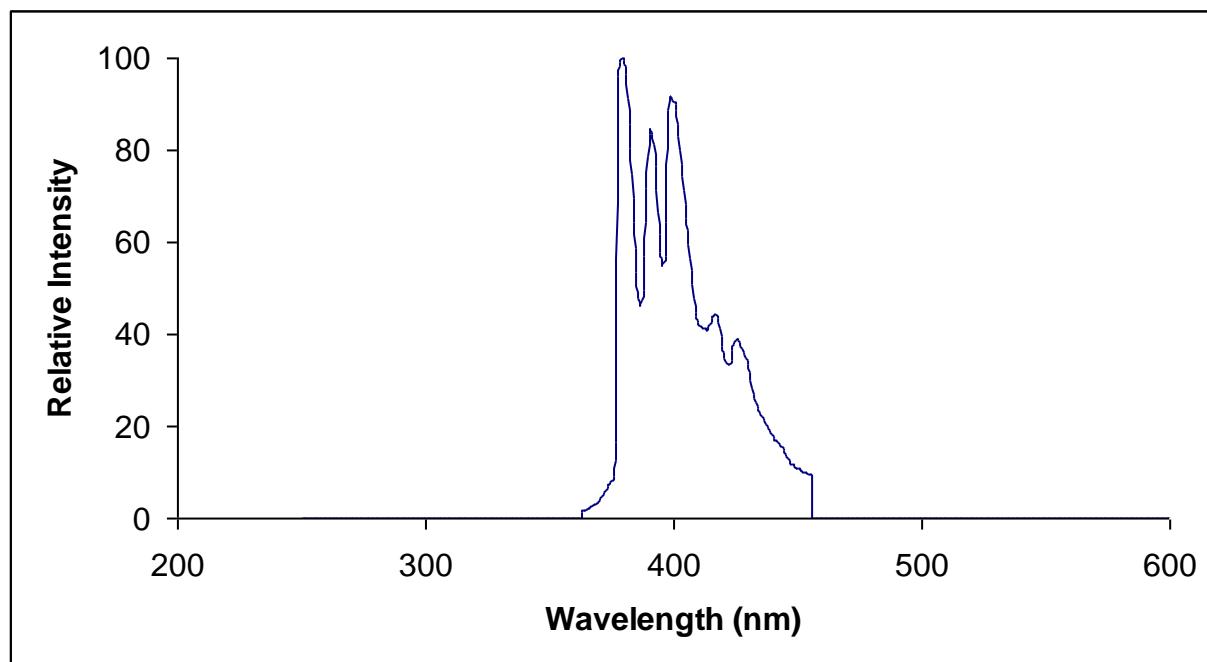
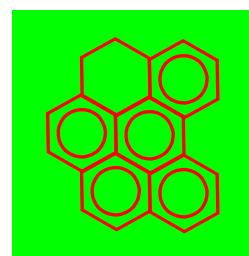
P06G

Benzo(r,s,t)pentaphene
Dibenzo(a,i)pyrene

λ exc.	393 nm
Formula	C ₂₄ H ₁₄
M.W.	302 u
H/H+C	0.368
m.p.	282°C
m.a.c. (266 nm)	2 (l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	Cyclohexane
source	Atlas pg. 796-II

P06L

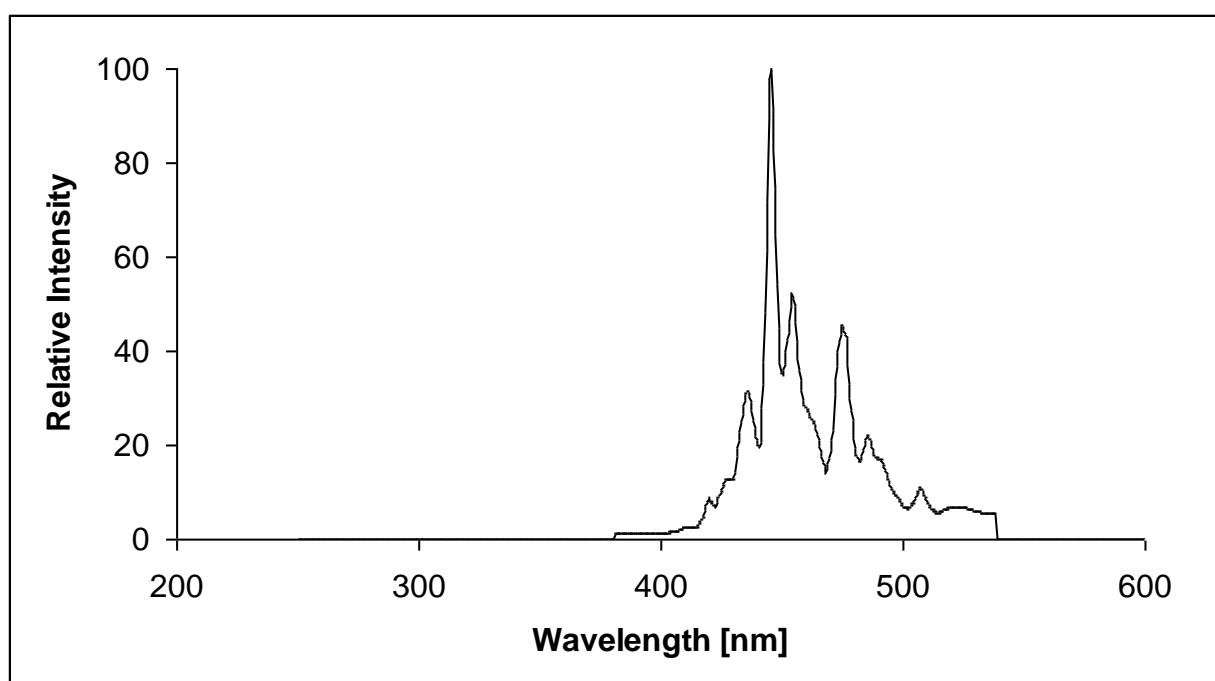
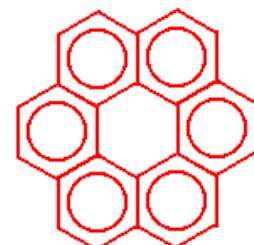
Dihydrobenzo[ghi]perylene



λ exc.	334 nm
Formula	$C_{22}H_{14}$
M.W.	278 u
H/H+C	0.389
m.p.	°C
m.a.c. (266 nm)	(l. $mol^{-1} cm^{-1} \times 10^{-4}$)
solvent	n-hexadecane
source	[14]

P07C

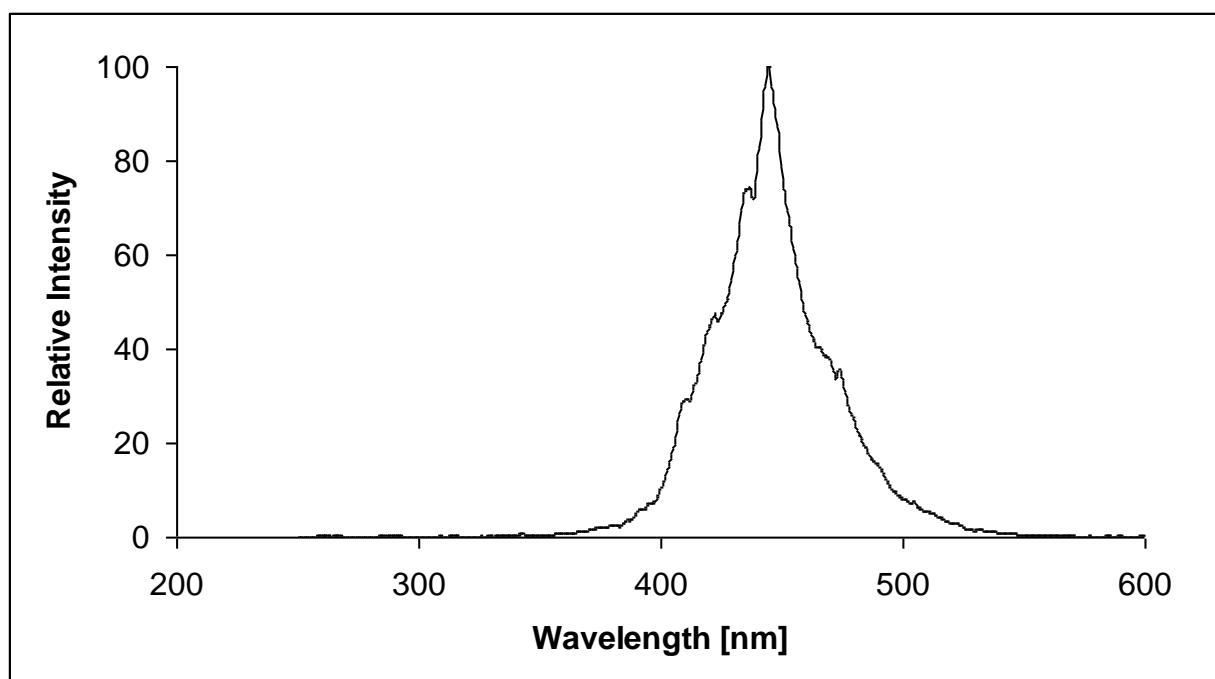
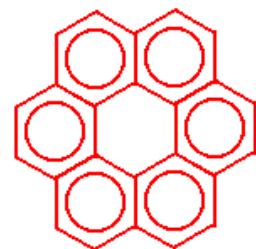
Coronene (1)



λ exc.	302 nm
Formula	$C_{24}H_{12}$
M.W.	300 u
H/H+C	0.333
m.p.	439°C
m.a.c. (266 nm)	0.45 ($l \cdot mol^{-1} cm^{-1} \times 10^{-4}$)
solvent	Cyclohexane
source	Atlas pg. 720-II

P07CG

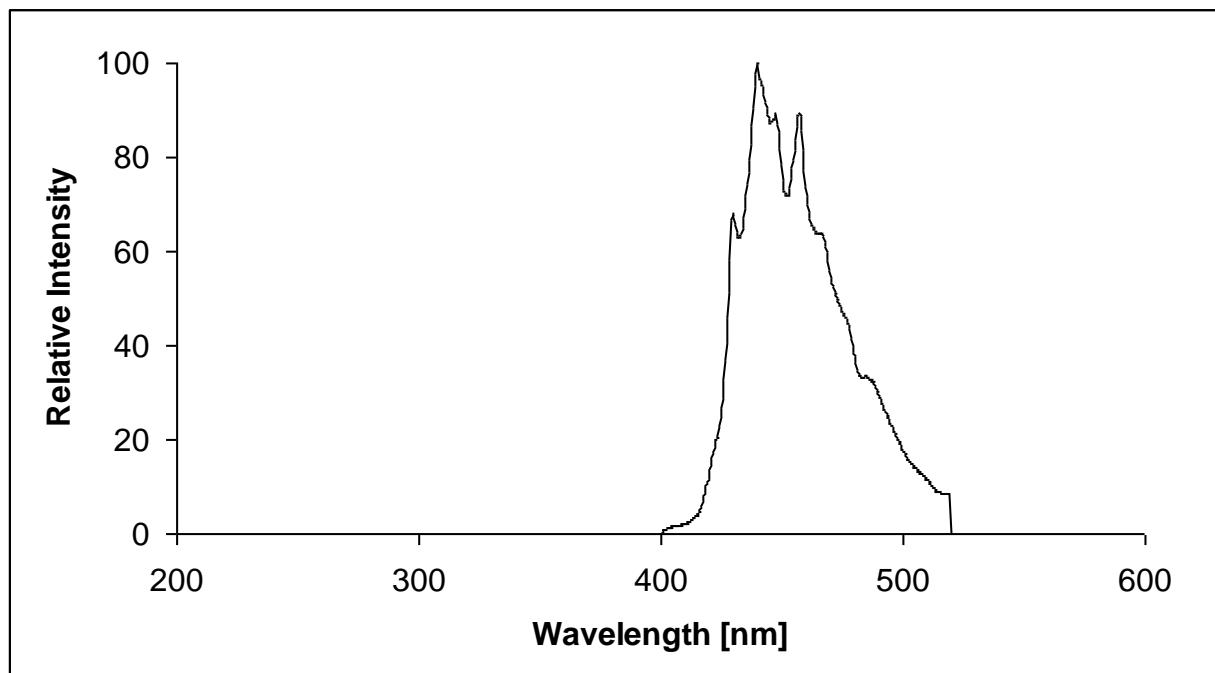
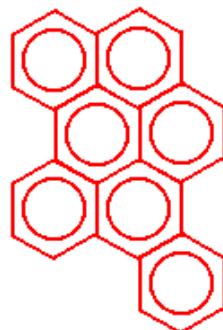
Coronene (2)



λ exc.	266 nm
Formula	$C_{24}H_{12}$
M.W.	300 u
H/H+C	0.333
m.p.	439°C
m.a.c. (266 nm)	$0.45 \text{ (l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4})$
solvent	Gas phase 700 K
source	CNPM

P07D

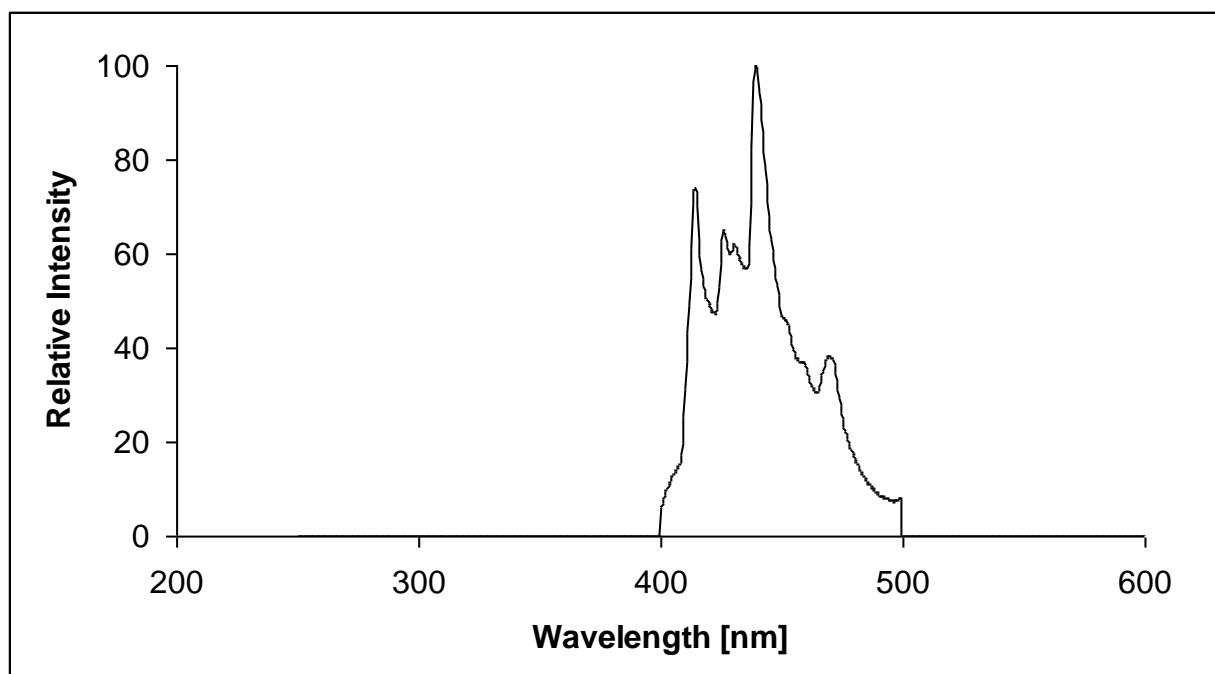
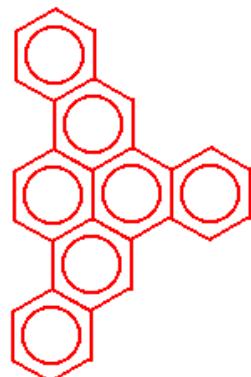
Dibenzo(e,ghi)perylene



λ exc.	330 nm
Formula	$C_{26}H_{14}$
M.W.	326 u
H/H+C	0.350
m.p.	°C
m.a.c. (266 nm)	(l. $mol^{-1} cm^{-1} \times 10^{-4}$)
solvent	n-hexadecane
source	[4]

P07E

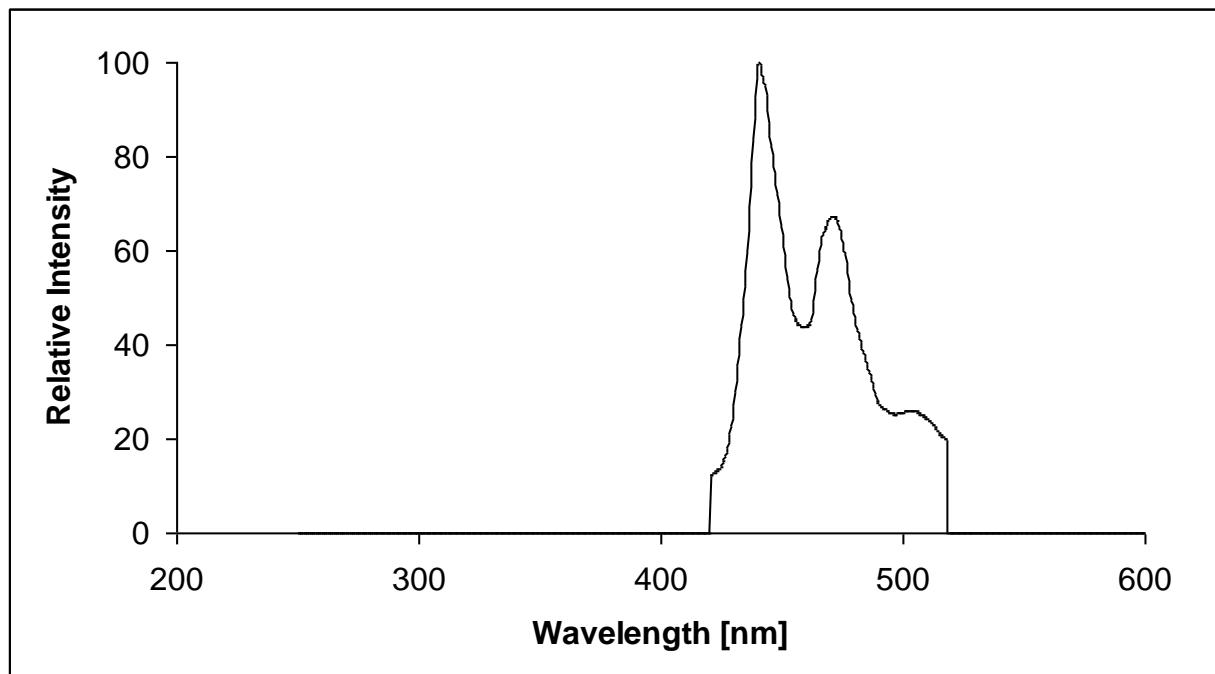
Dibenzo(h,rst)pentathene



λ exc.	366 nm
Formula	C ₂₈ H ₁₆
M.W.	352 u
H/H+C	0.346
m.p.	°C
m.a.c. (266 nm)	(l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	n-hexadecane
source	[5]

P07F

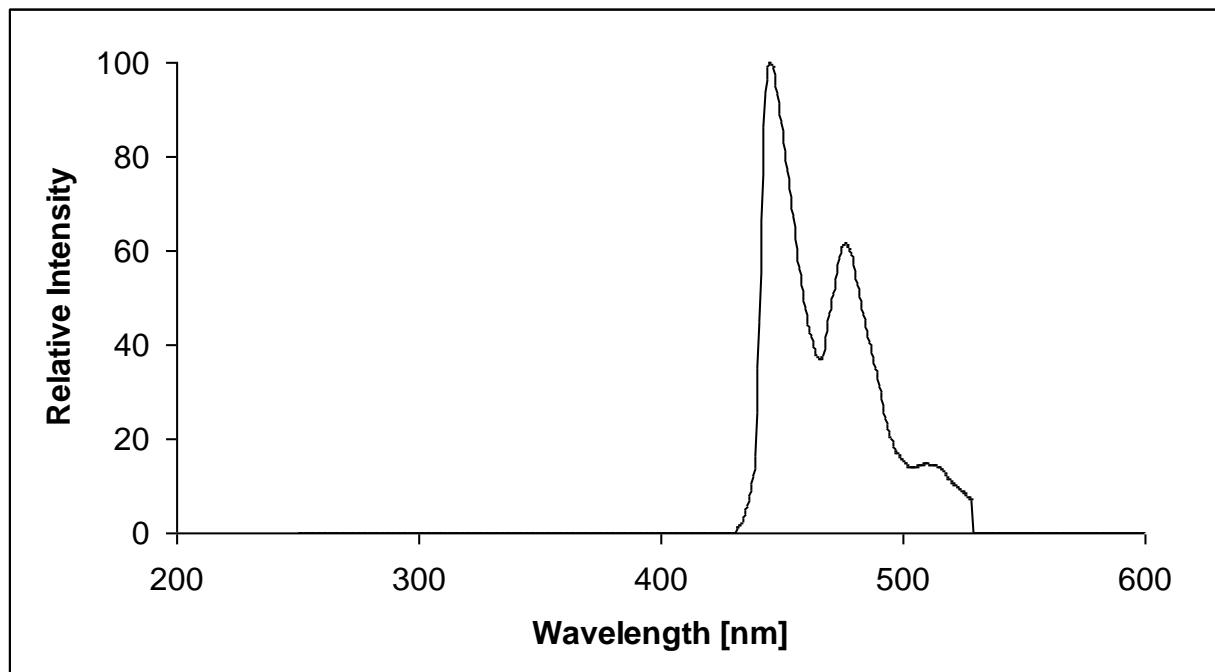
Dibenzo(fg,ij)pentathene



λ exc.	300 nm
Formula	C ₂₈ H ₁₆
M.W.	352 u
H/H+C	0.346
m.p.	°C
m.a.c. (266 nm)	(l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	n-hexadecane
source	[5]

P07G

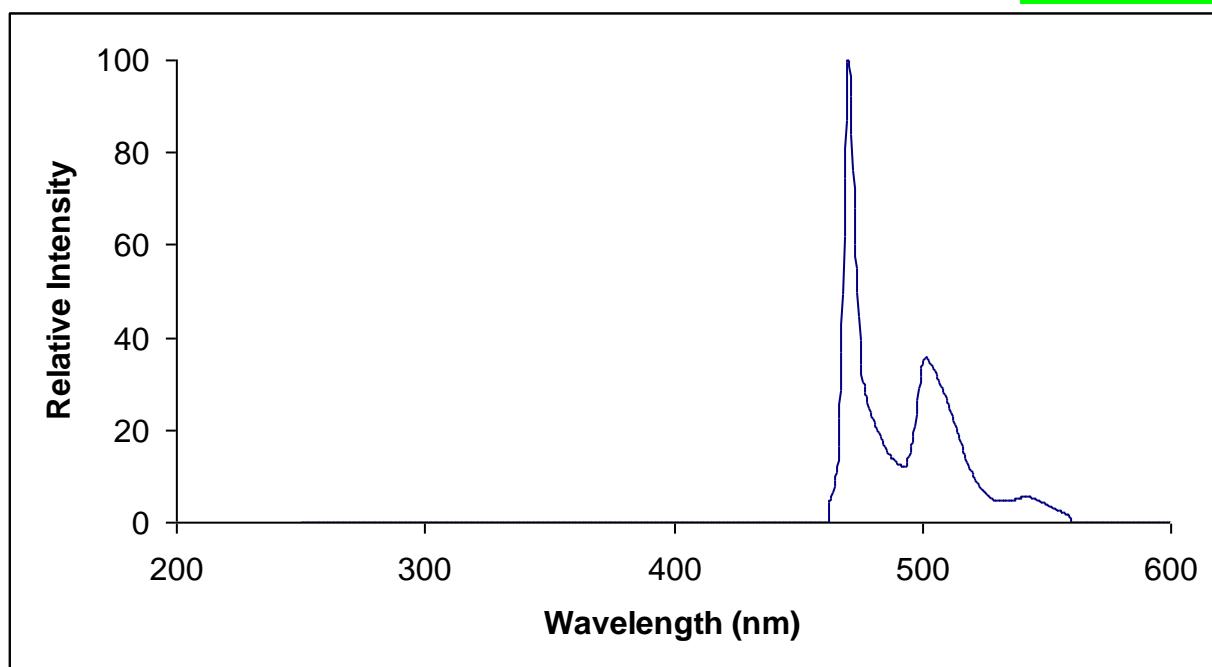
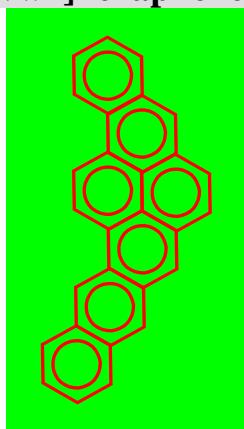
Benzo(a)naphtho(8,1,2-cde)naphthacene



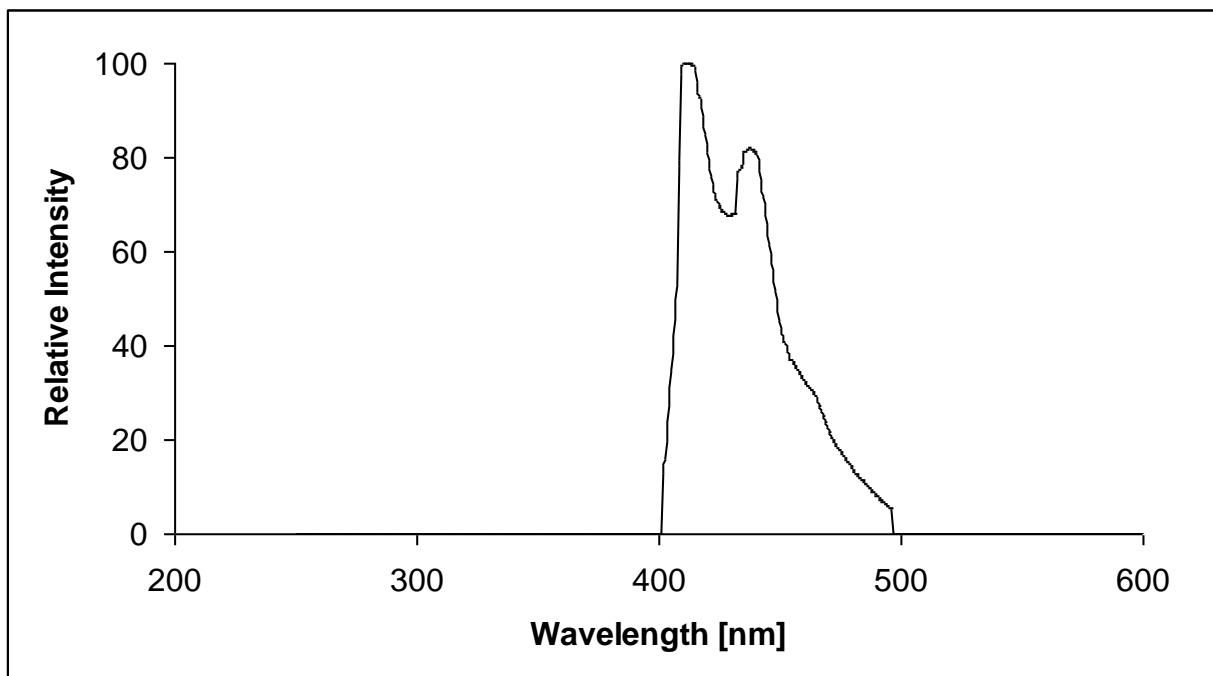
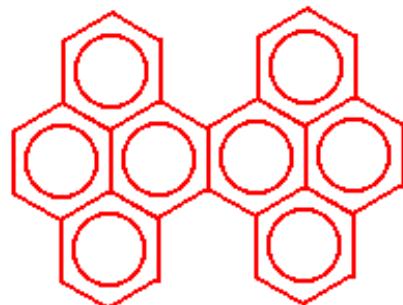
λ exc.	411 nm
Formula	$C_{28}H_{16}$
M.W.	352 u
H/H+C	0.364
m.p.	°C
m.a.c. (266 nm)	(l. $mol^{-1} cm^{-1} \times 10^{-4}$)
solvent	n-hexadecane
source	[4]

P07H

Benzo[vwx]hexaphene



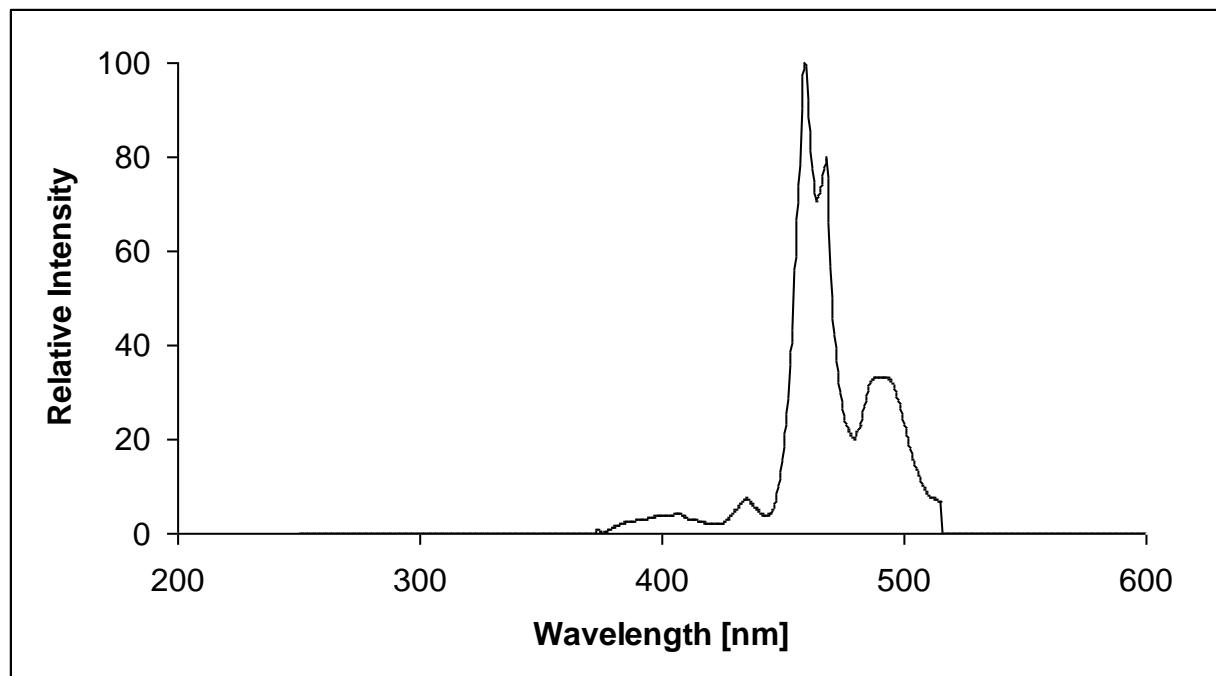
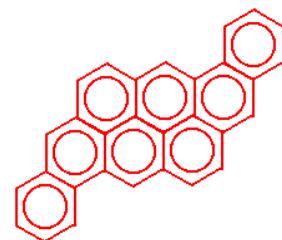
λ exc.	360 nm
Formula	C ₂₈ H ₁₆
M.W.	352 u
H/H+C	0.363
m.p.	°C
m.a.c. (266 nm)	(l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	n-hexadecane
source	[16]

P08N**Tetrabenzo(de,hi,mn,gr)naphthacene**

λ exc.	318 nm
Formula	$C_{30}H_{16}$
M.W.	376 u
H/H+C	0.348
m.p.	°C
m.a.c. (266 nm)	(l. $mol^{-1} cm^{-1} \times 10^{-4}$)
solvent	n-hexadecane
source	[3]

P08P

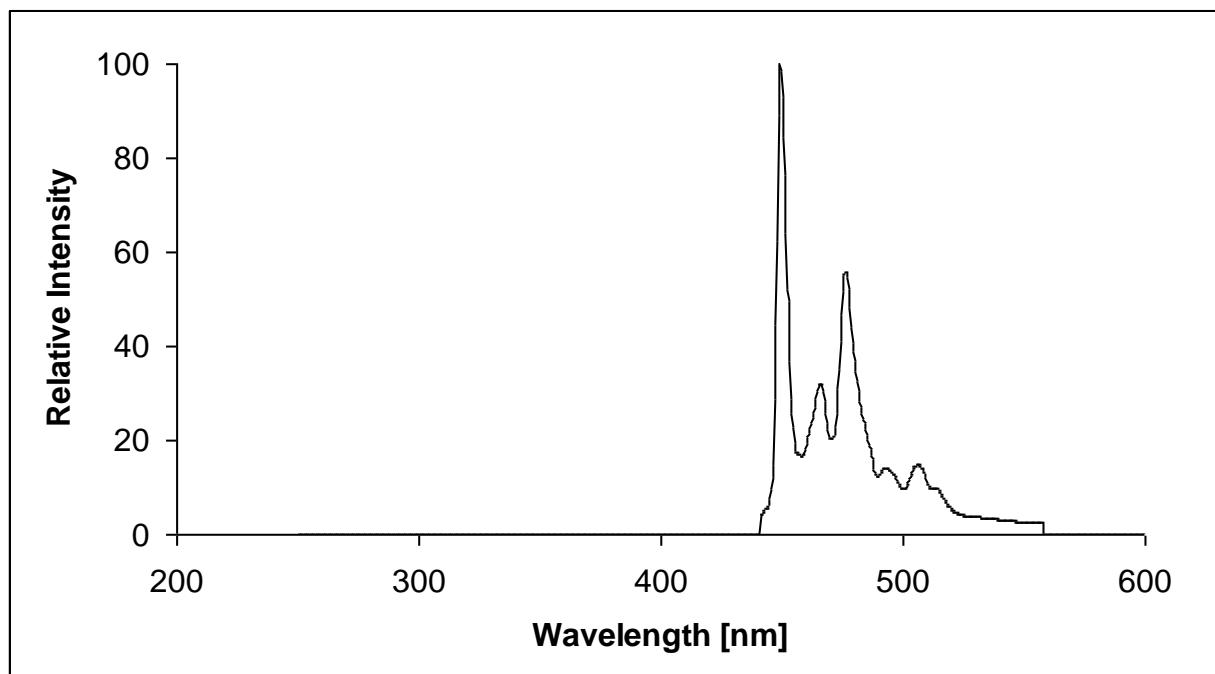
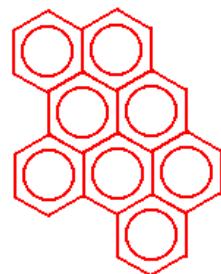
Pyranthrene



λ exc.	
Formula	$C_{30}H_{16}$
M.W.	376 u
H/H+C	0.348
m.p.	°C
m.a.c. (266 nm)	(l. $mol^{-1} cm^{-1} \times 10^{-4}$)
solvent	n-hexadecane
source	[5]

P08R

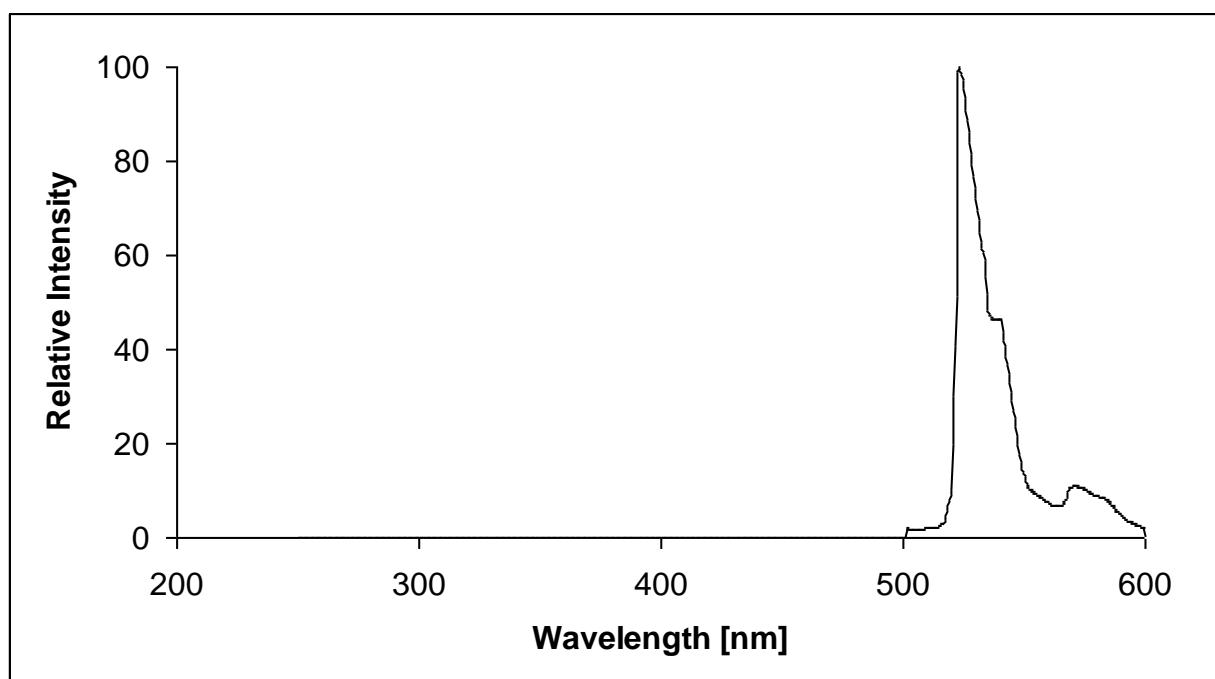
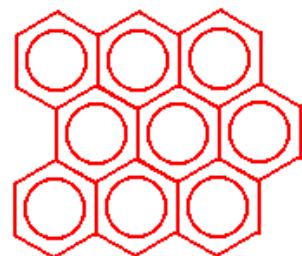
Phenanthro(5,4,3,2-efghi)perylene



λ exc.	365 nm
Formula	$C_{28}H_{14}$
M.W.	350 u
H/H+C	0.333
m.p.	°C
m.a.c. (266 nm)	(l. $mol^{-1} cm^{-1} \times 10^{-4}$)
solvent	n-hexadecane
source	[4]

P09C

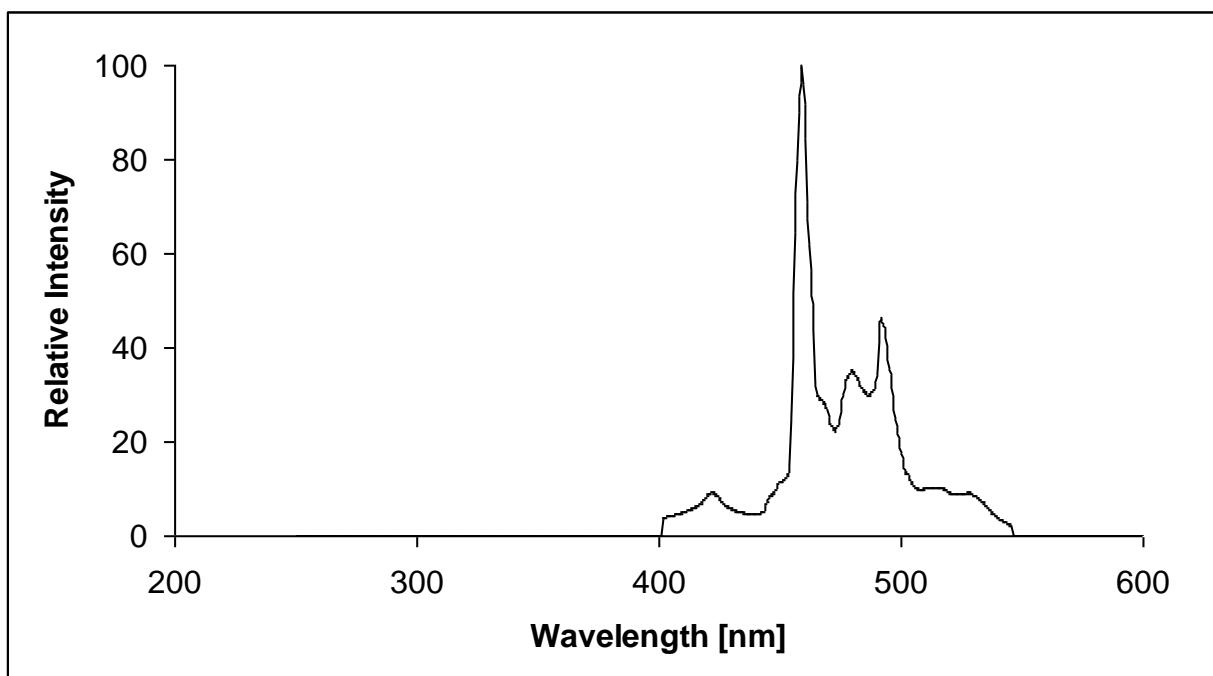
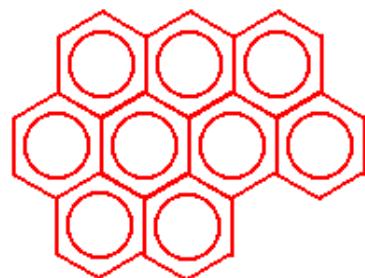
Dibenzo(bc,ef)coronene



λ exc.	336 nm
Formula	$C_{30}H_{14}$
M.W.	374 u
H/H+C	0.318
m.p.	°C
m.a.c. (266 nm)	(l. $mol^{-1} cm^{-1} \times 10^{-4}$)
solvent	n-hexadecane
source	[3]

P09D

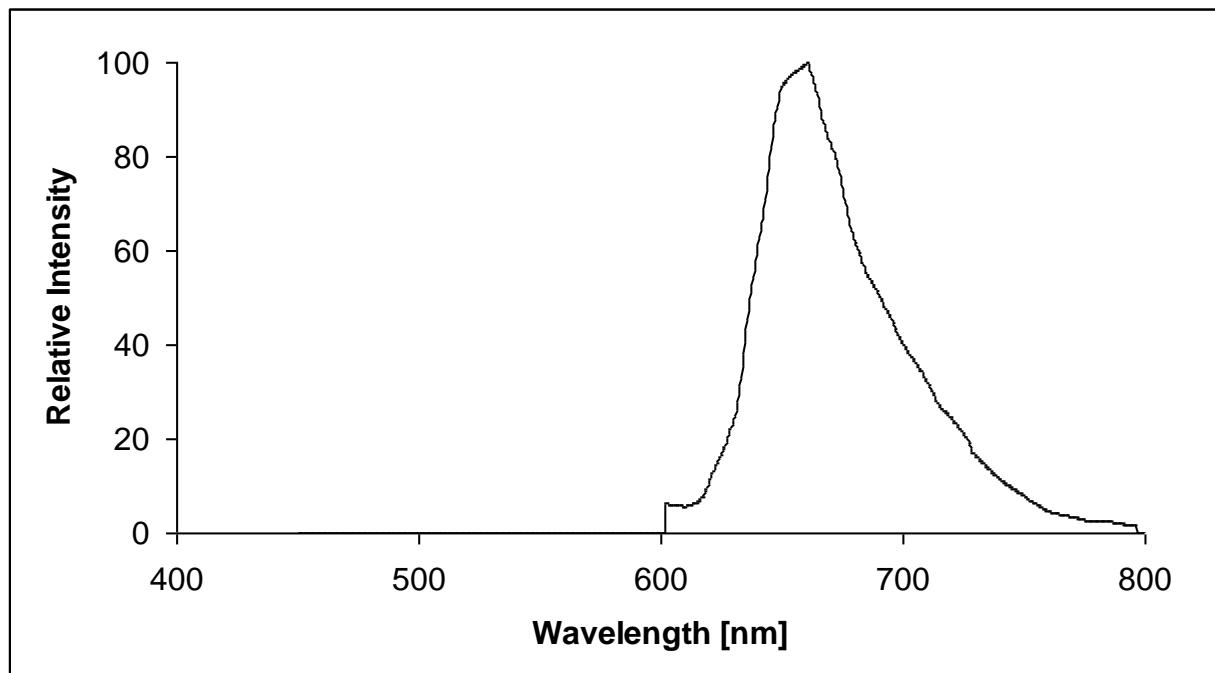
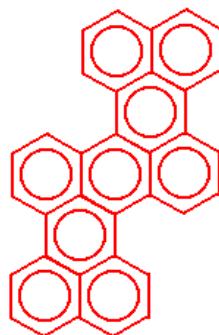
Naphtho(8,1,2-abc)coronene



λ exc.	322 nm
Formula	C ₃₀ H ₁₄
M.W.	374 u
H/H+C	0.318
m.p.	°C
m.a.c. (266 nm)	(l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	n-hexadecane
source	[3]

P09E

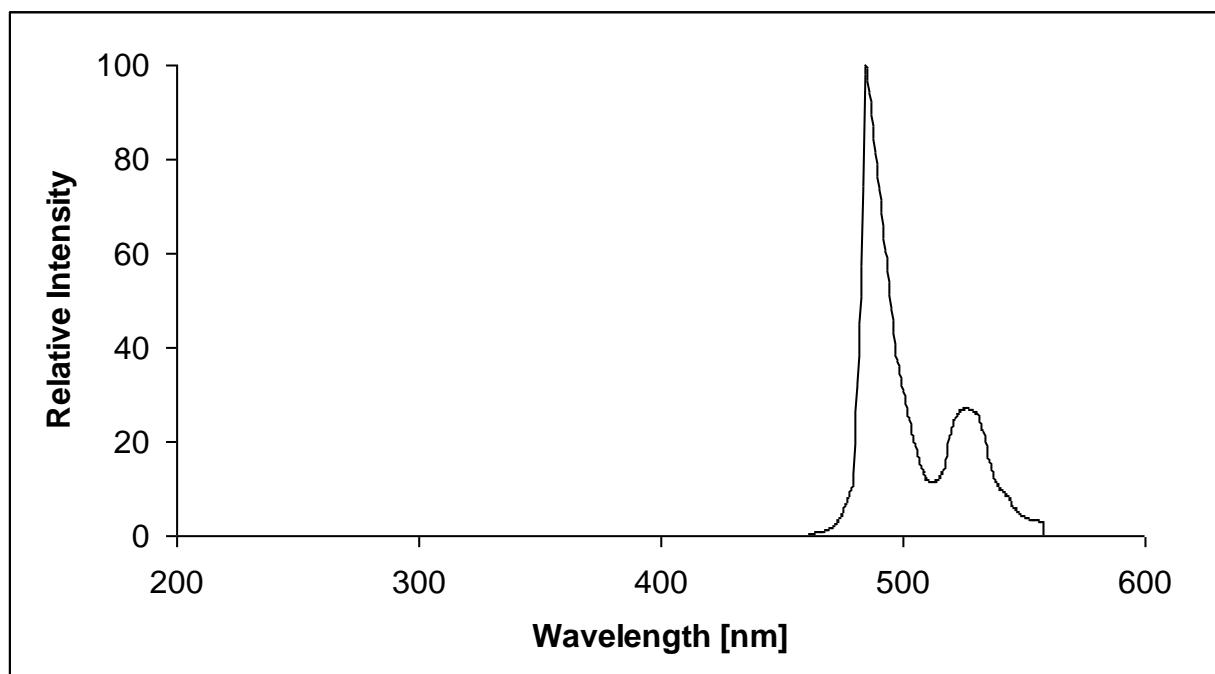
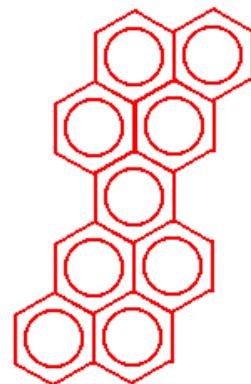
Tetrabenzo(de,hi,op,st)pentacene



λ exc.	375 nm
Formula	C ₃₄ H ₁₈
M.W.	426 u
H/H+C	0.346
m.p.	°C
m.a.c. (266 nm)	(l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	n-hexadecane
source	[5]

P09F

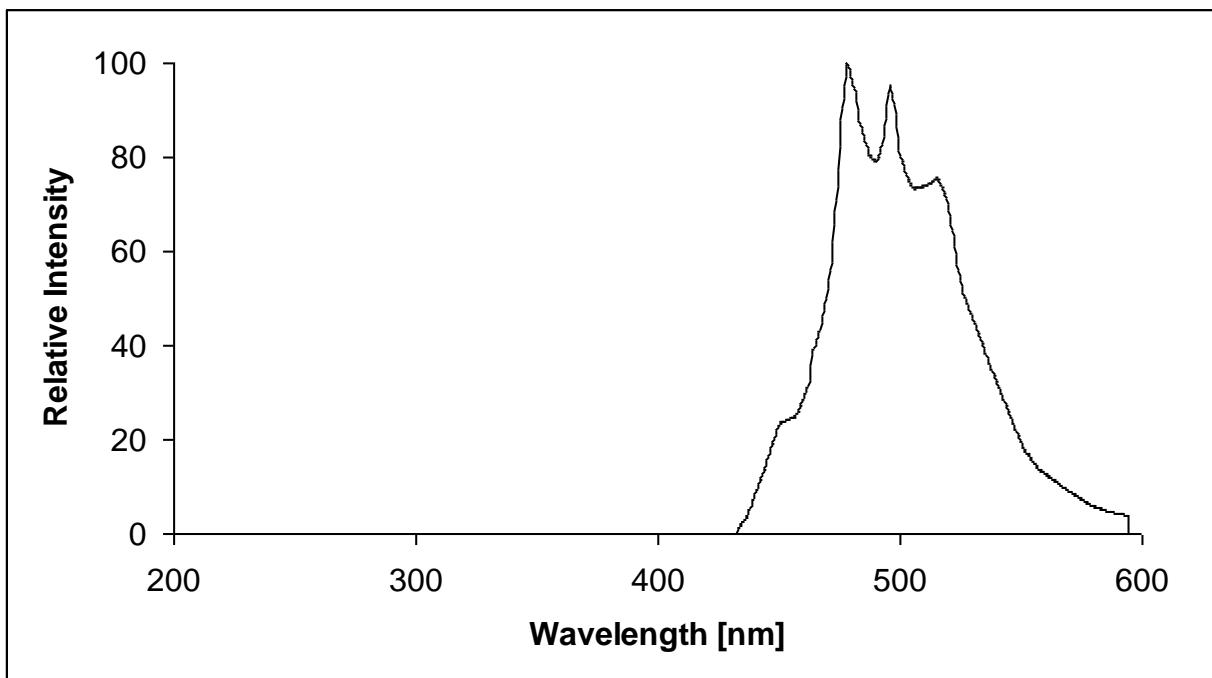
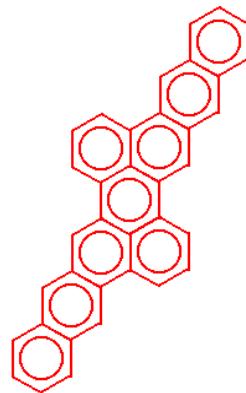
Dinaphtho(2,1,8,7-defg;2',1',8',7'-opqr)pentacene



λ exc.	410 nm
Formula	C ₃₂ H ₁₆
M.W.	400 u
H/H+C	0.333
m.p.	°C
m.a.c. (266 nm)	(l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	n-hexadecane
source	[5]

P09G

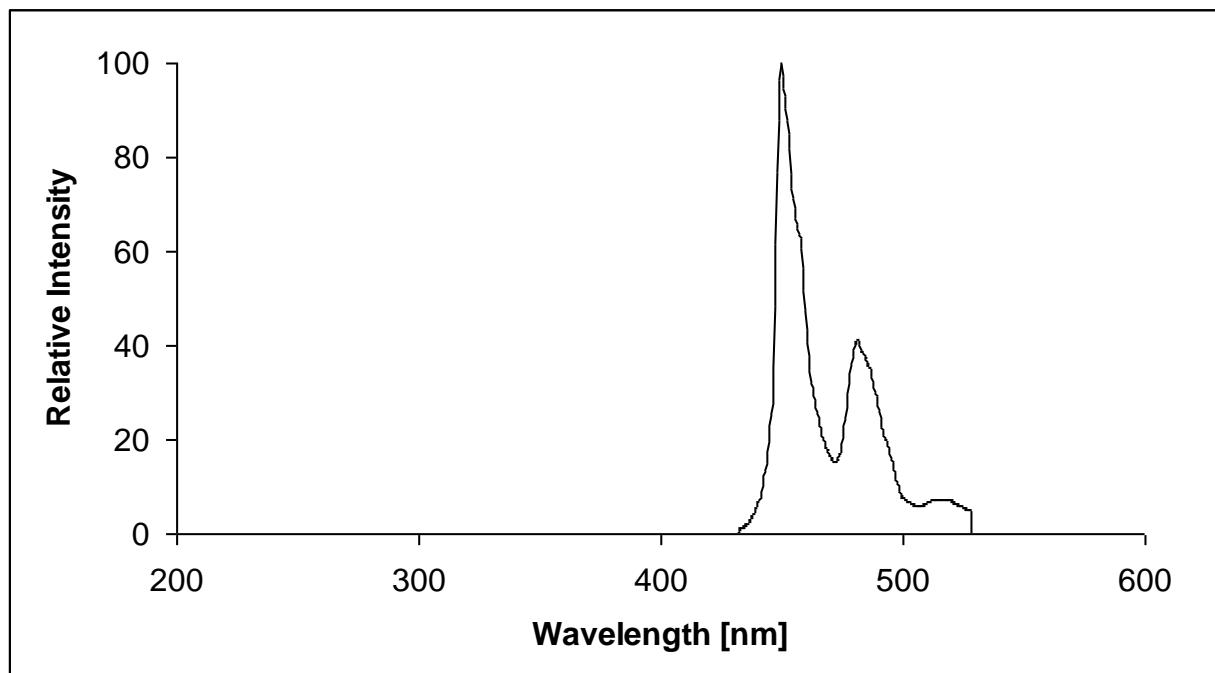
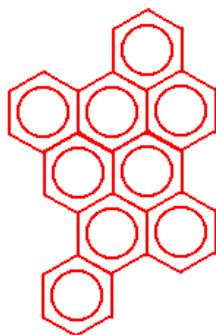
Dibenzo(hi,wx)heptacene



λ exc.	400 nm
Formula	$C_{36}H_{20}$
M.W.	452 u
H/H+C	0.357
m.p.	°C
m.a.c. (266 nm)	(l. $mol^{-1} cm^{-1} \times 10^{-4}$)
solvent	n-hexadecane
source	[16]

P09P

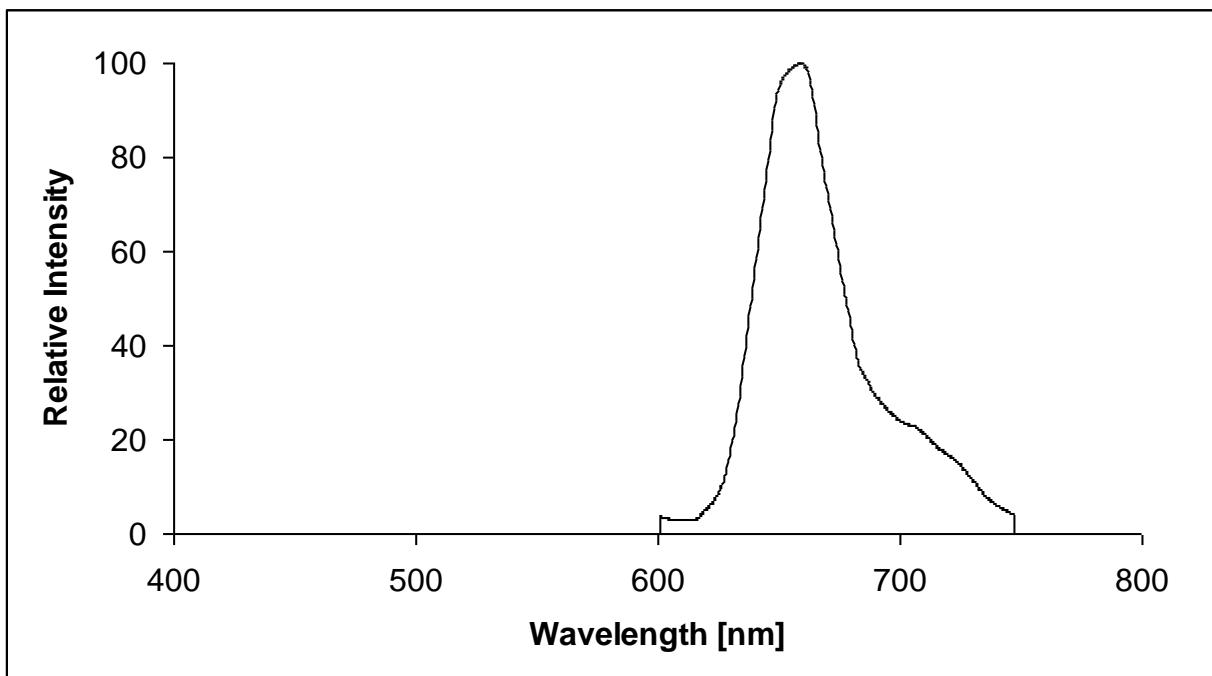
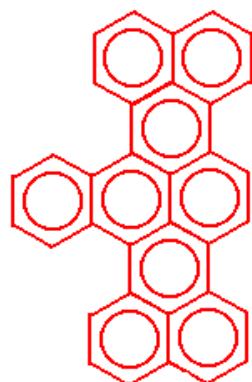
Dibenzo(cd,k)naphtho(8,1,2-fghi)perylene



λ exc.	321 nm
Formula	$C_{32}H_{16}$
M.W.	400 u
H/H+C	0.333
m.p.	°C
m.a.c. (266 nm)	(l. $mol^{-1} cm^{-1} \times 10^{-4}$)
solvent	n-hexadecane
source	[4]

P09R

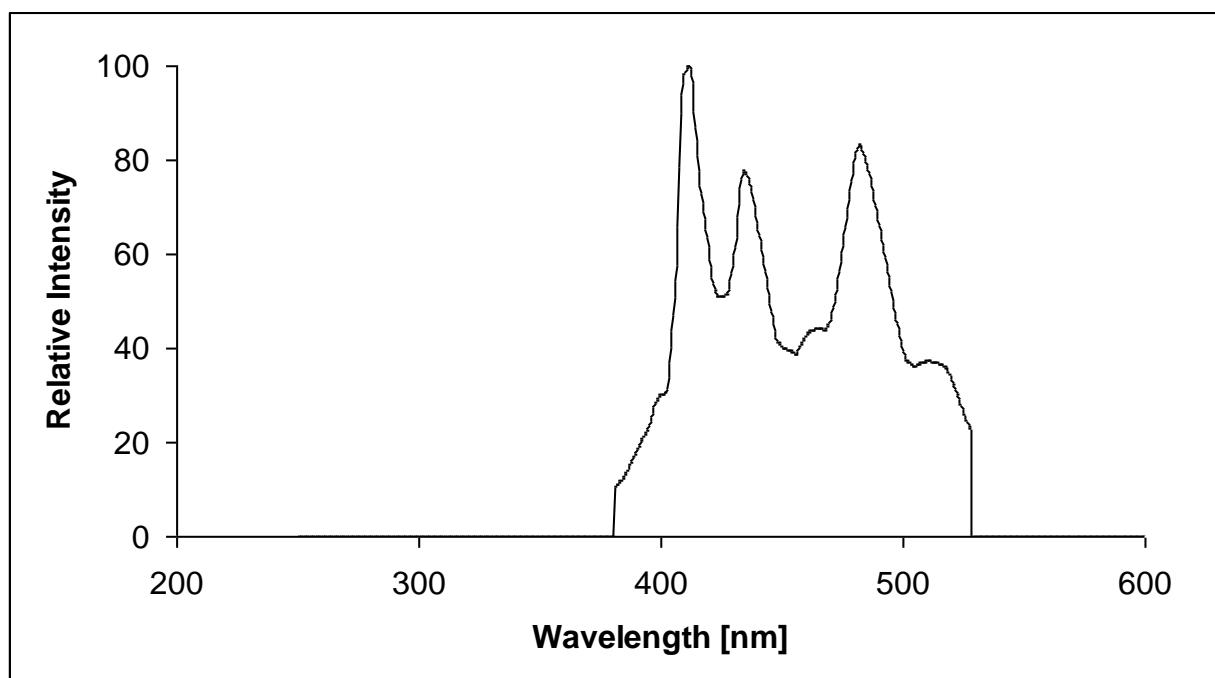
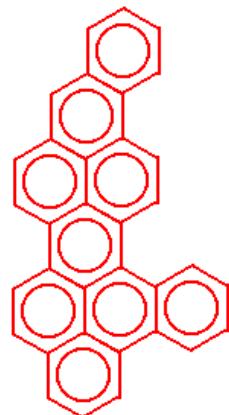
Dibenzo(de,kl)naphtho(1,2,3,4-rst)pentaphene



λ exc.	375 nm
Formula	$C_{34}H_{18}$
M.W.	426 u
H/H+C	0.346
m.p.	°C
m.a.c. (266 nm)	(l. $mol^{-1} cm^{-1} \times 10^{-4}$)
solvent	n-hexadecane
source	[5]

P09S

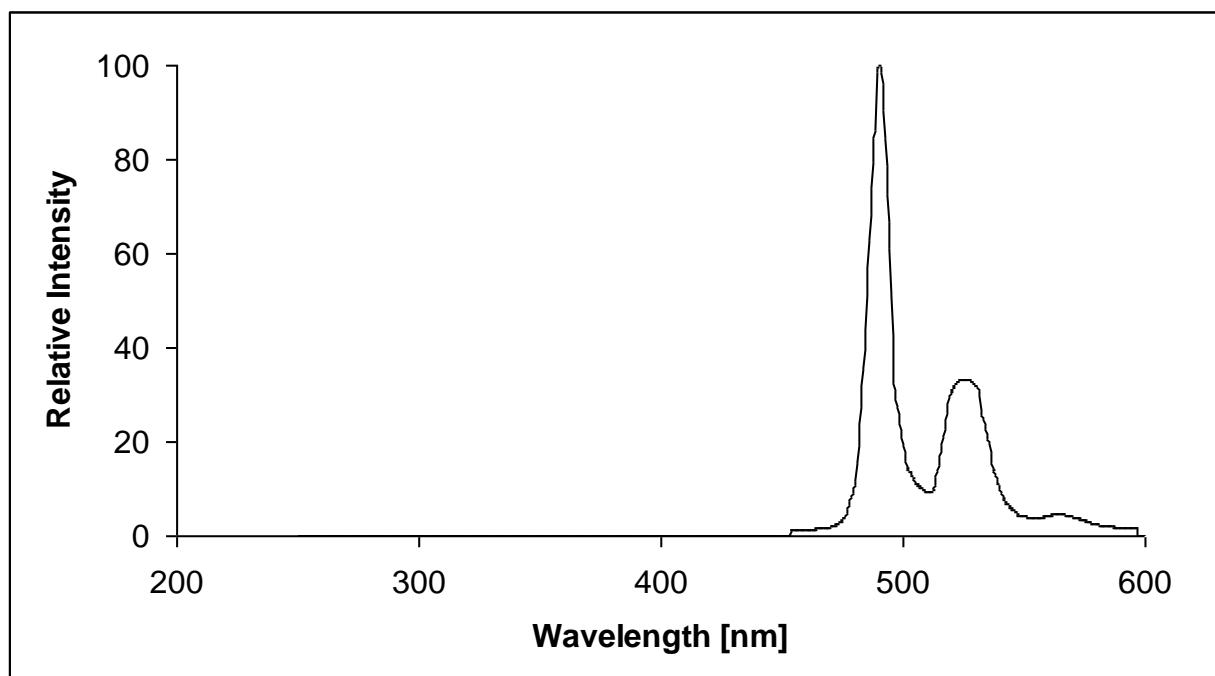
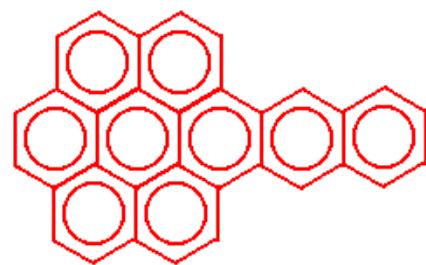
Dibenzo(a,rst)naphtho(8,1,2-cde)pentathene



λ exc.	300 nm
Formula	$C_{34}H_{18}$
M.W.	426 u
H/H+C	0.346
m.p.	°C
m.a.c. (266 nm)	(l. $mol^{-1} cm^{-1} \times 10^{-4}$)
solvent	n-hexadecane
source	[5]

P10C

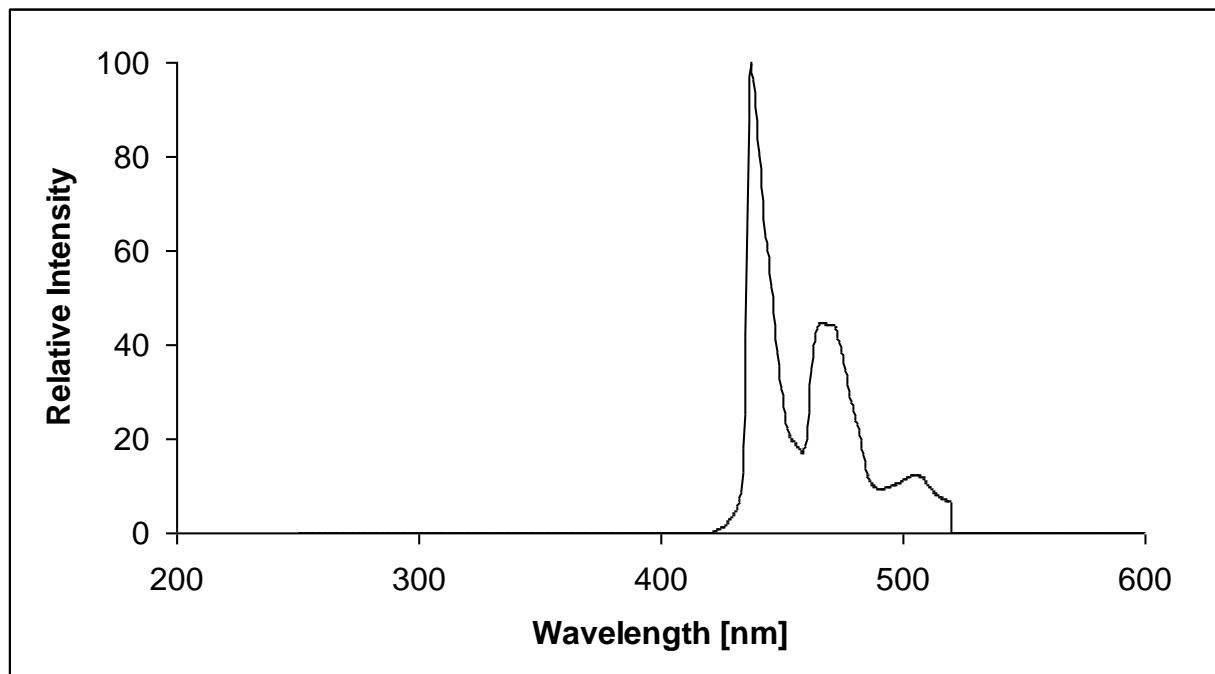
Anthra(2,3-a)coronene



λ exc.	357 nm
Formula	C ₃₆ H ₁₈
M.W.	450 u
H/H+C	0.333
m.p.	°C
m.a.c. (266 nm)	(l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	n-hexadecane
source	[3]

P10D

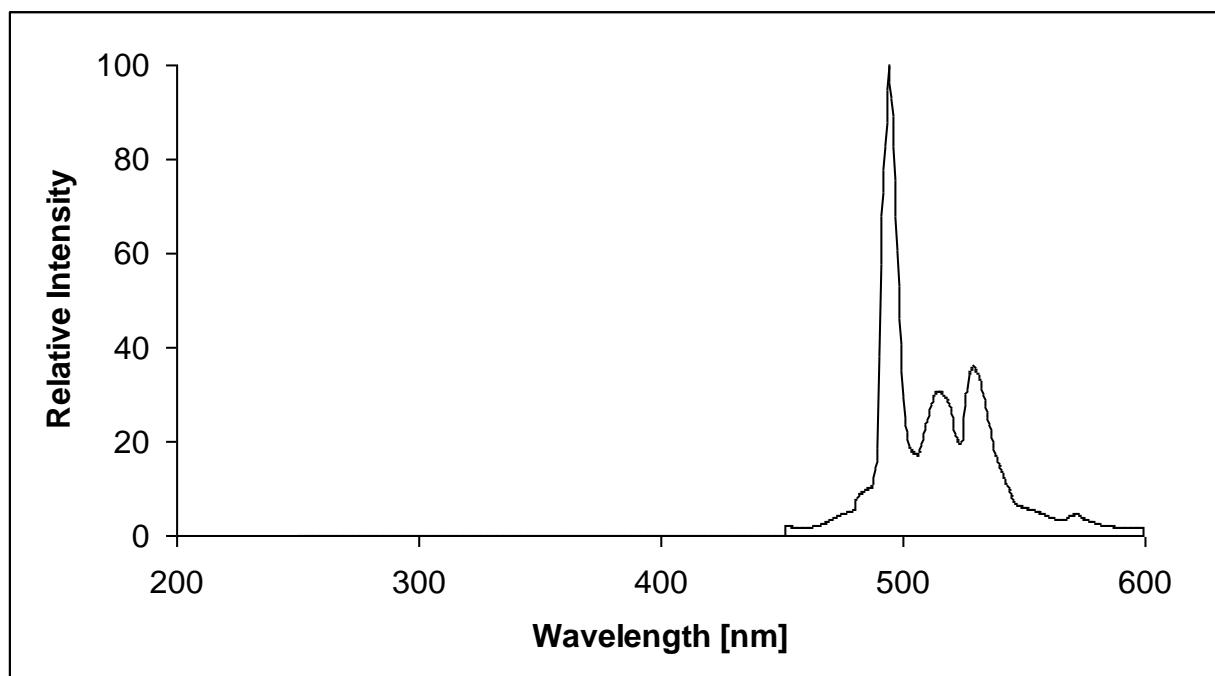
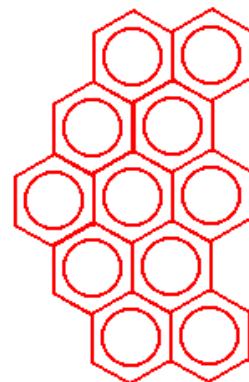
Benzo(rst)dinaphtho(2,1,8,7-defg;2',1',8',7'-ijkl)pentaphene



λ exc.	315 nm
Formula	$C_{34}H_{16}$
M.W.	424 u
H/H+C	0.320
m.p.	°C
m.a.c. (266 nm)	(l. $mol^{-1} cm^{-1} \times 10^{-4}$)
solvent	n-hexadecane
source	[5]

P11C

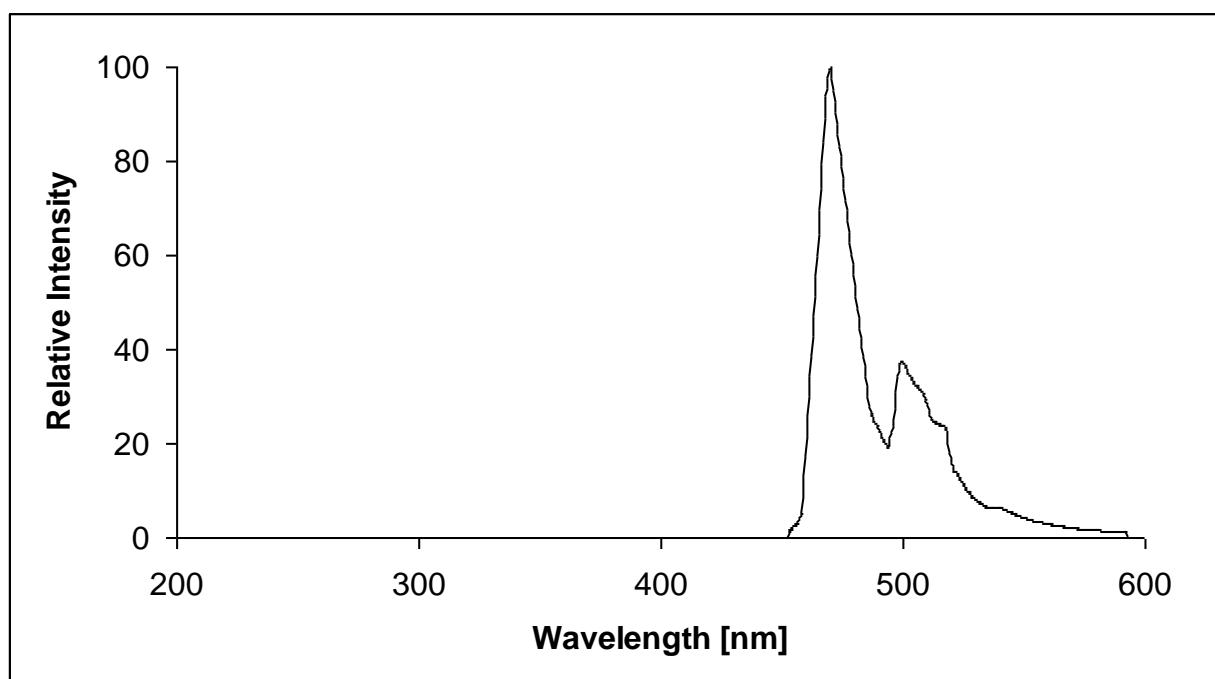
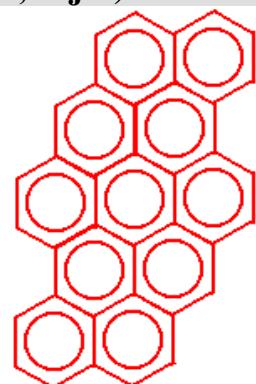
Dinaphtho(8,1,2-abc;2',1',8'-klm)coronene



λ exc.	425 nm
Formula	$C_{36}H_{16}$
M.W.	448 u
H/H+C	0.308
m.p.	°C
m.a.c. (266 nm)	(l. $mol^{-1} cm^{-1} \times 10^{-4}$)
solvent	n-hexadecane
source	[3]

P11D

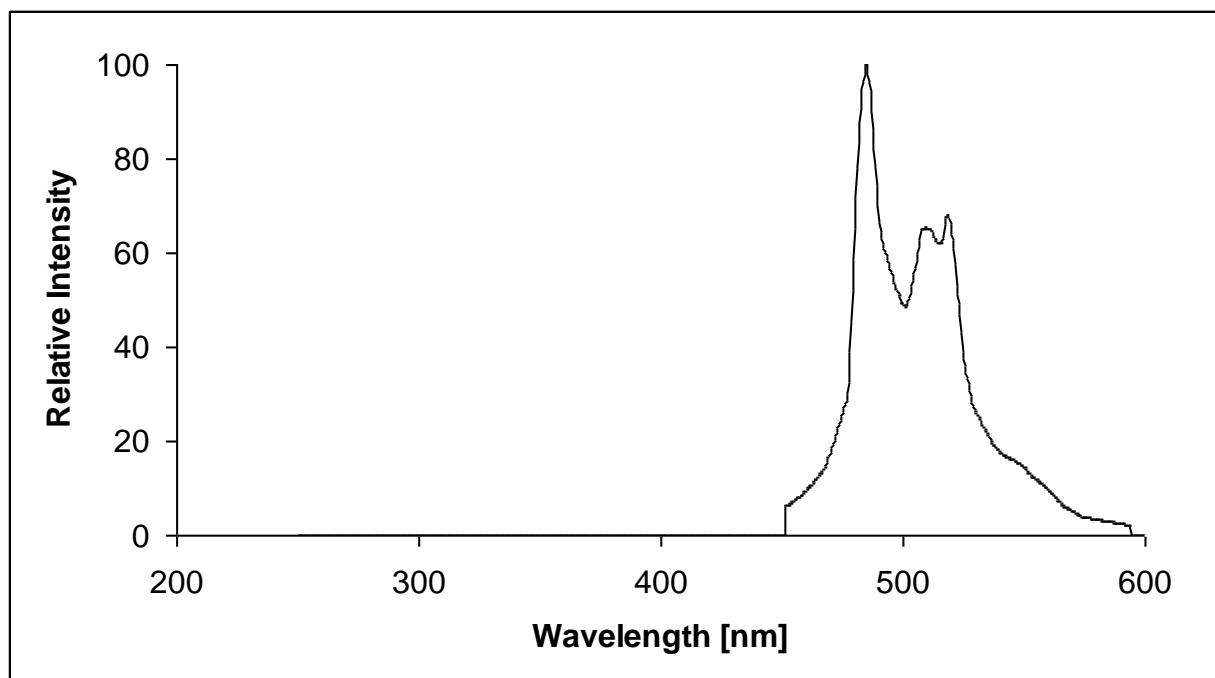
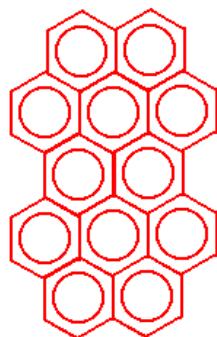
Dinaphtho(8,1,2-abc;2',1',8'-jkl)coronene



λ exc.	370 nm
Formula	$C_{36}H_{16}$
M.W.	448 u
H/H+C	0.308
m.p.	°C
m.a.c. (266 nm)	(l. $mol^{-1} cm^{-1} \times 10^{-4}$)
solvent	n-hexadecane
source	[3]

P12C

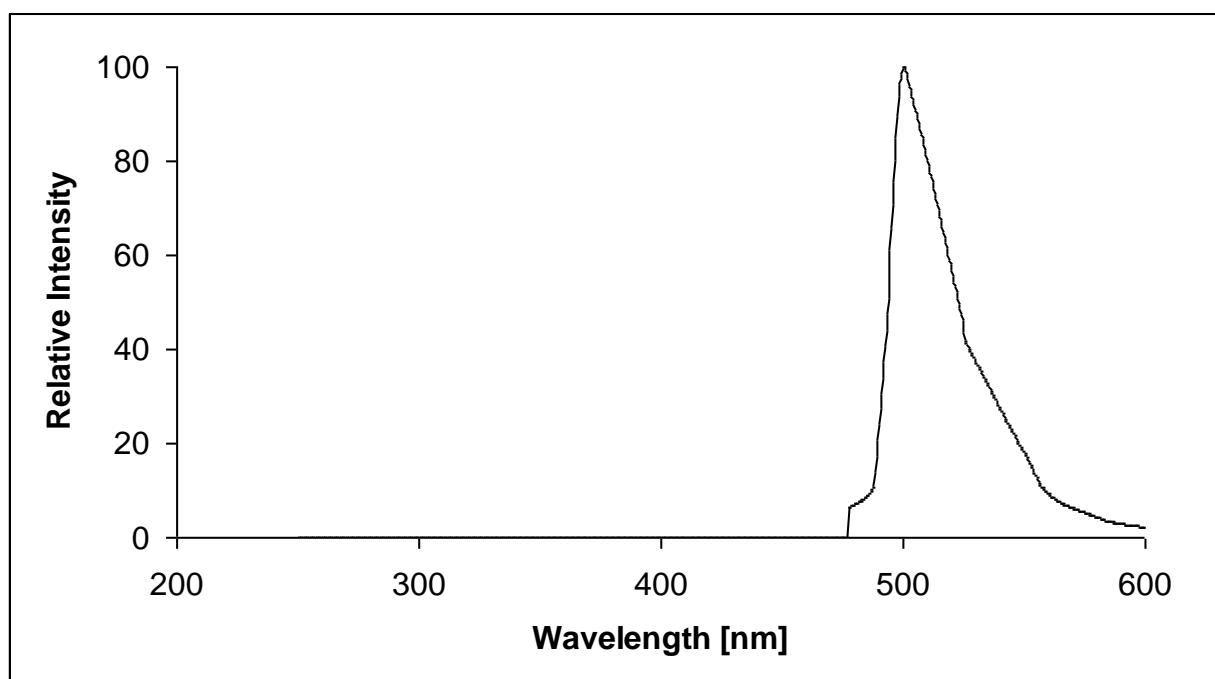
Naphth(2',1',8',7':4,10,5)anthra(1,9,8-abcd)coronene



λ exc.	425 nm
Formula	$C_{38}H_{16}$
M.W.	472 u
H/H+C	0.296
m.p.	°C
m.a.c. (266 nm)	(l. $mol^{-1} cm^{-1} \times 10^{-4}$)
solvent	n-hexadecane
source	[3]

P15C

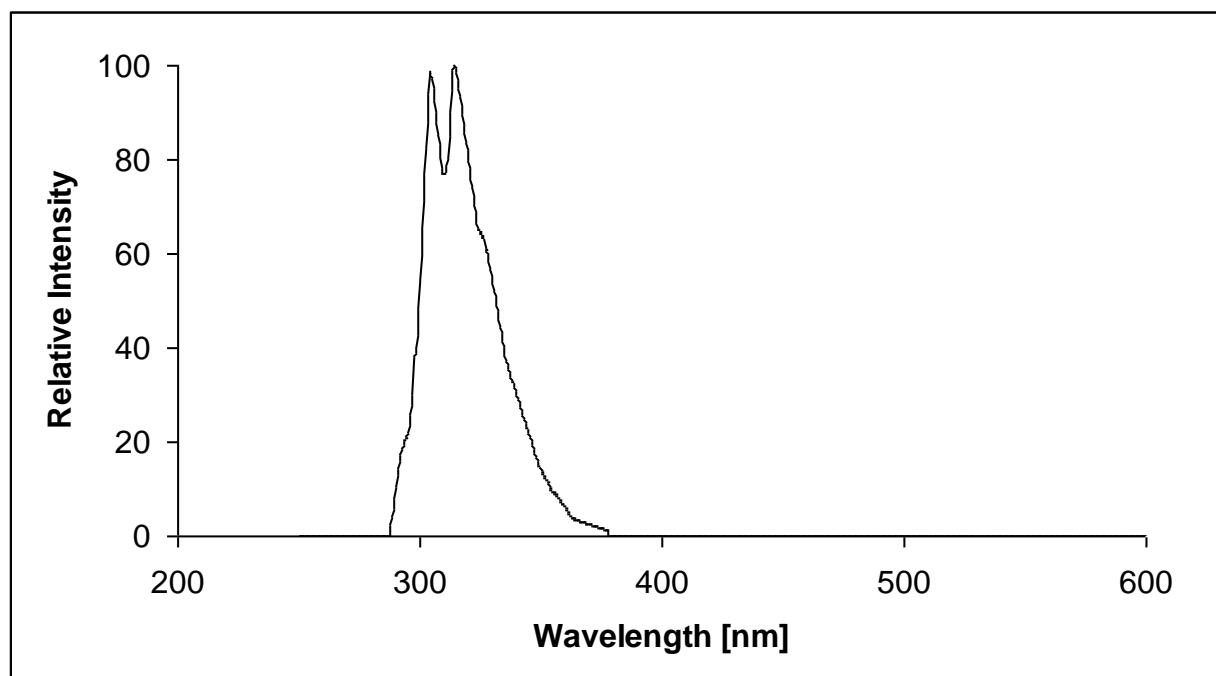
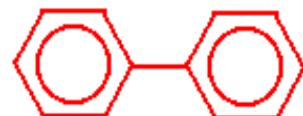
Benzo(1,2,3-bc;4,5,6-b',c')dicoronene



λ exc.	250 (440?) nm
Formula	C ₄₇ H ₂₀
M.W.	584 u
H/H+C	0.299
m.p.	°C
m.a.c. (266 nm)	(l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	Dichloromethane
source	[3]

PBP0

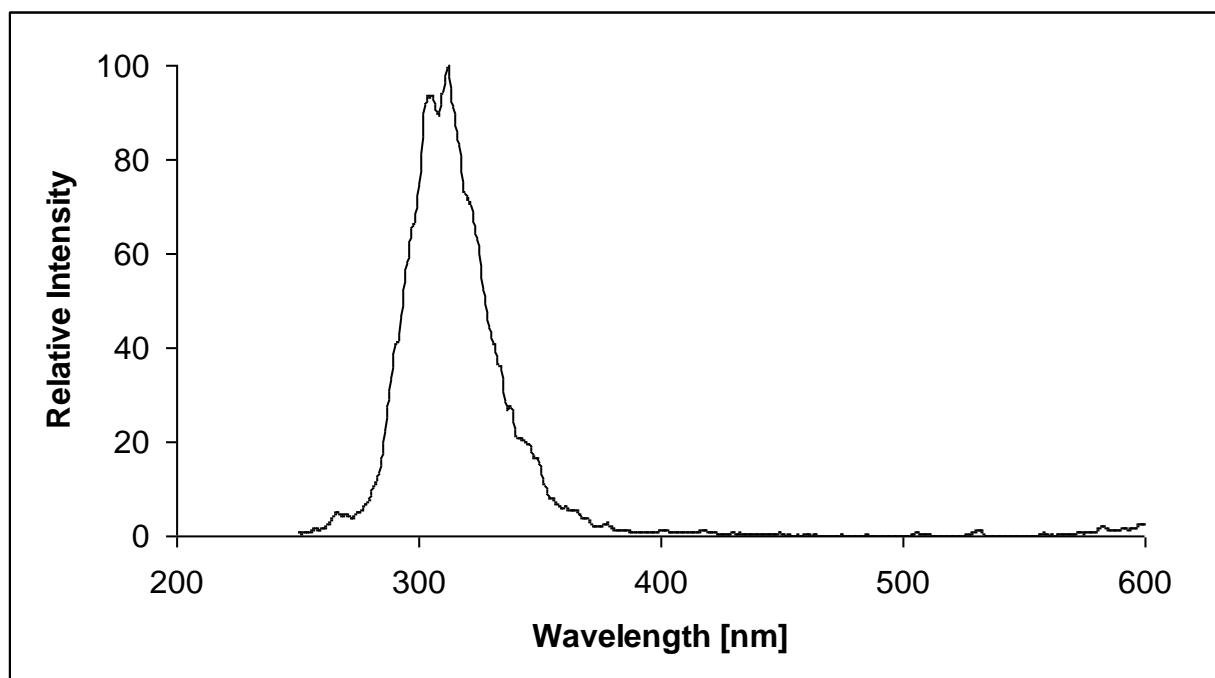
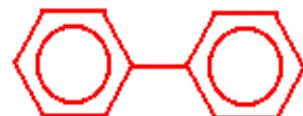
Biphenyl (1)



λ exc.	253.7 nm
Formula	C ₁₂ H ₁₀
M.W.	154 u
H/H+C	0.455
m.p.	°C
m.a.c. (266 nm)	0.7 (l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	Cyclohexane
source	Berlman

PBP0G

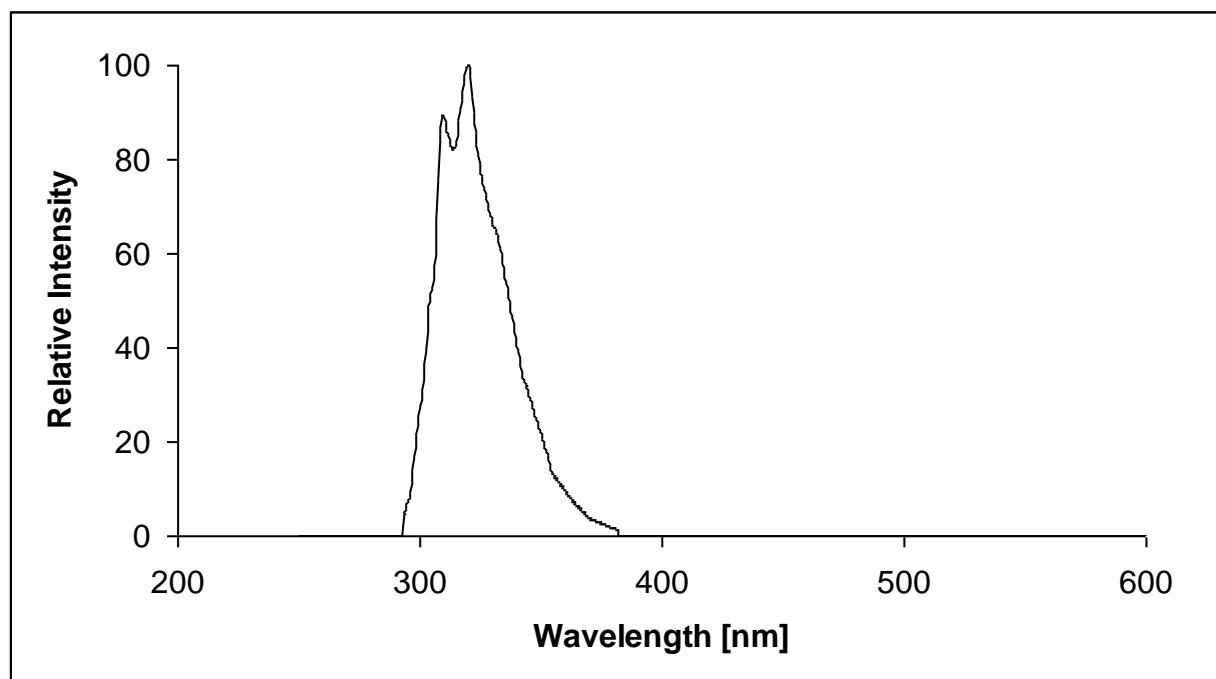
Biphenyl (2)



λ exc.	266 nm
Formula	C ₁₂ H ₁₀
M.W.	154 u
H/H+C	0.455
m.p.	°C
m.a.c. (266 nm)	0.7 (l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	Gas phase
source	CNPM

PBPB

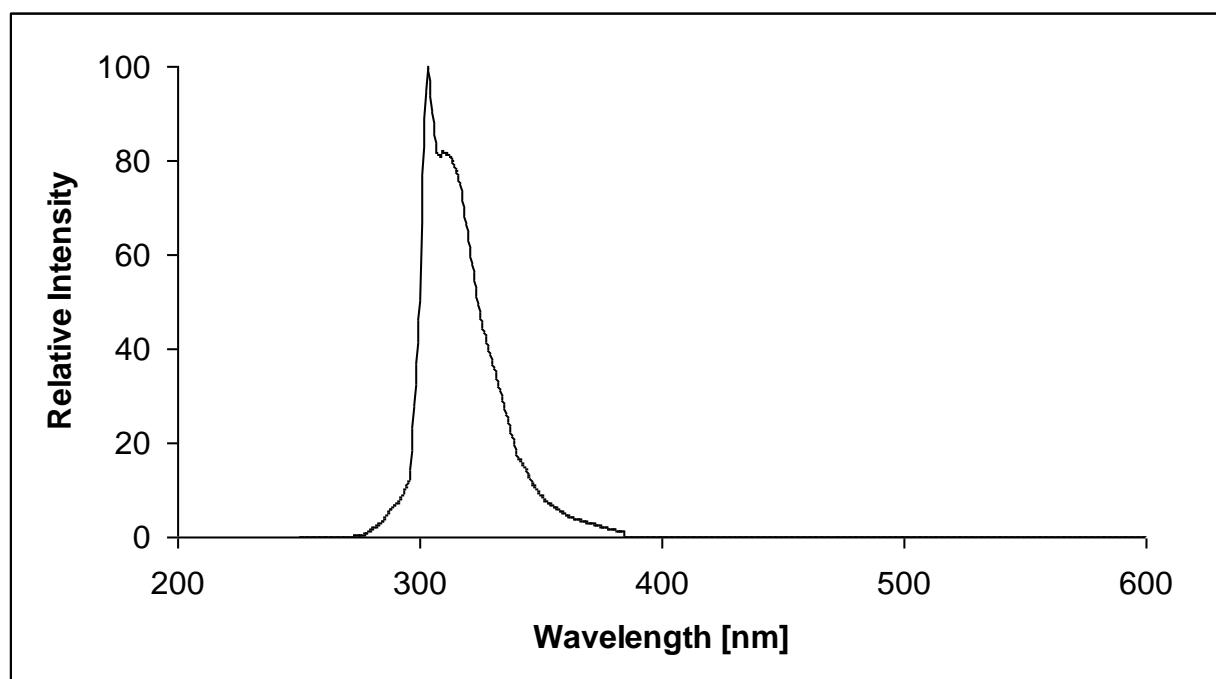
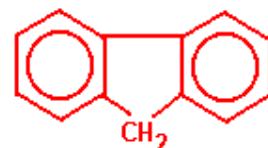
4 – Benzylbiphenyl



λ exc.	265 nm
Formula	C ₁₉ H ₁₆
M.W.	244 u
H/H+C	0.457
m.p.	°C
m.a.c. (266 nm)	1.65 (l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	Cyclohexane
source	Berlman

PBPF

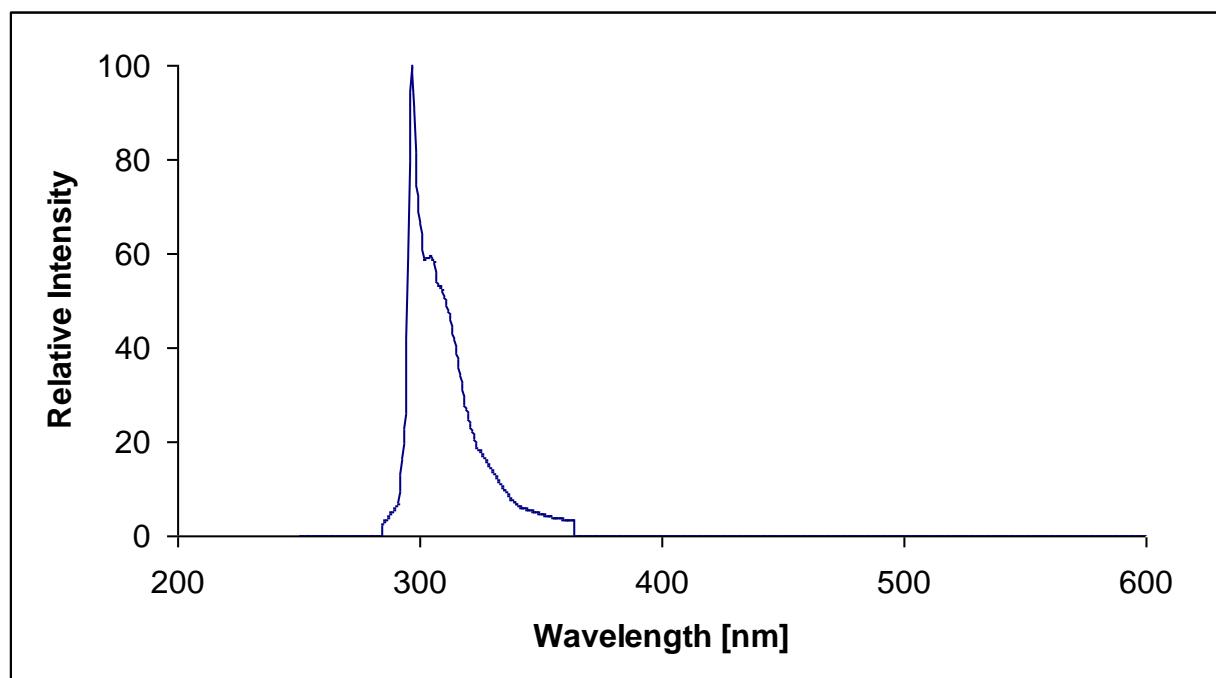
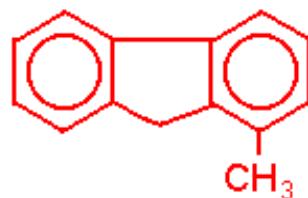
9H – Fluorene



λ exc.	262 nm
Formula	$\text{C}_{13}\text{H}_{10}$
M.W.	166 u
H/H+C	0.435
m.p.	115°C
m.a.c. (266 nm)	1.95 ($\text{l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4}$)
solvent	Cyclohexane
source	Atlas pg. 108-II

PBPFM1

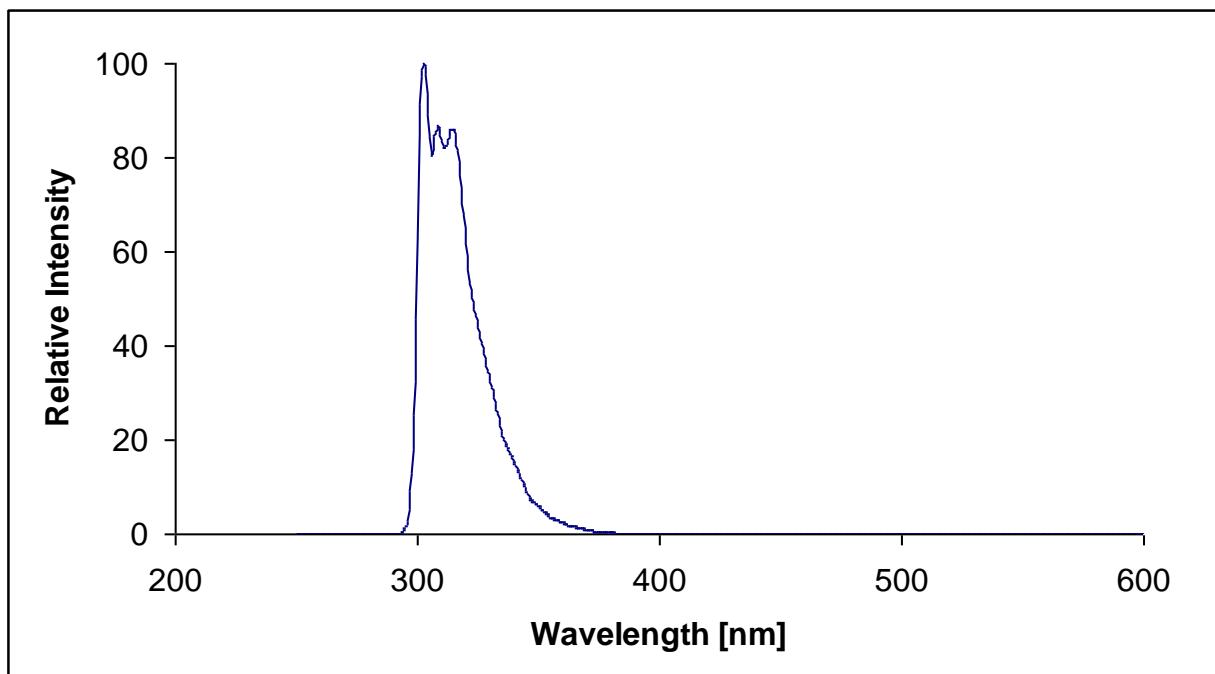
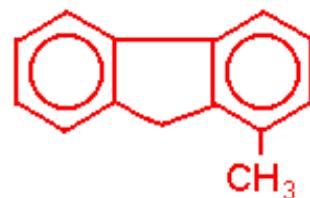
1 – Methylfluorene(1)



λ exc.	265.5 nm
Formula	$\text{C}_{14}\text{H}_{12}$
M.W.	180 u
H/H+C	0.462
m.p.	85°C
m.a.c. (266 nm)	$2.11 (\text{l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4})$
solvent	Cyclohexane
source	Atlas pag. 662 - III

PBPFM2

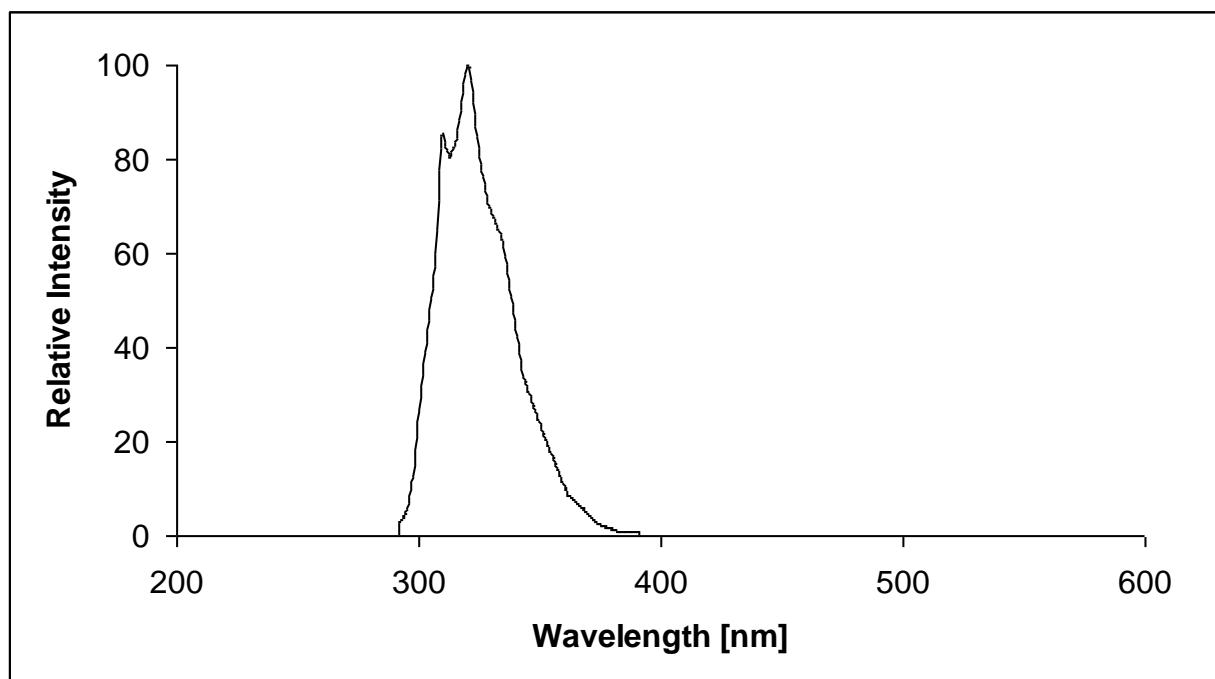
1 – Methylfluorene(2)



λ exc.	265.5 nm
Formula	$\text{C}_{14}\text{H}_{12}$
M.W.	180 u
H/H+C	0.462
m.p.	85°C
m.a.c. (266 nm)	2.11 ($\text{l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4}$)
solvent	Cyclohexane
source	Berlman

PBPM

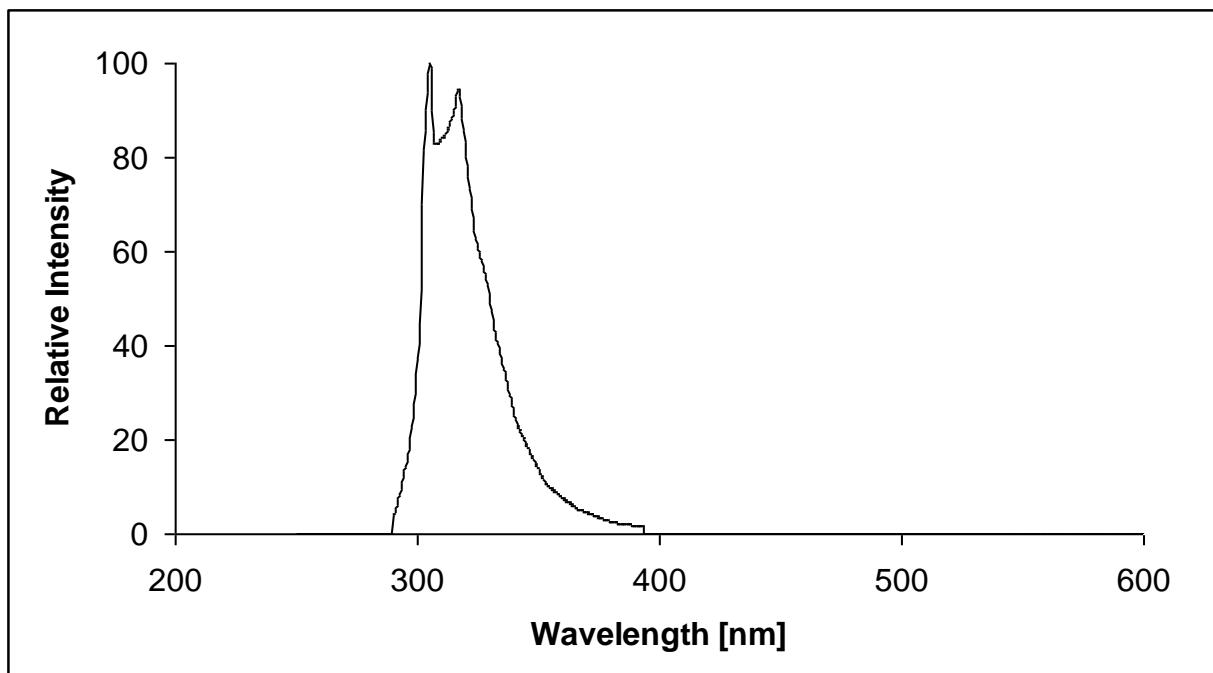
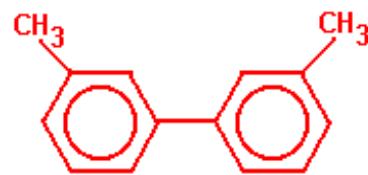
4 – Methylbiphenyl



λ exc.	265 nm
Formula	$\text{C}_{13}\text{H}_{12}$
M.W.	168 u
H/H+C	0.480
m.p.	°C
m.a.c. (266 nm)	$0.110 (\text{l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4})$
solvent	Cyclohexane
source	Berlman

PBPMJ

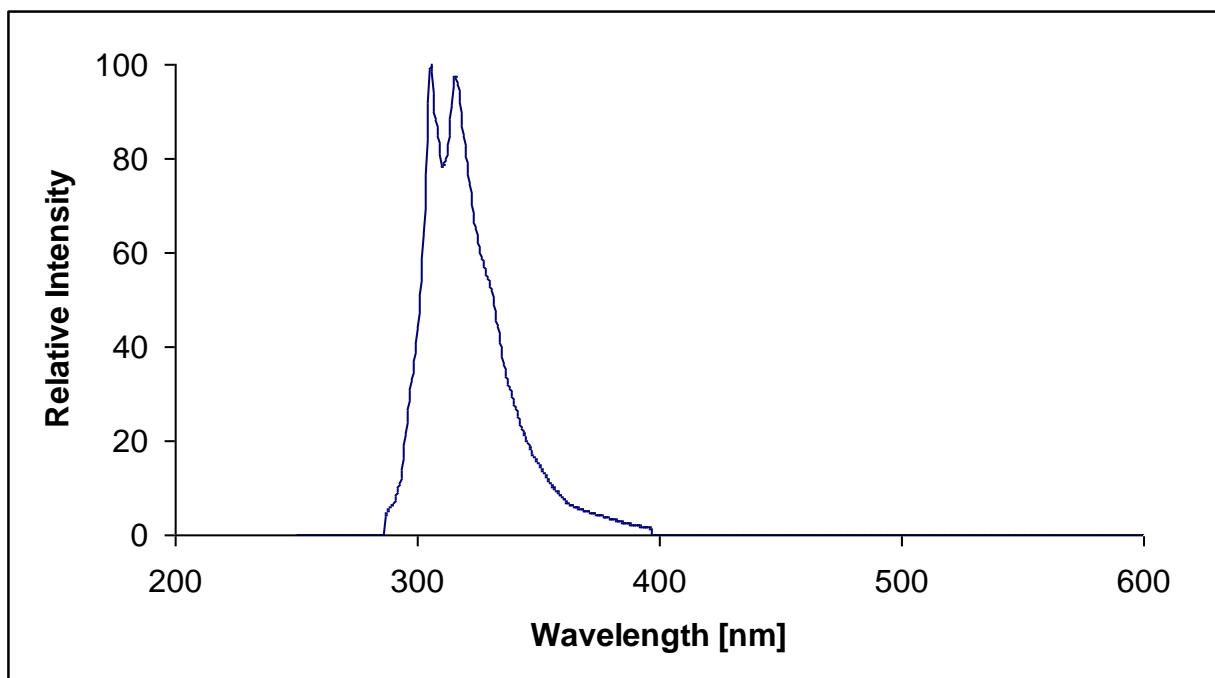
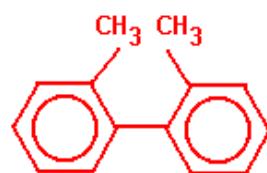
3,3' – Dimethylbiphenyl



λ exc.	265 nm
Formula	$\text{C}_{14}\text{H}_{14}$
M.W.	182 u
H/H+C	0.500
m.p.	°C
m.a.c. (266 nm)	$0.65 (\text{l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4})$
solvent	Cyclohexane
source	Berlman

PBPMK

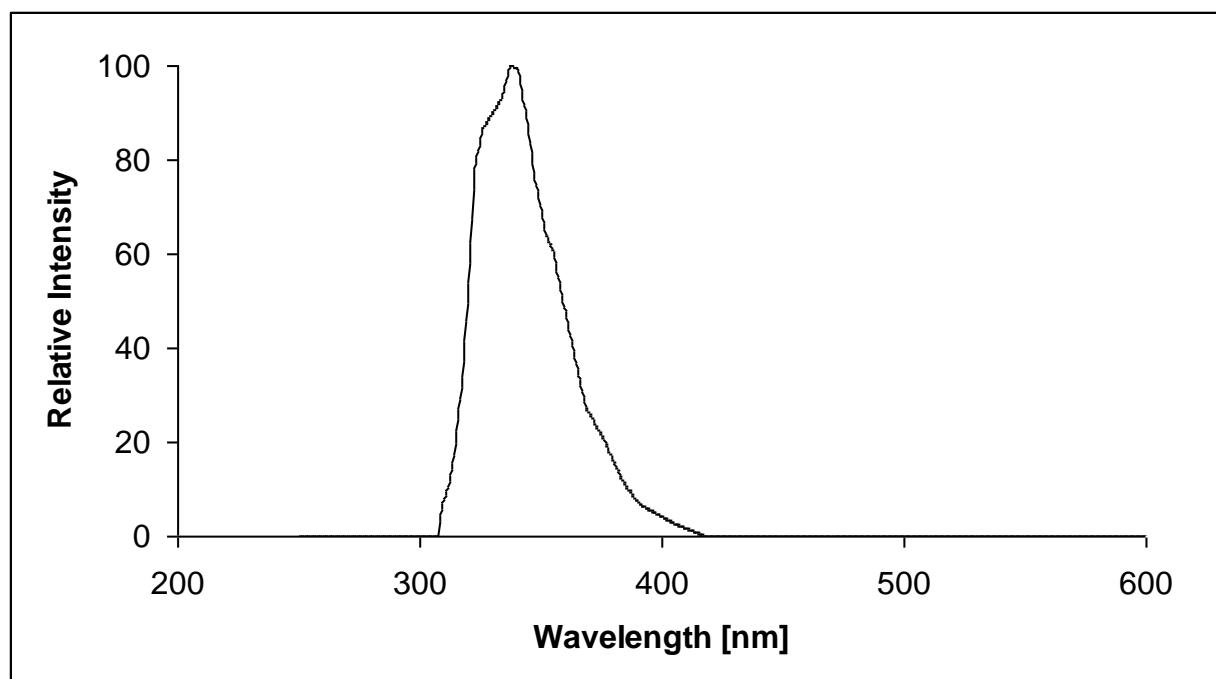
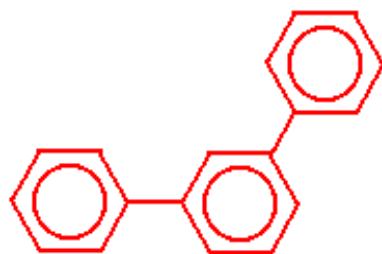
2,2' – Dimethylbiphenyl



λ exc.	265 nm
Formula	$C_{14}H_{14}$
M.W.	182 u
H/H+C	0.500
m.p.	°C
m.a.c. (266 nm)	0.105 ($l \cdot mol^{-1} cm^{-1} \times 10^{-4}$)
solvent	Cyclohexane
source	Berlman

PBPP

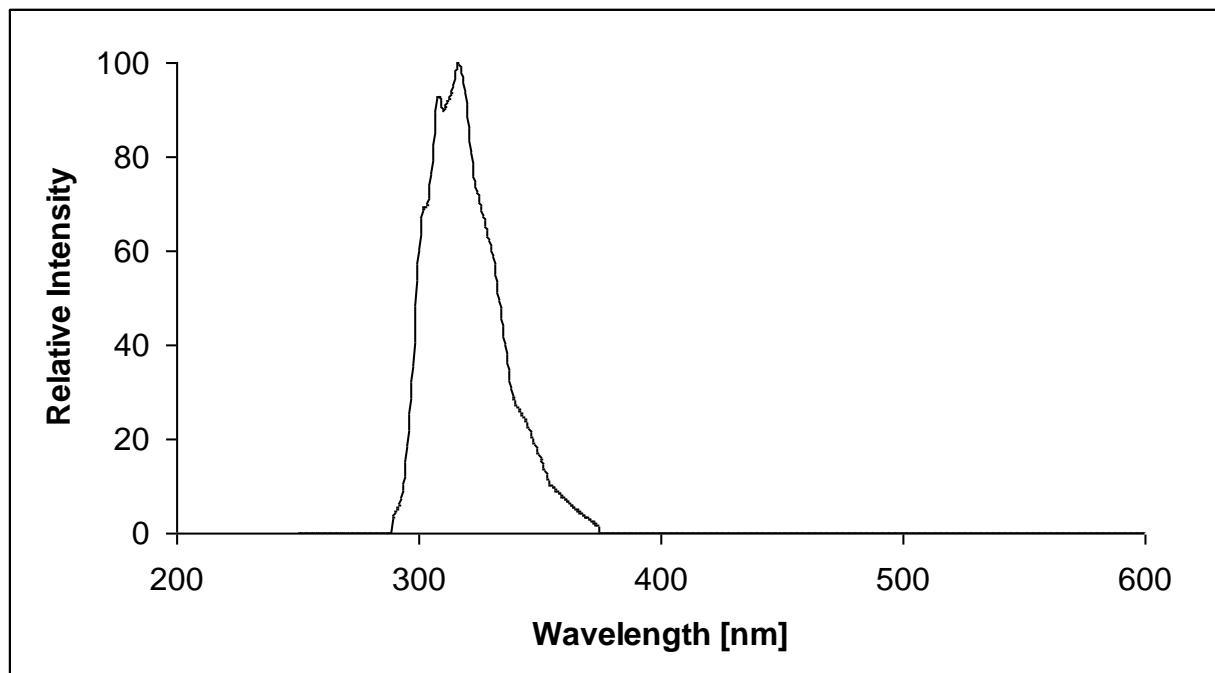
m-Terphenyl



λ exc.	253.7 nm
Formula	$C_{18}H_{14}$
M.W.	230 u
H/H+C	0.438
m.p.	°C
m.a.c. (266 nm)	$2.1 \text{ (l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4})$
solvent	Cyclohexane
source	Berlman

PBPR

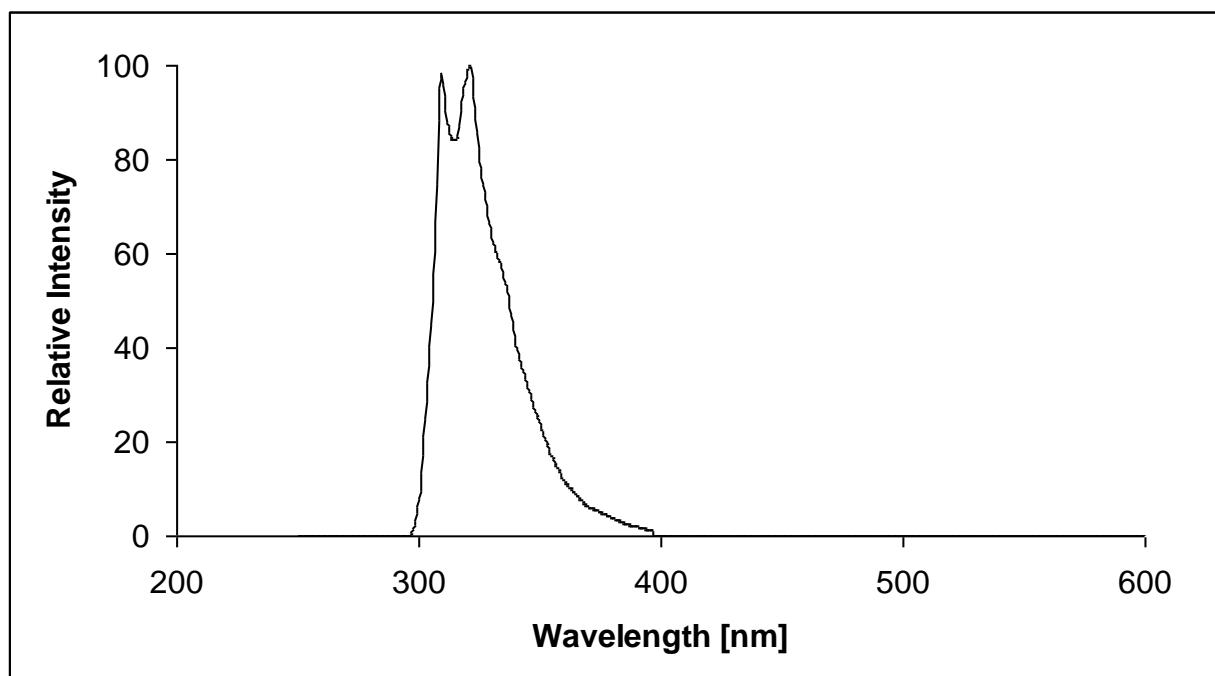
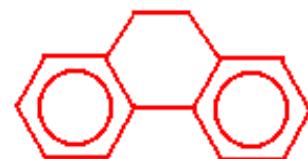
Monoisopropylbiphenyl



λ exc.	265 nm
Formula	$\text{C}_{15}\text{H}_{16}$
M.W.	196 u
H/H+C	0.516
m.p.	°C
m.a.c. (266 nm)	1 ($\text{l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4}$)
solvent	Cyclohexane
source	Berlman

PBPT

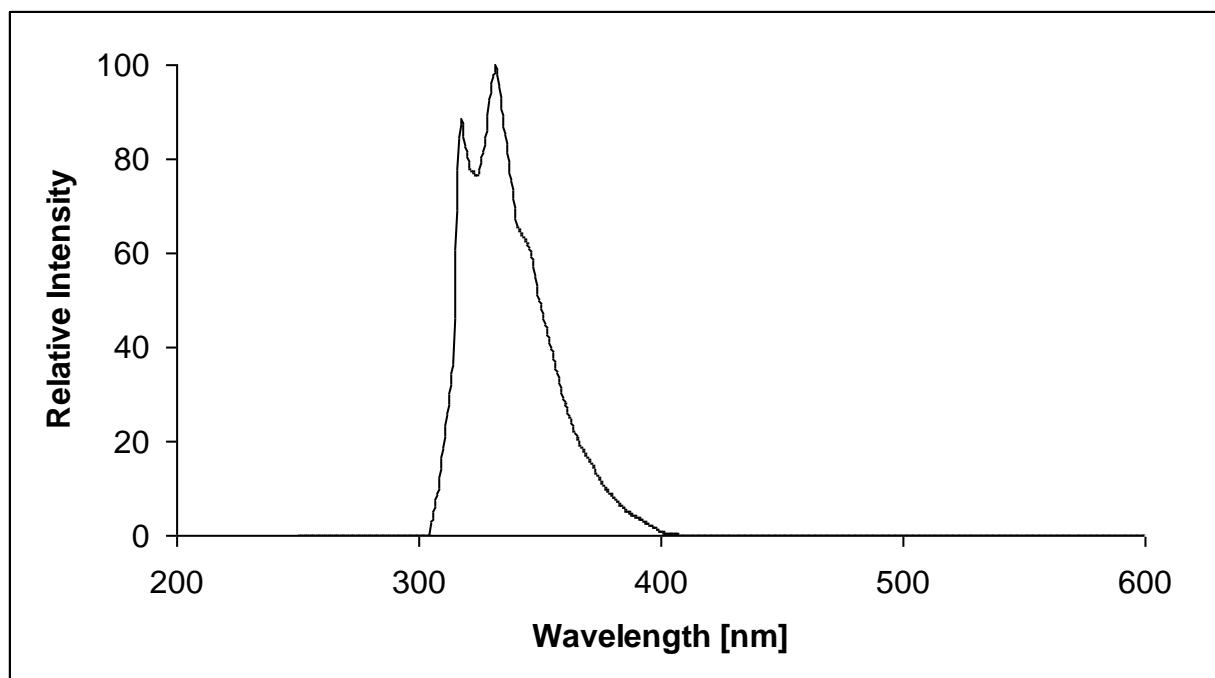
9,10 – Dihydrophenanthrene



λ exc.	265 nm
Formula	C ₁₄ H ₁₂
M.W.	180 u
H/H+C	0.462
m.p.	°C
m.a.c. (266 nm)	1.75 (l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	Cyclohexane
source	Berlman

PBPV

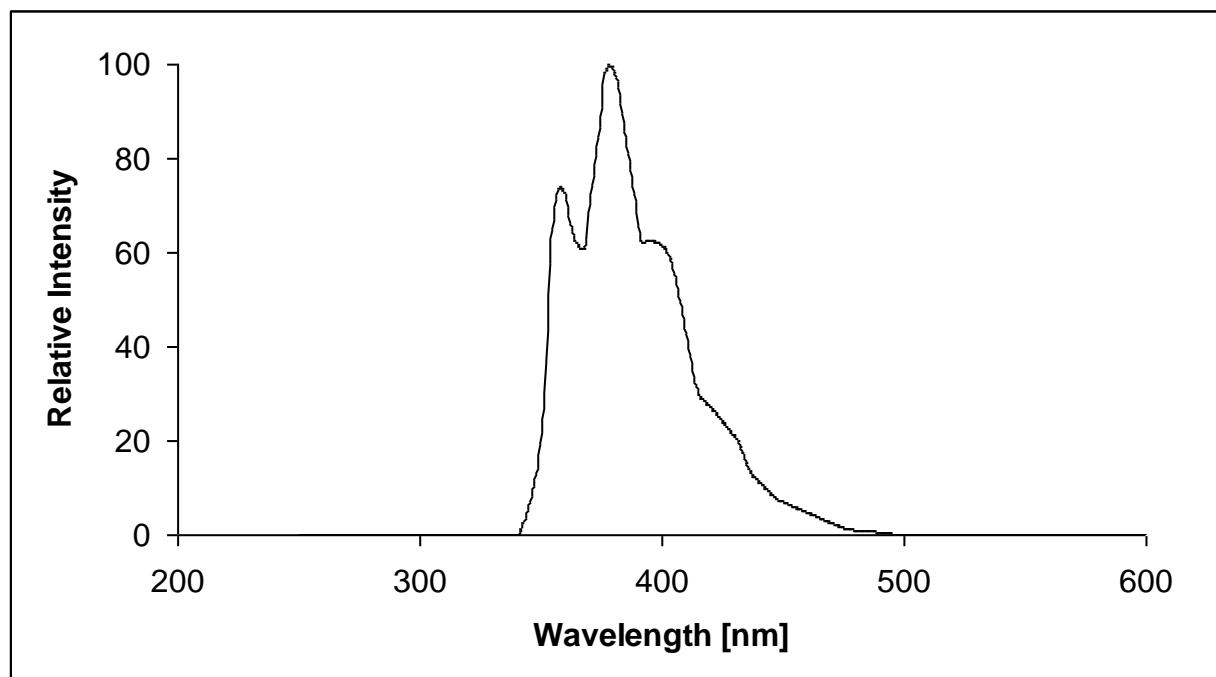
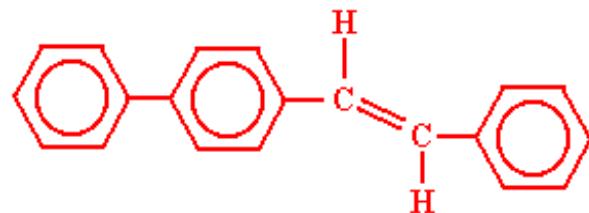
4 – Vinylbiphenyl



λ exc.	303 nm
Formula	C ₁₄ H ₁₂
M.W.	180 u
H/H+C	0.462
m.p.	°C
m.a.c. (266 nm)	2.55 (l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	Cyclohexane
source	Berlman

PBPZ

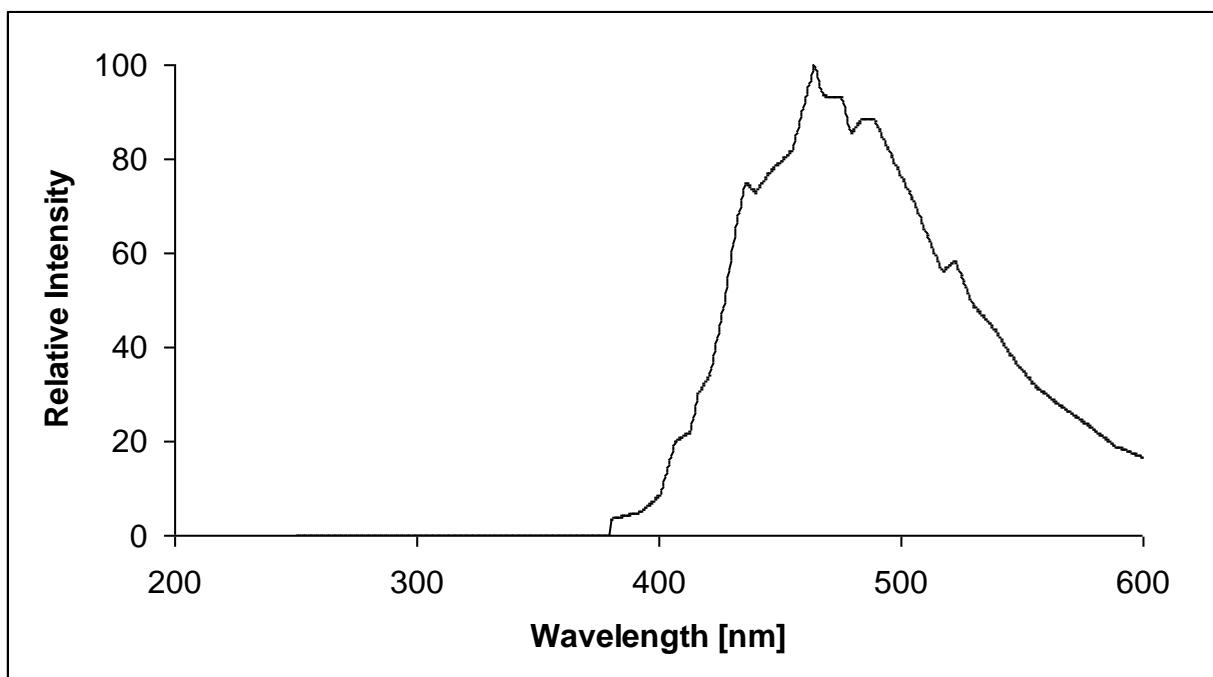
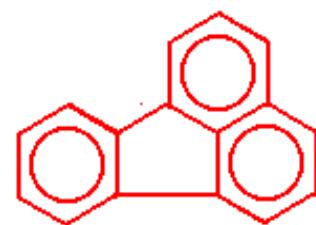
Trans-1-(4-biphenyl)-2-phenylethylene



λ exc.	313 nm
Formula	$C_{20}H_{16}$
M.W.	256 u
H/H+C	0.444
m.p.	°C
m.a.c. (266 nm)	0.4 ($l \cdot mol^{-1} cm^{-1} \times 10^{-4}$)
solvent	Cyclohexane
source	Berlman

PFL31

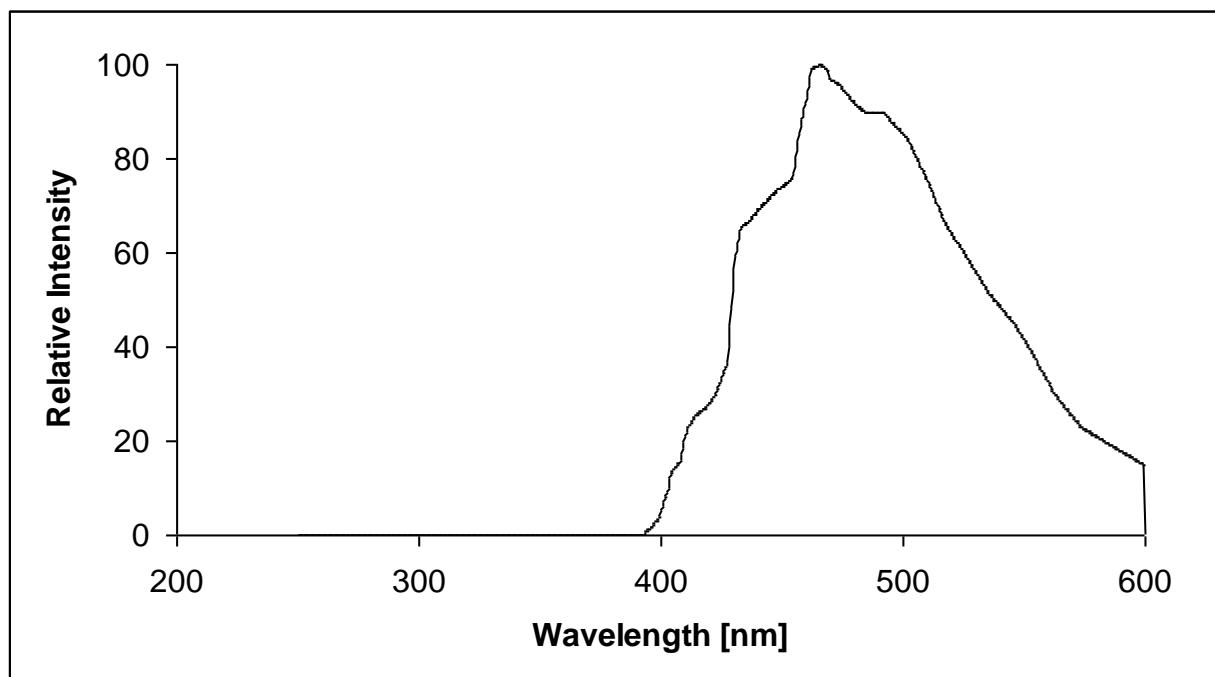
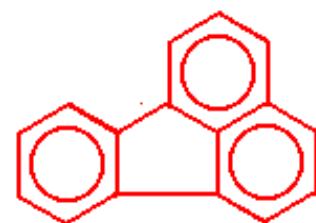
Fluoranthene(1)



λ exc.	288 nm
Formula	$C_{16}H_{10}$
M.W.	202 u
H/H+C	0.385
m.p.	108.8°C
m.a.c. (266 nm)	$0.95 \text{ (l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4})$
solvent	Cyclohexane
source	Atlas pg. 74-I

PFL32

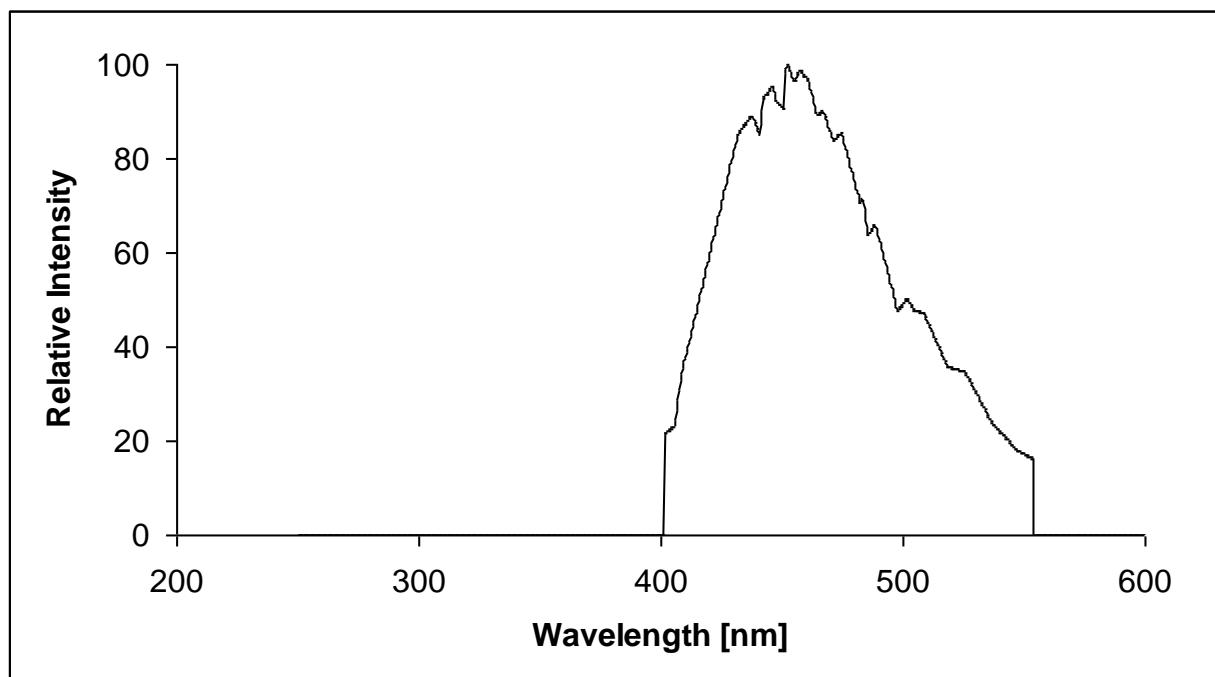
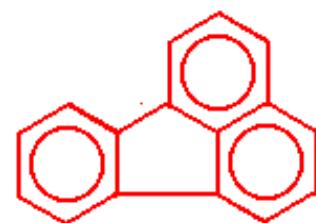
Fluoranthene(2)



λ exc.	365 nm
Formula	$C_{16}H_{10}$
M.W.	202 u
H/H+C	0.385
m.p.	°C
m.a.c. (266 nm)	(l. $mol^{-1} cm^{-1} \times 10^{-4}$)
solvent	Cyclohexane
source	Berlman

PFL3G1

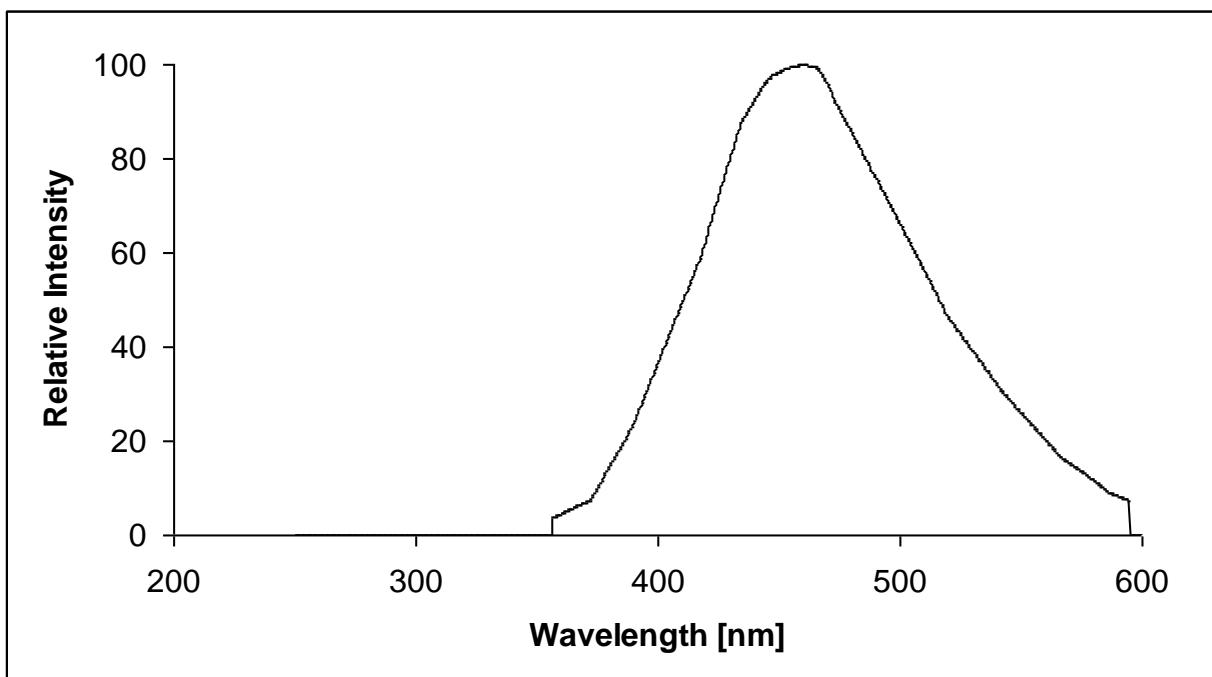
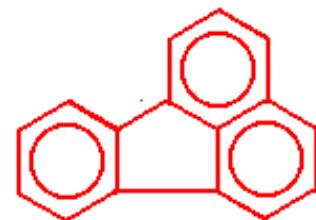
Fluoranthene(3)



λ exc.	337.1 nm
Formula	$C_{16}H_{10}$
M.W.	202 u
H/H+C	0.385
m.p.	108.8°C
m.a.c. (266 nm)	0.96 ($l \cdot mol^{-1} cm^{-1} \times 10^{-4}$)
solvent	Gas phase 298-378 K
source	[2]

PFL3G2

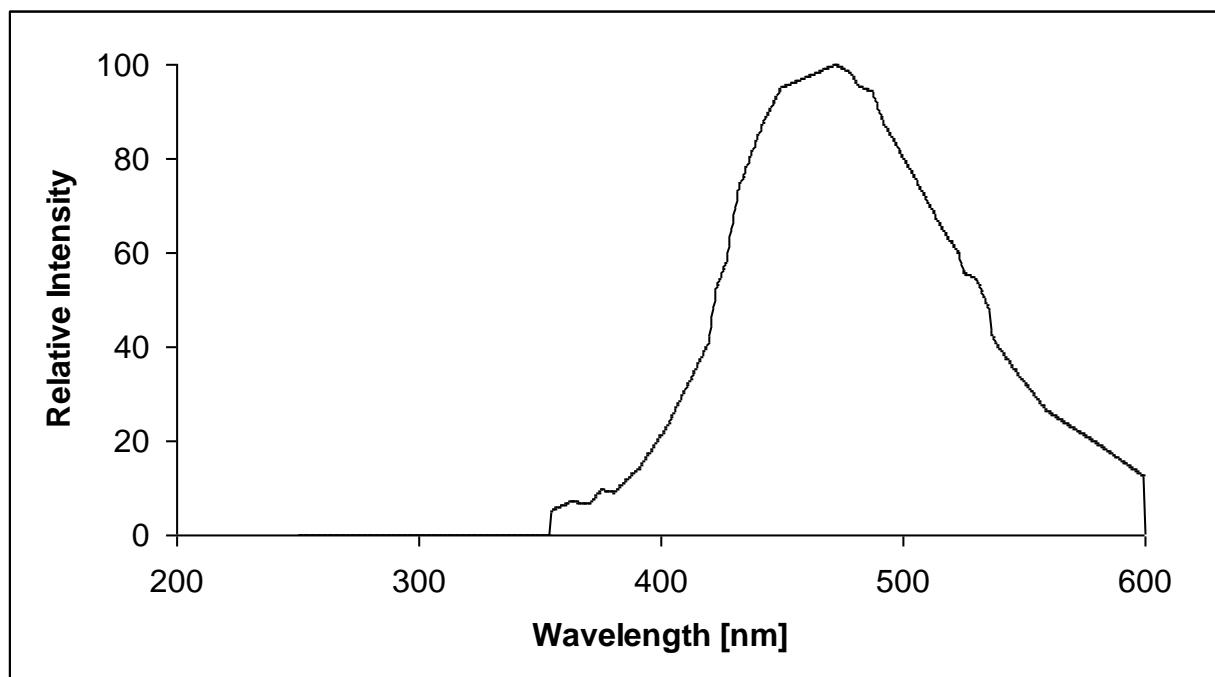
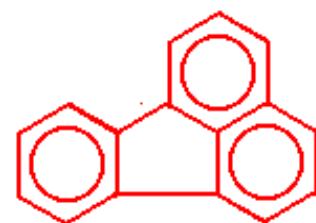
Fluoranthene(4)



λ exc.	337.1 nm
Formula	$C_{16}H_{10}$
M.W.	202 u
H/H+C	0.385
m.p.	108.8°C
m.a.c. (266 nm)	(l. $mol^{-1} cm^{-1} \times 10^{-4}$)
solvent	Gas phase 412-438 K
source	[12]

PFL3G3

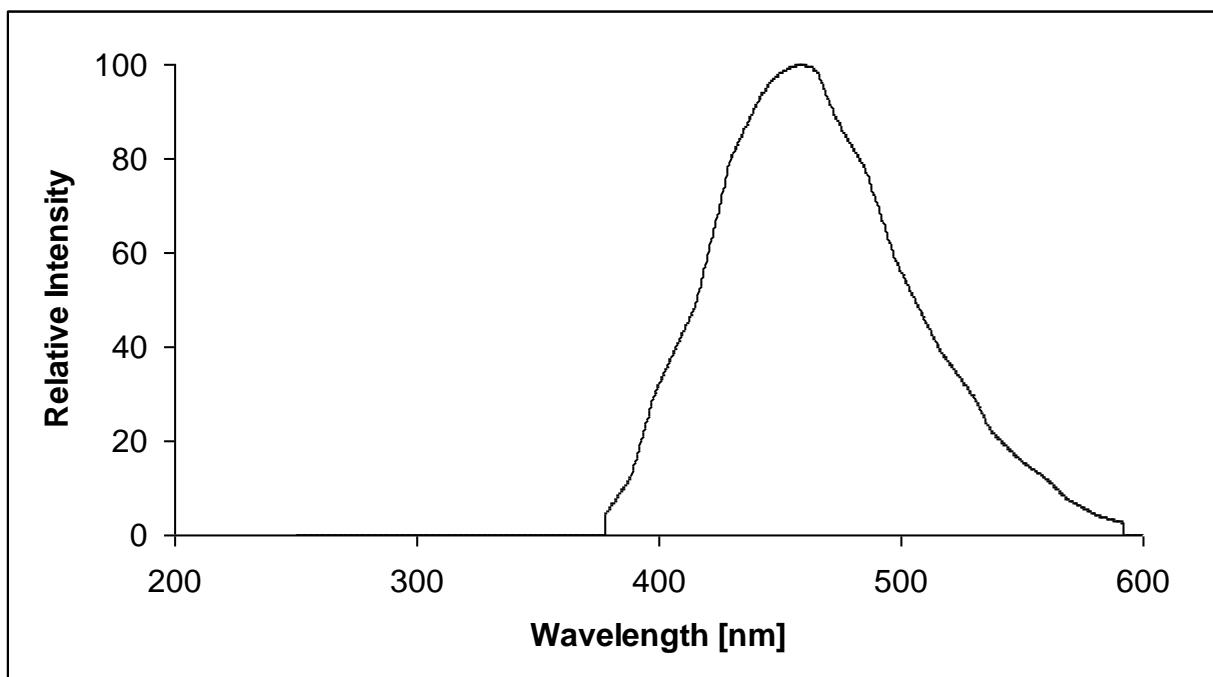
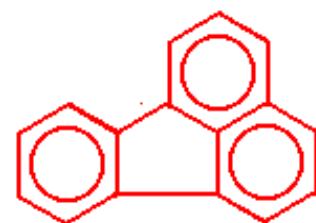
Fluoranthene(5)



λ exc.	337 nm
Formula	$C_{16}H_{10}$
M.W.	202 u
H/H+C	0.385
m.p.	108.8°C
m.a.c. (266 nm)	(l. $mol^{-1} cm^{-1} \times 10^{-4}$)
solvent	Gas phase 516 K
source	[1]

PFL3G4

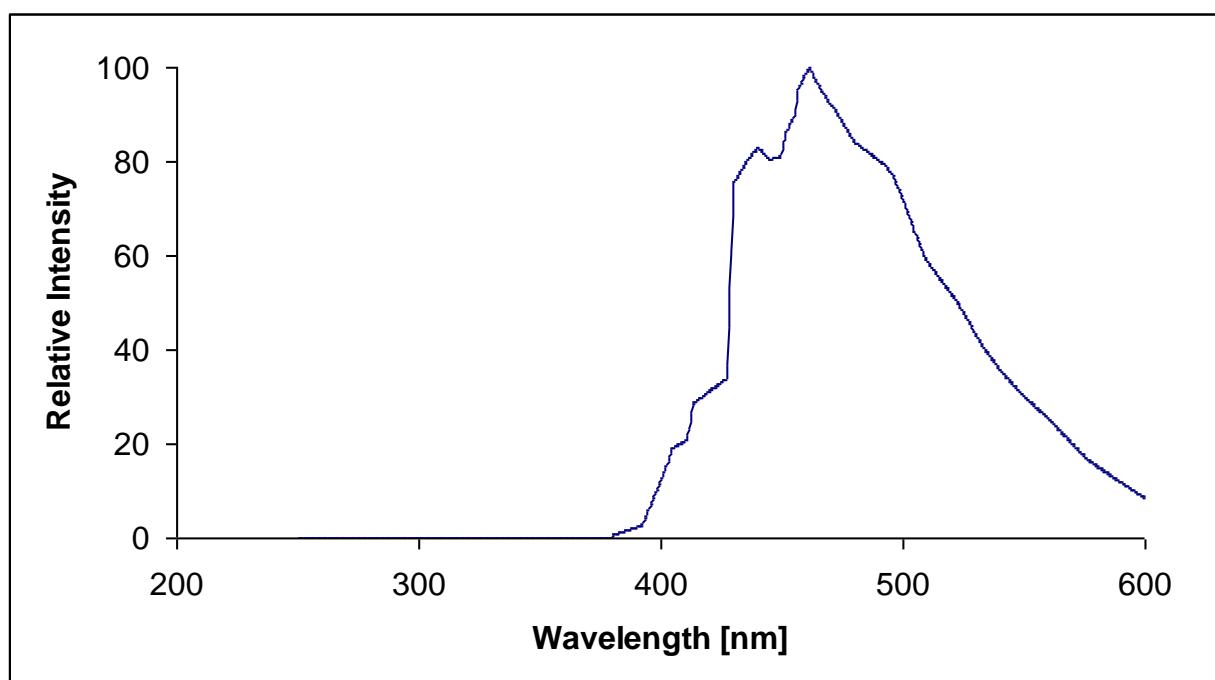
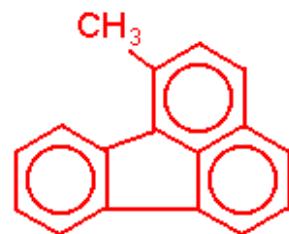
Fluoranthene(6)



λ exc.	337 nm
Formula	$C_{16}H_{10}$
M.W.	202 u
H/H+C	0.385
m.p.	108.8°C
m.a.c. (266 nm)	(l. $mol^{-1} cm^{-1} \times 10^{-4}$)
solvent	Gas phase 412-438 K
source	[12]

PFL3MA

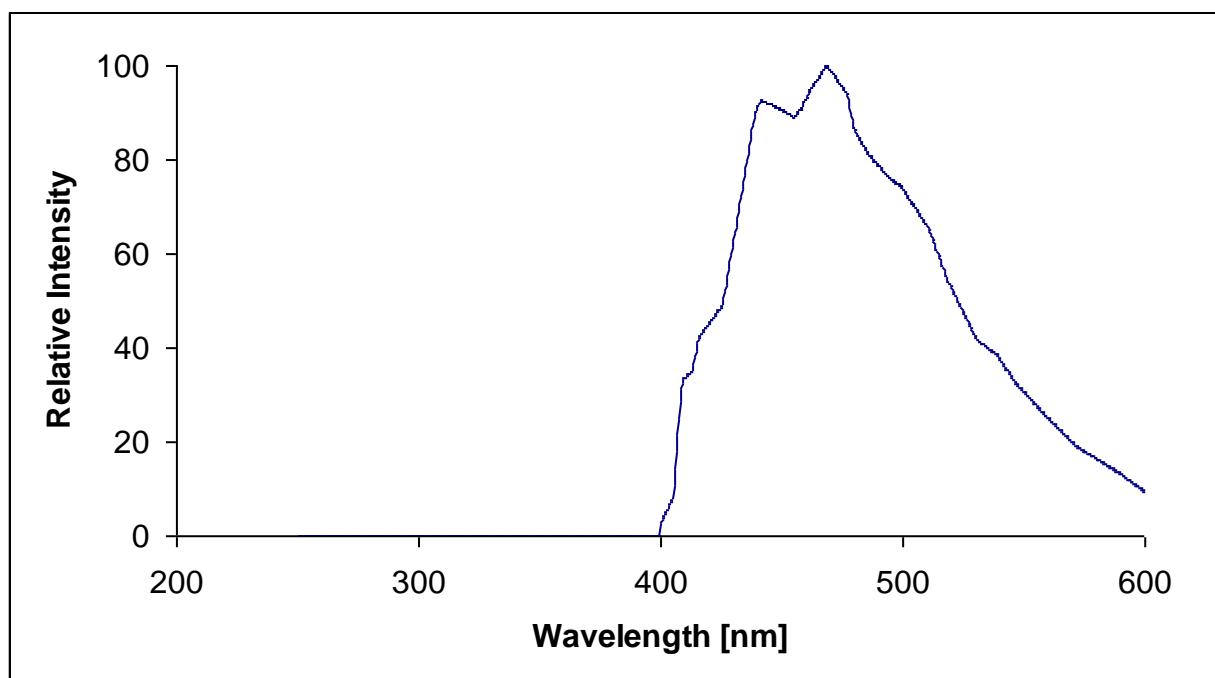
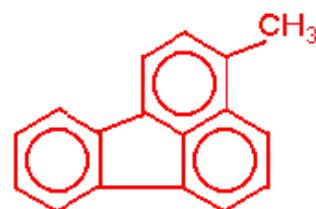
1 – Methylfluoranthene



λ exc.	345.5 nm
Formula	$\text{C}_{17}\text{H}_{12}$
M.W.	216 u
H/H+C	0.414
m.p.	72-73°C
m.a.c. (266 nm)	$1.5 (\text{l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4})$
solvent	Cyclohexane
source	Atlas pag. 842 -III

PFL3MB

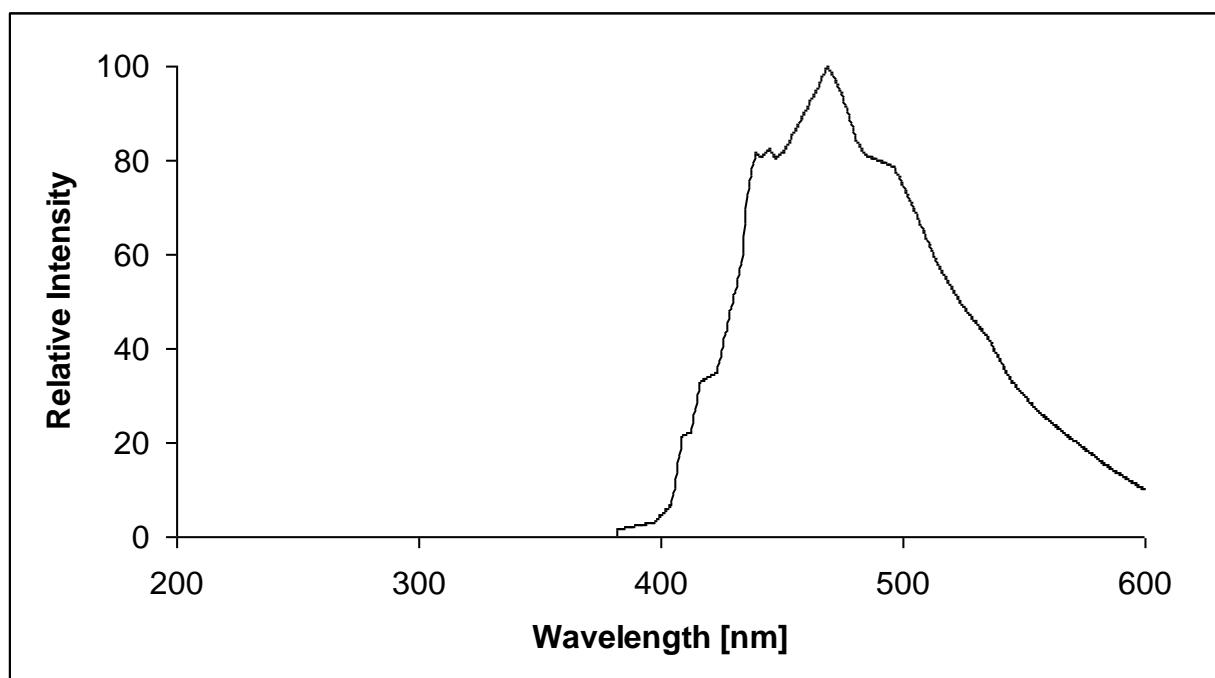
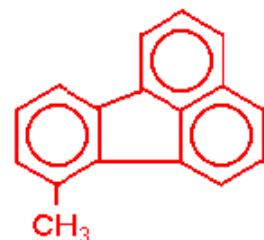
3 – Methylfluoranthene



λ exc.	345 nm
Formula	C ₁₇ H ₁₂
M.W.	216 u
H/H+C	0.414
m.p.	66-68°C
m.a.c. (266 nm)	0.8 (l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	Cyclohexane
source	Atlas pg. 860-III

PFL3MC

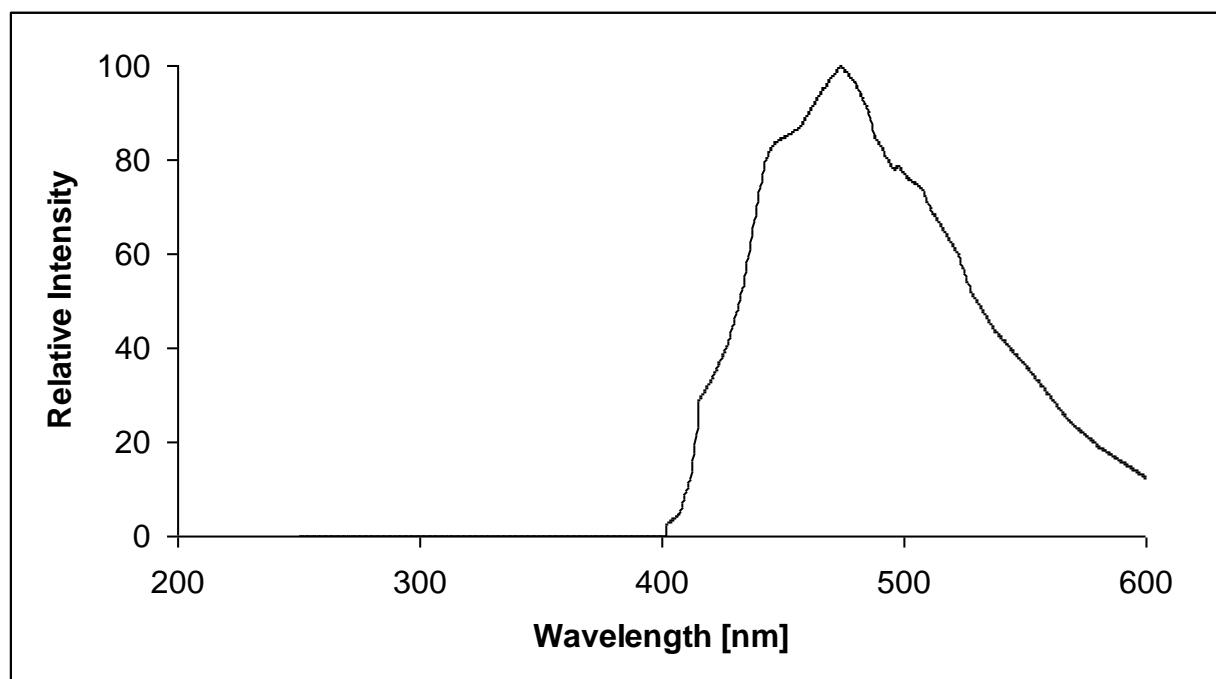
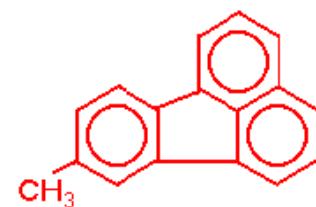
7 – Methylfluoranthene



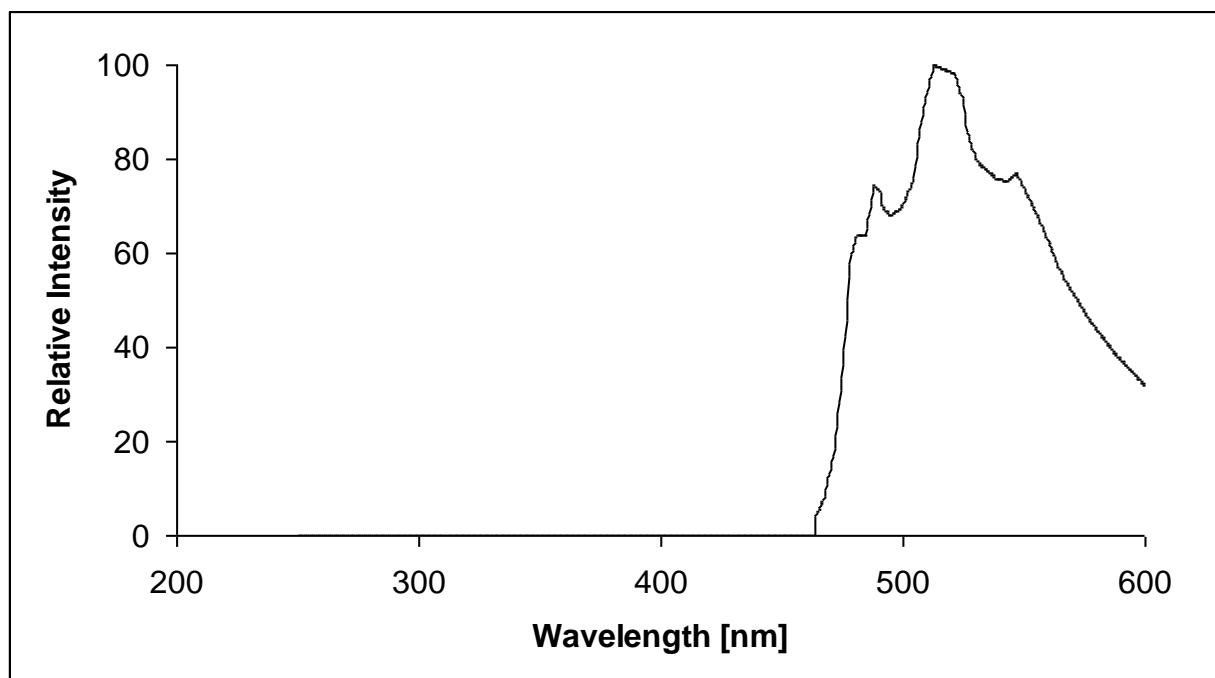
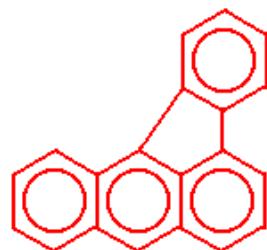
λ exc.	348 nm
Formula	$\text{C}_{17}\text{H}_{12}$
M.W.	216 u
H/H+C	0.414
m.p.	136-137°C
m.a.c. (266 nm)	$1.55 (\text{l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4})$
solvent	Cyclohexane
source	Atlas pg. 878-III

PFL3MD

8 – Methylfluoranthene



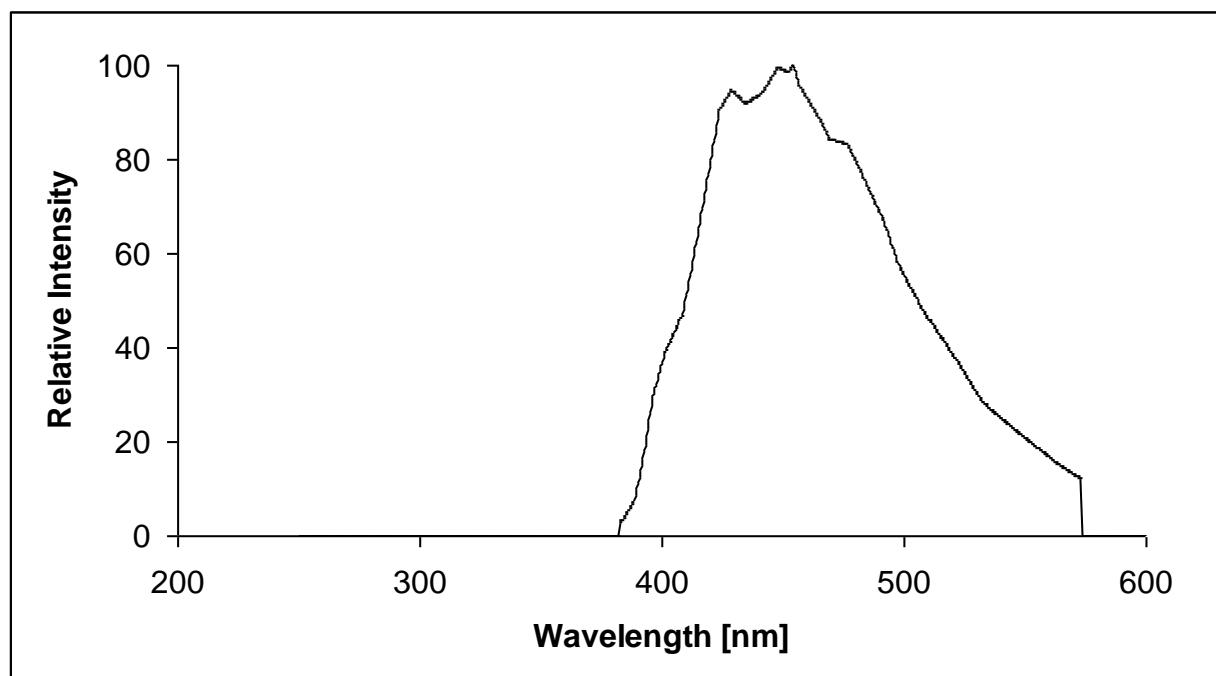
λ exc.	354 nm
Formula	C ₁₇ H ₁₂
M.W.	216 u
H/H+C	0.414
m.p.	93-94°C
m.a.c. (266 nm)	0.82 (l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	Cyclohexane
source	Atlas pg. 896-III

PFL4A**Benz(a)aceanthrylene**
Benzo(a)fluoranthene

λ exc.	388 nm
Formula	C ₂₀ H ₁₂
M.W.	252 u
H/H+C	0.375
m.p.	146.3°C
m.a.c. (266 nm)	5.89 (l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	Cyclohexane
source	Atlas pg. 460-I

PFL4B

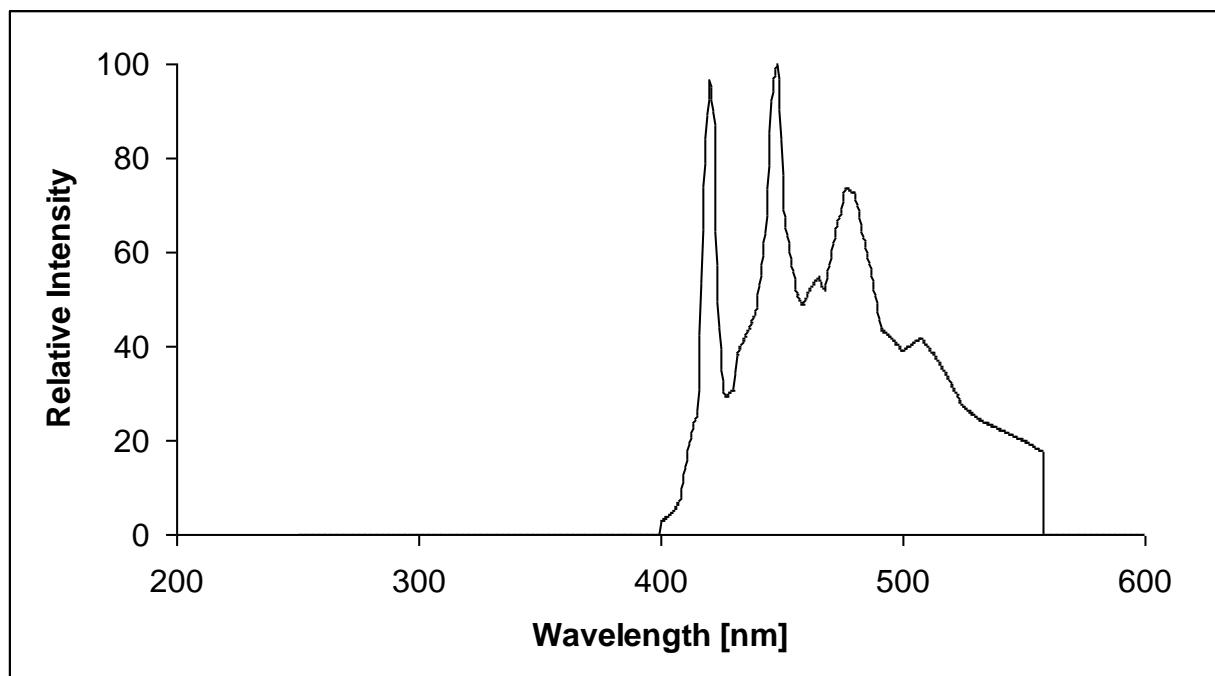
Benz(a)acephenanthrylene
Benzo(b)fluoranthene



λ exc.	302 nm
Formula	C ₂₀ H ₁₂
M.W.	252 u
H/H+C	0.375
m.p.	168.3°C
m.a.c. (266 nm)	2.5 (l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	Cyclohexane
source	Atlas pg. 478-I

PFL4H1

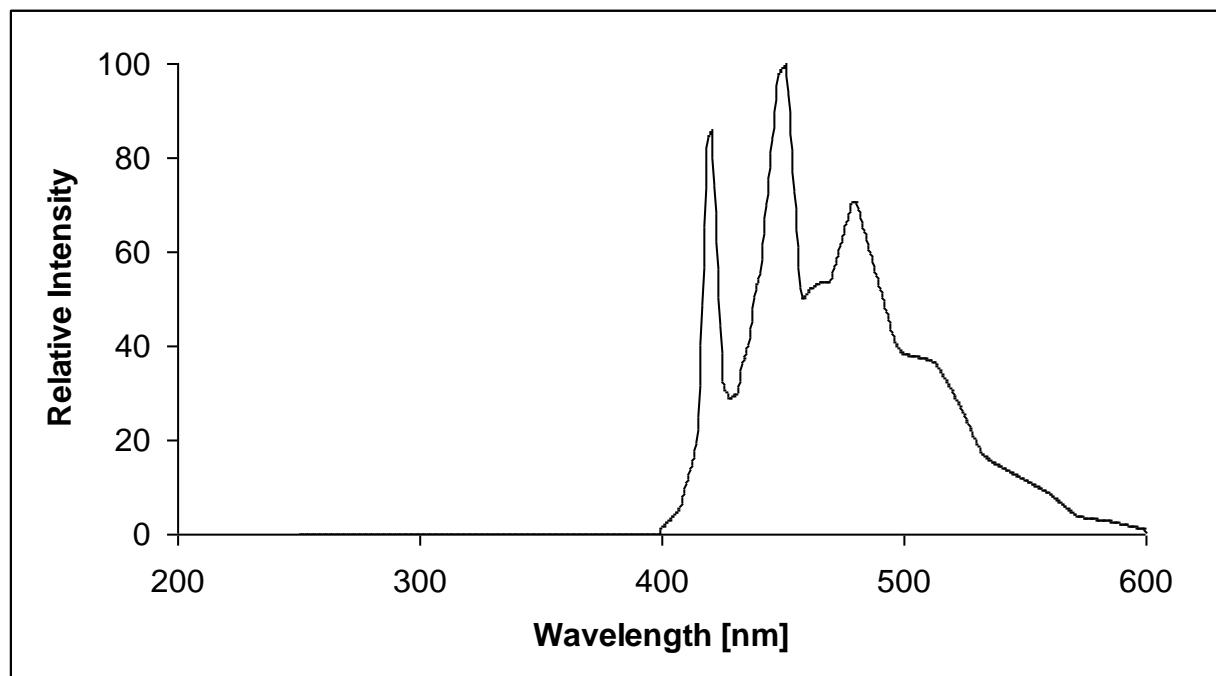
Benzo(ghi)fluoranthene (1)



λ exc.	291 nm
Formula	$C_{18}H_{10}$
M.W.	226 u
H/H+C	0.357
m.p.	128.4°C
m.a.c. (266 nm)	$1.9 \text{ (l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4})$
solvent	Cyclohexane
source	Atlas pg. 170-I

PFL4H2

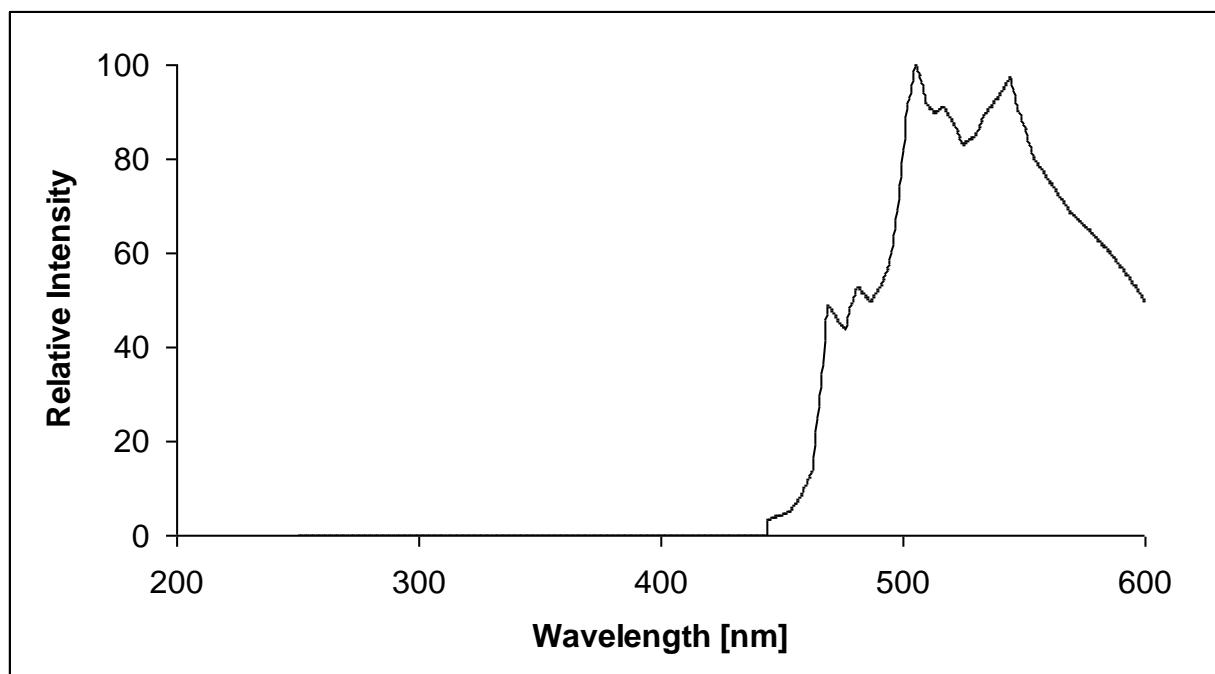
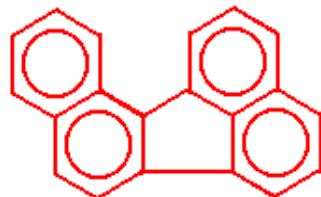
Benzo(ghi)fluoranthene (2)



λ exc.	365 nm
Formula	C ₁₈ H ₁₀
M.W.	226 u
H/H+C	0.357
m.p.	°C
m.a.c. (266 nm)	(l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	Cyclohexane
source	Berlman

PFL4J

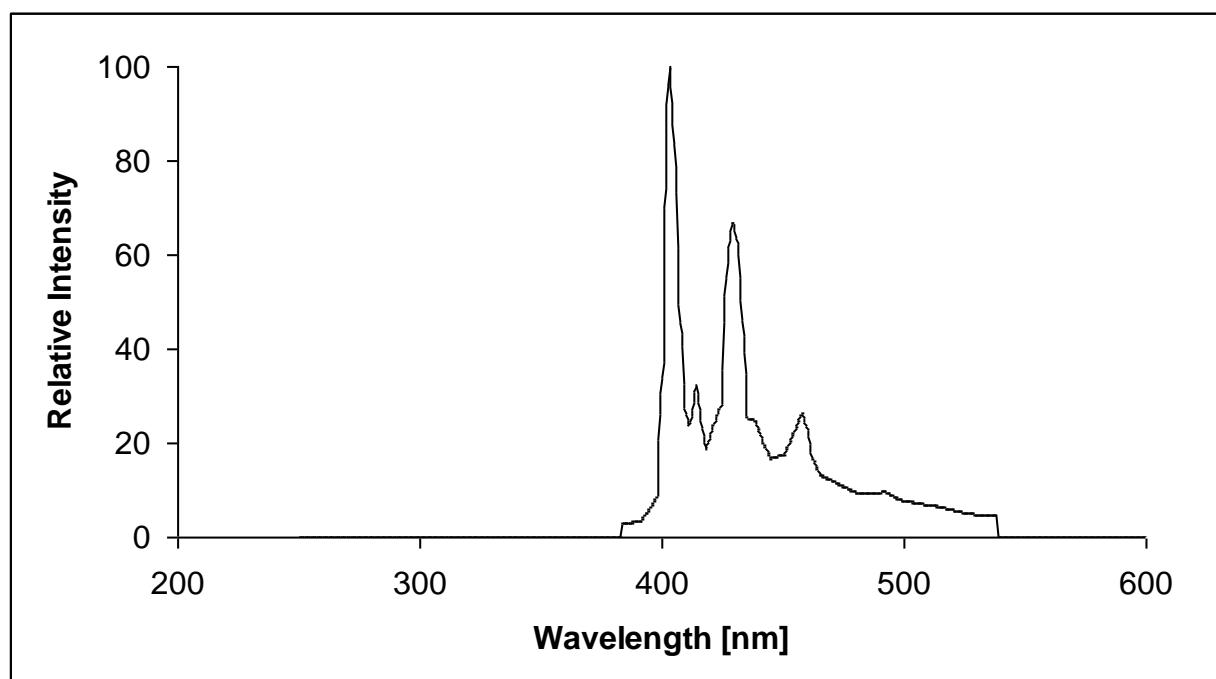
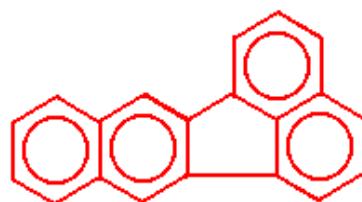
Benzo(j)fluoranthene



λ exc.	319 nm
Formula	$C_{20}H_{12}$
M.W.	252 u
H/H+C	0.375
m.p.	165.4°C
m.a.c. (266 nm)	$0.8 \text{ (l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4})$
solvent	Cyclohexane
source	Atlas pg. 496-I

PFL4K

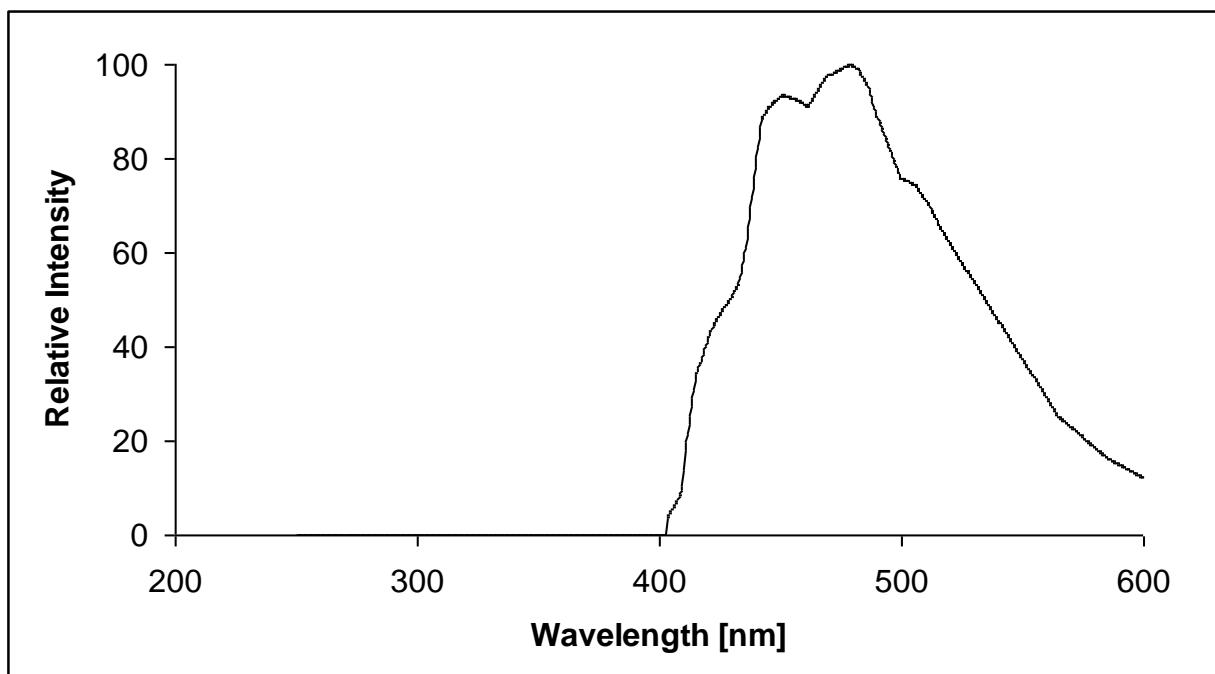
Benzo(k)fluoranthene



λ exc.	308 nm
Formula	$C_{20}H_{12}$
M.W.	252 u
H/H+C	0.375
m.p.	215.7°C
m.a.c. (266 nm)	$1.7 \text{ (l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4})$
solvent	Cyclohexane
source	Atlas pg. 514-I

PFL4P

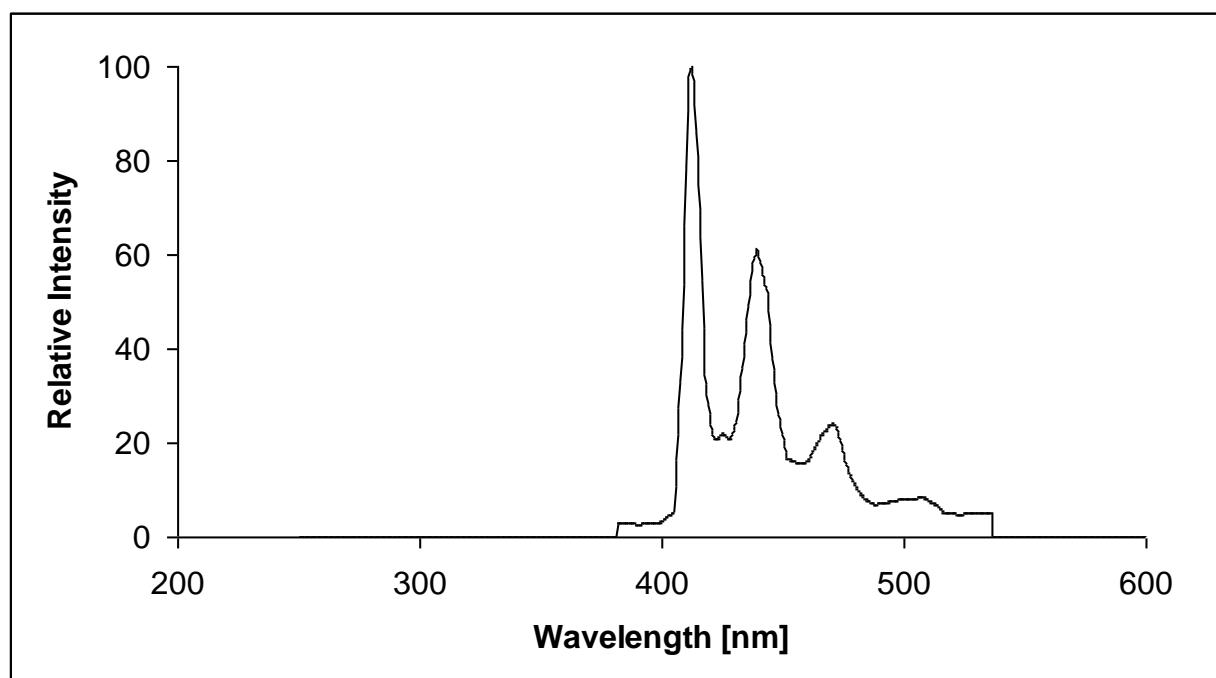
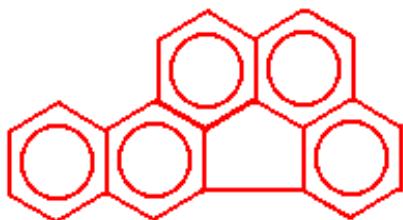
3 – Phenylfluoranthene



λ exc.	365 nm
Formula	$C_{22}H_{14}$
M.W.	278 u
H/H+C	0.389
m.p.	°C
m.a.c. (266 nm)	(l. $mol^{-1} cm^{-1} \times 10^{-4}$)
solvent	Cyclohexane
source	Berlman

PFL5B

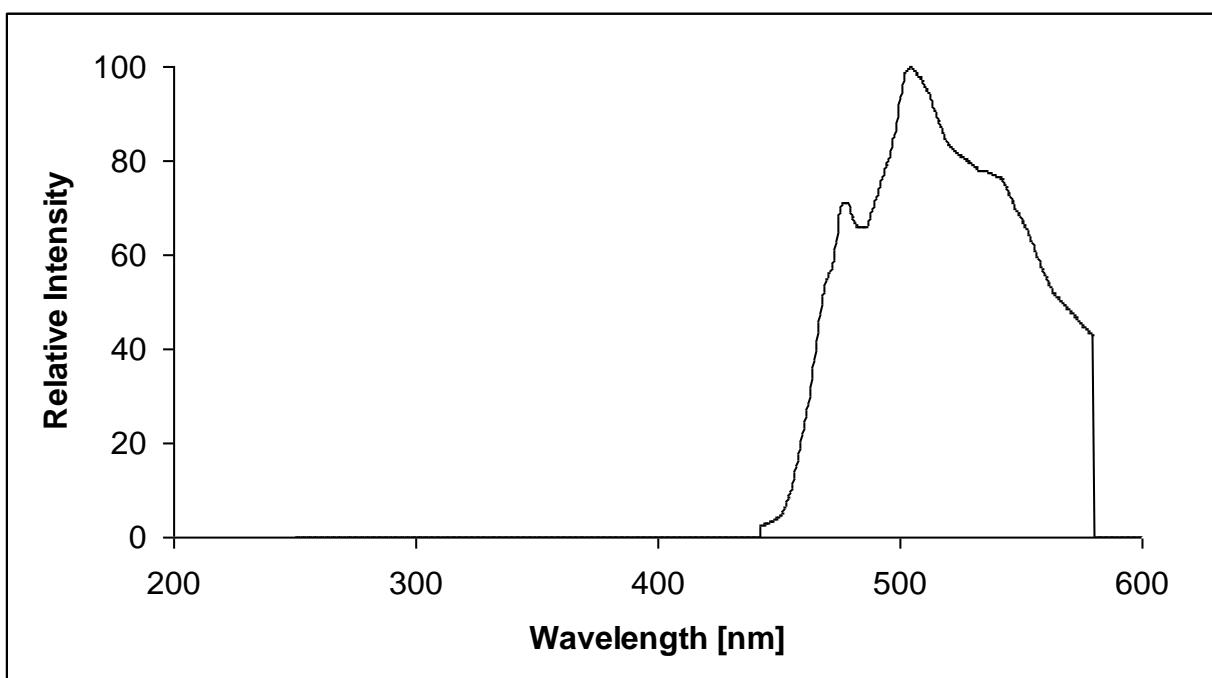
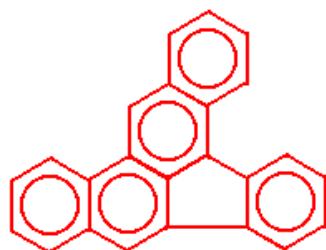
Indeno(4,3,2,1-cdef)chrysene
Bibenzo(k,mno)fluoranthene



λ exc.	300 nm
Formula	$C_{22}H_{12}$
M.W.	276 u
H/H+C	0.353
m.p.	150°C
m.a.c. (266 nm)	$4.29 \text{ (l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4})$
solvent	Cyclohexane
source	Atlas pg. 572-II

PFL5C1

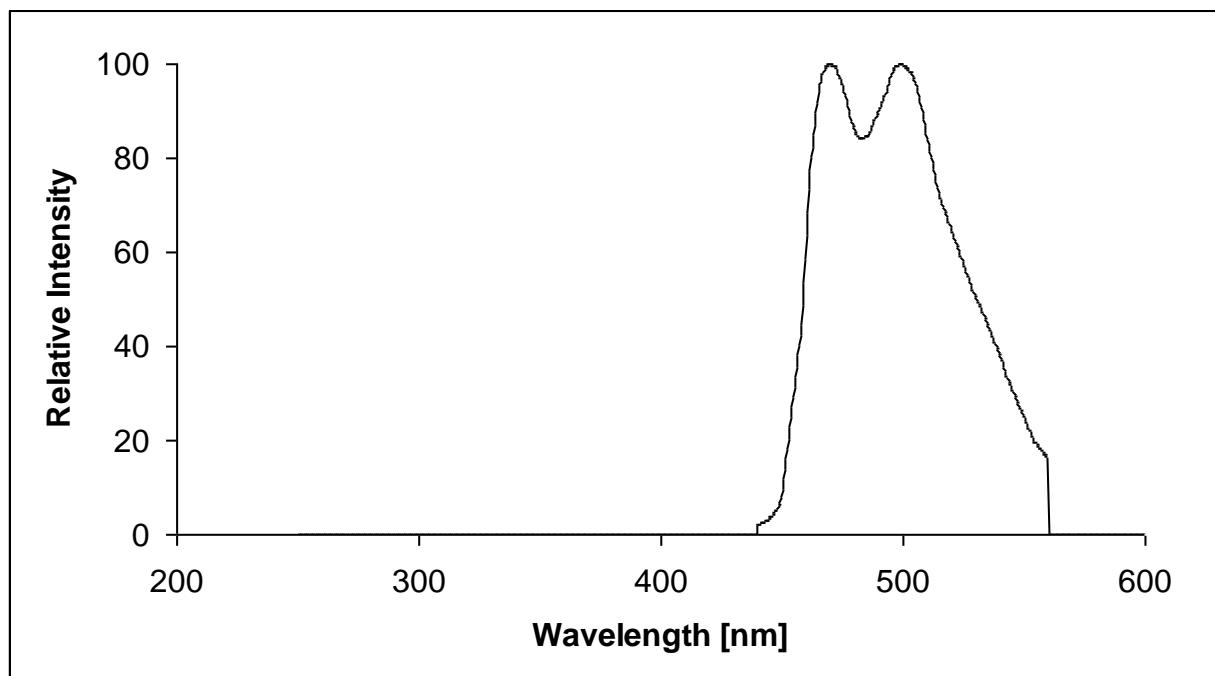
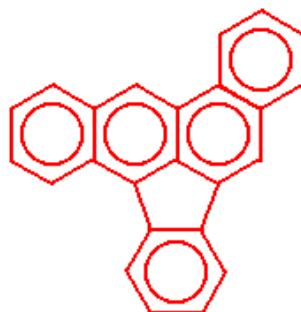
Dibenz(a,e)aceanthrylene (1)
Dibenzo(a,e)fluoranthene (1)



λ exc.	300 nm
Formula	$C_{24}H_{14}$
M.W.	302 u
H/H+C	0.368
m.p.	232°C
m.a.c. (266 nm)	$4.4 \text{ (l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4})$
solvent	Cyclohexane
source	Atlas pg. 738-II

PFL5C2

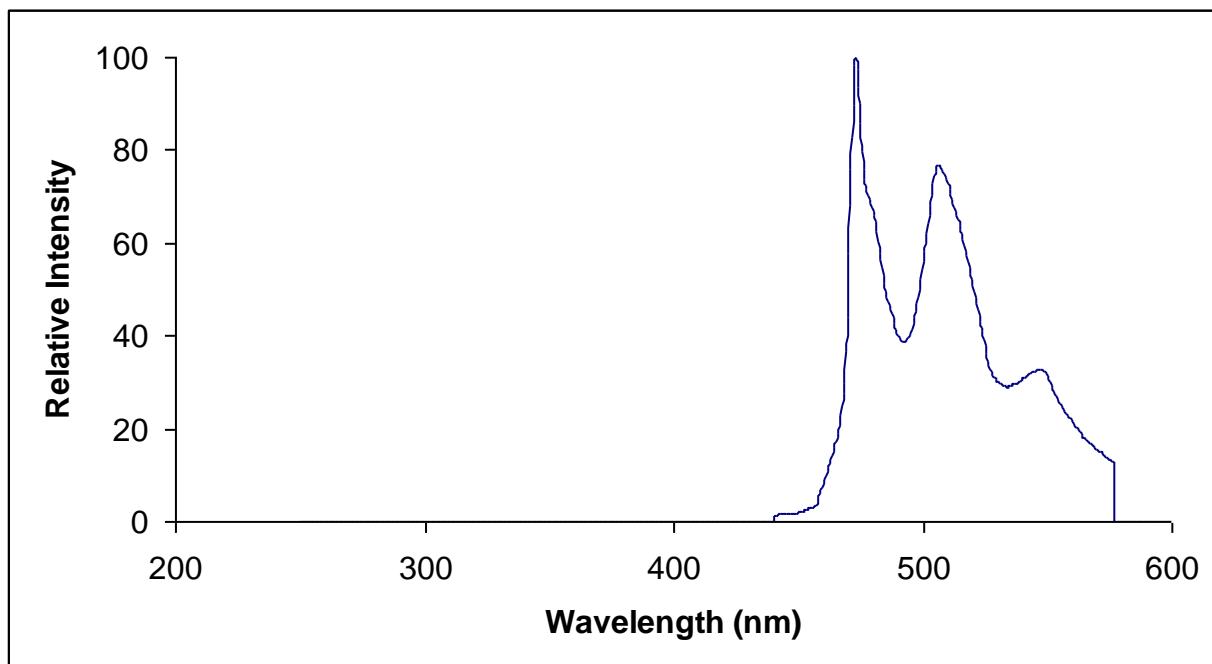
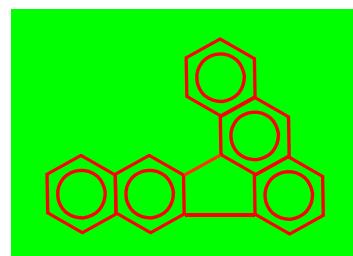
Dibenz(a,e)aceanthrylene (2)
Dibenzo(a,e)fluoranthene (2)



λ exc.	390 nm
Formula	$C_{24}H_{14}$
M.W.	302 u
H/H+C	0.368
m.p.	232°C
m.a.c. (266 nm)	($l \cdot mol^{-1} cm^{-1} \times 10^{-4}$)
solvent	n-hexadecane
source	[6]

PFL5D

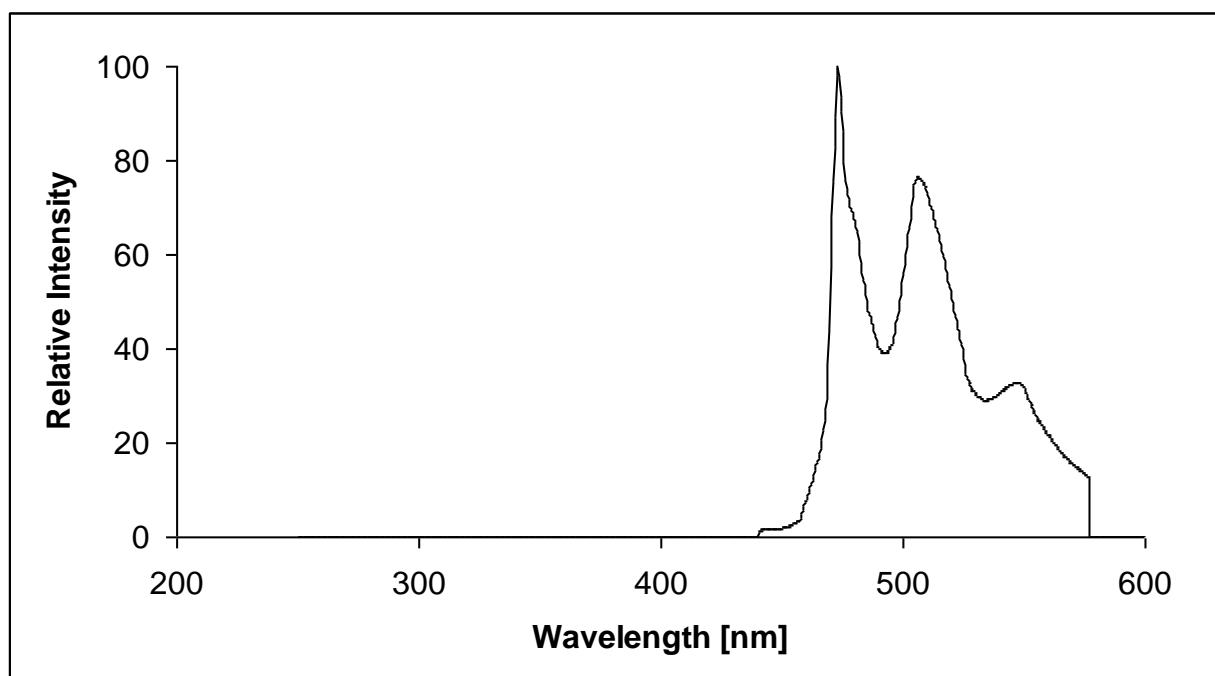
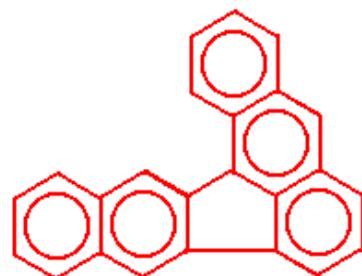
Dibenzo(a,k)fluoranthene
Naphth(2,3-a)aceanthrylene



λ exc.	415 nm
Formula	$C_{24}H_{14}$
M.W.	302 u
H/H+C	0.368
m.p.	229 °C
m.a.c. (266 nm)	$7.11 \text{ (l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4})$
solvent	Cyclohexane
source	ATLAS pag. 756-II

PFL5D

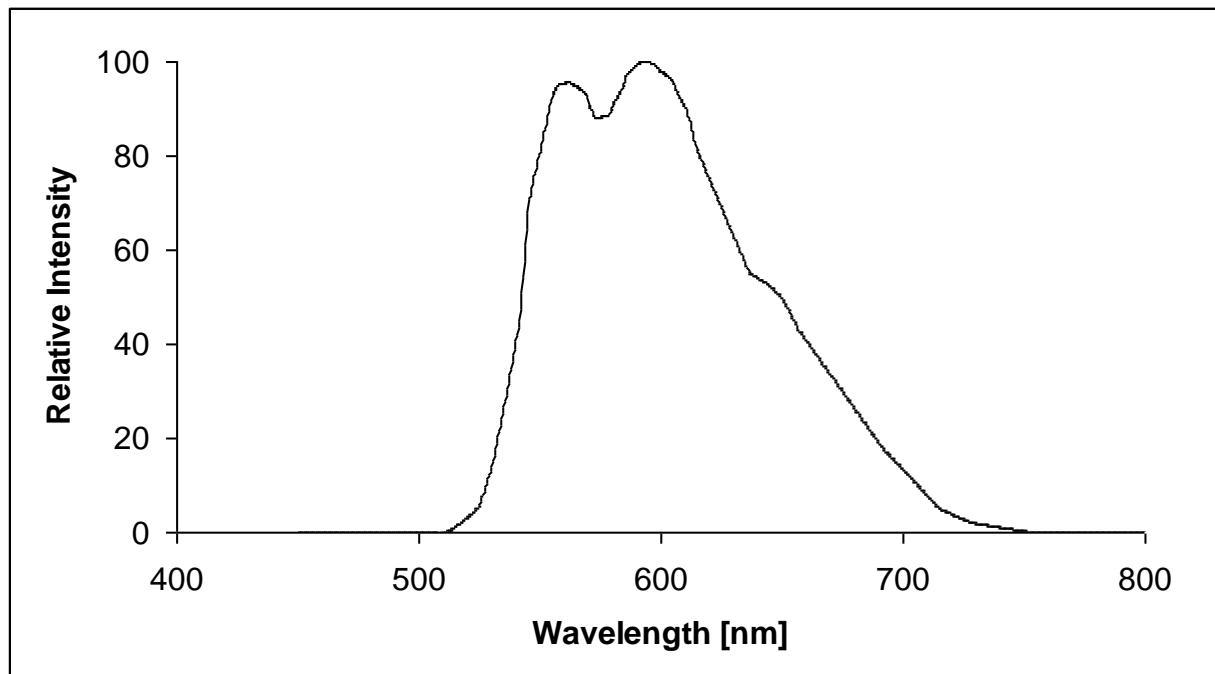
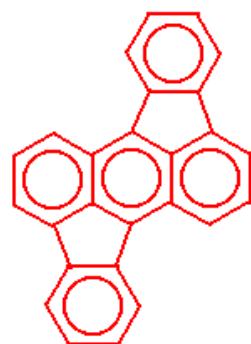
Naphth(2,3-a)aceanthrylene
Dibenzo(a,k)fluorathene



λ exc.	415 nm
Formula	C ₂₄ H ₁₄
M.W.	302 u
H/H+C	0.368
m.p.	229°C
m.a.c. (266 nm)	7.11 (l. mol ⁻¹ cm ⁻¹ × 10 ⁻⁴)
solvent	Cyclohexane
source	Atlas pg. 756-II

PFL5F

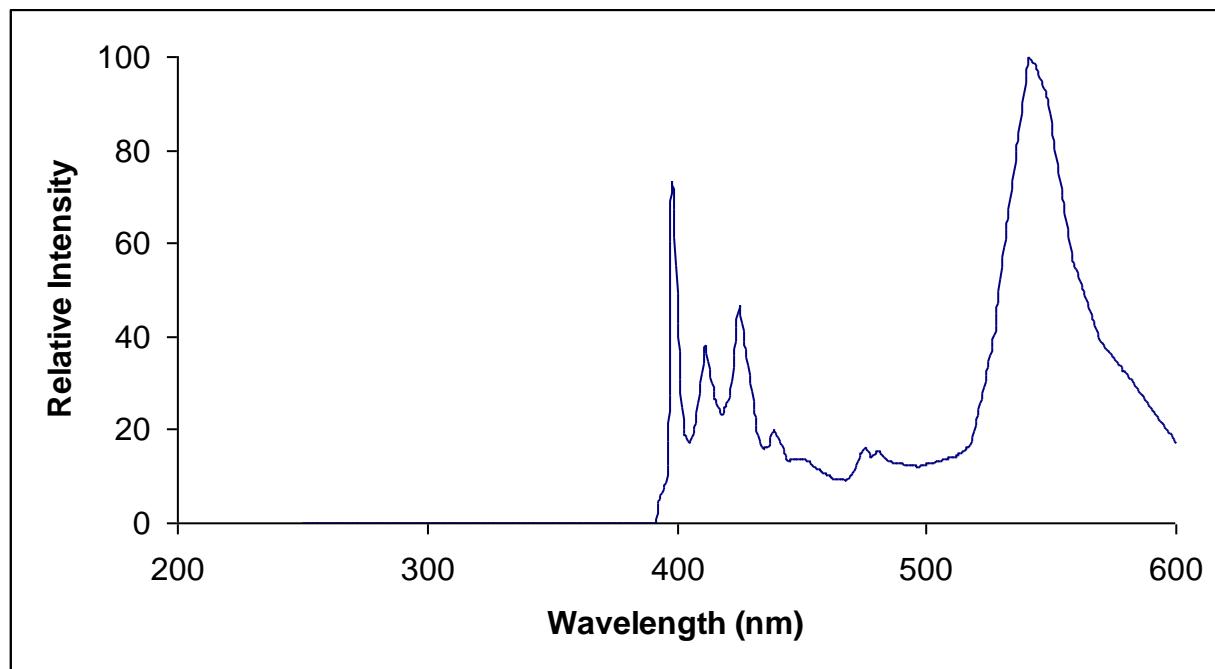
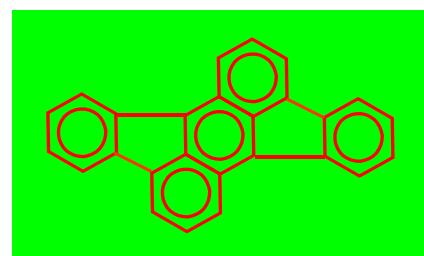
Rubicene



λ exc.	435 nm
Formula	$C_{26}H_{14}$
M.W.	326 u
H/H+C	0.350
m.p.	°C
m.a.c. (266 nm)	(l. $mol^{-1} cm^{-1} \times 10^{-4}$)
solvent	Benzene
source	Berlman

PFL5F2

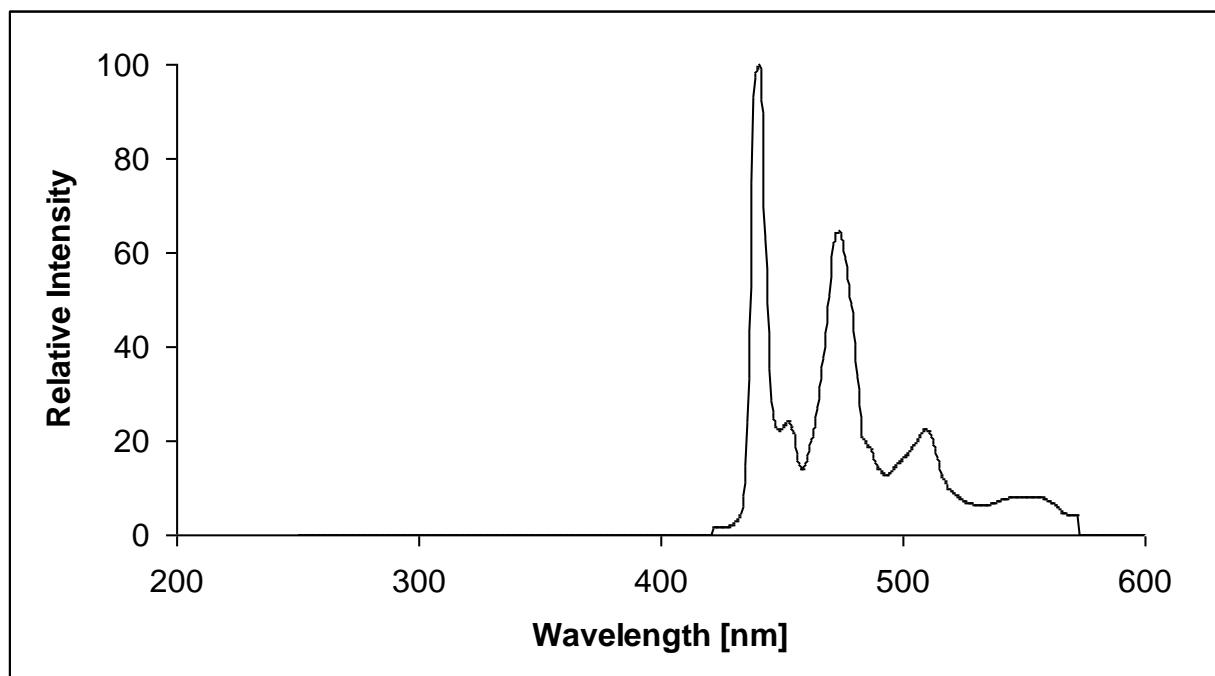
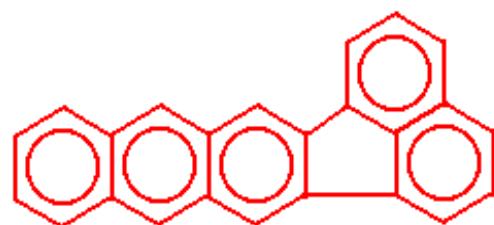
Rubicene



λ exc.	370 nm
Formula	$C_{26}H_{14}$
M.W.	326 u
H/H+C	0.350
m.p.	229 °C
m.a.c. (266 nm)	(l. $mol^{-1} cm^{-1} \times 10^{-4}$)
solvent	n-hexadecane
source	[13]

PFL5N

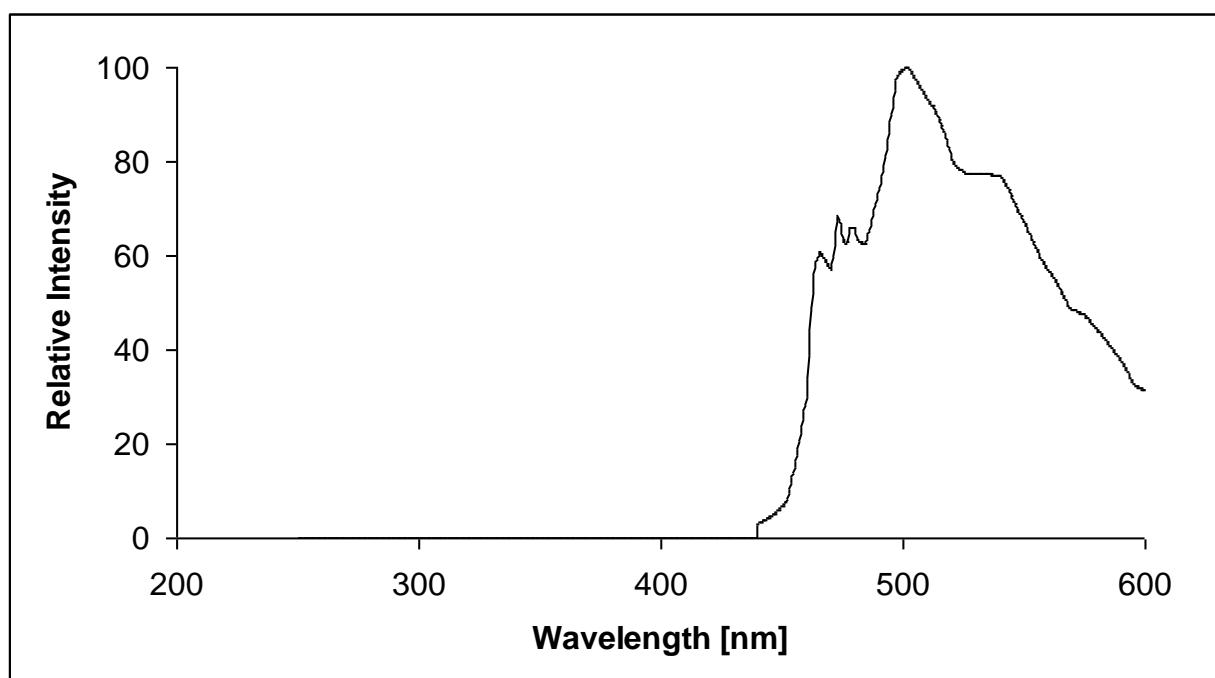
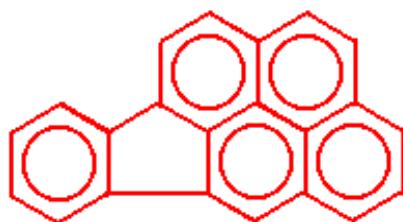
Naphtho(2,3-k)fluoranthene



λ exc.	384.5 nm
Formula	$C_{24}H_{14}$
M.W.	302 u
H/H+C	0.368
m.p.	299-300°C
m.a.c. (266 nm)	$2.75 \text{ (l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4})$
solvent	Cyclohexane
source	Atlas pg. 814-II

PFL5P1

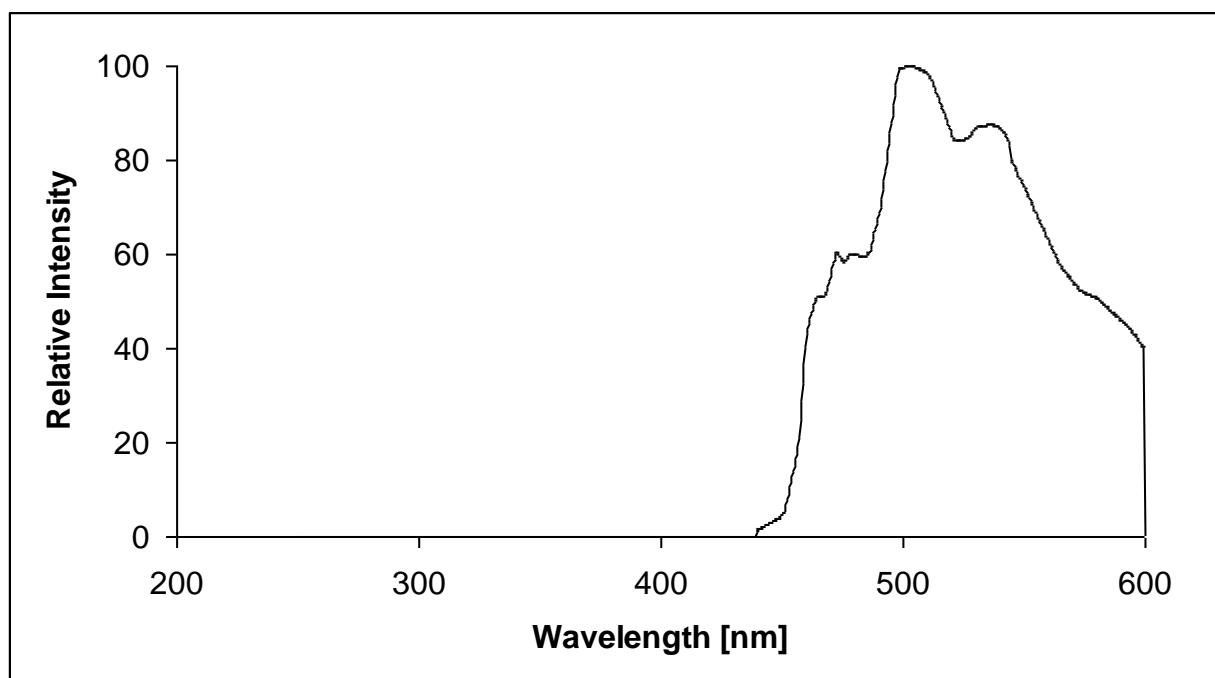
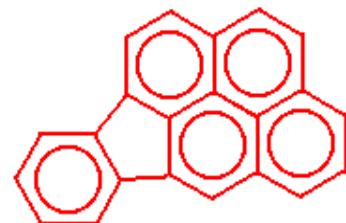
Indeno(1,2,3-cd)pyrene(1)
O-Phenylene-pyrene



λ exc.	360.5 nm
Formula	$C_{22}H_{12}$
M.W.	276 u
H/H+C	0.353
m.p.	163.6°C
m.a.c. (266 nm)	$2.4 \text{ (l. mol}^{-1} \text{ cm}^{-1} \times 10^{-4})$
solvent	Cyclohexane
source	Atlas pg.650-I

PFL5P2

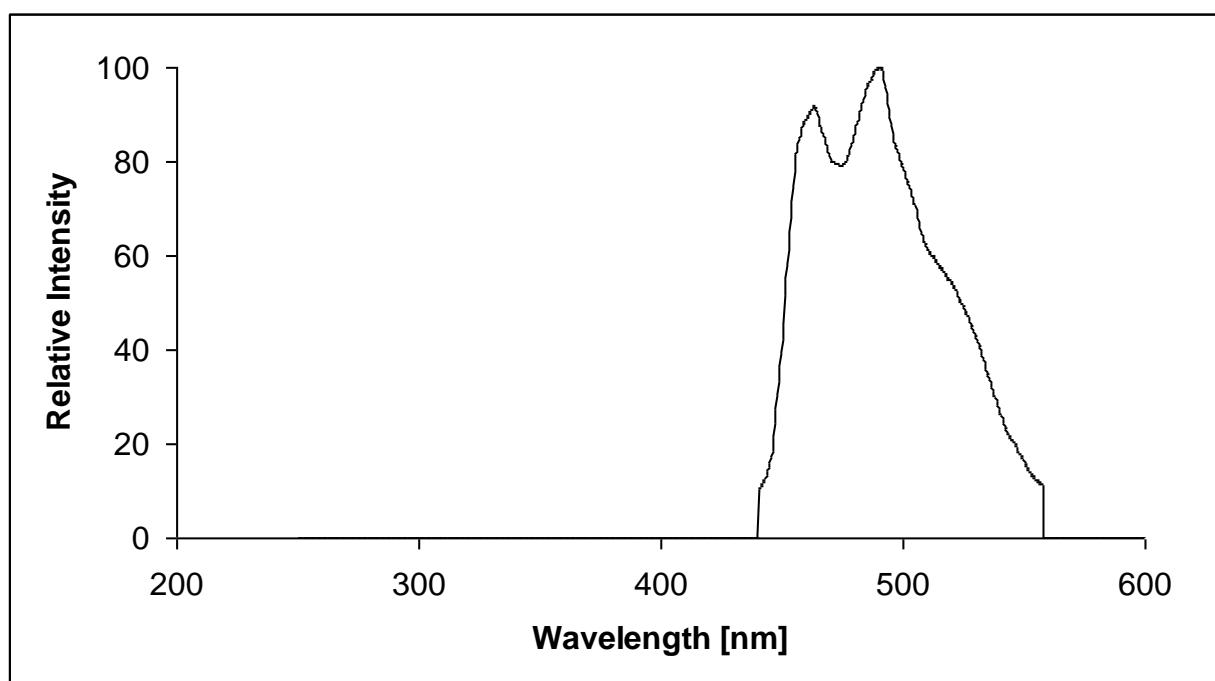
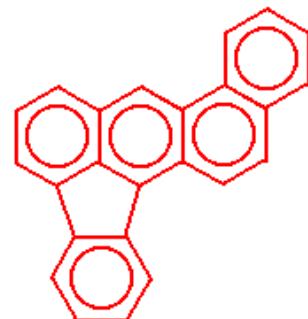
**Indeno(1,2,3-cd)pyrene(2)
O-Phenylene-pyrene**



λ exc.	365 nm
Formula	$C_{22}H_{12}$
M.W.	276 u
H/H+C	0.353
m.p.	°C
m.a.c. (266 nm)	(l. $mol^{-1} cm^{-1} \times 10^{-4}$)
solvent	Cyclohexane
source	Berlman

PFL5S

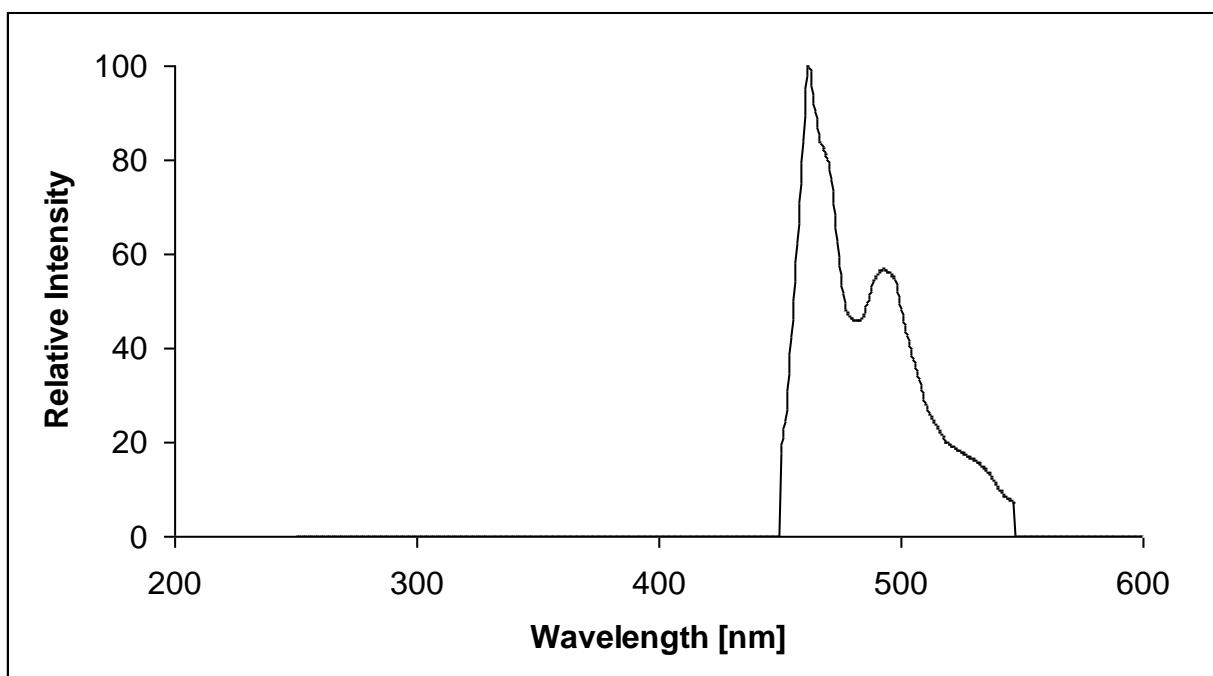
Naphtho(2,1-a)fluoranthene



λ exc.	400 nm
Formula	$C_{24}H_{14}$
M.W.	302 u
H/H+C	0.368
m.p.	°C
m.a.c. (266 nm)	(l. $mol^{-1} cm^{-1} \times 10^{-4}$)
solvent	n-hexadecane
source	[6]

PFL6C

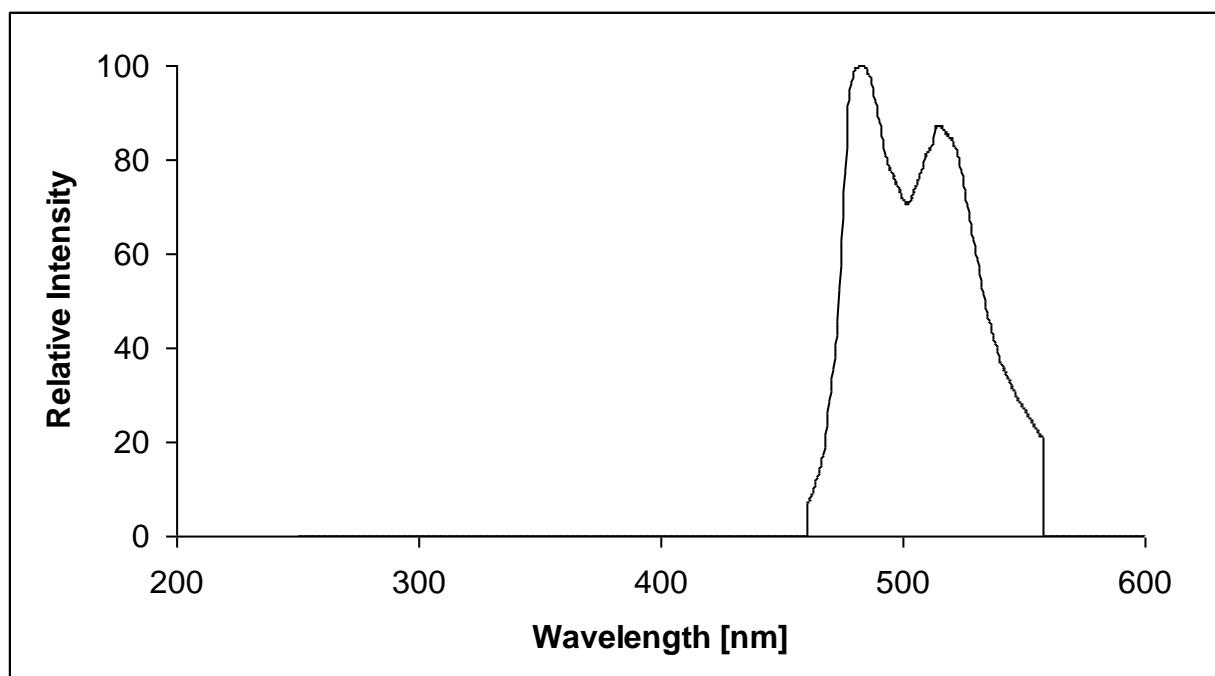
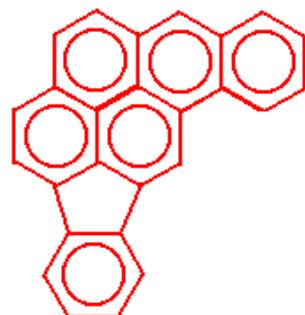
Benz(def)indeno(1,2,3-hi)chrysene



λ exc.	406 nm
Formula	$C_{26}H_{14}$
M.W.	326 u
H/H+C	0.350
m.p.	°C
m.a.c. (266 nm)	(l. $mol^{-1} cm^{-1} \times 10^{-4}$)
solvent	n-hexadecane
source	[6]

PFL6D

Benz(def)indeno(1,2,3-qr)chrysene



λ exc.	408 nm
Formula	$C_{26}H_{14}$
M.W.	326 u
H/H+C	0.350
m.p.	°C
m.a.c. (266 nm)	(l. $mol^{-1} cm^{-1} \times 10^{-4}$)
solvent	n-hexadecane
source	[6]